POROUS ELECTRODES I: NUMERICAL SIMULATION USING RANDOM NETWORK AND SINGLE PORE MODELS

by

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A Random Network model is introduced to stimulate the porous metal-electrolyte interface. This is the first time that an attempt has been made at defining a model for this system that maintains the random nature of the rough topology. A previously utilized model for this system - the single pore model is explored and extended. The models are compared, and the results of the Random Network model are found to be in qualitative agreement with the single pore models.
ABSTRACT

A Random Network model is introduced to simulate the porous metal-electrolyte interface. This is the first time that an attempt has been made at defining a model for this system that maintains the random nature of the rough topology. A previously utilized model for this system - the single pore model is explored and extended. The models are compared, and the results of the Random Network model are found to be in qualitative agreement with the single pore models.

Keywords: Electrode Impedance Battery Topography
INTRODUCTION

The porous metal electrode is a subject of intense theoretical and applied interest. Although widely utilized in electrochemical systems, especially in battery technology, very little is known in detail about the porous interface from first principles, aside from a general knowledge of how the porous electrode’s behavior deviates from flat-electrode behavior. Specifically, it would be useful to know how the complicated topology of the porous metal electrode contributes to the electrical properties of the interface. Previous attempts at modeling the porous electrode are presented in the review article by DeLevie [1]. The macroscopic model [2] treats the electrode-electrolyte system as a superposition of two continua, one of the electrode matrix and one of the solution matrix that fills all the unoccupied space. The solution is assigned an effective conductivity, and the interface is considered to have an effective capacitance per unit area. The single pore model is based on a one-dimensional representation of a pore as a transmission line (see figure 5) which represents a pore of "average pore length". These models, however, do not account for the random distribution of pores in a true porous solid, nor for a complicated surface topology. We have chosen to introduce a three dimensional model of the rough interface based on the random-network lattice that has proven to be so useful in representing conduction in inhomogeneous materials and related percolation problems [3,4,5]. A.c. impedance measurements are widely used experimentally for the in-situ characterization of electrochemical systems under equilibrium conditions. In conducting these measurements on porous electrodes, the complex topology of the electrode in various states of charge yields ambiguous data, not subject
to a unique interpretation. The computer simulation allows us to examine the impedance of the system as a function of given charge accumulation modes at the electrode-electrolyte interface. The model is not intended to portray the d.c. operation of the electrode, but rather to provide a means for understanding the interface. Using an IBM/370 computer, we construct our model and calculate its complex impedance, $Z(\omega)$, in the frequency range of one Hertz to one Megahertz. Based on this impedance spectrum, we model the system in terms of passive elements [6].

In this paper, we will present our model in detail, discuss preliminary results, and compare these results with the "DeLevie model". We will also discuss further work being done to utilize this model for a more detailed characterization of the interface.
I. THE "RANDOM" NETWORK MODEL

We consider a simple system, in which a 1 cm$^3$ porous metal electrode is immersed in electrolyte, along with a counter-electrode. Only the bottom (1 cm$^3$) surface of the electrode is exposed to the electrolyte. The electrode is allowed to discharge, allowing for a build-up of a semiconductor on the metal surfaces exposed to the electrolyte. To explore the electrical behavior of this system, we note that we may define a local conductivity, $\sigma(F)$, the bulk conductivity at point $F$ in our system. Rather than attempt to solve for the current distribution using the electrostatic equations for the continuous case, we transform the problem into one requiring the solution of a discrete set of Kirchoff's Law equations (a finite difference approximation to the continuum problem) [3]. We subdivide our bulk system into a regular cubic mesh of points $(r_i)$ ($i=1,S$) and assign to each branch of the mesh a conductivity that represents the conductivity of the surrounding bulk in that region of space. Let the conductance of the circuit branch that connects node $i$ and $j$ be denoted as $g_{ij}$, the voltage at node $i$ be $V_i$, and the net current into node $i$ be $I_i$. We then have $(S-1)$ equations of the form:

$$I_j g_{ij} \left( V_i - V_j \right) = I_i \quad (1)$$

or in matrix notation:

$$G \cdot V = I$$

where $G$ is the conductance matrix ($G_{ii} = I_j g_{ij}$ and $G_{ij} = -g_{ij}$), $V$ is a vector containing the voltages on each node with respect to ground, and $I$ is a vector containing the net current inputs to each node. We may solve these simultaneous equations for the $V$'s on all the nodes,
and for the total impedance of the system. A detailed presentation of
the algorithm that we have employed follows.

