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AFML-TR-75-17

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ML-101 THERMAL CONTROL COATING SPACEFLIGHT EXPERIMENT

ELASTOMERS AND COATINGS BRANCH
NONMETALLIC MATERIALS DIVISION

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AUGUST 1975

TECHNICAL REPORT AFML-TR-75-17
FINAL REPORT FOR PERIOD JANUARY 1972 - JANUARY 1974

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This report was submitted by the Elastomers and Coatings Branch, Nonmetallic Materials Division, Air Force Materials Laboratory, Air Force Systems Command, Wright-Patterson Air Force Base, Ohio under job order 73400703. Daniel E. Prince (AFML/MBE) was the laboratory project engineer.

This technical report has been reviewed and is approved for publication.

Daniel E. Prince

Daniel E. Prince
Project Monitor

FOR THE COMMANDER

M. L. Minges

M. L. Minges
Chief, Elastomers and Coatings Branch
Nonmetallic Materials Division

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Block 20 (continued)

Based on selected data from over 5000 revolutions covering a period of one year, it was found that all the coatings initially degraded to a greater degree than expected, possibly due to contamination. The most stable coatings were optical solar reflectors and the least stable was a white α -Al₂O₃ pigmented coating. An experimental fabric material showed greater stability than state-of-the-art white coatings.

alpha

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AFML-TR-75-17

FOREWORD

This report was submitted by the Elastomers and Coatings Branch, Nonmetallic Materials Division, Air Force Materials Laboratory, Air Force Systems Command, Wright-Patterson Air Force Base, Ohio. The work was performed under Project 7340, Job Order 73400703. Daniel E. Prince (AFML/MBE) was the laboratory project engineer.

Contributions by Lt. M. Blessing, AFML/MBE, to data reduction and analysis were greatly appreciated.

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LIST OF SYMBOLS

<u>SYMBOL</u>	<u>DESCRIPTION</u>	<u>UNITS</u>
A_e	Area of edge of sample	ft^2
A_n	Cross-sectional area of sample	ft^2
C_p	Specific heat of sample	$\text{BTU/lbm-}^\circ\text{R}$
F_{e-m}	Shape factor from edge of sample to Mylar holder	---
F_{e-s}	Shape factor from edge of sample to space	---
G	Solar heating constant	$\text{BTU/ft}^2\text{-hr}$
H_p	Incident planetary heating flux	$\text{BTU/ft}^2\text{-hr}$
H_r	Incident planetary albedo heating flux	$\text{BTU/ft}^2\text{-hr}$
H_s	Incident solar heating flux	$\text{BTU/ft}^2\text{-hr}$
I	Current	Amps
K	Thermal Conductance of sample	$\text{BTU/hr-}^\circ\text{R}$
M	Mass of sample	lbm
Q_{ne}	Net heat exchange rate between sample and Mylar	BTU/hr
Q_{rs}	Total solar heat incident on sample edge	BTU/hr
T_s	Temperature of sample	$^\circ\text{R}$
T_m	Temperature of adjacent Mylar	$^\circ\text{R}$
T_n	Temperature of adjacent samples	$^\circ\text{R}$
T_{ref}	Temperature of sample holder base	$^\circ\text{R}$
α_s	Absorptance of sample	---
α_m	Absorptance of adjacent Mylar	---
ϵ_e	Emittance of sample edge	---
ϵ_{th}	Emittance of sample	---
ϵ_m	Emittance of Mylar holder	---
ϵ_n	Emittance of adjacent sample edges	---
σ	Stephan-Boltzmann constant, 0.1714×10^{-3}	$\text{BTU/ft}^2\text{-}^\circ\text{R}^4$

SECTION I
INTRODUCTION

The temperature control of spacecraft is one of the most challenging technical problems confronting spacecraft designers today. The ultimate objective of thermal design is to ensure that the spacecraft will operate within the prescribed temperature ranges defined by the temperature limitations of the spacecraft's materials and components, such as detectors, batteries, solar cells, and electronics, which operate properly and efficiently only if maintained within specific temperature ranges.

The surface temperature of a spacecraft depends primarily upon the solar radiation absorptance (α_s) of the surface, the radiation of solar energy from the surface or surface emittance (ϵ), and the generation of heat in the body of the spacecraft. Secondary parameters that influence the surface temperature of a spacecraft are material properties such as thermal conductivity and specific heat of components. Solar absorptance is defined as the ratio of solar energy absorbed by a surface to that absorbed by a blackbody, while emittance is the ratio of the energy radiated from a surface to that radiated from a blackbody at the same temperature.

Two techniques are used for regulating spacecraft temperatures: active systems and passive systems. Active systems usually employ electrical power and moving parts which can be adjusted providing a constantly varying surface in terms of optical properties. Passive systems, in contrast, require neither electrical power nor moving parts and rely on surfaces with appropriate thermophysical properties; namely, solar absorptance, α_s and infrared emittance (ϵ). Regardless of what combination of systems is used, the thermophysical properties of the entire spacecraft surface must be known since it is the average surface properties which determine the spacecraft equilibrium temperature.

Because of the large role surface coatings play in the passive temperature control of spacecraft, extensive programs have been conducted

to develop space-stable, low α_s/ϵ coatings. Future spacecraft requirements call for the development of coatings with an initial solar absorptance of less than 0.10 and initial emittance of more than 0.90, with long-term stability (5-10 years) in the space environment; i.e., increase in α_s of less than 10%. Low α_s/ϵ surfaces can be divided into three divisions: (1) those represented by diffusely reflecting white pigmented coatings, or paints; (2) those represented by second surface mirrors that obtain their low solar absorptance from the metallic reflection of an aluminum or silver mirror and their emittance from a selectively transmitting substrate such as quartz or FEP Teflon; and (3) those represented by newly emerging white fabric cloth materials.

Many of the previous materials have been extensively subjected to in situ laboratory simulation exposures, however, most important to the thermal designer is knowledge of actual spaceflight data and its correlation with laboratory data. This report describes an Air Force Materials Laboratory coatings spaceflight experiment whose primary objective was to determine the effects of the space environment on the equilibrium-radiative properties of selected candidate temperature control coating systems. Its secondary objective was to correlate the actual space degradation with simulated laboratory data.

SECTION II
SATELLITE ORBITAL PARAMETERS AND EXPERIMENTS

The Space Test Program (STP) is a Department of Defense program for the spaceflight of research and development payloads as well as operational payloads. The STP P72-1 flight vehicle was launched from Vandenberg Air Force Base on an Atlas-F/BII combination. Liftoff occurred at 2012.5 Z on 2 October 1972. The satellite orbit was sun-synchronous with a nominal inclination of 98.34 degrees (polar) and a nominal circular altitude of 400 nautical miles. The initial orbital parameters and the orbital parameters as of September 30, 1973 are compared to the planned nominal parameters for the 72-1 vehicle in Table I.

TABLE I
SATELLITE ORBITAL PARAMETERS

<u>Flight Parameters</u>	<u>Planned Nominal</u>	<u>Actual (Initial)</u>	<u>Actual (Sep 1973)</u>
Maximum altitude (nm)	409.19	411.41	411.21
Minimum altitude (nm)	398.13	400.37	399.27
Eccentricity	0.00041375	0.00039	0.00055995
Inclination (deg)	98.348	98.424	98.433
Period (min:sec)	99:40.158	99:39.465	99:39.217
Rev. No.		1	5191

The 72-1 satellite included a payload of six experiments. These are listed in Table II together with the type of experiment, principal investigator, and appropriate organization.

TABLE II
SATELLITE PAYLOADS

<u>Experiment No.</u>	<u>Investigator & Organization</u>	<u>Description</u>
ML- 01	D. E. Prince Air Force Materials Laboratory	Thermal Control Coatings
RTD-802	A. Y. Harper ABMDA-Huntsville	Radar calibration target
ARPA-501	J. B. Reagen Lockheed Research Laboratory	Gamma-spectrometer
NNRL-114	C. S. Weller Naval Research Laboratory	Extreme UV radiation
SSD-988A	D. R. Croley, Jr. Aerospace Corporation	Low altitude particles
SSD-988B	P. Rothwell AFCRI: Hanscom Field	Low altitude particles

SECTION III
STP FLIGHT P72-1 SATELLITE SYSTEMS

1. ELECTRICAL POWER SYSTEM

The satellite electrical power is supplied by a solar cell array that generates electrical energy required to operate the satellite subsystem and experiments and to charge two nickel-cadmium batteries that supply electrical power during solar eclipse. The electrical power is distributed directly from the main power bus or switched by a relay control box by uplink command.

2. TELEMETRY, TRACKING, AND COMMAND SYSTEM

The telemetry system transmits all data in PCM/PM at 8 or 64 Kbps real-time and recorded data at 64 or 512 Kbps. A 2-watt SGLS reference transmitter is used for the 8 Kbps rate and a 10-watt data transmitter is used for the 64 and 512 Kbps data rate. The processor furnishes multiplexing, analog to digital conversion (where required) formatting, clocking, and synchronizing signals for the TT&C systems including the programmer. Three tape recorders are supplied for reliability and high data capacity for unusually high peak load periods. All other systems are dual redundant. A command system which is also dual redundant, includes the Command Receiver Demodulator and the Decoder/Programmer which provides 375 MHz system real-time uplink command capability and stored program routines for flexible payload and telemetry system control.

3. STABILIZATION, ORIENTATION, AND ATTITUDE DETERMINATION

The STP P72-1 satellite operates in a sun-synchronous orbit with the spin axis nominally normal to the orbit plane within five degrees with a goal of one degree. The satellite spin speed is controlled within 12 ± 2 rpm.

The Satellite Attitude Control System (ACS), makes use of magnetic torque coils to control the attitude and spin rate of the highly stable spinning vehicle. Body fixed earth sensors provide data for primary

determination of attitude and rate. Ground commands control the torque coil current to provide correction of observed errors. Wobble damping is provided by a passive fluid loop damper.

Measurement of the earth's magnetic field is provided by two three-axis magnetometers mounted on the forward antenna boom. The associated Magnetometer Electronics Assembly provides proportional outputs that are multiplexed into the telemetry stream. The outputs are used as a polarity sense switch for spin control and provide data for the backup attitude and spin rate determination.

A dual winding iron-core precession coil is used to maintain the spin axis within five degrees of the orbit normal. The coil is operated continuously to provide orbit regression rate. Periodic updates by ground control commands to the multitap precession coil eliminate any buildup of errors caused by random disturbance torques or bias in the satellite magnetic field. Attitude information is provided by a redundant horizon sensor system. A spin rate control coil, identical to the precession coil, is provided to update the satellite spin speed on command from the ground. The coil is automatically commutated by radial magnetometer output signals. Backup capability is provided by cross strapping a backup magnetometer through the PCM Processors to the dual spin coil windings.

SECTION IV
EXPERIMENTAL PACKAGE

The complete basic design and initial evaluation data are contained in References 1 and 2. Sample holders used on the 72-1 satellite are Numbers 7 and 12. Each sample holder weighs approximately 11 ounces (300 grams), occupies approximately 40 cubic inches, and will accommodate five test samples and one reference sample. The samples or thermal control coating specimens are 15/16 inch in diameter with the thickness depending on coating applied and substrate. The sensor disks used for the sample holders are Trans-Sonics T4086-3 platinum resistance thermistors cemented in stainless steel sensor disks. Each platinum thermistor has been calibrated to an accuracy of $\pm 0.5^{\circ}\text{C}$ at temperatures of -193°C , 0°C , and 100°C . The constants for the temperature resistance relationship of platinum are supplied with each sensor and are used in translating resistance to temperature. A sensor (Trans-Sonics T4086-3) is attached to the bottom surface of the holders as a temperature reference for correction of conduction losses from the sample sensors.

The sensors are supported upon 50 layers of super-insulation (1/4 mil Mylar aluminized on one side) by a continuous Dacron cord under tension. The individual samples and sensors are thus thermally isolated from each other with the primary conduction loss through the thin sensor leads. A temperature sensor is placed adjacent to the electrical connector to permit correction of flight data for this loss. Ten pounds of tension is placed on the Dacron thread used to bind the sensors in place during fabrication. Each sensor is held at three places by the cord and is cemented to the cord at these points. The sensors rest upon a three mil sheet of Mylar (aluminized side down) which acts as a cover for the Dacron cord.

The measurement of the resistance of the sensors is obtained by supplying an external voltage (90 millivolts) to the sensors and measuring the voltage drop across each sensor. The voltage drop across each thermistor is in turn a linear function of resistance which in turn is

a linear function of temperature for this range. Through the calibration data for each thermistor, the temperature can be determined. Room temperature calibration was performed in both real-time and tape recorder modes in the pre-launch satellite acceptance test phase. This calibration data obtained prior to launch together with the installed flight hardware is being used for the data reduction and then calculation of the desired thermophysical values.

Figure 1 shows the ML-101 sample holder with six coatings attached and Figure 2 shows the ML-101 sample holders attached to the partially completed 72-1 satellite.

SECTION V
EXPERIMENTAL THERMAL CONTROL COATINGS

A total of 12 thermal control coating samples were used on two sample holders for the ML-101 spaceflight thermal control coating experiment. The sample holders are designated 7 and 12 on all experiment references.

7-1, ZnO (Silicate Coated)/Methyl Silicone (S-13G)

Formulation, processing, and application was performed by IITRI* in accordance with IITRI Report 4-6053-5-7; Batch No. C-133. The S-13G was sprayed to a thickness of 8.0 + 1.0 mils.

7-2, SiO₂ Cloth Fabric

2024-T3 unclad 15/16 inch diameter 0.032 inch thick aluminum substrate was prepared as follows: (1) faced on 240 SiC stationary grinder, (2) prepolished with jeweler's rouge and buffed, and (3) final polished using Buehler's Magomet (M_gO) on Microcloth. The silica cloth fabric is a 12 harness double layer satin weave woven from 450/1/0 Suprasil yarn. Polyvinylalcohol sizing was removed by heating, and filament diameter size was six microns. The fabric construction consists of 150 ends/inch by 240 picks/inch. The fabric thickness is nominally 0.005 inch and weighs 4.65 oz/yd². The fabric was bonded to the polished aluminum substrate with General Electric SR-585 silicone adhesive and the entire system vacuum outgassed at 375°F.

7-3, Aluminized 5 mil FEP Teflon Second Surface Mirror

This material was manufactured by G. T. Schjeldahl, Northfield, Minnesota, Type G400900, lot number 12-33 and was supplied to the AFML through the courtesy of Mr. A. E. Hultquist, Lockheed Missiles and Space Company of Sunnyvale, California. The material was bonded to a 15/16 inch diameter 2024 NaOH etched aluminum disc in the following manner. Both the front side of the disc and the back side of the vacuum

* IIT Research Institute, Technology Center, Chicago, Illinois.

deposited aluminum were primed with General Electric SS4004 clear silicone primer and then 3M Company's Number 465 adhesive transfer tape was used on both the disc and the back side of the vacuum deposited aluminum layer and pressure applied by hand rubbing with 0.125 inch Teflon sheet.

7-4, Zn₂TiO₄/Methyl Silicone

Pigment synthesis, coating formulation, and application was performed by the Illinois Institute of Technology using Owens Illinois 650 methyl silicone resin as a binder material. Coating thickness was 5 ± 1.0 mils.

