



*Ni-Cd Battery
Separator Improvement
Based upon Mr. Paul Scardaville's research
and Crane testing*

**DSCR and NAVAIR Sponsored
program to develop a Ni-Cd
battery separator system that
will increase battery safety and
life to highest levels**



*Ni-Cd Battery
Separator Improvement*

**2009 Joint Service Power Expo
New Orleans, LA
6 May 2009**

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- **Background: Problems that prompted program**
 - **Tests:**
 - Gurley airflow (time to pass air volume)
 - Rewet-ability in KOH (soak 15%, rewet 30%)
 - Temperature-Rise & Float Charge (TR&F)
 - **Comparisons: Results of TR&F**
 - Wetting agents
 - Absorbers
 - Gas barriers

- **Production Battery Performance began to decrement**
 - Celgard increased average porosity of gas barrier by tightening tolerance toward high porosity limit
 - Kimberly-Clark (KC) dropped melt-blown polypropylene (mbPP) absorbers.*
 - As manufacturers reached end of their KC supply, battery performance decrement accelerated.
- (Reason: “Wayfos A” no longer available)**

High/Low Gurley Testing (2004)

Type M81757/16, KC mbPP Absorber

- **Celgard gas barriers with Celgard standard wetting agent**
 - One with 37 Gurley-second (G-s) porosity
 - One with Celgard 3400 (24 G-s)
- **Both performed essentially same in TRF & Life Cycling.**
- **No difference in post life capacities**

Sulfur Contamination

Sulfur in electrolyte was believed to cause a permanent decrement in capacity.

Source of the sulfur was determined to be from water-soluble dispersants that were used to apply the wetting agent to the mbPP absorber.

Surfactants Suspected

- **Dissections exhibited separator dryout and poor rewet ability in production batteries made after the 1980's**
- **Investigated why wetting agent appeared to leaving gas barrier**
- **Determined Celgard Inc was applying a wetting agent that was fugitive**

Surfactant Comparisons



Performed a wetting agent rewet-ability using membrane with Celgard's & 2 candidate W.A.s* from Mr. Paul Scardaville's search

Test: Samples soaked in 15% KOH solution, air dried and returned to 30% solution for rewet.

Results of soak durations to 12 months

A) Celgard 3400: Lost rewet ability in 1 day

B) Surfonic L24-4 (alcohol/ether): Rewet

C) Deforest HP-739* (anionic ester): Rewet

*HP-739 is a clone of 1970- 2000 Wayfos A & has same CAS #

Type M81757/16 batteries with Celgard 37 G-s porosity gas barrier

Separators:

- #1: Manufacturer's absorber & gas barrier with fugitive wetting agent (N3400G1-P)**
- #2: Grafted H&V MBPP and N3400G1-P**
- #3: Grafted H&V MBPP and nonsoluble wetting agent on gas barrier, DePHOS HP 739 (CAS # 12645-31-7)**

Baseline Conclusions

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- **Porosity in 20-40 G-s range has large impact on charge stability***
 - **Wetting agents**
 - Nonionic (Huntsman) was unusable
 - Fugitive afforded no safety*
 - Dispersant residues were generally harmful
 - **Coated & grafted absorbers have same performance**
 - *Influenced by wetting agent transfer