We define a three dimensional cubic lattice, \( A(i,j,k) \), where \( i, j \) and
\( k \) range from 1 to \( N \), and \( N^3 = S \), the total number of nodes. \( N \) is
chosen so that it is sufficiently large to approximate an infinite system.
Construction of the electrode is simulated using a random number
generator to place "metal" at various points in the lattice, until the
desired porosity is reached. A check is made of the resulting electrode
to insure that it is continuous and that no pieces of metal are
"hanging" in mid-air. Any disjointed pieces of metal are removed, and
are randomly replaced on the remaining available lattice points. This
process is repeated until a continuous electrode of the desired porosity
is obtained. Lattice plane \( k=1 \) is defined as electrolyte, to represent
the solution; the electrode itself begins at lattice plane \( k=2 \). The
electrolyte is allowed to "seep into" the pores in the electrode until all
accessible pores are filled, by searching for continuous electrolyte
paths from the \( k=1 \) plane throughout the electrode. Any point on the
lattice which has been left empty (i.e. no metal was placed there, and
electrolyte was precluded from penetrating to that site) is considered to
be "air".

Each lattice point represents a node in a three dimensional circuit
network (See figure 1), and each pair of nodes defines a circuit branch
in the network. The impedance of each branch is determined by the
characteristics of the two surrounding nodes. Thus, for example, if
\( A(1,1,1) \) is electrolyte and \( A(1,2,1) \) is also electrolyte, the circuit
branch defined by those nodes will consist of two series resistors of
resistance \( R_e \) (the resistance of a microscopic section of the
electrolyte). If $A(1,1,2)$ is metal and $A(1,2,2)$ is metal, then the circuit branch defined by those two nodes will be two series resistors of value $R_m$ (the resistance of a microscopic section of the metal). Finally, the branch defined by $A(1,1,1)$ - electrolyte, and $A(1,1,2)$ - metal, is represented by an $R_m$ and an $R_e$ resistor in series, with the addition of a parallel $R$-$C$ element in series with them to represent the semiconductor-electrolyte interface that results (see Figure 2). Any circuit branches leading into nodes that are defined as "air" are taken to be of infinite resistance.

The values that we have chosen to use as our unit impedances for the individual components of metal, electrolyte, and semiconductor are presented in Figure 2. These values were obtained by taking the values of bulk impedance for a $1 \text{ cm}^3$ sized sample, (we chose numbers characteristic of Zn and ZnO) and scaling them down to the magnitude of a unit pore size- approximately $10\mu$ in diameter. We define a characteristic time constant of the interface, $\tau = R_p C$. These components are assembled into a system, as described above (see figure 3) and the total impedance of the system is calculated. The total impedance is then rescaled by an appropriate factor that normalizes the whole system to a $1 \text{ cm}^3$ size. Thus, an attempt is made to obtain impedance data with numbers representative of that which might be encountered in a real system.

Once the electrical network has been defined, the computer sets up the Kirchoff's law equations. Due to the fact that our model includes reactive circuit elements, the quantities $G$, $V$, and $I$ in equations (1) are complex, and may be represented as $2S-1$ equations of the form:
\[
\begin{pmatrix}
Gr & -Gi \\
Gi & Gr
\end{pmatrix}
\begin{pmatrix}
Vr \\
Vi
\end{pmatrix}
=
\begin{pmatrix}
Ir \\
II
\end{pmatrix}
\] (2)

where the subscripts \( r \) and \( i \) stand for the real and imaginary components respectively.

A known current is sent uniformly into all the bottom nodes (so as to eliminate the "edge effects" that would result if the current were sent only into one node), and the impedance of the system is calculated between one node on the electrolyte plane \((1,1,1)\) and one node of the top electrode plane \((N,N,N)\) by solving for the \( V \)'s on all the nodes.

For a system of \( N=15 \) (a lattice of size \( 15 \times 15 \times 15 \) with 3375 nodes) there are up to 6748 equations to be solved. This is a formidable computer problem. Indeed, the task would be prohibitive without taking advantage of the fact that the conductance matrix, \( G \), is a sparse-symmetric coefficient matrix.