7-5, PV-100

This material is a rutile TiO₂ pigment in a silicone-alkyd binder, better known as Vita-Var heat absorption paint, formulation PV-100, 15966 white. The coating was prepared on 0.040 inch thick, 15/16 inch diameter 2024 aluminum disks after they were cleaned by solvent wiping followed with a sodium hydroxide etch and water rinse. Strontium chromate primer was applied per MIL-P-23377B to a thickness of 0.0005 inch. The PV-100 topcoat was then sprayed to a thickness of 0.002 inch and air-dried.

7-6, 3M Black Velvet

This is a two component polyester-epoxy paint, type 401-C10 prepared by dispersing three parts of Part A, Lot No. VC-39-125 and one part of Part B, Lot No. 16. This was then spray applied onto 0.032 inch aluminum discs which had been sanded, solvent wiped, and primed with G. E.'s SS4044 primer (thickness less than 0.001 inch). Thickness of the coating was 0.006 ± 0.001 inch.

12-1, TiO₂/Methyl Silicone

This coating consisted of a TiO₂ pigment (Dupont R-992, Lot No. 9980) in purified RTV 602 (G1638-46 EFG from Hughes Aircraft Company under Contract F33615-71-C-1155) polydimethylsiloxane. A pigment to binder ratio of 150/100 was used. Thickness was 5.5 ± 0.5 mils.

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12-2, Optical Solar Reflector, 0.008

This is a second surface mirror with the front surface consisting of 0.008 inch of Corning 7940 quartz, and the back surface coated with vacuum deposited silver, overcoated with Inconel and a final dielectric back cover. These samples were supplied to AFML through the cooperation of Mr. E. N. Borscn, Aerospace Corporation and Mr. A. E. Hultquist, Lockheed Missile and Space Company. This material was bonded to the aluminum disc using the same procedure as sample 7-3.

12-3, Optical Solar Reflector 0.050

Same material as 12-2 except the Corning 7940 quartz was 0.050 inch thick. The same bonding procedure used on sample 7-3 was utilized.

12-4, Eu_2O_3 /Methyl Silicone

This coating consisted of a europium oxide pigment (American Potash and Chemical, Code 1014, Lot No. D1031, silicate treated) in purified RTV 602 (G1638-46EFG from Hughes Aircraft under Contract F33615-71-C-1155) polydimethylsiloxane. A pigment to binder ratio of 270/100 was used. Thickness was 8.0 ± 1.0 mils.

12-5, $\alpha\text{Al}_2\text{O}_3$ /Methyl Silicone

This coating was made from an α -alumina pigment (#1013 from Martin-Marietta Company under Contract F33615-71-C-1410) in unpurified commercial RTV 602, Lot #242, polydimethylsiloxane. A pigment to binder ratio of 200/100 was used. Coating thickness was 8.0 ± 1.0 mils.

12-6, 3M Black

A duplicate 3M Black per sample 7-6.

7-7 and 12-7 Reference Temperatures

Junction inside each sample holder.

54-6 Calibrated Resistor

A 499 ohm resistor in sample holder number 7, used as a check on voltage variations.

SECTION VI
OPTICAL PROPERTY DETERMINATION AND ENVIRONMENTAL EXPOSURE

Table III provides initial data for each coating, including mass, emittance, solar absorptance, and specific heat. The solar absorptance (α_s) value for each coating was determined utilizing a Beckman DK-2A spectrophotometer with a Gier-Dunkle Integrating Sphere and a xenon energy source (Figure 3). Integration of solar absorptance was accomplished using 25 points in conjunction with F. S. Johnson Solar Spectrum. In-air infrared spectral reflectance of each coating was measured in-house utilizing a Gier-Dunkle Model DB100 Portable Infrared Reflectometer (Figure 4). Subtracting the infrared reflectance from unity gave the total normal emittance. The theory employed by this instrument has been described by Nelson, et al (Reference 3).

Figure 5 depicts the AFML In Situ Exposure Device for simulation of ultraviolet-vacuum space conditions. The system was operated generally in the 10^{-7} torr region during radiation exposure. A xenon lamp ultraviolet source was used with an ultraviolet sun rate of approximately 1.25. The coating reflectance values were measured in situ before and after ultraviolet radiation exposure utilizing a Beckman DK-2A spectrophotometer with an integrating sphere inside the exposure chamber. Figures 6 through 15 present typical reflectance curves, before and after 500 EUVSH, for the coating samples flown. The pre-test and post-test solar absorptance values are also given for each coating.

TABLE III
INITIAL DATA INPUTS

Coating	Holder No.	Mass (lb _m)	ϵ_N	Initial Solar Absorbance	Specific Heat
S-13G	7-1	0.0119	0.89	0.20	0.15
Silica Fabric	7-2	0.0109	0.85	0.25	0.15
FEP/Al	7-3	0.0117	0.80	0.14	0.15
Zn ₂ TiO ₄ /OI 650	7-4	0.0115	0.87	0.17	0.15
PV-100	7-5	0.0114	0.85	0.23	0.15
3M Black	7-6	0.0107	0.92	0.985	0.17
Reference	7-7	N/A	N/A	N/A	N/A
TiO ₂ /M-Silicone	12-1	0.0109	0.87	0.19	0.15
OSR (0.003)	12-2	0.0114	0.80	0.073	0.15
OSR (0.050)	12-3	0.0137	0.81	0.075	0.15
Eu ₂ O ₃ /M-Silicone	12-4	0.0110	0.92	0.13	0.15
α -Al ₂ O ₃ /M-Silicone	12-5	0.0110	0.90	0.11	0.15
3M Black	12-6	0.0107	0.92	0.985	0.17
Reference	12-7	N/A	N/A	N/A	N/A
Calibrated Resistor	54-6	N/A	N/A	N/A	N/A

SECTION VII
COMPUTER DATA ANALYSIS

1. HEAT BALANCE EQUATIONS

The heat balance equations, using the list of symbols, for a single sample of a thermal control coating on this experiment are:

$$Q \text{ (input)} = Q \text{ (direct solar)} + Q \text{ (conduction)} \quad (1)$$

$$+ Q \text{ (sensor dissipation)} + Q \text{ (reflected solar)}$$

$$+ Q \text{ (albedo)} + Q \text{ (earth emission)}$$

$$Q \text{ (output)} = Q \text{ (emitted)} + Q \text{ (exchange with surroundings)} \quad (2)$$

$$+ Q \text{ (storage)}$$

The individual terms of Equations 1 and 2 are:

$$Q \text{ (direct solar)} = \alpha_s A_n H_s \quad (3)$$

$$Q \text{ (conduction)} = K(T_s - T_{ref}) \quad (4)$$

$$Q \text{ (sensor dissipation)} = I^2 R \text{ loss in sensor} \quad (5)$$

$$Q \text{ (reflected solar)} = \text{reflected sunlight from the Mylar upper surface and other areas of surroundings} = Q_{rs} \quad (6)$$

$$Q \text{ (albedo)} = \alpha_s A_n H_r \quad (7)$$

$$Q \text{ (earth emission)} = \epsilon_{th} A_n H_p \quad (8)$$

$$Q \text{ (emitted)} = \epsilon_{th} A_n \sigma T_s^4 \quad (9)$$

$$Q \text{ (exchange with surrounds)}$$

$$Q_{(ne)} = A_e \epsilon_e F_{e-s} \sigma T_s^4 + A_c F_{e-m} \sigma \epsilon_e \epsilon_m (T_s^4 - T_m^4) \quad (10)$$

$$+ \sum A_e F_{e-m} \sigma \epsilon_e \epsilon_n (T_s^4 - T_n^4)$$

$$Q \text{ (storage)} = MC_p dT_s / d\theta \quad (11)$$

Combining Equations 1 - 11 results in:

$$\begin{aligned} & a_s A_n H_s + K(T_s - T_{ref}) + Q_{rs} + a_s A_n H_r + \epsilon_{th} A_n H_p \\ & = \epsilon_{th} A_n \sigma T_s^4 + A_e \epsilon_e F_{e-s} \sigma T_s^4 + A_e F_{e-m} \sigma \epsilon_e \epsilon_m (T_s^4 - T_m^4) \end{aligned} \quad (12)$$

where $I^2 R$ and $\sum A_e F_{e-m} \sigma \epsilon_e \epsilon_m (T_s^4 - T_m^4)$ are considered negligible terms. Letting $T_m = \left\{ \left[a_m (H_s + H_r) / \epsilon_m + H_p \right] / \sigma \right\}^{1/4}$ and solving for $MC_p \frac{dT_s}{d\theta}$ results in the following:

$$\begin{aligned} MC_p \frac{dT_s}{d\theta} &= A_n (H_s + H_r) a_s + A_e F_{e-m} \epsilon_e a_m (H_s + H_r) \\ &+ (A_e F_{e-m} \epsilon_e \epsilon_m + A_n \epsilon_{th}) H_p + Q_{rs} + K T_{ref} - \sigma (A_n \epsilon_{th} \\ &+ A_e F_{e-m} \epsilon_e \epsilon_m + A_e \epsilon_e F_{e-s}) T_s^4 - \frac{K T_s}{MC_p} \end{aligned} \quad (13)$$

Dividing by MC_p and defining the following:

$$AA = \frac{A_n (H_s + H_r) a_s + A_e F_{e-m} \epsilon_e a_m (H_s + H_r)}{MC_p} \quad (14)$$

$$BB = \frac{A_e F_{e-m} \epsilon_e \epsilon_m H_p + K T_{ref} + Q_{rs} + H_p A_n \epsilon_{th}}{MC_p} \quad (15)$$

$$BB = \frac{K}{MC_p} \quad (15)$$

$$CC = \frac{\sigma (A_n \epsilon_{th} + A_e F_{e-m} \epsilon_e \epsilon_m + A_e \epsilon_e F_{e-s})}{MC_p} \quad (16)$$

upon substitution into Equation 13 gives:

$$\frac{dT_s}{d\theta} = AA - BB T_s - CCT_s^4 \quad (17)$$

From Equation 15, it can be seen that BB is a constant dependent only upon property factors. Likewise CC in Equation 16 is dependent upon the property factor of the particular sample in question. The term AA in Equation 14 contains the heat inputs, property factors, and the solar absorptance term which is the quantity to be calculated.

The resulting differential equation (Equation 17) cannot be solved analytically because it is nonlinear due to the T_s^4 term. It can be solved by La Place transform methods after the fourth order T_s^4 term is linearized by an expansion such as the Taylor series expansion with higher order terms being neglected. However, errors result by neglecting these higher order terms.

Another more accurate method is to use the Runge-Kutta integration technique with the aid of the computer which is the method chosen and explained in Section 2.

2. RUNGE-KUTTA INTEGRATION TECHNIQUE AND COMPUTER PROGRAM MB4

The fourth order Runge-Kutta integration method solves the differential equation (Equation 17) by giving a solution set of values of the dependent variable, T_s , for several increments of the independent variable, θ .

The method uses a variable step integration with the interval of integration being specified. An interval of 0.001 was used in this analysis. This variable step integration is designed to keep the maximum local relative error of the dependent variable less than 1×10^{-6} . The method integrates from one point to another. The initial point is assumed to be at $\theta = 0$ for the independent variable while the dependent value of temperature at $\theta = 0$ is given as T_{s0} .

A program to solve the differential equation of the form of Equation 17 using Runge-Kutta integration techniques was written and appears in Appendix I. The program integrates from some initial condition (temperature T_{s0} at time T zero) to some final condition

(temperature T_{sck} at time T check). The program then checks to see if the temperature found from the integration is the same as that given from telemetry readings to within $0.1^{\circ}R$. If the temperature does not agree the parameter AA is then adjusted by 1.0 unit until the temperature agrees. This adjustment of AA by 1.0 unit was found to be the best factor to use. Should the function begin to oscillate the agreement interval is increased to $0.4^{\circ}R$ and AA adjusted until that interval of agreement is met.

After the integration is completed and the equation fit to match that received from telemetry data, the values of AA (fit parameter) and corresponding values of temperature calculated from integration and compared to T_{sck} are printed out. The solution set of temperatures found for the fitted differential equation from time zero to some later time is printed out. Then these can be checked against further points from telemetry for a good overall fit if desired.

The portion of the T_s versus θ telemetry curve that is fit, is the initial portion where the effect of H_r should be minimized and the points chosen for the fit should be the same for all the curves in which the α_s are to be compared. The portion of the curve used in this analysis begins with time zero and temperature T_{s0} at 0.155 hours into the orbit after just passing into the sun. The final point for which the curve is fit is at time 0.47 hours later or 0.625 hours into the orbit. The temperature T_{sck} is the temperature from the T_s versus θ curve which occurs at this time and is used as a check temperature for fitting the differential equation solution to the telemetry curve. Although this relatively large interval allows more time for the effect of changing H_r to enter the system, it permits more smoothing of integrated data.

The reason the initial point of 0.155 hours into the orbit was chosen was that some of the telemetry curves to be compared did not have telemetry data until that time in the orbit. This is due to tracking acquisition that occurred on certain revolutions.

During the actual calculation, values of initial temperature T_{so} and check temperature T_{sck} for each sample for a given revolution are obtained. These are supplied to the program in the following order: Sample 7-6, 7-1, 7-2, 7-3, 7-4, 7-5, 12-1, 12-2, 12-3, 12-4, and 12-5.

The program then uses the data from the black sample 7-6 to fit the integrated equation to the telemetry data and obtain the amount of heat incident upon each of the samples. This is done by calculating the amount of heat from the equation for a fixed alpha (0.985). This can be done because the solar absorptance of the black samples should not change appreciably with exposure time. This amount of total heat input is then used to calculate the alphas for each of the nonblack samples after they are fit to their corresponding telemetry curves. The alpha values are then printed out along with coating property information used in the calculations.

3. ASSUMPTIONS

(a) The heat capacity, C_p , was assumed to be constant and the values given in Table III were used as the correct values for each sample.

(b) The value of the total solar heat incident on the sample edge, Q_{rs} , was constant at 0.0375 BTU/hr.

(c) The shape factor from the edge of the sample to Mylar holder, F_{e-m} , and the shape factor from edge of sample to space, F_{e-s} , were constant at 0.5.

(d) The planetary heating flux, H_p , was 18.1 BTU/hr-ft².

(e) The solar heating flux, H_s , and the planetary albedo heating flux, H_r , were constant and found from sample 7-6 for each revolution.

(f) The constant H_r assumption is made because only the initial portion of the temperature vs time curve is used where increasing H_r should have the smallest effect.

(g) The term $\sum A_e F_{e-m} \sigma \epsilon_e \epsilon_n (T_s^4 - T_m^4)$ is zero, indicating negligible interaction between adjacent sample edges.

(h) Sensor dissipation or I^2R loss of the platinum resistors is taken as zero since the term at 500 ohms and 100 mv results in a temperature rise of about $1/4^\circ\text{F}$, which is well within our temperature error limits.

4. DEVELOPMENT OF TIME-TEMPERATURE PROFILES

The computer program presented in Appendix II was utilized to convert raw digital data tapes for each required revolution directly into time-temperature curves. The program utilizes ground based sensor calibration data and sample holder conduction coefficients to convert millivolts to resistance to temperature. The program also plots directly the reference temperature and the resistance of the reference calibrated resistor. Typical plots for all sensors are presented in Figures 16 through 28.

SECTION VIII
DISCUSSION OF RESULTS

Table IV presents the solar absorptance changes for the coatings investigated. Column one indicates the initial laboratory measured solar absorptance, column two the initial in space calculated solar absorptance (calculated during revolution number 46 after the satellite had stabilized), and column three the calculated in space solar absorptance after one year (5191 revolutions, 2160 EUVSH).