TR&F Cycling Test

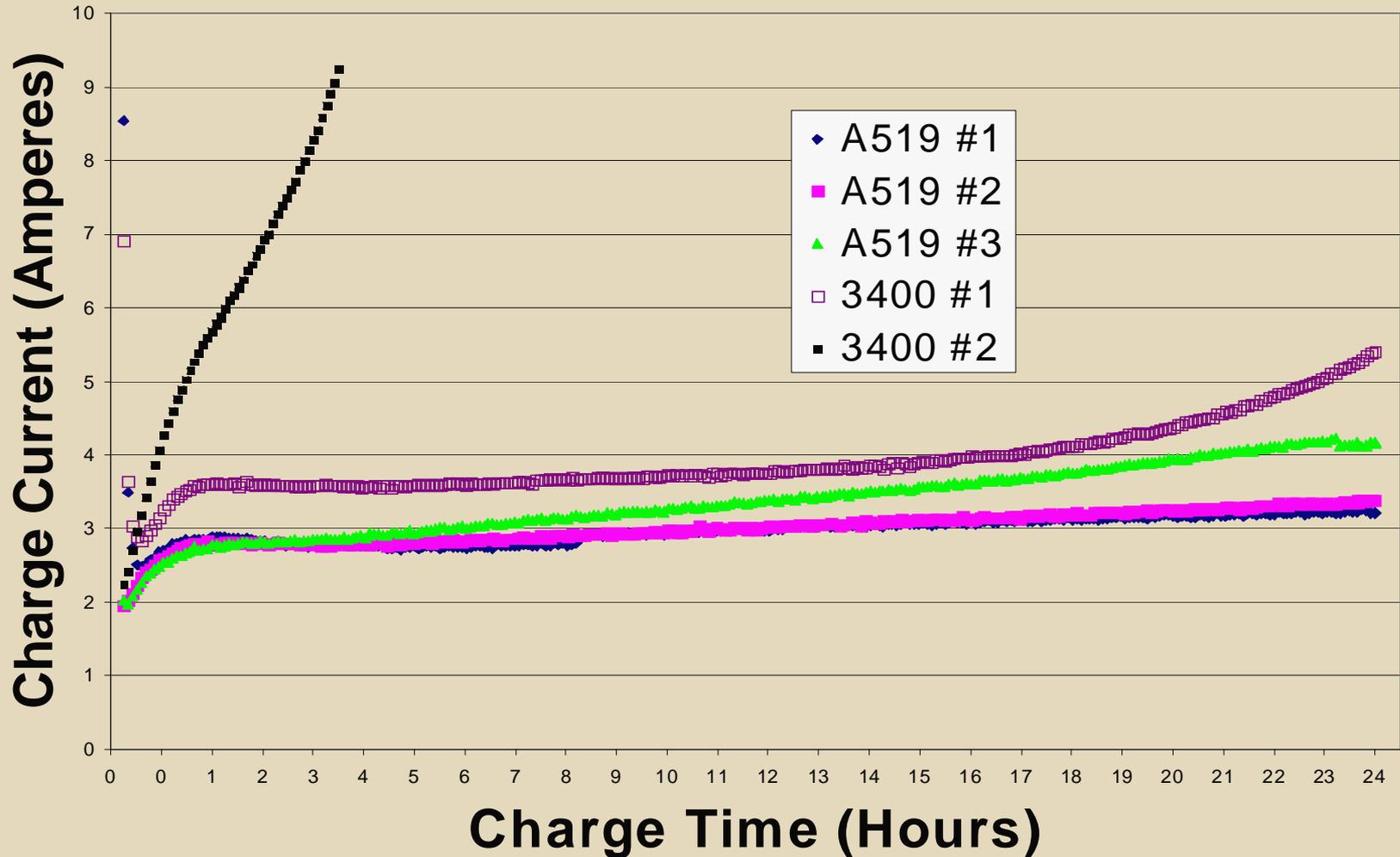
Type M81757 35Ah Batteries

- **Initial charge: 2-Step CC with water addition**
- **TRF cycles:**
 - Stabilize battery in Chamber @ 120°F
 - 315A discharge to 14.4V or 5 minutes
 - 24-hour CP @ 28.5V
 - Repeat -315A and CP charges (M–F)
 - Sat AM: Rest open circuit and return to amb.
 - Sun PM: Repeat sequence above
 - Water additions: As needed

- **35Ah Batteries using Woven Nylon absorber and different gas barriers**
 - **3400: 24G-s porosity and fugitive wetting agent**
 - **A519: 37G-s porosity & insoluble wetting agent DePHOS HP 739
(CAS # 12645-31-7)**