A number of techniques exist for the solution of this class of problems. A widely used method for solving the simultaneous equations in the Random Resistor lattice is the Gauss Seidel iteration procedure with over-relaxation \([5]\). This is a very economical method, in both execution time and storage space. A serious drawback of this technique is the relatively strict requirements that it places on the coefficient matrix in order for convergence to be achieved, namely that the matrix be either diagonally dominant, or at least positive definite \([7]\). For Kirchoff's law problems with purely resistive components, diagonal dominance is assured, since the diagonal elements of the conductance matrix are simply the sum of the off diagonal elements. In our problem, with complex impedances, we find that at frequencies for

- 7 -
which the real conductance approaches the same order of magnitude as
the imaginary conductance, the Gauss-Seidel procedure does not
converge. This is due to the form of the equation in (2) where there
are many more off diagonal elements due to the $G_i$ values. When these
imaginary components are large, the matrix is no longer diagonally
dominant, nor positive definite.

For most of our computations, we have resorted to the use of a
Gaussian elimination routine which takes advantage of both the
sparseness and the symmetry of the $G$ coefficient matrix [8]. Only
the nonzero elements in the upper half triangle of the matrix are
stored, and an efficient pivoting strategy is chosen to minimize non-
zero matrix fill during the pivoting and to minimize the number of
multiplications required in the solution. In practice, the technique is
approximately an order of magnitude more expensive to use than Gauss-
Seidel in both speed and storage requirements. The advantage of using
the Gaussian elimination technique is that a solution is guaranteed for
almost any problem. The solutions obtained from the Gauss-Seidel
procedure (in the range in which convergence is reached) are in
complete agreement with the results obtained using the modified
Gaussian elimination technique.

Figure 3 presents a system constructed with a porosity of 0.30, and
the resulting impedance of the system as a function of frequency is
plotted in figure 4. We have identified three basic regions of interest
in the frequency range spanned by our results. Region I is the low
frequency range, in which our system seems to behave as a parallel R-
C combination (See figure 4). Region II is in the mid-frequency range,
with an impedance spectrum that cannot be represented using passive
elements, and Region III is representative of a series R-C combination. The variation of the impedance data as a function of system size is presented in Figure 5 from size N=5 to N=15. For the results presented here, N=15 was used to ensure proper statistical representation of an infinite system.

II. THE SINGLE-PORE MODEL

It is interesting to compare, at least qualitatively, the results from our three dimensional model with the results obtainable from the single pore model previously mentioned [1].

De Levie, following on the heels of other researchers (Daniel' Beck, [9] et. al.), suggested that each pore in a porous electrode may be thought of as having a uniformly distributed electrode and electrolyte resistance throughout its length. As such, the single pore may be modeled as in figure 6a, where R is the resistance per unit length of the electrolyte solution inside the pore, and Z is the impedance of the electrode-electrolyte interface. The current-voltage relationships of this circuit may be expressed as a differential equation and a solution may be obtained for the total impedance of the circuit as [1]:

\[ Z_\infty = (RZ)\frac{1}{2} \coth(\rho L) \]

where R and Z are defined above, \(\rho = (R/Z)^{\frac{1}{2}}\), and L is the length of the pore. This is the main, and currently utilized [10], result from this model. The main characteristics of the calculation are: (1) Although flat electrode impedances vary as a function of Z, porous surface impedances will be dependant on \((RZ)^{\frac{1}{2}}\), and (2) the contribution of the electrode surface deep inside the pore is negligible, \(1/\rho\) becoming
the characteristic "penetration depth". If \( \rho L >> 1 \) the pore behaves like a semi-infinite one.