One of the most important features to be noted in Table IV is that the initial in space solar absorptances of the two OSR's are twice those measured in the laboratory. This indicates some form of contamination on the surfaces. This contamination, from all indications, was present even during revolution number one, indicating contamination during liftoff as well as contamination during the initial revolutions from outgassing of volatile constituents from the satellite materials.

The effects of contamination on the initial in space solar absorptance values for the other coatings appear much smaller with the greatest change being for FEP/Al. The OSR coatings and the FEP/Al coating were, undoubtedly, affected the most because of their low initial solar absorptance which produced low surface temperatures, therefore attracting condensable contaminants. Also their low initial solar absorptance values were the most vulnerable to changes caused by small amounts of contaminant. The two coatings, $\text{Eu}_2\text{O}_3/\text{m-silicone}$ and $\alpha\text{-Al}_2\text{O}_3/\text{m-silicone}$ both had initial in space solar absorptances which were lower than those measured in the laboratory. The reasons for this are not clear, but these initial calculated values were used for determining the total solar absorptance changes in Figures 37 and 38.

Figures 29 through 38 present a comparison of laboratory simulated data and flight data for all the coatings flown, except for 3M black velvet. These figures also indicate the rate of change in solar absorptance over the one year exposure period. The flight $\Delta \alpha_s$

in these figures is the change in coating solar absorptance when compared to the initial calculated in space solar absorptance, not the initial laboratory values. For example, although the OSR(0.008) increased by 0.087 from laboratory to initial in space, it only increased another 0.025 after one year exposure in space and this is the $\Delta \alpha_s$ reported.

Each flight curve represents the compilation and analysis of 200 revolutions of data between revolution 1 and revolution 5191. The data scatter was extremely small, primarily because of the satellite stability in space and the large amount of data received per unit time (in this experiment each sample was pulsed every 0.9 seconds), which allowed for excellent computer data averaging. The actual plots of time-temperature curves from the raw data tapes showed minimal data scatter and maximum consistency from one revolution to the next. Over 800 data tapes were received during the first year which allowed for the selection of revolutions of primary interest.

From Figures 29 through 38 it can be seen that, in all instances, after 500 hours, the flight coatings were degrading more rapidly than the laboratory tested coatings. As indicated earlier, this can best be explained by the synergistic effect of contamination. If, as it appears, all surfaces were contaminated, the degradation of the coatings consisted of a combination of coating solar degradation plus solar degradation of the contaminant on the coating surface. Results from the Air Force Materials Laboratory D024 thermal control coatings experiment aboard SKYLAB, where samples were returned to earth from space and evaluated, clearly indicated the detrimental effects of contamination on thermal control coating stability. In all instances where contamination occurred during the SKYLAB exposure, solar degradation of the coatings increased much more rapidly than expected from laboratory tests.

Figures 30, 31, 35, 36, and 37 for SiO_2 Fabric, FEP/Al, two OSR's and Eu_2O_3 , all indicate an initial solar absorptance increase during the early revolutions with a gradual leveling of degradation with continued exposure. This result indicates a trend toward saturation of degradation

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for these coatings, however, additional data collection beyond one year is needed to confirm this as well as determine the degradation saturation point for all of the coatings.

One further note relative to the figures is that in Figure 38 the ordinate has been increased, because the degradation of the $\alpha\text{-Al}_2\text{O}_3$ coating was much more severe than any other coating.

TABLE IV
INITIAL VS SPACE SOLAR ABSORPTANCE

Coating	Initial Laboratory Solar Absorptance	Initial In Space Solar Absorptance	In Space Solar Absorptance (one year)
S-13G	0.20	0.19	0.33
Silica Fabric	0.25	0.27	0.35
FEP/Al	0.14	0.18	0.22
Zn ₂ TiO ₄ /OI650	0.17	0.19	0.33
PV-100	0.23	0.27	0.42
TiO ₂ /M-Silicone	0.19	0.19	0.33
OSR (0.008)	0.073	0.16	0.185
OSR (0.050)	0.075	0.16	0.195
Eu ₂ O ₃ /M-Silicone	0.13	0.08	0.16
α-Al ₂ O ₃ /M-Silicone	0.11	0.08	0.28

SECTION IX
CONCLUSIONS AND RECOMMENDATIONS

1. CONCLUSIONS

a. Some form of contamination, most likely from outgassing materials from the satellite, caused flight degradation of the thermal control coatings to be more severe than ground simulated degradation.

b. Of all the coatings flown, the Optical Solar Reflectors (OSR's) were the most stable. No significant difference in degradation was observed for the two OSR's of different thicknesses. This would indicate that the solar degradation was primarily a surface phenomenon rather than a bulk phenomenon.

c. FEP/Al was very stable with a $\Delta\alpha_s = 0.04$ after one year at 400 nautical miles (EUVSH $\approx 2,160$ hrs).

d. Of the white coatings flown, $\text{Eu}_2\text{O}_3/\text{m-silicone}$ was the most stable ($\Delta\alpha_s = 0.08$) and $\text{Al}_2\text{O}_3/\text{m-silicone}$ was the most unstable ($\Delta\alpha_s = 0.20$).

e. The silica fabric material flown was more stable than the white coatings with a $\Delta\alpha_s$ of 0.08 after one year exposure.

2. RECOMMENDATIONS

a. This data could also be used to calculate changes in coating emittance. This was not done due to time limitations.

b. Additional thermal control coating experiments should be flown at or near synchronous orbit to more closely simulate the environment of operational Air Force satellites.

c. The STP 72-1 satellite has continued to function satisfactorily even after one year of orbit life. Data from the ML-101 experiment should continue to be analyzed to determine the effects of long-term space exposure on these thermal control coatings.

d. Correlation of the ML-101 degradation data with data obtained from the AFML D024 experiment aboard the manned SKYLAB orbital workshop should be attempted.

REFERENCES

1. RTD-TDR-63-4269, Design and Construction of Sample Holders for Orbital Temperature Control Coatings Experiment, Vol. 1, Design, Analysis, and Test Results, TRW Systems, April 1964.
2. RTD-TDR-63-4269, Calibration Data and Drawings, Vol. 2, TRW Systems, April 1964.
3. K. E. Nelson, E. E. Luedke, and J. T. Bevans, Journal of Spacecraft Rockets, Vol. 3, No. 5, p. 758, 1966.

APPENDIX I
COMPUTER PROGRAM MB4

XP()	=	Theta printout and storage array.
YP()	=	TS(1) printout and storage array.
TS (1)	=	Temperature of sample at time theta.
P (1)	=	Highest derivative value.
ACHECK ()	=	Storage array for AA values.
TSCHEK ()	=	Storage array for TS (1) values calculated and compared to TSCK.
AA	=	Constant in differential equation which is changed to fit different equation solution to telemetry curve in $\left(\frac{dT_s}{d\theta} = AA - BBT_s - CCT_s^4\right).$
BB, CC	=	Other differential equation constants.
NSC	=	Sample identifier label check for sample 7-6.
TCHECK	=	Time value at which differential equation solution check is to be made with curve from telemetry data.
REV	=	Revolution number.
TREF	=	Reference sample temperature.
G	=	Solar heating constant.
NEXEC	=	Counter of number of times calculation loop is executed.
NS	=	Sample identifier label.
XMASS	=	Mass of sample.
CP	=	Heat capacity of sample.
ES	=	Emittance of sample.
TSO	=	Initial value of temperature of sample for integration.
TSCK	=	Temperature at which check is to be made of integrated equation temperature with telemetry data value.
AS	=	Solar absorptance.
HS	=	Solar heat input.
HR	=	Earth albedo heat input.
HP	=	Planetary heat from earth.

TIMEE	= Upper limit on time for integration.
QRS	= Total solar heat incident on sample edge.
AN	= Cross-sectional area of sample.
AE	= Area of edge of sample.
FES	= Shape factor edge sample to space.
FEM	= Shape factor edge sample to Mylar.
EE	= Emittance of edge of sample.
AM	= Absorptance of adjacent Mylar.
EM	= Emittance of Mylar holder.
ZK	= Thermal conductance of sample.
SIG	= Stephan-Boltzmann constant.
N	= Order of equation to be integrated.
ICHECK	= Check variable for oscillation of function, if oscillating, agreement interval increased to 0.4°F.
THETA	= Time.
DTHETA	= Increment for integration.
NCOUNT	= Counter.
NCT	= Counter for printing out every 10th integration solution point.
NPLT	= Counter of number of solution points.
SUBROUTINE F(THETA, TS, P)	= Function subroutine for differential equation.
RKDF	= Runge-Kutta subroutine.
IER	= Error flag.

```

C PROGRAM RUNG(INPUT,OUTPUT)
PROGRAM SOLVES DIFFERENTIAL EQUATION BY RUNGE KUTTA INTEGRATION
DIMENSION XP(200),YP(200)
DIMENSION TS(11),P(11)
DIMENSION ACHECK(100),TSCHEK(100)
COMMON AA,BB,CC

```

```

C DATA NSG/4M 7-6/
C CHANGE THE FOLLOWING 3 CARDS FOR EACH PUNNING OF PROGRAM
DATA TCHECK/47,REV/12900./
DATA TREF/502./
DATA G/450./

```

```

C 00 8436 NEXEC =1,11
READ 8500,NS,XYMASS,CPAES
FORMAT(I7,A%,T28,F5%,Y38,F3.2,T46,F3.2)
IF(NS.EQ.NSC) GO TO 8660
READ 8501,TSO,TSCK,AS
FORMAT(I16,F4.0,T27,F4.0,T36,F2.1)
GO TO 8661
8660 READ 9505,TSO,TSCK,AS
8505 FORMAT(I16,F4.0,T27,F4.0,T36,F4.3)
8661 MS=.318*G
HZ=.101*G
MP=18.1

```

```

C TIMEE=.60
QRS=.0375

```

```

C AN=.0048
AE=.0011
FEM=.5
FES=FEM
EE=.85
AM=.34
EM=.76
7K=6.E-05

```

```

SIC=1.714E-09
AA=(AN*AS*(NS*HR))AE*FEM*EE*AM*(MS*RT)AE*FEM*EE*EM*MP*7K*TRF
1 *QRS*HP*AN*ES)/(XYMASS*CP)
C OVERTURE THE STARTING VALUE OF AA
BB=ZK/(XYMASS*CP)
CC=SIG*(AN*ES*AE*FEM*EE*EM*AE*EE*FFS)/(XYMASS*CP)
N=1

```

```

C ITEST=0
NCOUNT=0
DELTS=0.1
OELTS2=0.4
ICHECK=0
THETA=0.
TS(1)=TSO

```

```

C 130 IF(ITEST.EQ.0.0) GO TO 46
PRINT 199
FORMAT(I11,10X,THETA*,14X,*TS*)
THETA=0.
TS(1)=TSO

```

```

C 46 OTHETA=.001

```

```

60      NCT=0
        NPLT=1
        CALL F(THETA,TS,P)
        CALL RKD(THETA,TS,N,OTHETA,IER)
        NCI=NOT+1
        IF(IIEI.EQ.1) GO TO 111
        IF(THETA.GE.TCHECK) GO TO 110
        GO TO 60
110     DTI=(TS(1)-TSCK)
        IF(A9SDIF).LE.DELTS) GO TO 120
        NCOUNT=NCOUNT+1
        IF(DIF19.97.97)
70      AA=AA+1.
        IF(IICHECK.EQ.2) DELTS=DELTS2
        ICHECK=1
        ACHEGK(INGOUNT)=AA
        TSCHEK(INGOUNT)=TS(1)
        GO TO 130
75      AA=AA-1.
        IF(IICHEGK.EQ.1) DELTS=DELTS2
        ICHEGK=2
        ACHEGK(INGOUNT)-1A
        TSCHEK(INGOUNT)=TS(1)
        GO TO 130
80      ITEST=1
        GO TO 130
111     IF(INGT.LT.10) GO TO 60
        XP(NPLT)=THETA
        YP(NLPT)=TS(1)
        NPLT=NPLT+1
90      IF(THETA.LE.TIMEE) GO TO 10
        NPLT=NPLT-1
        PRINT 112,(ACHEGK(IL),TSCHEK(IL),L=1,NCOUNT)
        FORMAT(2X,AA=*,1PE11.4,2X,TS=*,1PE11.4,2X,AA=*,1PE11.4,2X,
112     *TS=*,1PE11.4,2X,AA=*,1PE11.4,2X,TS=*,1PE11.4)
        PRINT 12,(XP(IL),YP(IL),L=1,NLPT)
95      FORMAT(2X,1P,1E11.4)
        PRINT 1640,NPLT,NCOUNT,AA
        PRINT 1640,NPLT,NCOUNT,AA
        PRINT 893,ITEST
        FORMAT(2X,ITEST = *,IC)
100     PRINT 7901,REV,G,MS,HR,MP,TSCG,TSCK
        FORMAT(//,10X,*,FOR REVOLUTION*,F6.0//,5X,*,1PE11.4,2X,*,MS=*,
        *2X,*,1PE11.4//)
        ZCK=*,1PE11.4//)
105     IF(NS.EQ.NSC) GO TO 1113
        AS=(XMASS*CP*AA-ZK*TRF-QRS-(AE*FEM*E*EH*AN*ES)*HP)/(AN*(HR+MS))
        1 - *FEM*EE*AM/AN
        PRIN 7900,NS,XMASS,CP,ES,BB,CC,AA,AS
        FORMAT(2X,*,FOR SAMPLE*,A6//,4X,*,MASS = *,F6.4,2X,*,CP = *,F4.2,
12X,*,ES = *,F6.2//,6X,*,BB = *,1PE11.4,2X,*,CC = *,1PE11.4,2X,
        *AA = *,1PE11.4,2X,*,AS = *,1PE11.4//)
        GO TO 8436
1113    G=(XMASS*CP*AA-(AE*FEM*E*EH*AN*ES)*HF-ZK*TRF-QRS)/(L*19*(AN*AS+A
        1E*FEM*EE*AM))
        MS=.318*G
    
```

PROGRAM RUNG 74/74 OPT=0 TRACE DEBUG FTN 4.2:P380 05/20/75 10.31.43. PAGE 4

```

115      HR=.101*G
      PRINT 1114,G,HS,HR,HP
1114    FORMAT(2X,0G=,F4.0,2X,HS=,F6.2,HR=,F6.2,HP=,F6.1)
8436    CONTINUE
      STOP
      END
120
    
```

SUBROUTINE F 74/74 OPT=0 TRACE DEBUG FTN 4.2:P380 05/20/75 10.31.49. PAGE 1

```

5      SUBROUTINE F (THETA,TS,P)
      DIMENSION TS(1),P(1)
      COMMON AA,BB,CC
      P(1)=AA-BB*TS(1)-CC*TS(1)**4
      RETURN
      END
    
```

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APPENDIX II
COMPUTER PROGRAM DAN

This appendix lists the computer program utilized to convert raw digital data tapes directly into time-temperature profile curves.