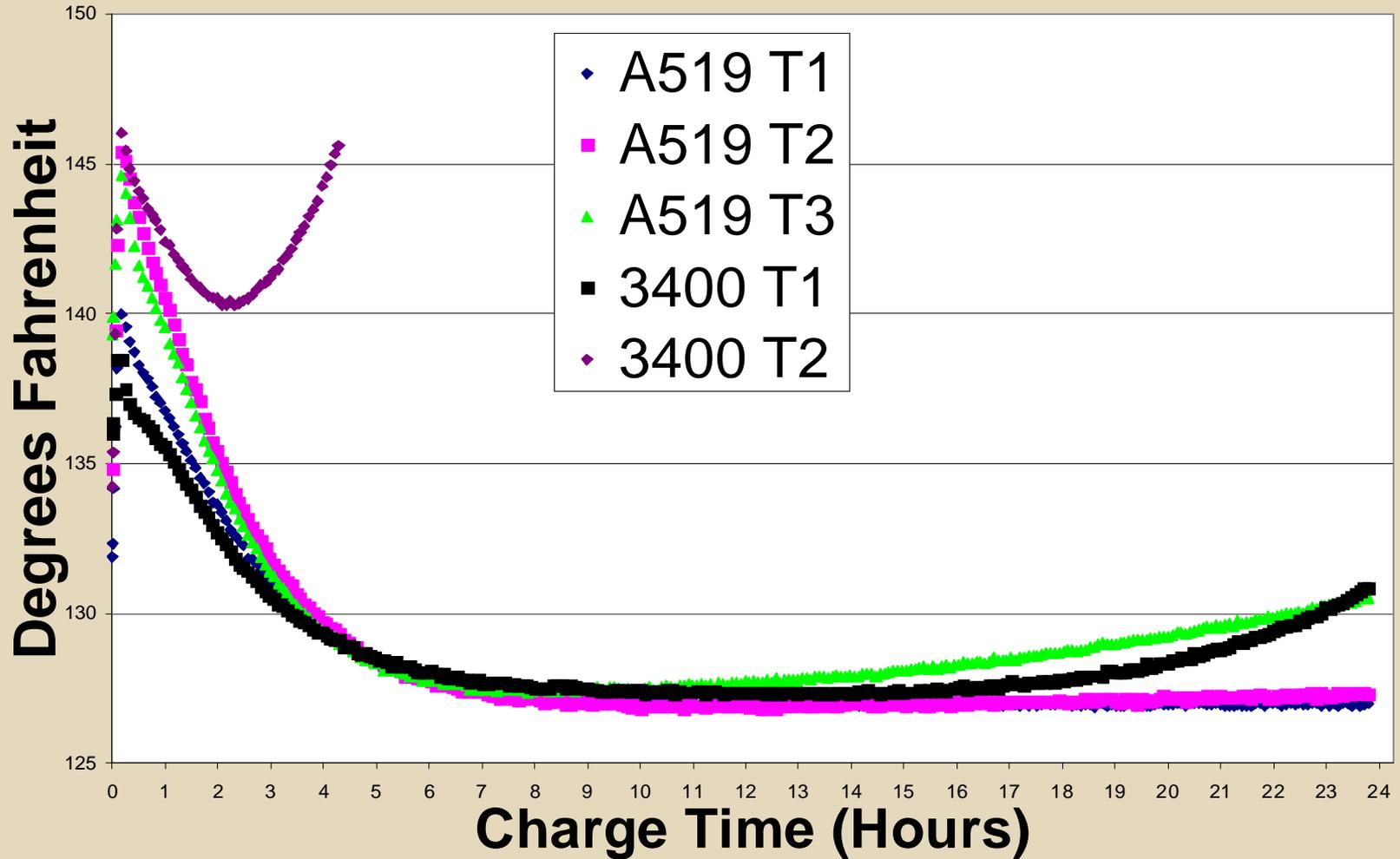
Charge Current TR&F Cycling

W.N. & A519 VS. W.N. & 3400



Battery Temperature TR&F Cycling

W. N. & A519 VS. W. N. & 3400



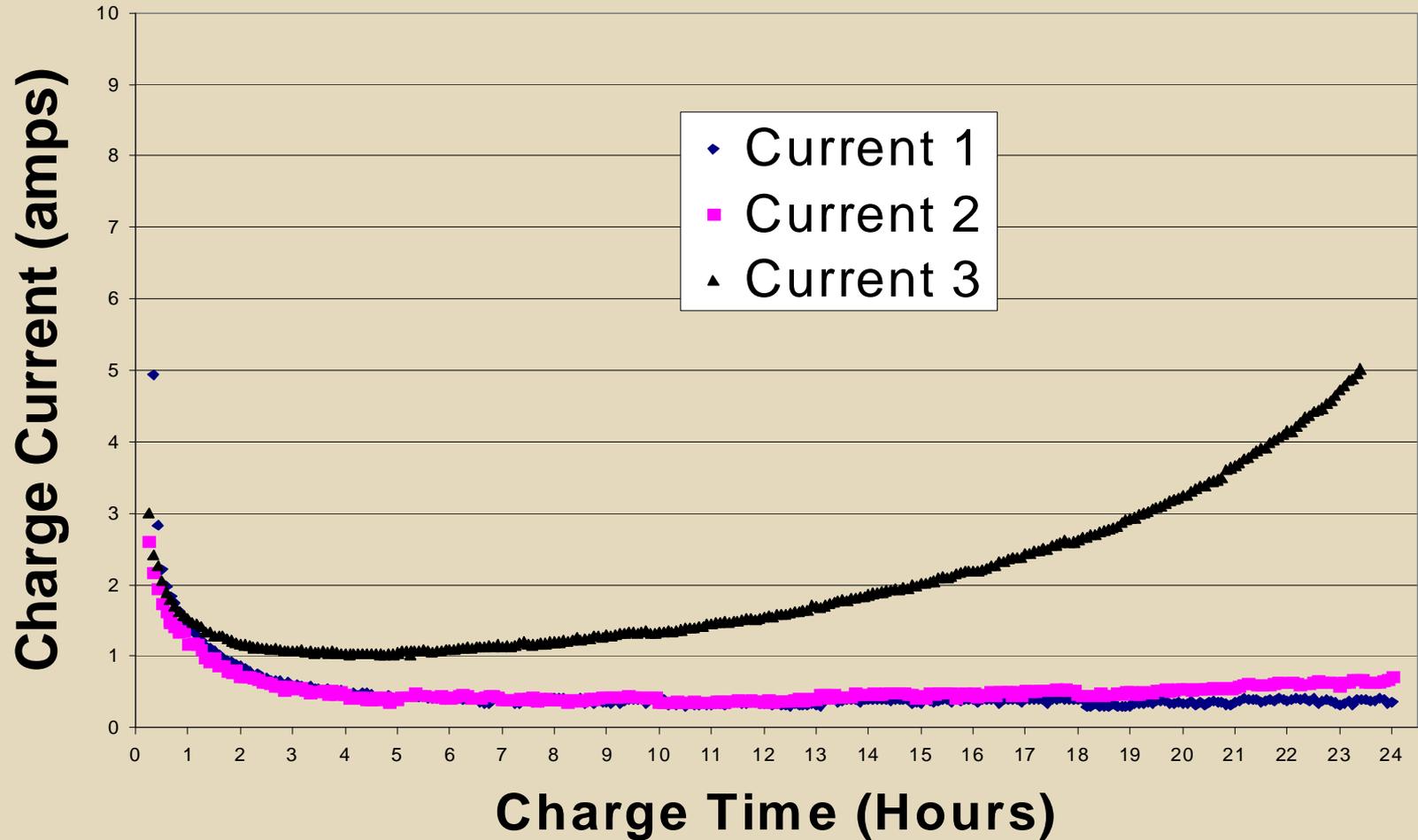
Wetting Agent Mobility

Effect on gas barrier performance

- **Gas barrier (N3400G1-P)**
 - Porosity: 37G-s
 - Wetting agent: Fugitive - Celgard proprietary
- **Absorbers**
 - Grafted mbPP
 - Coated mbPP (CAS # 12645-31-7)

TR&F Charge Current (PL)

