DETERMINATION OF MS2 BACTERIOPHAGE STABILITY
AT HIGH TEMPERATURES
USING THE INTEGRATED VIRUS DETECTION SYSTEM (IVDS)

Charles H. Wick
Ilya Elashvili

RESEARCH AND TECHNOLOGY DIRECTORATE

Patrick E. McCubbin

OPTIMETRICS, INC.
Bel Air, MD 21015-5203

Amnon Birenzvige

GEO-CENTERS
GEO-CENTERS, INC. - GUNPOWDER BRANCH

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**Abstract**

This is the third report in a series of reports that investigates the survivability of the MS2 virus in harsh environments. The first two reports study the effects of pH on the virus. In this report, we study the effect of elevated temperatures on the virus. The stability of the virus was studied by counting the number of virus particles using the Integrated Virus Detection System (IVDS). Test results show that the virus particles are relatively stable for a limited time (up to 0.50 hr) at temperatures as high as 63 °C. Longer exposure or exposure to higher temperatures, results in quick breakdown of the virus particles.

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Sandra J. Johnson

19b. **Telephone Number** (include area code)

(410) 436-2914

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14. **Abstract**

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EXECUTIVE SUMMARY

This report characterizes the MS2 bacteriophage sample, MS2D, under conditions of elevated temperatures at various times. The sample was analyzed using the detection stage of the Integrated Virus Detector System (IVDS).
PREFACE

This work was started in September 2004 and completed in June 2005.

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1. INTRODUCTION

The stability of viruses under different environmental conditions has always been a problem for microbiologists. Measuring this stability requires subjecting the virus to a harsh environment and monitoring the decay of the number concentration of virus particles with time. Until recently, measuring this stability had been exceedingly difficult. However, recently a new device—the Integrated Virus Detection System (IVDS), which can characterize and measure the number concentration of viruses, has been developed at the U.S. Army Edgewood Chemical & Biological Center (ECBC). The IVDS relies on measuring physical characteristics (size) of the virus instead of bacteriological means. This allows us to measure the number concentration of virus particles quickly. The IVDS is described in ECBC-TR-018. This report is the third in a series of reports that describe studies on the survivability of MS2 Bacteriophage under different environmental conditions. This report discusses the fate of MS2 at high temperatures. Two previous reports (ECBC-TR-472 and ECBC-TR-473) dealt with the fate of MS2 at high and low pHs.

2. EXPERIMENTAL PROCEDURES

A sample of MS2 stock solution (2 ml) was obtained from Dr. Deborah Kuzmanovic of the National Institute of Standards and Technology (NIST). The stock solution was diluted with 100 ml of distilled water and filtered by the Ultra Filtration (UF) subsystem of the IVDS using 100-K Dalton filters (for details on the IVDS and its different subsystems, the reader is referred to ECBC-TR-018 and ECBC-TR-468). The filtration system’s function is to remove any material with molecular weight smaller than the filtration system is set for (100K Daltons in this case), such as growth media, salt molecules, and proteins from the solution and leave a concentrated virus solution. The concentrated MS2 solution was added to about 20 ml of 20 mM ammonium acetate solution. Ammonium acetate is needed to increase the conductivity of the solution to allow it to be injected into the test module of the IVDS. The MS2 solution was concentrated again by the UF subsystem to a total of 2.5 ml of clean solution. The bacteriophage solution was then subjected to elevated temperatures. Periodically samples were extracted, quenched in an ice bath, and scanned by the IVDS.

3. RESULTS AND DISCUSSIONS

Analysis result of the original MS2 stock solution (after purification and concentration) is shown in Figures 1 and 2. Figure 1 shows the counts of the MS2 at around 24 nm. Figure 2, which is an expended scale of Figure 1, shows a peak at around 14 nm, which is a breakdown protein of the virus.
Figure 3 shows the number of virus particles counted as a function of temperature. At each temperature the residence time was 10 min. We can see that the number of virus particles decline somewhat (by about 25%) at temperatures of 62 °C-63 °C after 10 min. Further increase of temperature to 64 °C causes a sharp decline in the number of surviving virus particles, and the number of surviving particles continue to decrease as the temperature increased. The unexpected increase in the number of virus particles at 70 °C is probably due to some error.

Figure 4 shows the number of particles counted as a function of time when the MS2 stock solution was exposed to 63 °C. There is some drop in the number of virus particles in the first 30 min. In the next 30 min, there is a sharp drop in the number of particles. This reduction in the number of virus particles continues, until after 2 hr, <10% of the virus particles survive.

Figure 5 shows the number of particles counted after exposure to 70 °C for different time. There is about a 50% decline in the number count after 1 min of exposure. Exposure of up to 7 min does not change the survival rate of the virus particles. However, longer exposure (10 min) results in a sharp reduction in the survival rate.

Figures 6 and 7 show the concentration of lower molecular weight particles respectively in four particle size bins. In both cases, initially, we see a mix of break down products in all four size bins. With time, the number of particles in all four bins diminishes, probably as a result of further breakdown to a molecule of smaller molecular weight.

4. CONCLUSIONS

The Integrated Virus Detection System (IVDS) has proven itself as a useful tool to investigate the stability of virus particles in harsh environments. In this study, we have shown that MS2 can survive for a limited time (up to 30 min) at temperatures as high as 63 °C. Longer exposure or exposure to higher temperatures results in a quick drop in the survivability of the virus particles.
Figure 1. MS2 Initial Sample (6 Scan Average)

Figure 2. MS2 Initial Sample Showing Fragments at ~14 nm (6 Scan Average)
Stability of MS-2 as a function of temperature

Figure 3. Stability of MS2 as a Function of Temperature

Survivability of MS-2 at 63 degrees (centigrades)

Figure 4. Number of MS2 Particles as a Function of Time at 63 °C
Figure 5. Survivability of MS2 at 70 °C

Figure 6. Concentration of Breakdown Product at 63 °C
Figure 7. Concentration of Breakdown Products at 70 °C
LITERATURE CITED