PROGRAM DAN

```

170      RTT1=RT1(K)
          RTT2=RT2(K)
          CALL JOHN(RT,RTT1,RTT2,T)
          S124(J)= T * Y + 492.
          K=K+1
          RT= X*FLOAT(ISN(I+43))
          RTT1=RT1(K)
          RTT2=RT2(K)
          CALL JOHN(RT,RTT1,RTT2,T)
          S125(J)=T*Y+492.
          K=K+1
          RT= X*FLOAT(ISN(I+47))
          RTT1=RT1(K)
          RTT2=RT2(K)
          CALL JOHN(RT,RTT1,RTT2,T)
          S126(J)=T*Y+492.
          K=K+1
          RT=X*FLOAT(ISN(I+5))
          RTT1=RT1(K)
          RTT2=RT2(K)
          CALL JOHN(RT,RTT1,RTT2,T)
          C S127 REPRESENT 38-1
          S127(J)=T*Y+492.
          S54(J)=X*FLOAT(ISN(I+26))
          200 CONTINUE
          210 IF(LCT.EQ.0) PRINT 210,(IH(I),I=1,5)
          FORMAT(1H137X5(3YA10))
          LCT=LCT+20
          A1 = 0.
          A2 = 0.
          A3 = 0.
          A4 = 0.
          A5 = 0.
          A6 = 0.
          A7 = 0.
          A11 = 0.
          A12 = 0.
          A13 = 0.
          A14 = 0.
          A15 = 0.
          A16 = 0.
          A17 = 0.
          A18 = 0.
          DO 500J=1,10
          A1 = A1 +S701(J)
          A2 = A2 +S702(J)
          A3 = A3 +S703(J)
          A4 = A4 +S704(J)
          A5 = A5 +S705(J)
          A6 = A6 +S706(J)
          A7 = A7 +S707(J)
          A11= A11+S121(J)
          A12= A12+S122(J)
          A13= A13+S123(J)
          A14= A14+S124(J)

```

PROGRAM DAN

```

225      A15= A15+S125(J)
          A16= A16+S126(J)
          A17= A17+S127(J)
230      A54= A54+S54(J)
          A2 = A2 /10.0
          A3 = A3 /10.0
          A4 = A4 /10.0
          A5 = A5 /10.0
          A6 = A6 /10.0
          A7 = A7 /10.0
          A11= A11/10.0
          A12= A12/10.0
          A13= A13/10.0
          A14= A14/10.0
          A15= A15/10.0
          A16= A16/10.0
          A17= A17/10.0
          A54 = A54/10.0
240      IF(LC.EQ.0) WRITE (7,210)
          IF(LC.EQ.0) WRITE (7,201)
          FORMAT(1M067H 7-1 12-1 7-2 12-2 7-3 12-3 7-4
1.51H 7-4 12-4 7-5 12-5 7-6 12-6 12-7
2.28H 54-6 DAY HR H S MS)
245      WRITE (7,202) A1,A2,A3,A4,A5,A6,A7,A11,A12,A13,A14,A15,A16,A17,
          1A54,0AY(5),HRS(5),MIN(5),SEC(5),HILSEC(5)
          FORMAT(1M06F6.1,6F7.1,F8.1,6F7.1,5XF5.1,3X13.3I3,I4)
          LC=LC+2
250      IF(LC.GE.50) LC=0
          IF(LC.EQ.1) GO TO 299
          IF(IFIL.EQ.1) GO TO 299
          IF(IFIL.EQ.3) GO TO 600
          GO TO 299
255      600 CONTINUE
          IC=1
          IR=IR+1
          XR(IR,IC1) = A1
          XR(IR,IC+ 1) = A2
          XR(IR,IC+ 2) = A3
          XR(IR,IC+ 3) = A4
          XR(IR,IC+ 4) = A5
          XR(IR,IC+ 5) = A6
          XR(IR,IC+ 6) = A7
          XR(IR,IC+ 7) = A11
          XR(IR,IC+ 8) = A12
          XR(IR,IC+ 9) = A13
          XR(IR,IC+10) = A14
          XR(IR,IC+11) = A15
          XR(IR,IC+12) = A16
          XR(IR,IC+13) = A17
          XR(IR,IC+14) = A54
          ISEC=SEC(5)
          IHRS=HRS(5)
          IMIN=MIN(5)
          MTIME=0

```


PROGRAM DAN CDC 6600 FTN V3.0-367A OPT=1 02/05/74 21.41.28. PAGE 7

```
CALL PLOT(22.0,0.0,-3)
CONTINUE
CALL PLOTE
END
```

SUBROUTINE JOHN CDC 6600 FTN V3.0-367A OPT=1 02/05/74 21.41.28. PAGE 1

```
SUBROUTINE JOHN (PT,R0,P100,T)
TRY=100.0*((RT-R0)/(R100-R0))
A=TRY/100.0
Y=100.0*((RT-P0)/(P100-P0))+1.493*(A-1.0)*A+0.111*(A-1.0)*((TRY**3)/
110.0**6)
TMT=T-TRY
IF(ABS(TMT).LE.10) GO TO 2
TRY=TRY+((T-TRY)/2.)
GO TO 1
RETURN
END
```

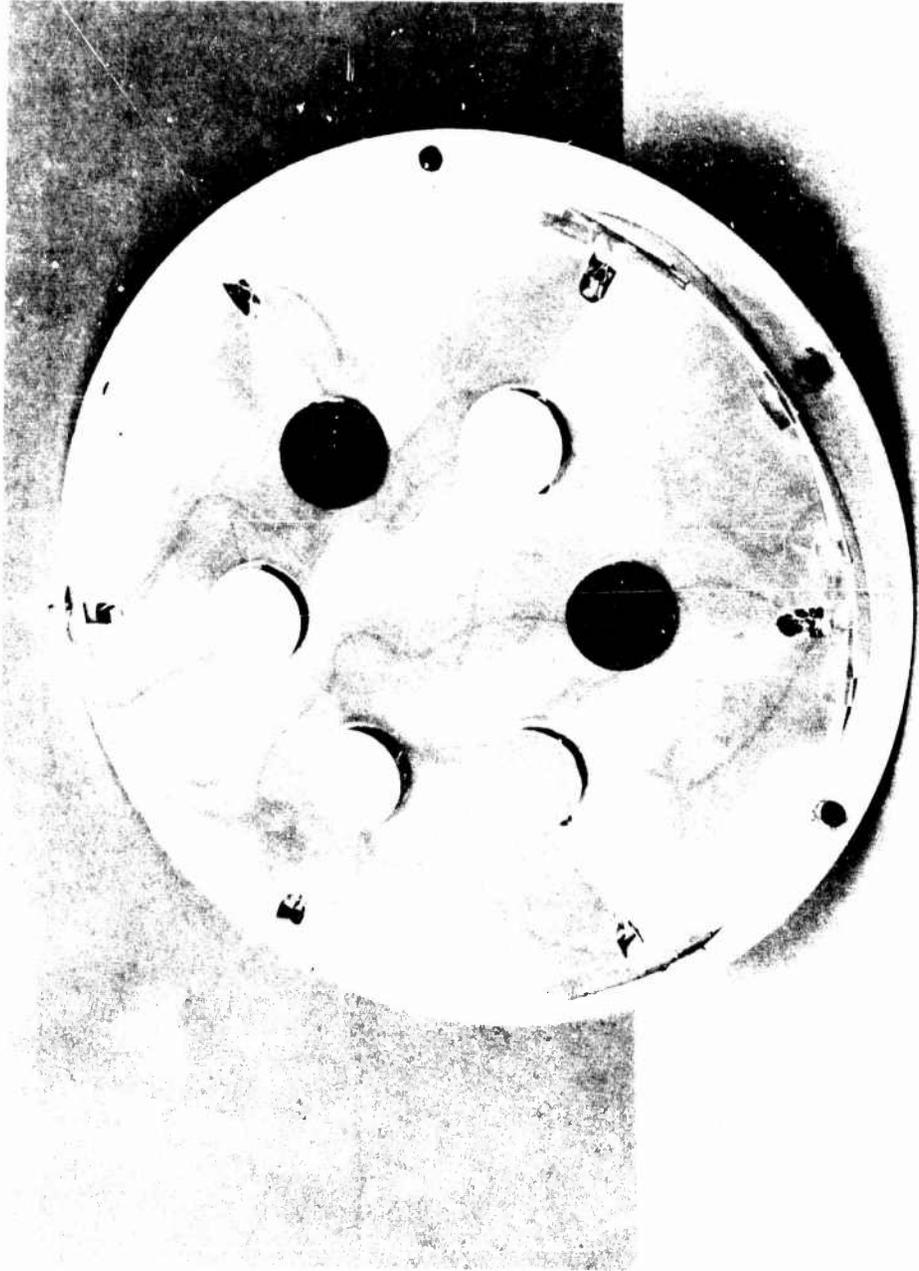


Figure 1. ML-101 Sample Holder

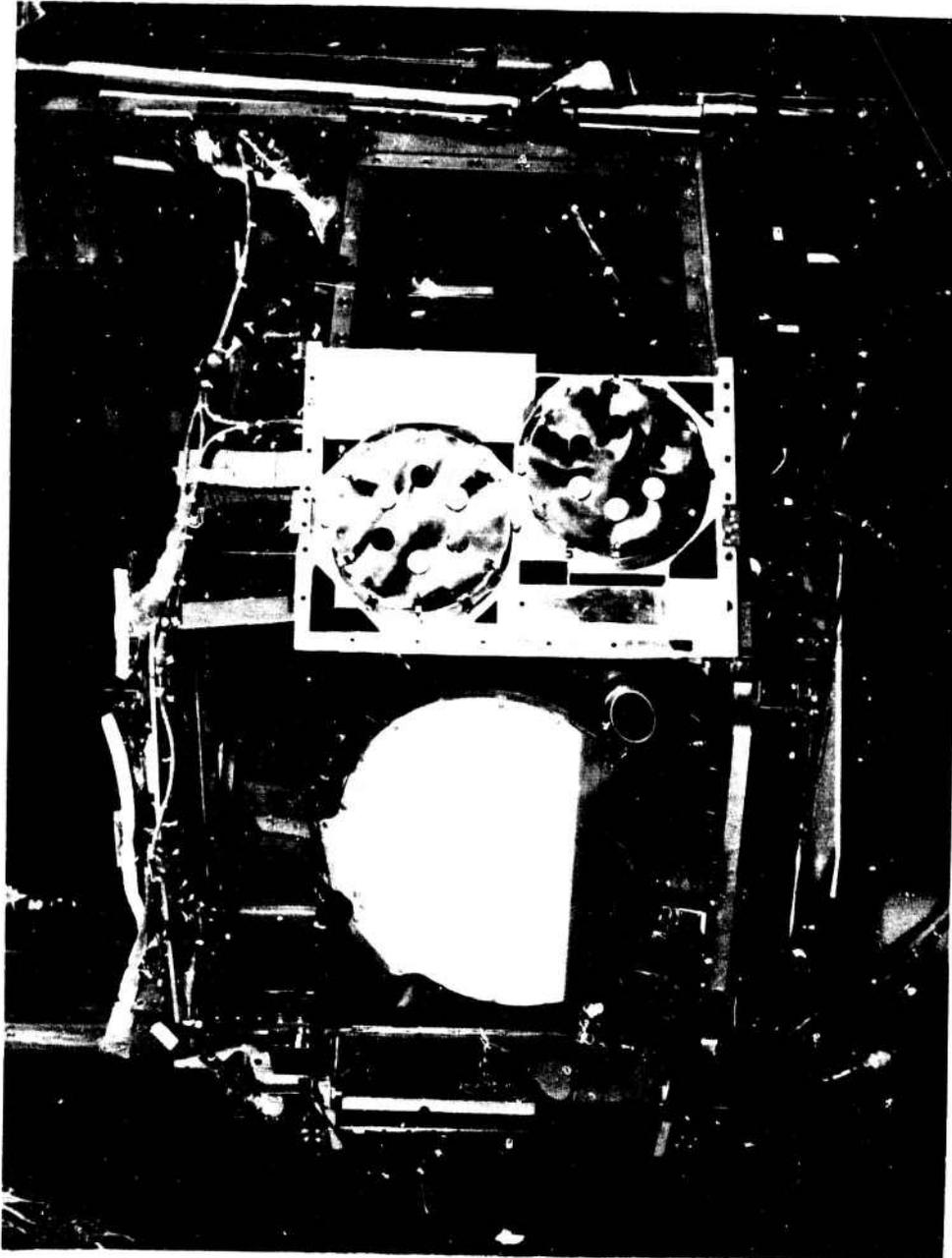


Figure 2. ML-101 Sample Holders Attached to 72-1 Satellite

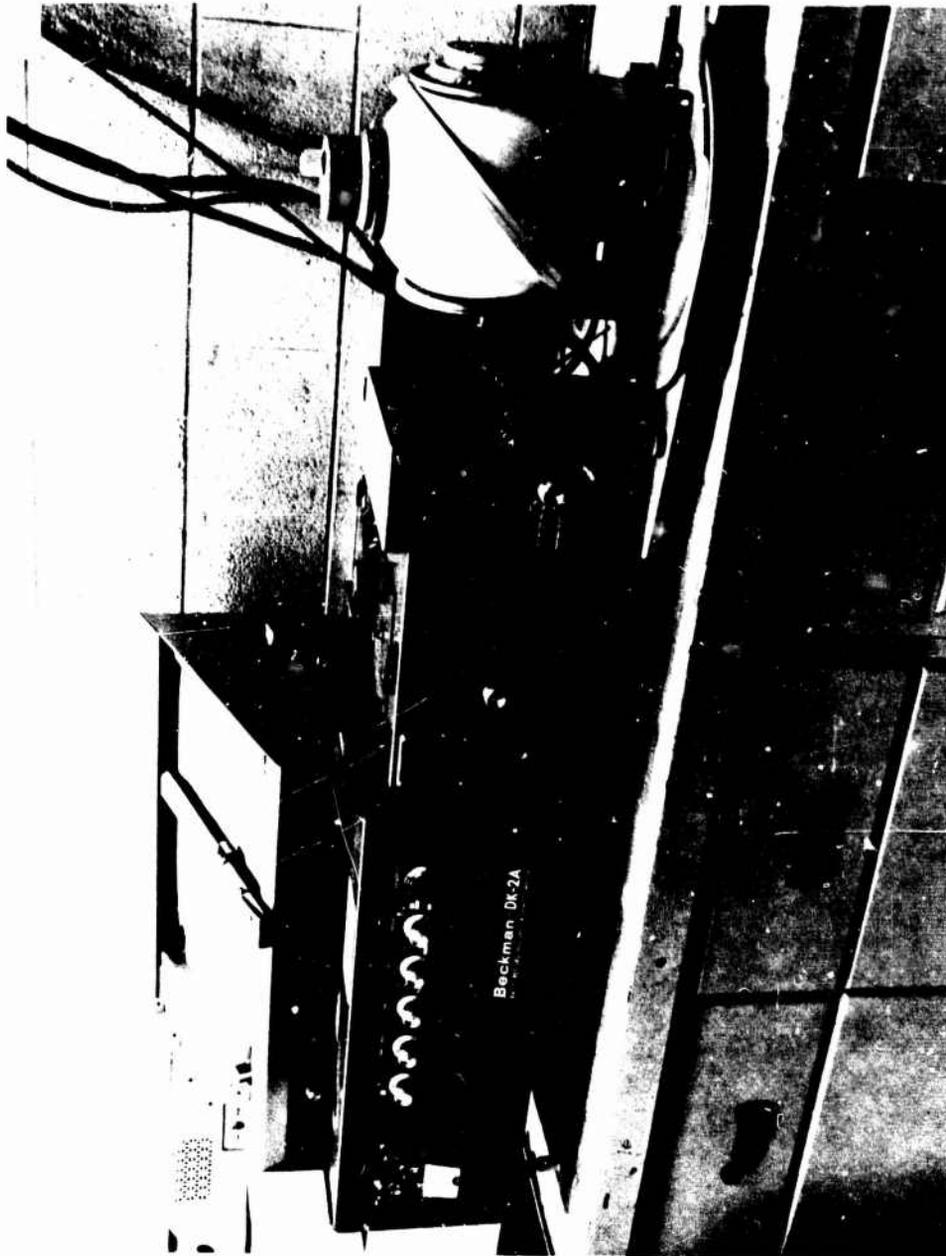


Figure 3. Beckman DK-2A Spectrophotometer with a Gier-Dunkle Absolute Integrating Sphere