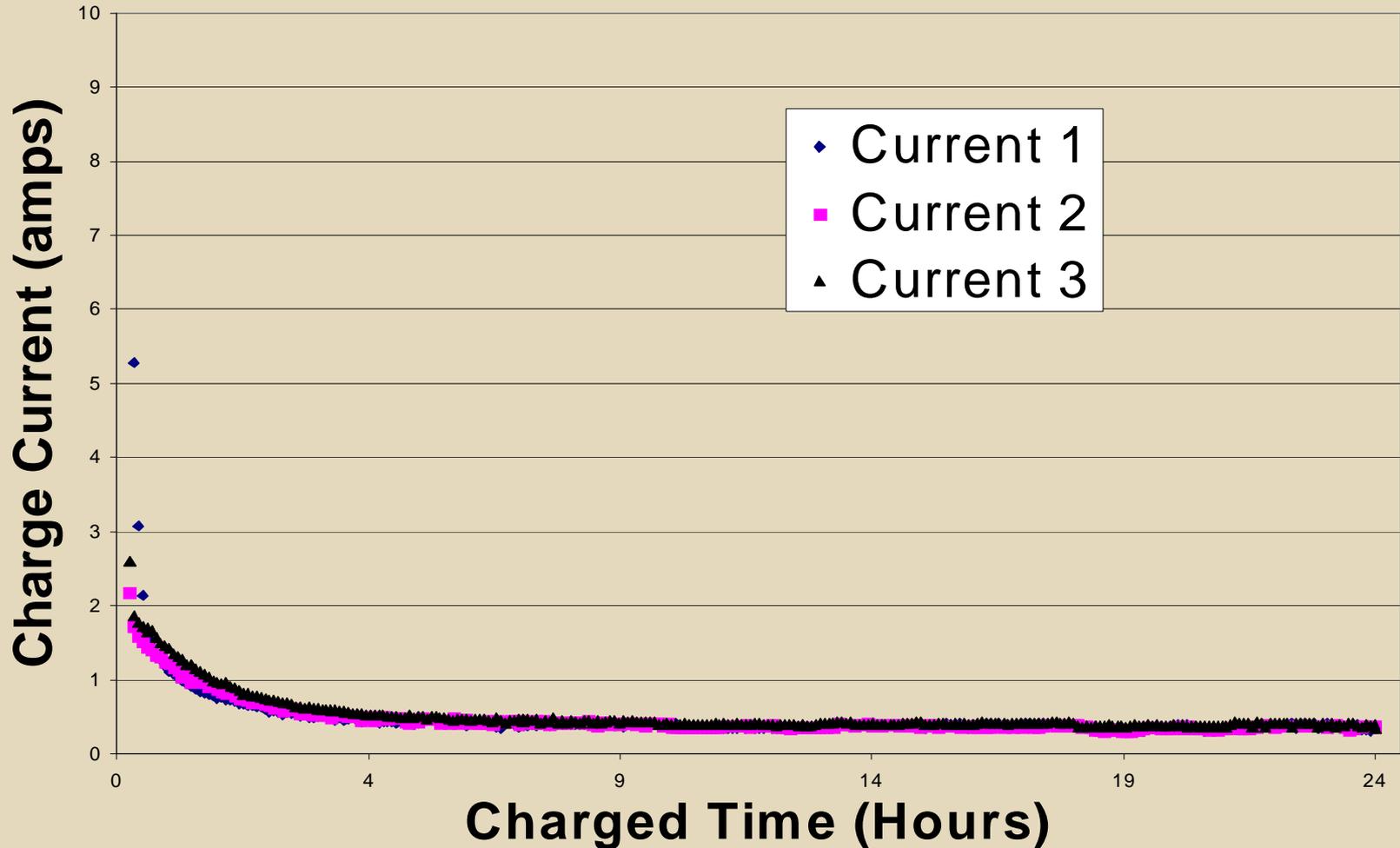
Grafted Absorber, N3400G1-P



Distribution Statement A: Approved for Public Release; Unlimited Distribution

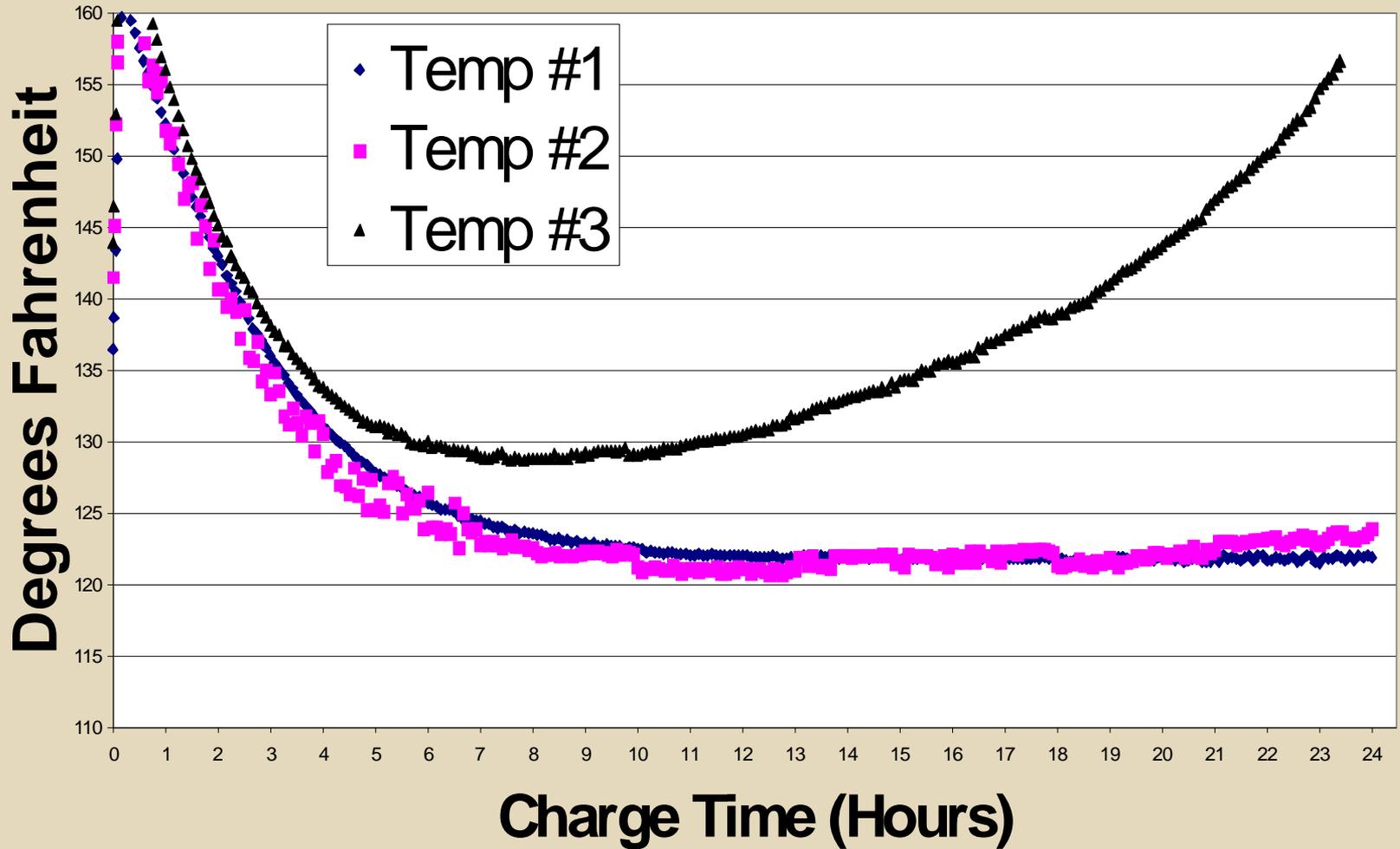
TR&F Charge Current (PL)