In order to utilize these results for comparison with the Random Network Model, we must introduce a value for \( Z \), and we must also combine the single pore transmission-line impedances to form a complete rough electrode. The former is accomplished by defining the electrode-electrolyte interface as a standard parallel R-C model, as depicted in figure 6b. We have used the same numerical values for the individual components as those in the Random Network Model (here they are shown already normalized to the 10\( \mu \) single pore size). As a result,

\[
Z = r_s + \frac{R_p}{1 + \omega^2 \tau^2} - j \frac{\omega T R_p}{1 + \omega^2 \tau^2}
\]  

(4)

where \( \tau = R_p C \). Substituting (4) into (3) and simplifying, we get:

\[
Z_0 = \sqrt{\frac{R P}{2}} \left\{ \left[ (\beta + \alpha)^2 + \frac{2 \sin 2\theta}{e^{2x} - e^{-2x}} (\beta - \alpha)^2 \right] + j \left[ (\beta - \alpha)^2 - \frac{2 \sin 2\theta}{e^{2x} - e^{-2x}} (\beta + \alpha)^2 \right] \right\}
\]

(5)

where we have defined the following quantities:

\[
P = \frac{R_p}{1 + \omega^2 \tau^2} \]

\[
\alpha = 1 + \frac{r_s}{P}
\]

\[
\beta = (\alpha^2 + \omega^2 \tau^2)^{1/2}
\]

\[
x = L \left( \frac{R_p}{P} \right)^{1/4} \frac{1}{\sqrt{2\beta}} (\beta + \alpha)^{1/4}
\]

\[
\theta = L \left( \frac{R_p}{P} \right)^{1/4} \frac{1}{\sqrt{2\beta}} (\beta - \alpha)^{1/4}
\]
A log-log plot of the relationship between $Z$ and $f$ (for a single pore) is presented in figure 7. It is interesting to note the similarity in shape between the impedance spectrum of the single-pore model (even though we are only considering a single pore) and that of our Random Network Model (figure 4). The advantage of considering a single pore at the present stage is that it is the only system for which we have an analytic result (equation 5), which may be used to derive expressions for the impedance in different frequency ranges.

We define three frequency regimes:

(a) The low frequency range ($\omega t << 1$):

\[
\text{Re}(Z_0) = K \\
\text{Im}(Z_0) = \omega t
\]

where $K$ (independent of frequency) is given as:

\[
K = \frac{e^{2x} - e^{-2x}}{\sqrt{RR_p} e^{2x} + e^{-2x} + 2}
\]

and $x$ is:

\[
x = L \left(\frac{R}{R_p}\right)^{\frac{1}{2}}
\]

The notable features of equation (7a) are the frequency independence of the Real part and the slope 1 dependence (on the log-log plot) in the imaginary part of the impedance (see region I on figure 7).

(b) The intermediate frequency range ($\omega t >> 1$ but $\frac{r_g \omega t}{R_p} << 1$):

\[
\text{Re}(Z_0) = \text{Im}(Z_0) = \sqrt{\frac{RR_p}{2}} \frac{1}{\sqrt{\omega t}}
\]
This results in the slope -1/2 line in region II of figure 7.

(c) The high frequency range \((\omega t >> 1 \text{ AND } r_s\omega t/r_p >> 1)\):

\[
\text{Re}(Z_0) = \sqrt{R_s} \\
\text{Im}(Z_0) = \frac{k\sqrt{R_s}}{\omega t} \frac{R_p}{\omega t}
\]

which results in a frequency independent real impedance, and the slope -1 line in the imaginary impedance as can be seen in region III of figure 7.

In addition to the frequency limits, it was also assumed that \(R_s/R_p << 1\), and that \(R > r_s\) by at least an order of magnitude (which is usually the case).

To combine the single pores into a unified metal-electrolyte system, we extend De Levie's model slightly. We define an x-y plane and divide it into \(N^2\) squares, and randomly place a rectangular-width pore on the x-y plane with its length extending in the z direction. The length of the pore, \(L\), is also randomly chosen to be from 1 to \(N\), and the pore width, \(W\), is selected so that there is a Gaussian distribution about \(W = 2\) (to prevent a few very wide pores from "taking over" the whole electrode). More random sized pores are randomly placed on the x-y plane until the desired porosity is reached. We may then solve for the total impedance of the system by adding up all the parallel impedances due to the individual pores, using equation (5) to represent the impedance of each pore. The impedances of the non-porous sections of the electrode are also added on to the total impedance.

The results, for a porosity of 0.3 and \(N=40\), are presented in figure 8. The criteria for choosing \(N\) was based on considerations similar to
those used to choose $N$ in the Random Network model, namely that the results be independent of $N$ and statistically independent of the random number. This phase of the calculation was performed on an IBM PC.