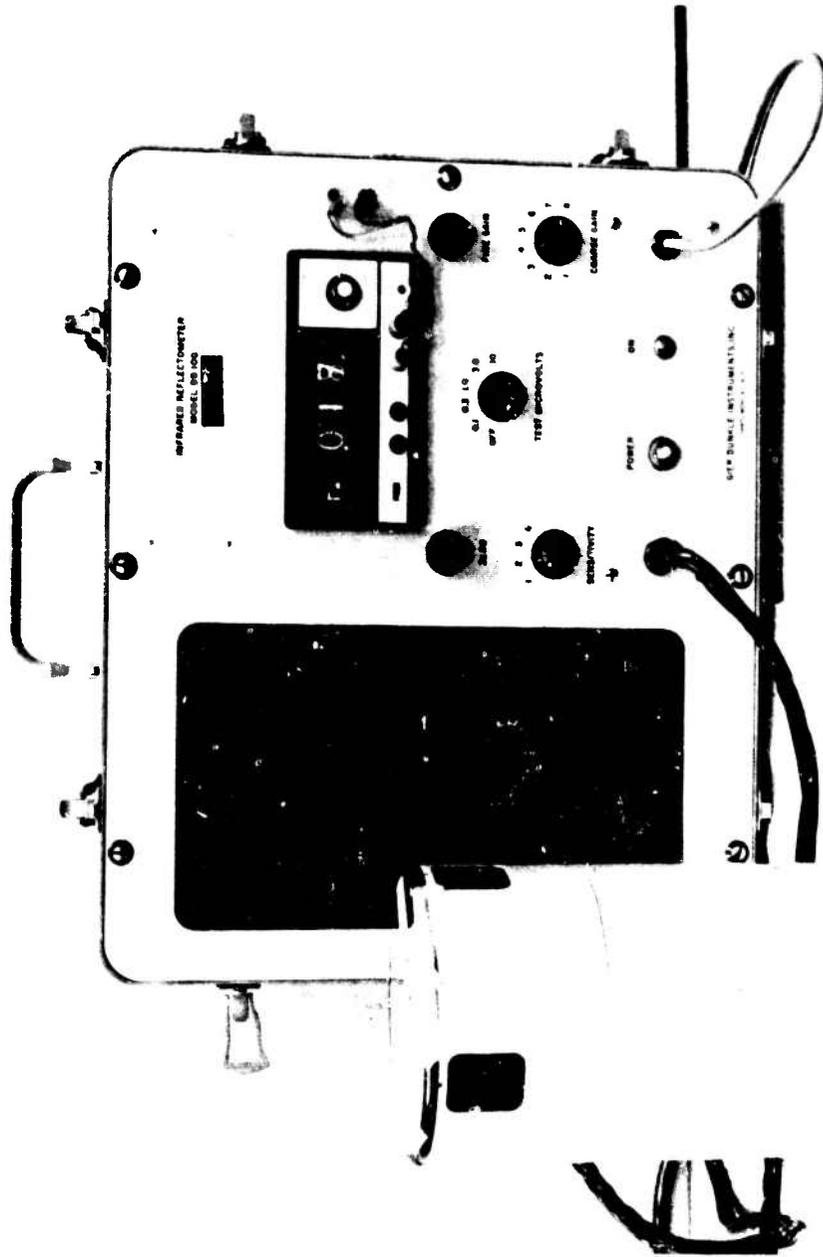


Figure 4. Gier-Dunkle Model DB100 Portable Infrared Reflectometer

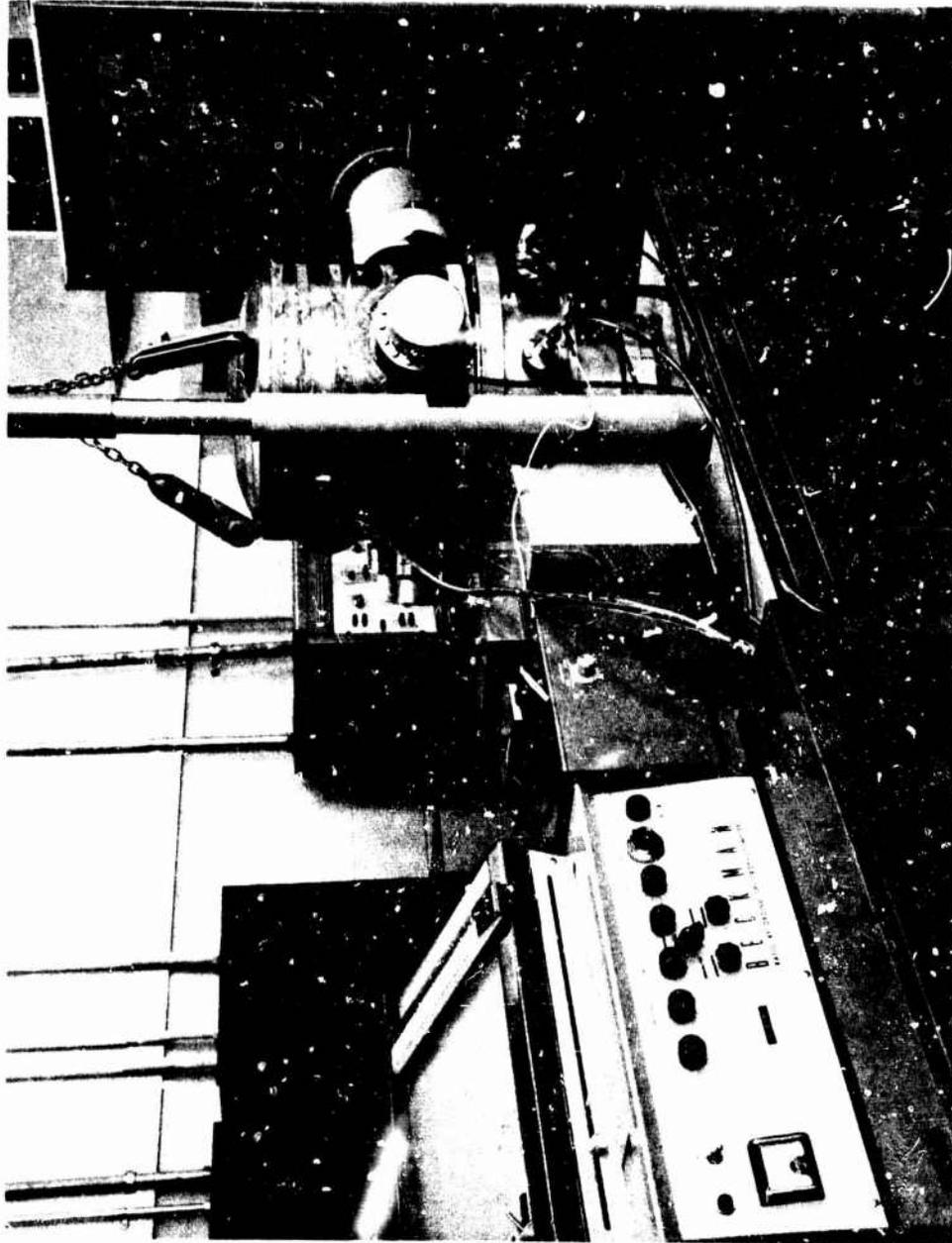


Figure 5. AFML In-Situ Exposure Device

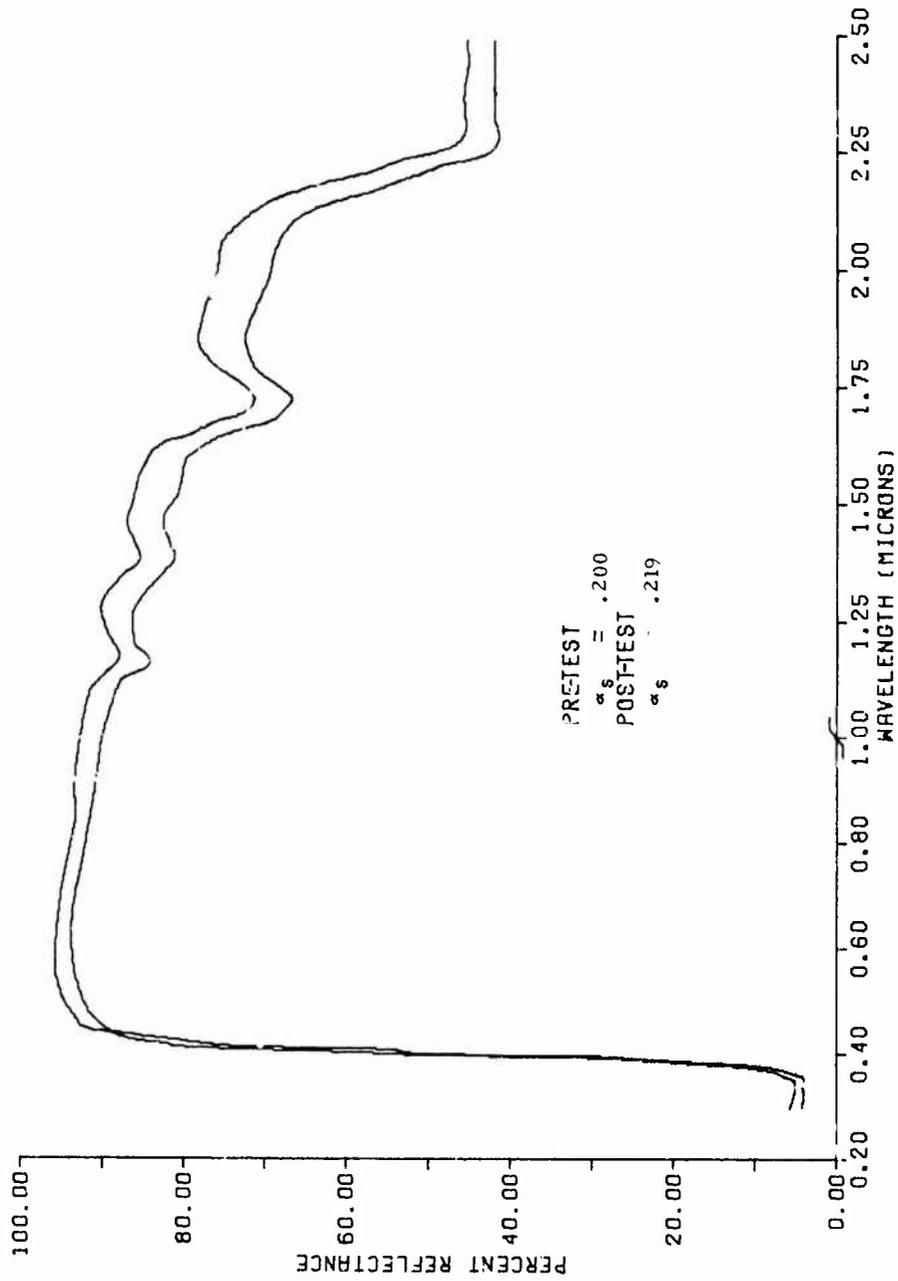


Figure 6. ML-101 Simulation Experiment S-156 (500 EUVSH)

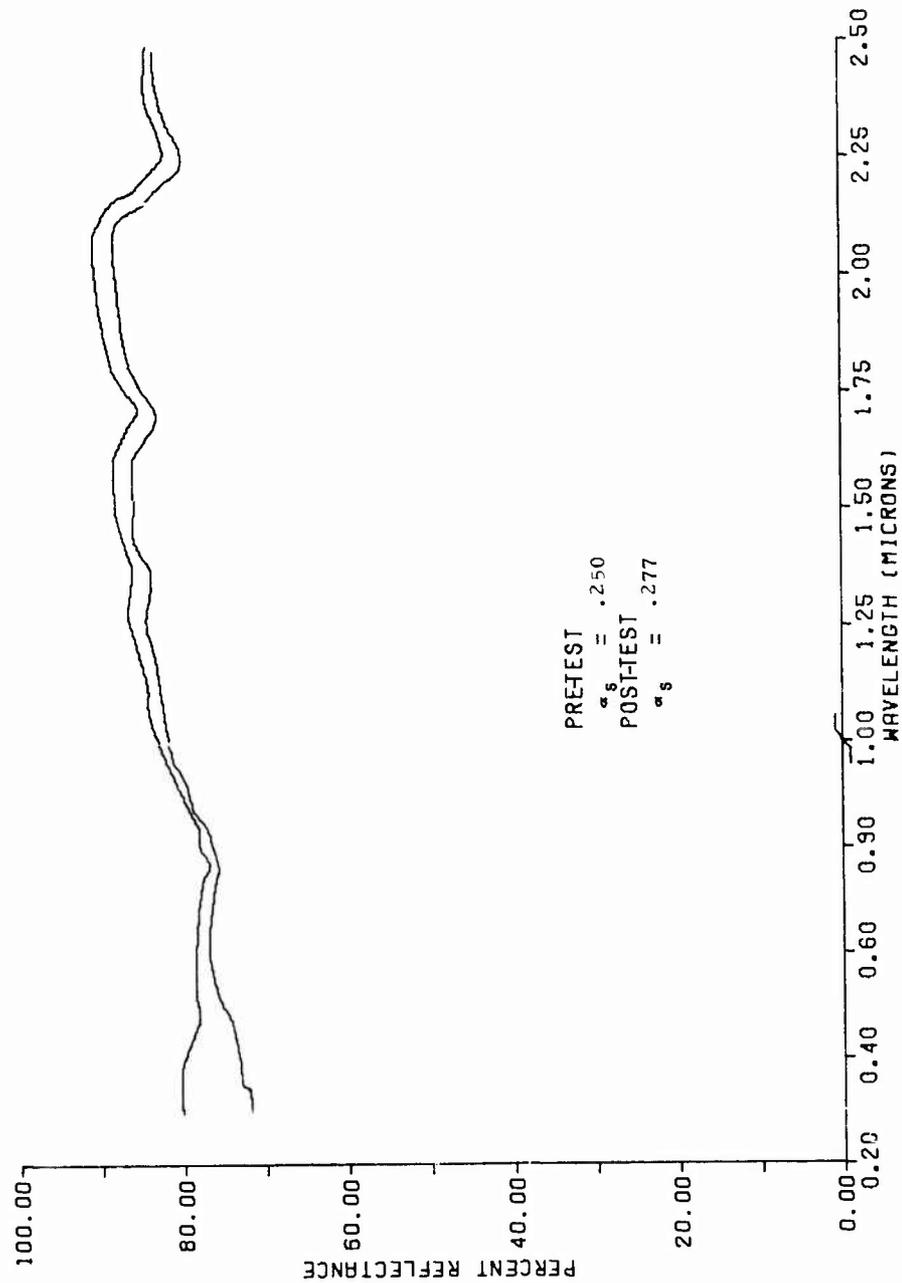


Figure 7. ML-101 Simulation Experiment SiO₂ Fabric (500 EUVSH)