Coated Absorber, N3400G1-P



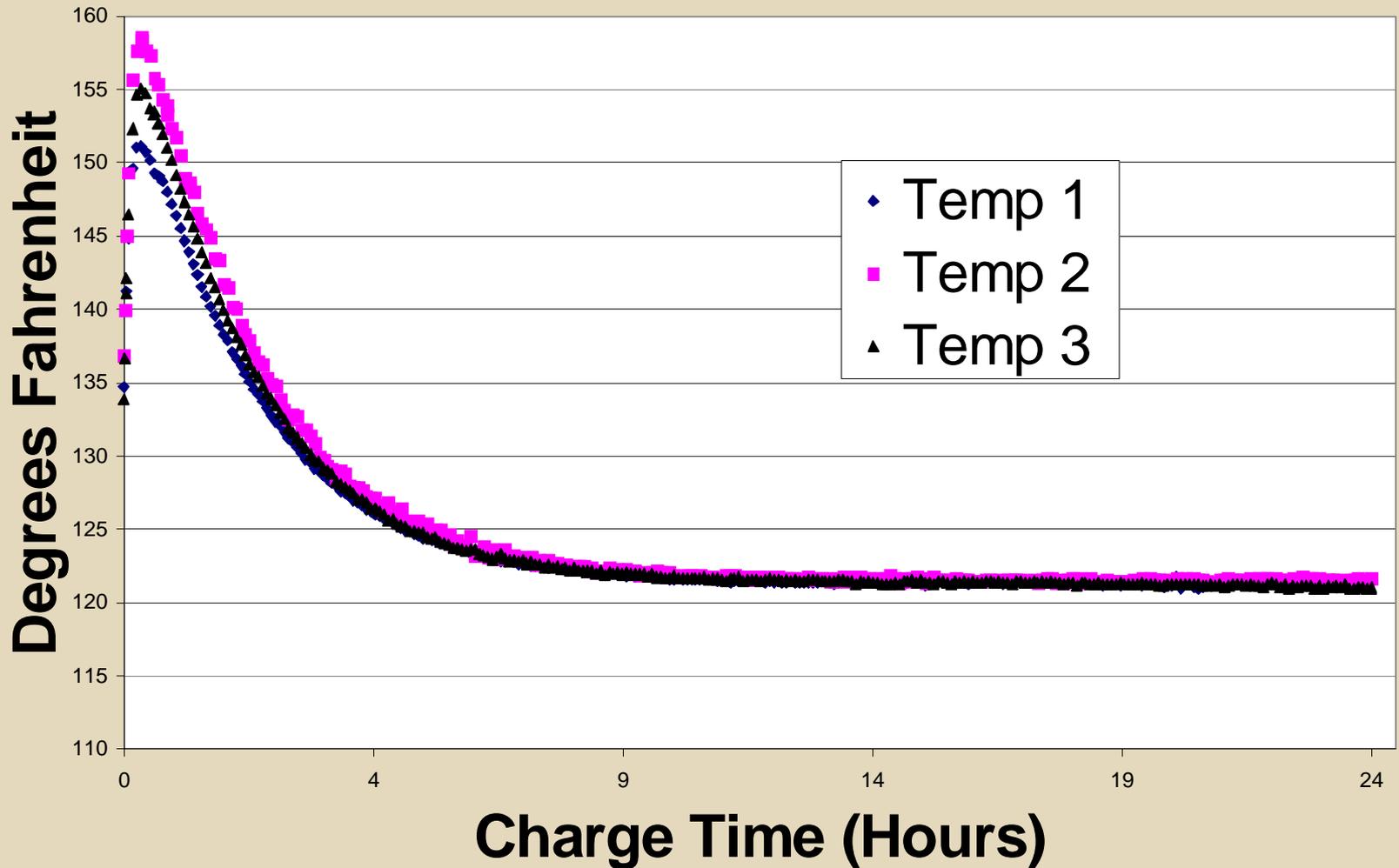
TR&F Charge Temperature (PL)

Grafted Absorber, N3400G1-P



TR&F Charge Temperature (PL)

Coated Absorber, N3400G1-P



TR&F Conclusions

- **Gas barrier porosity does not control charge stability if wetting agent is absent.**
- **Anionic wetting agent (CAS 12645-31-7) on absorber “caused” charge stability. It appears W.A. can transfer from absorber to the gas barrier.**
- **The wetting agent in the pores IS the gas barrier.**