III. DISCUSSION OF RESULTS

There is a qualitative agreement between the Random Network model and the single-pore model. A comparison of the passive elements representation of the two systems (figure 4 and figure 8) also yields order-of-magnitude quantitative agreement. One discrepancy between the two models is the definite shift in the location of the "mid-frequency peak" in Figure 8 as compared to the Random Network and single-pore results. A major difference between the two models in construction, is the fact that in the Random Network the pores interact with each other as in a true porous electrode, while in the De Levie model each pore is kept separate from the others, and is a separate channel. The favorable comparison of the two models would seem to bear out De Levie's assertion that "the disregard of crosslinks [between the pores] presumably does not introduce a significant error [1]. Nevertheless, this can be a useful starting place for an explanation of a discrepancy between the two.

The behavior of the system impedance in the Random Network model may be somewhat inferred from the values of the constituent components that make up the system. The low-frequency (real and imaginary) impedance in figure 4 reflects the values of $R_p$ and $C$ of the unit (semiconductor) interface, although the 300Ω value of $Re(Z)$ is noticeably lower than the 1KΩ $R_p$. In the high-frequency regime, $C$ dominates, totally shunting out $R_p$, the only contribution to the Real impedance
being $R_e$. This provides us with an indication that the total behavior of the porous-metal system may be inferred from the values of the components that make up the complicated surface topography. Our ultimate goal is to generate a detailed picture of the macroscopic interface from the impedance data of the microscopic components that make up the system. This will be considered further in a subsequent paper, along with an exploration of the physical significance of the passive elements model of the entire system.

We will also examine the correlation between the single pore analytic results (in the various frequency limits) and the Random Network model. Only a single porosity has been considered thus far, but an exploration of the Random Network model at different porosities is in order. Although some rough surfaces may exhibit a conduction percolation threshold [11,12], we do not expect to see any classical percolation in our system. This is due to the fact that we have constrained the system to be above the percolation threshold by the requirement that the metal electrode be continuous. We would, however, expect to see some less dramatic system dependence on porosity.
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REFERENCES


LIST OF FIGURES

1. Representation of the 3-D Lattice in terms of Circuit elements. Figure 1b shows our model in terms of a checkerboard pattern of bulk components of metal, electrolyte, semiconductor, and air. Figure 1a depicts our representation of (a corner of) Figure 1b in terms of a network of circuit elements. $Z_{sc}$ is the impedance of the semiconductor layer on the interface (see Figure 2).

2. Equivalent circuits used to represent the microscopic components of the system (metal, electrolyte, semiconductor, etc.).

3. The Random Network lattice used to generate the data for Figure 4. At this porosity of .3, the electrolyte does not fully penetrate the lattice. (Actually, the penetration of the electrolyte into the pores seems to conform to classical percolation theory. At porosities above .35 the liquid starts to penetrate through the electrode).

4. Impedance spectrum from the Random Network model at a single porosity (0.3), and a single random number. Three frequency regimes are identified, and we have modelled two regimes with a simple passive element model. The bars on each curve represent the variations in the data when various random number seeds are used to construct the lattice. For each random number seed, the shape of the impedance spectrum is almost identical.

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6. The De Levie model of a single pore. Figure (6a) presents the general transmission-line model. (6b) presents the value we have chosen for $R$, the electrolyte impedance, and our specific model for $Z$, the electrode-electrolyte interface inside the pore.

7. Impedance spectrum from the DeLevie model - single pore. The sections of the curve labelled (1) are areas in which the imaginary impedance plot is of slope=1 on the log-log plot, and the impedance in these ranges may be modeled by the two passive elements models shown. The section of the curve labelled (2) has a slope=1/2; See equations 7a - 7c.

8. Impedance data from Delevie model - randomly distributed pores. Porosity = 0.3, $N=40$. 

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Figure 1
Metal \[ R_m = 6 \times 10^{-6} \Omega \]  
Electrolyte \[ R_e = 5 \Omega \]  
Semiconductor \[ R_s = 0.035 \Omega \]  
\[ R_p = 1 \text{K}\Omega \]  
\[ C = 1 \mu\text{F} \]  

Figure 2
Figure 5
(6a)

\[ R = 5 \times 10^6 \Omega \]

\[ R_p = 10^9 \Omega \]

\[ C = 1 \text{pF} \]

(6b)
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