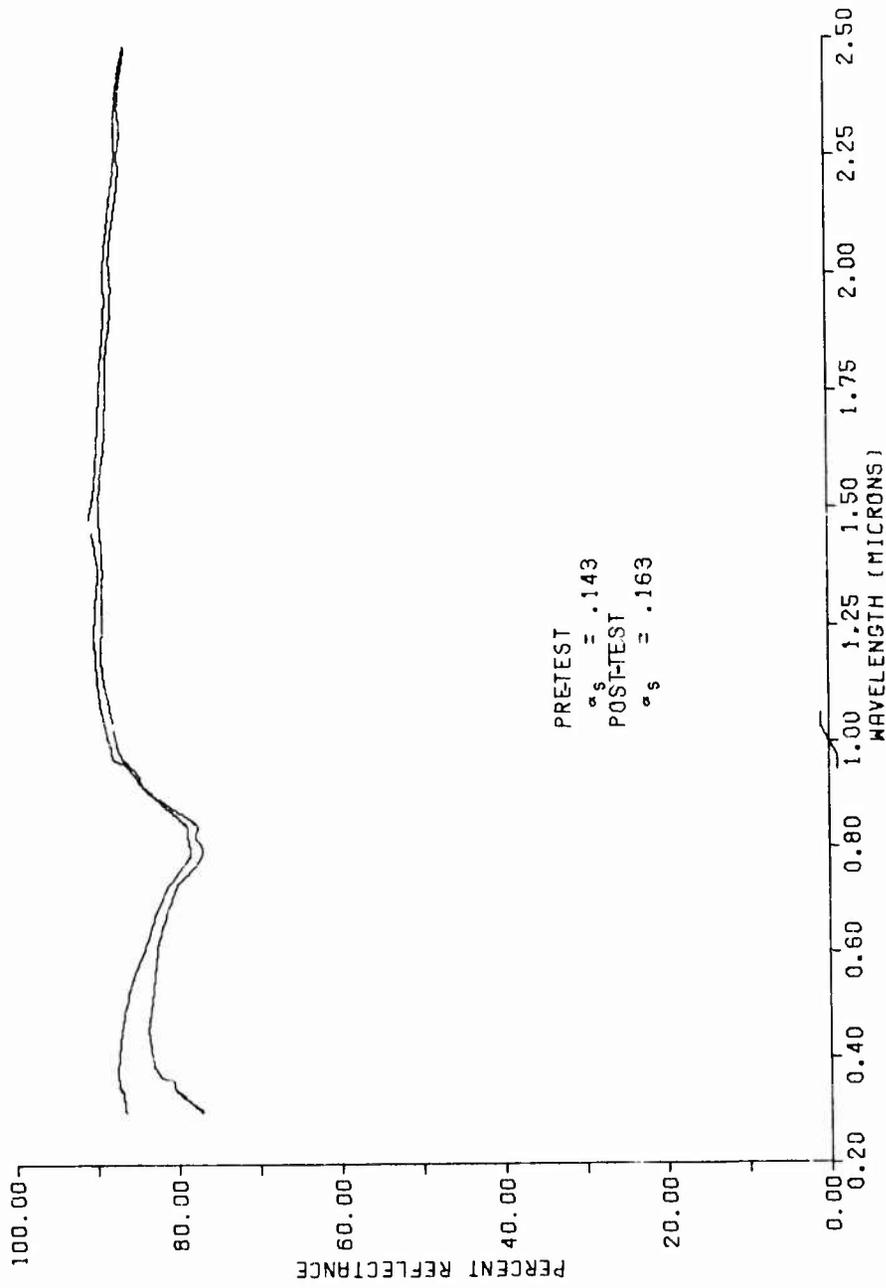


Figure 8. ML-101 Simulation Experiment FEP/A1 (500 EUVSH)

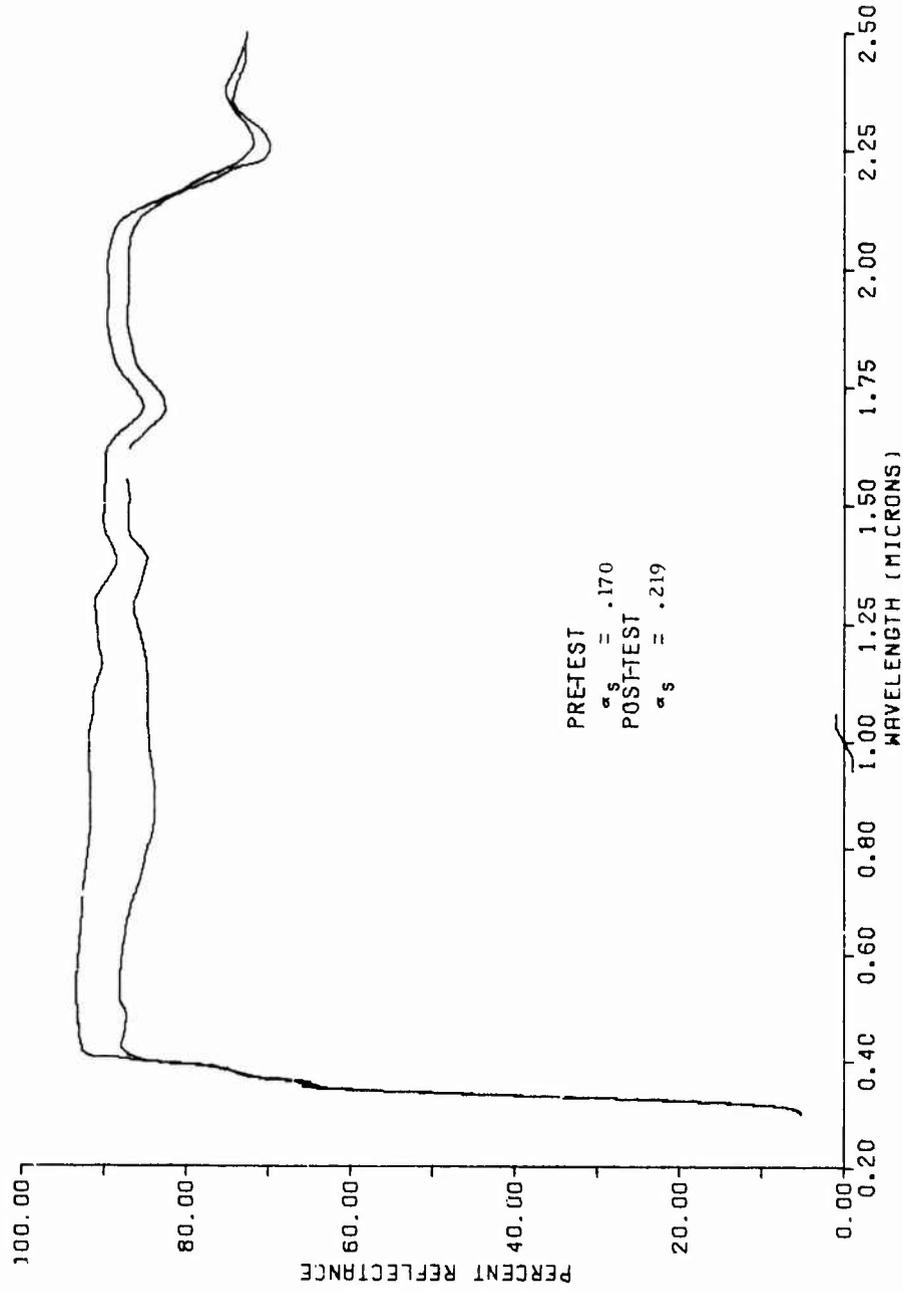


Figure 9. ML-101 Simulation Experiment $Zr, TiO_4/OI650$
(500 EUVSH)

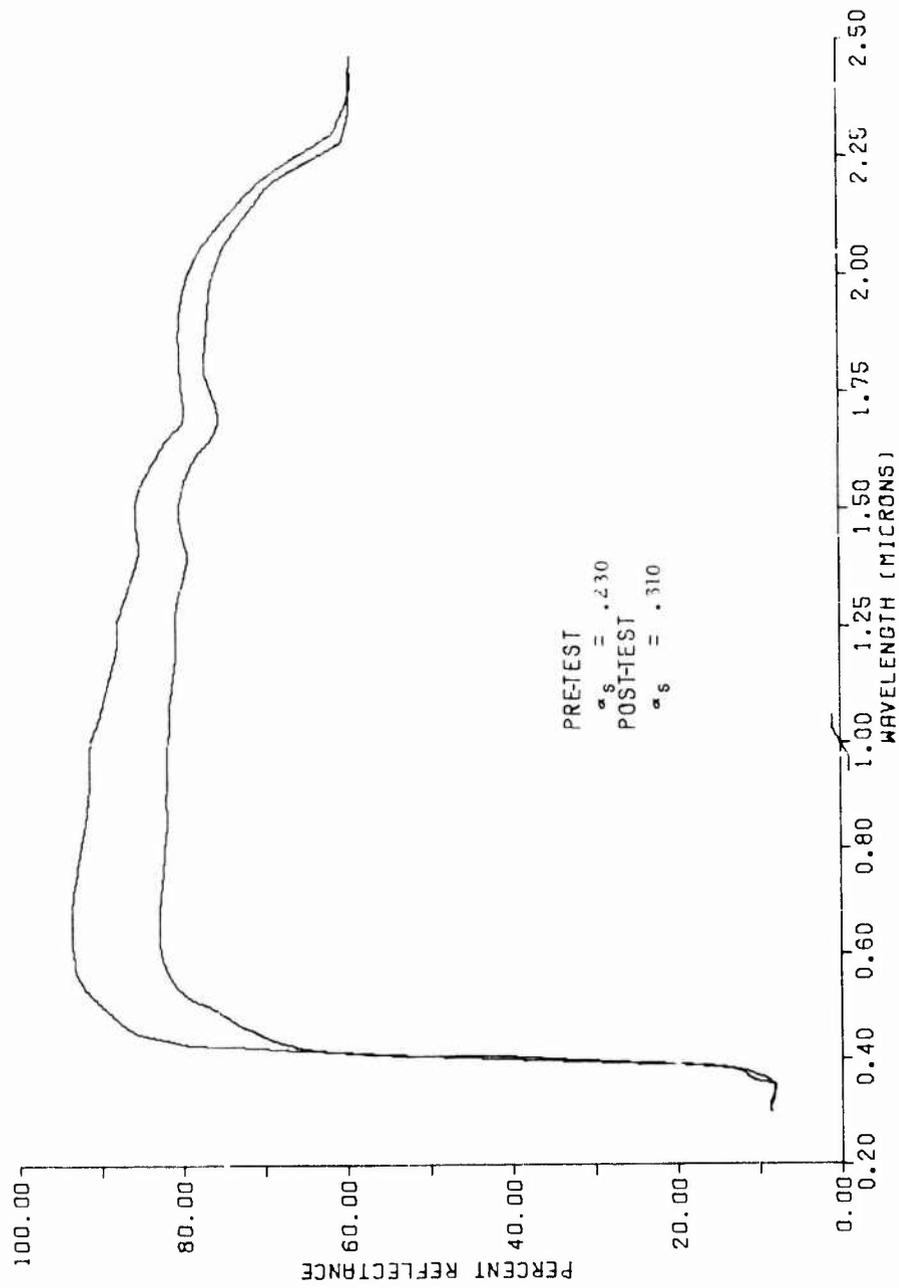


Figure 10. ML-101 Simulation Experiment PV-100 (500 EUVSH)

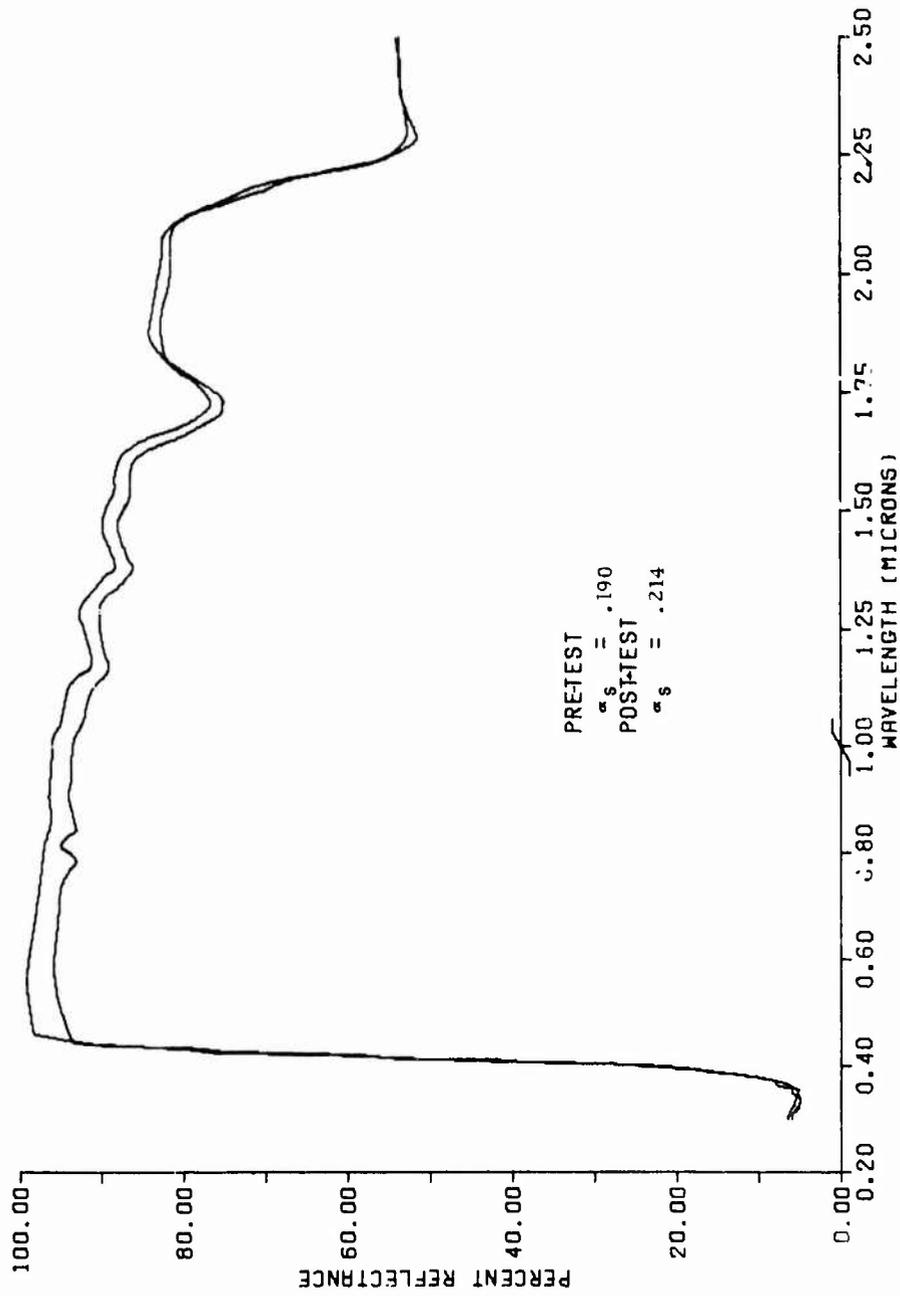


Figure 11. ML-101 Simulation Experiment TiO₂/RTY602 (500 EUVSH)