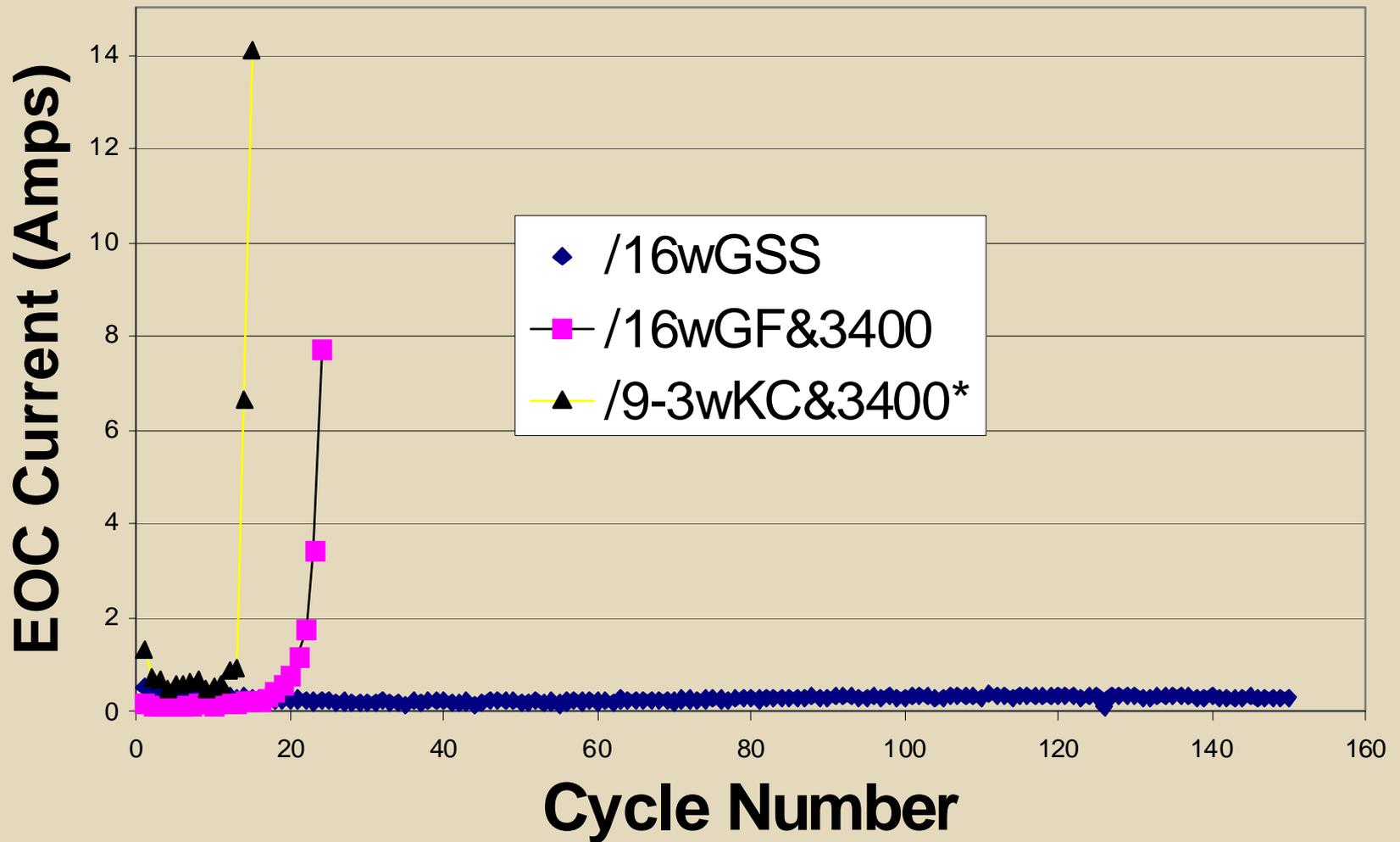
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How Good is Good?

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- **Type M81757/16 battery with GSS was subjected to continuous TR&F cycling**
 - **Results:**
 - **Battery's charge stability remained completely stable throughout test.**
 - **Testing was terminated after 226 days on test and completing 150 TR&F cycles**