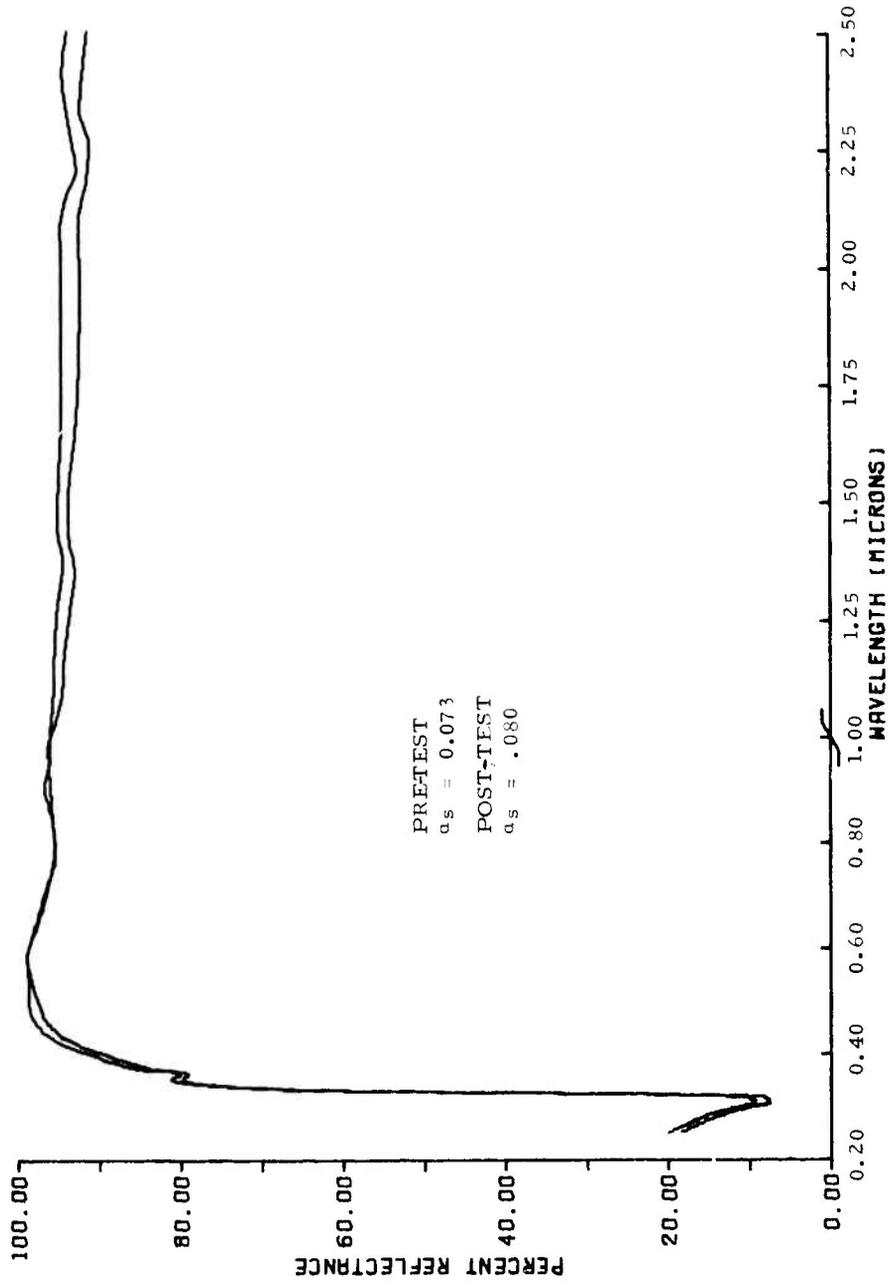


Figure 12. ML-101 Simulation Experiment .008" OSR
(500 EUVSH)

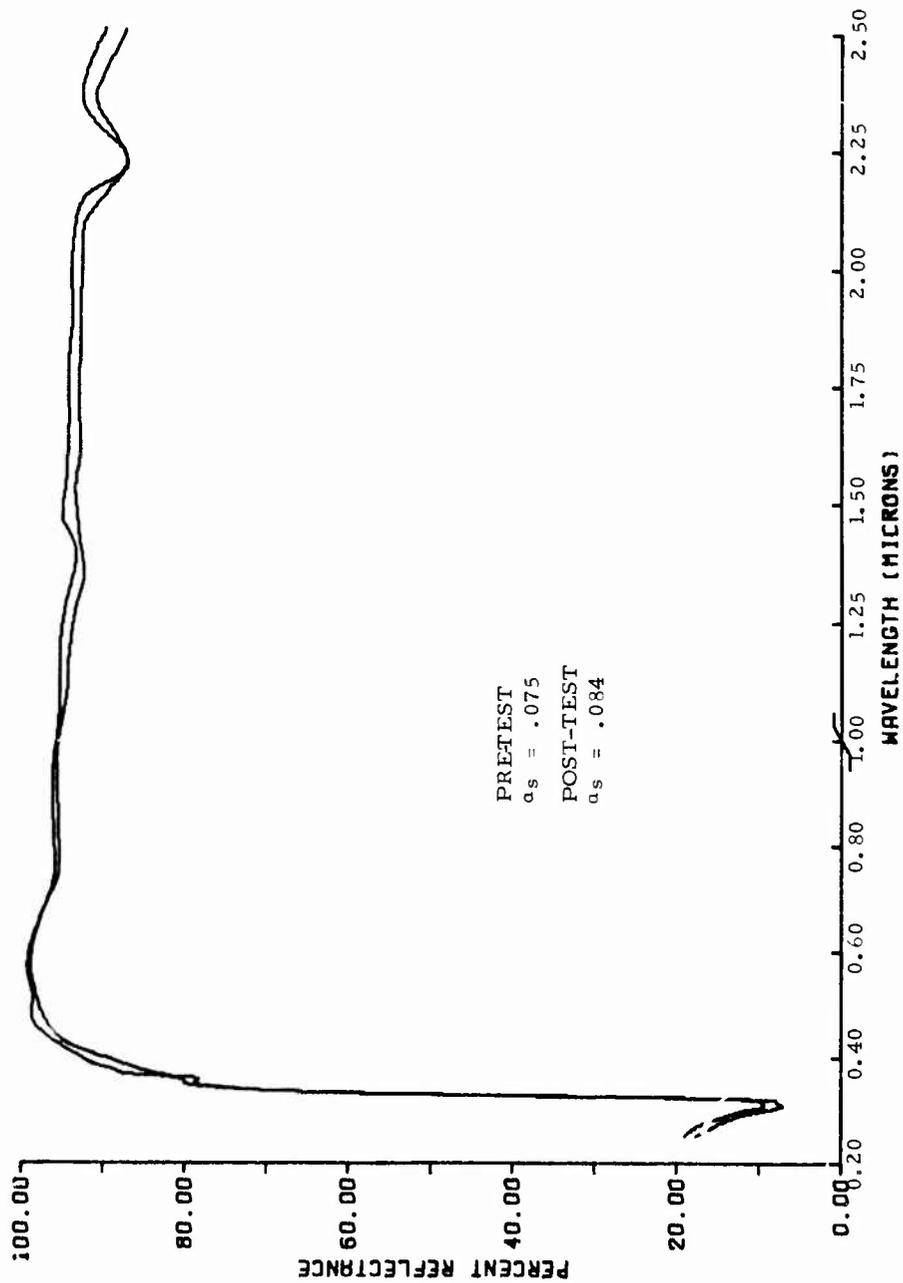


Figure 13. ML-101 simulation Experiment .050" OSR (500 EUVSH)

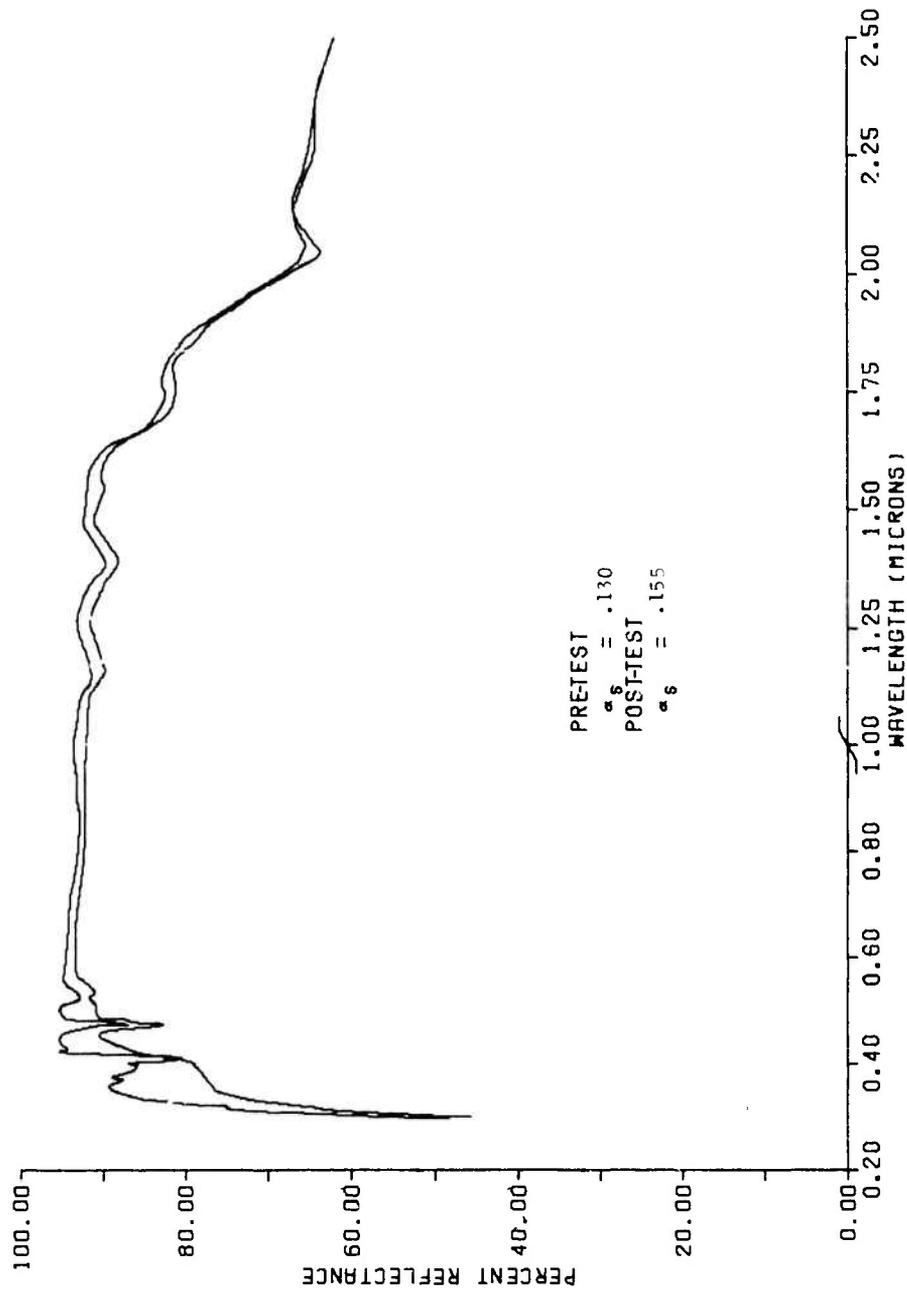


Figure 14. ML-101 Simulation Experiment $\text{Eu}_2\text{O}_3/\text{RTV602}$ (500 EUVSH)

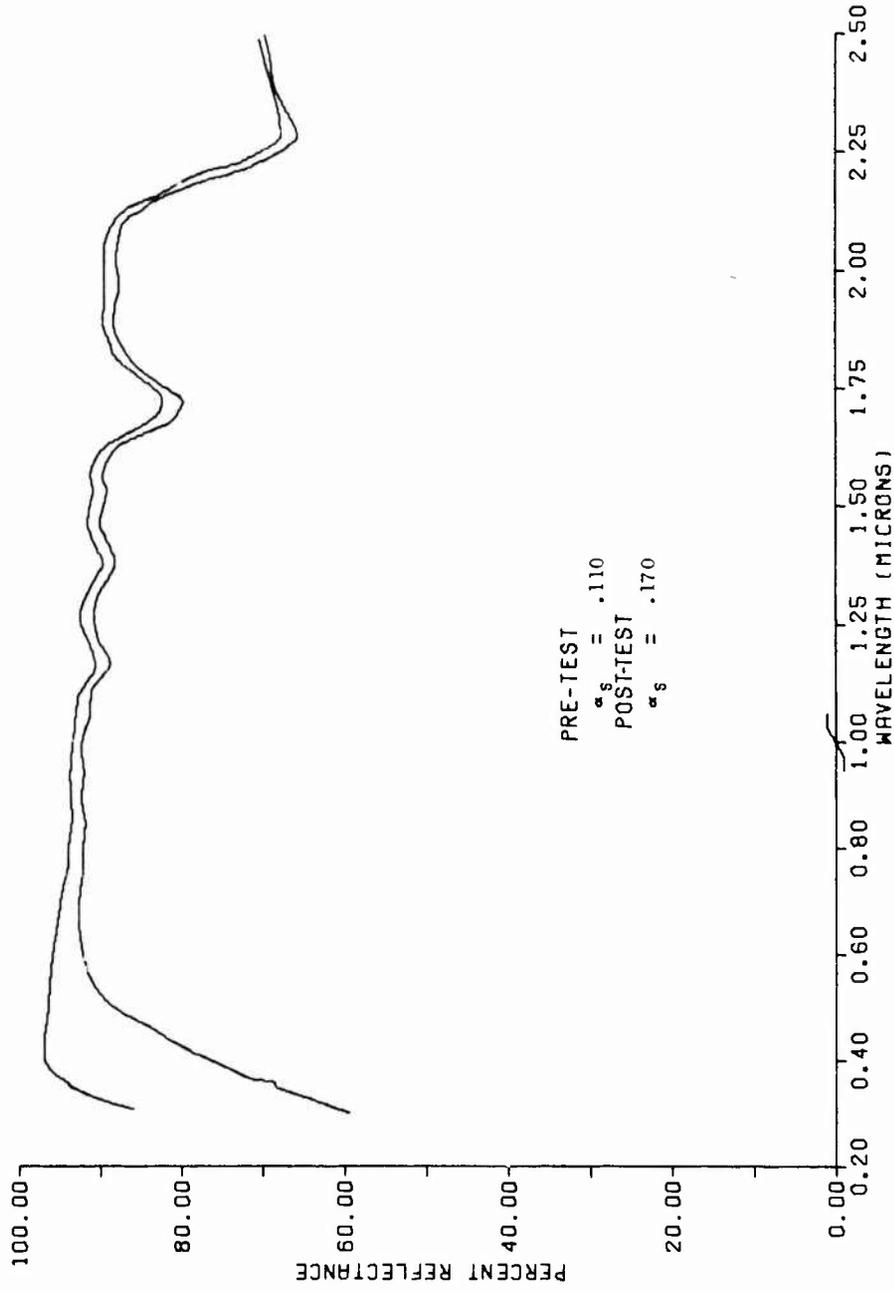


Figure 15. ML-101 Simulation Experiment Al₂O₃/RTV602 (500 EUVSH)