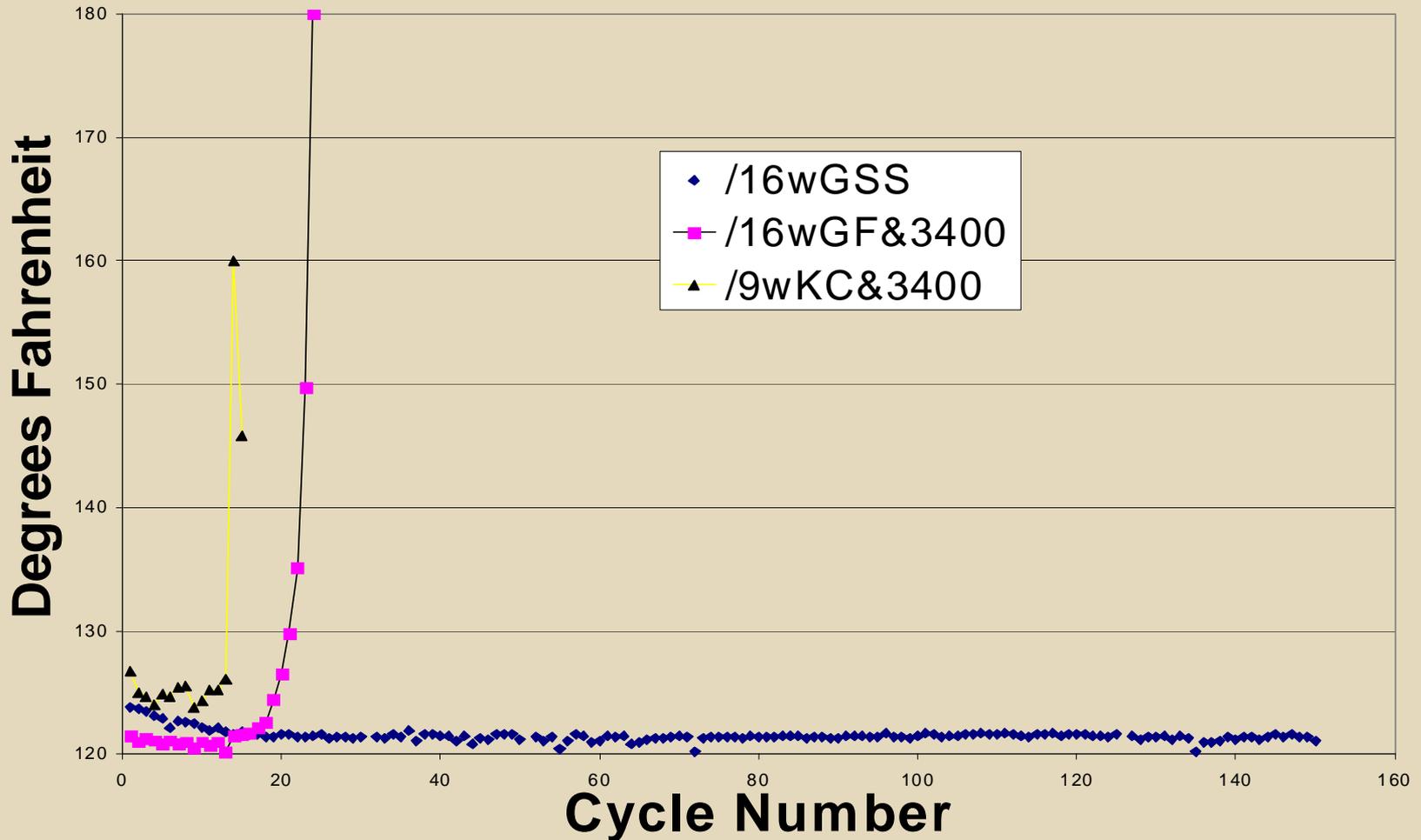
TR&F Cycling Comparisons

EOC Currents

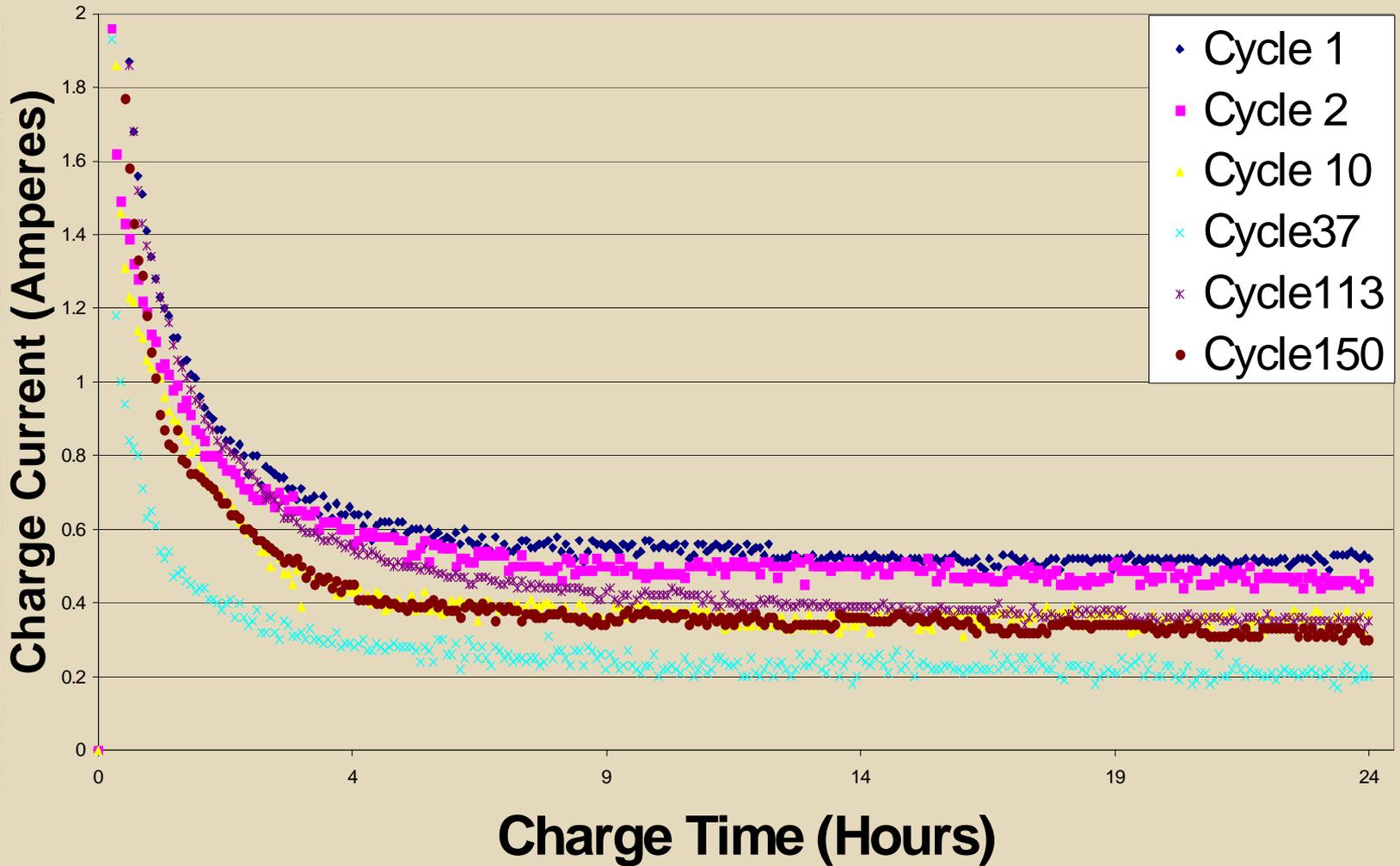


TR&F Cycling Comparisons

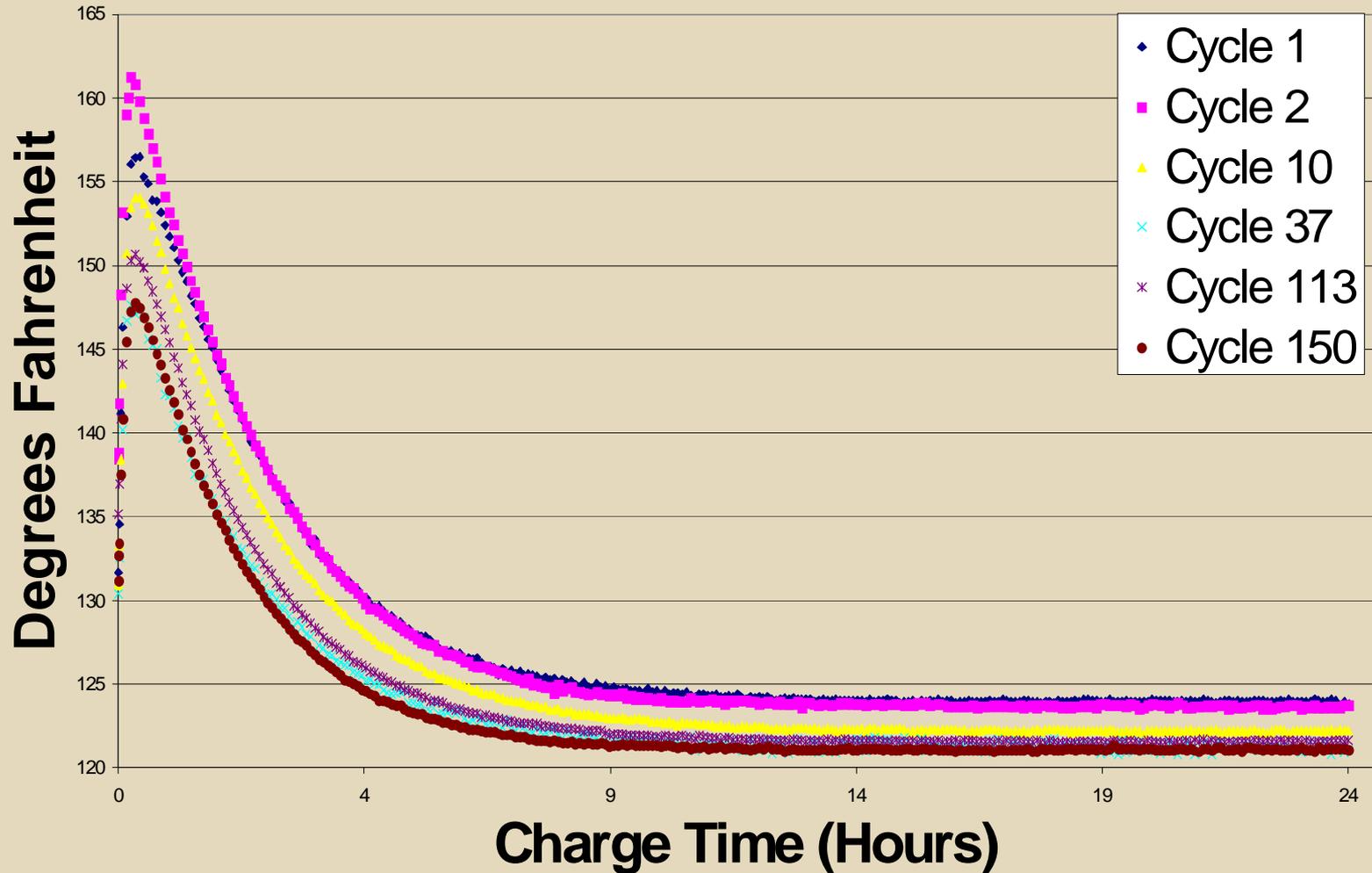
EOC Battery Temperatures



Charge Current Individual TR&F Cycles



Battery Temperature Individual TR&F Cycles



Specifying a Separator System

- **Wetting agent**
 - **Anionic**
 - **Unaffected by charge V using special test cell**
 - **Insoluble in electrolyte**
 - **Dispersant must leave no residue that can disperse into electrolyte**

Specifying a Separator System

- 
- **Gas Barrier**
 - **Polyolefin membrane**
 - **Thickness: 1mil ± 0.1 mil**
 - **Maximum Resistance: 18-millionohm-sq. inch**
 - **Porosity (35 to 40 G-s)**
 - **High enough for low resistance**
 - **Low enough to keep wetting agent in pores.**

Absorber:

- Hydrophilic (W.A. coated preferred)**
- Highly absorbent**
- High tortuosity for better protection (mbPP)**
- Weight: Governed by performance**



Any Questions?

Speaker POC Info



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TR&F Cycle Data by Days on Test GSS: A519 & TRC0950KG