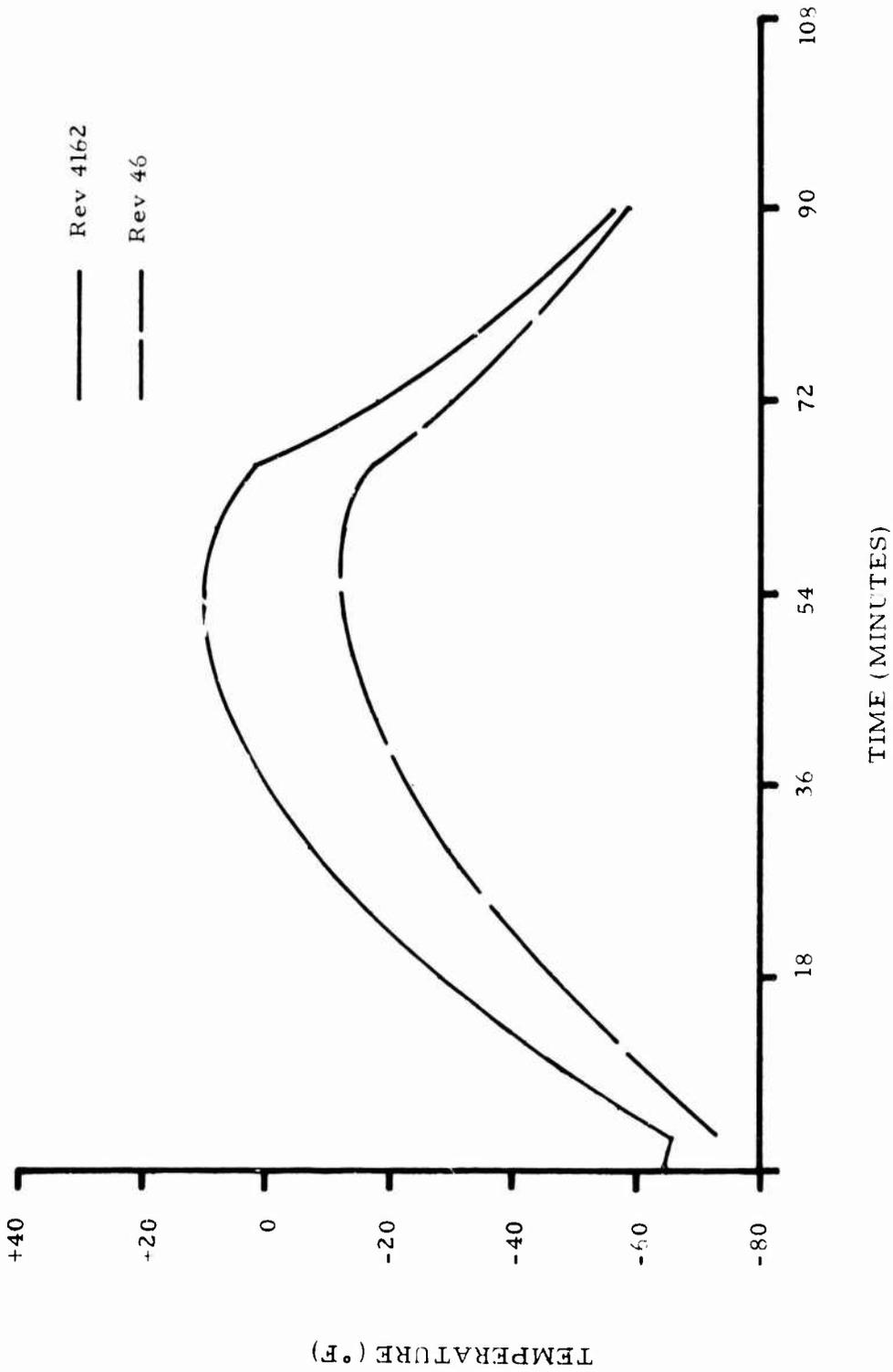


Figure 16. Sensor Temperature vs Time for S-13G During Revolutions 46 and 4162

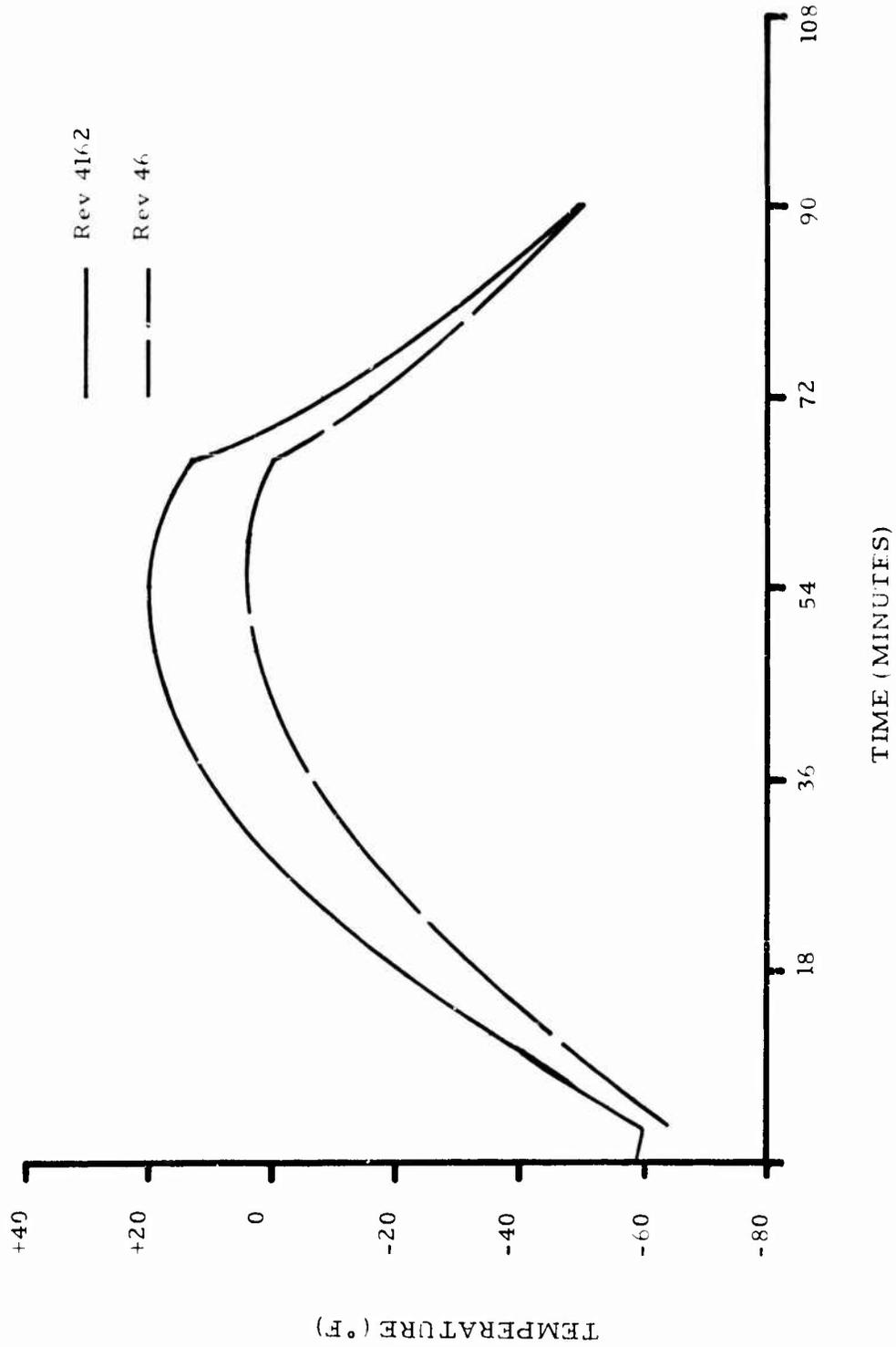


Figure 17. Sensor Temperature vs Time for SiO₂ Fabric During Revolutions 46 and 4162

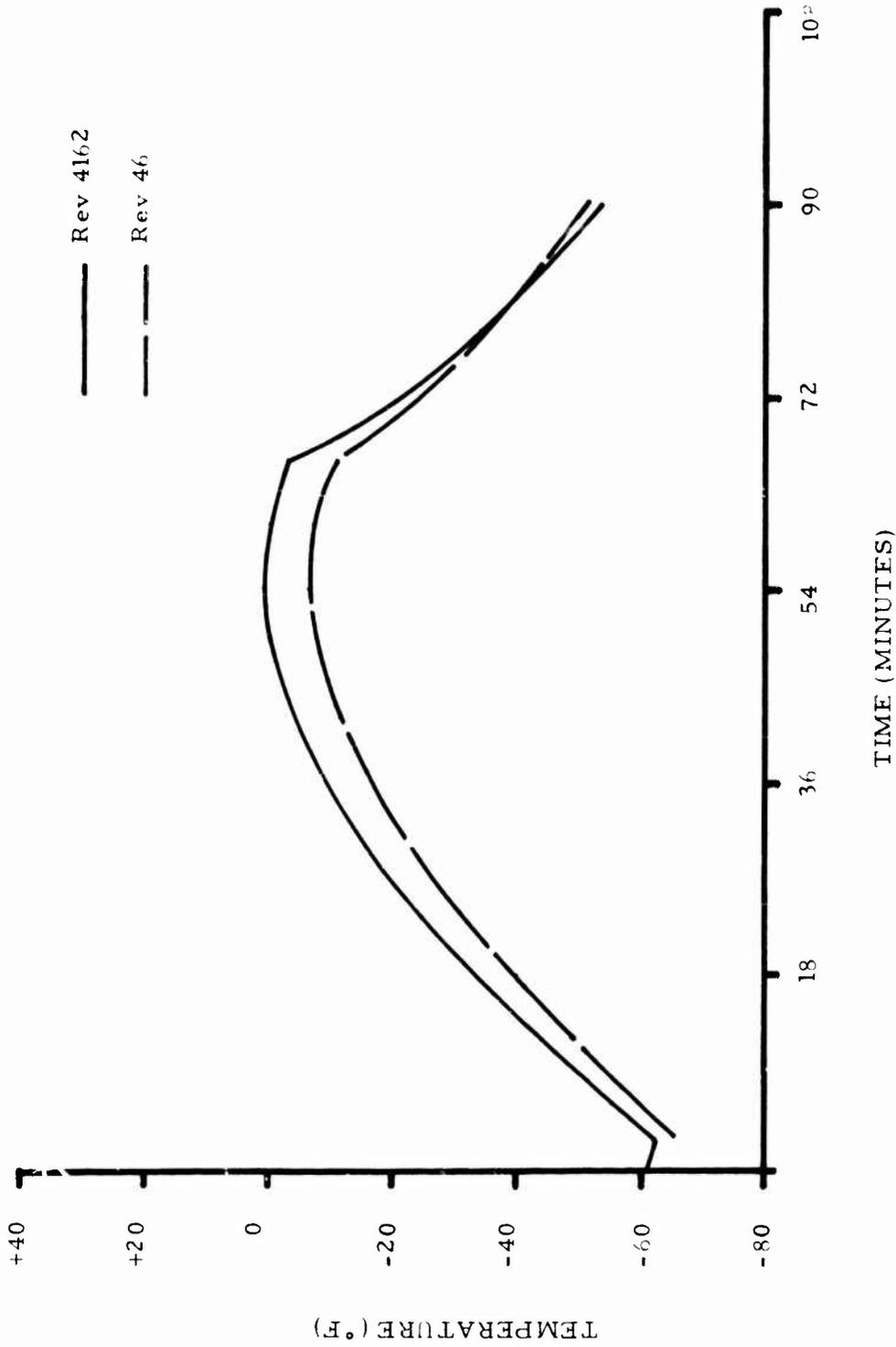


Figure 18. Sensor Temperature vs Time for FEP/A1 During Revolutions 46 and 4162

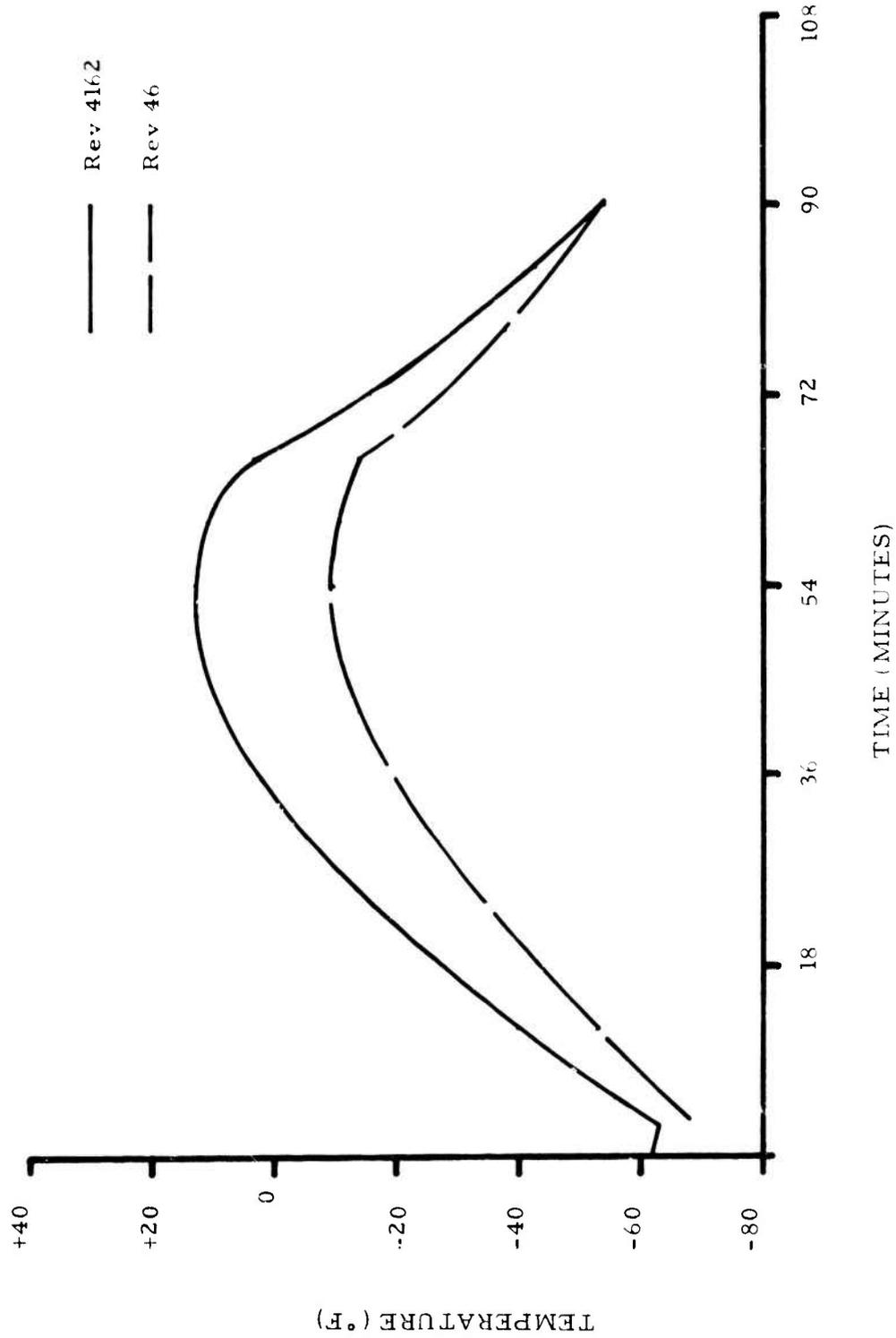


Figure 19. Sensor Temperature vs Time for $Zn_2TiO_4/01650$ During Revolutions 46 and 4162

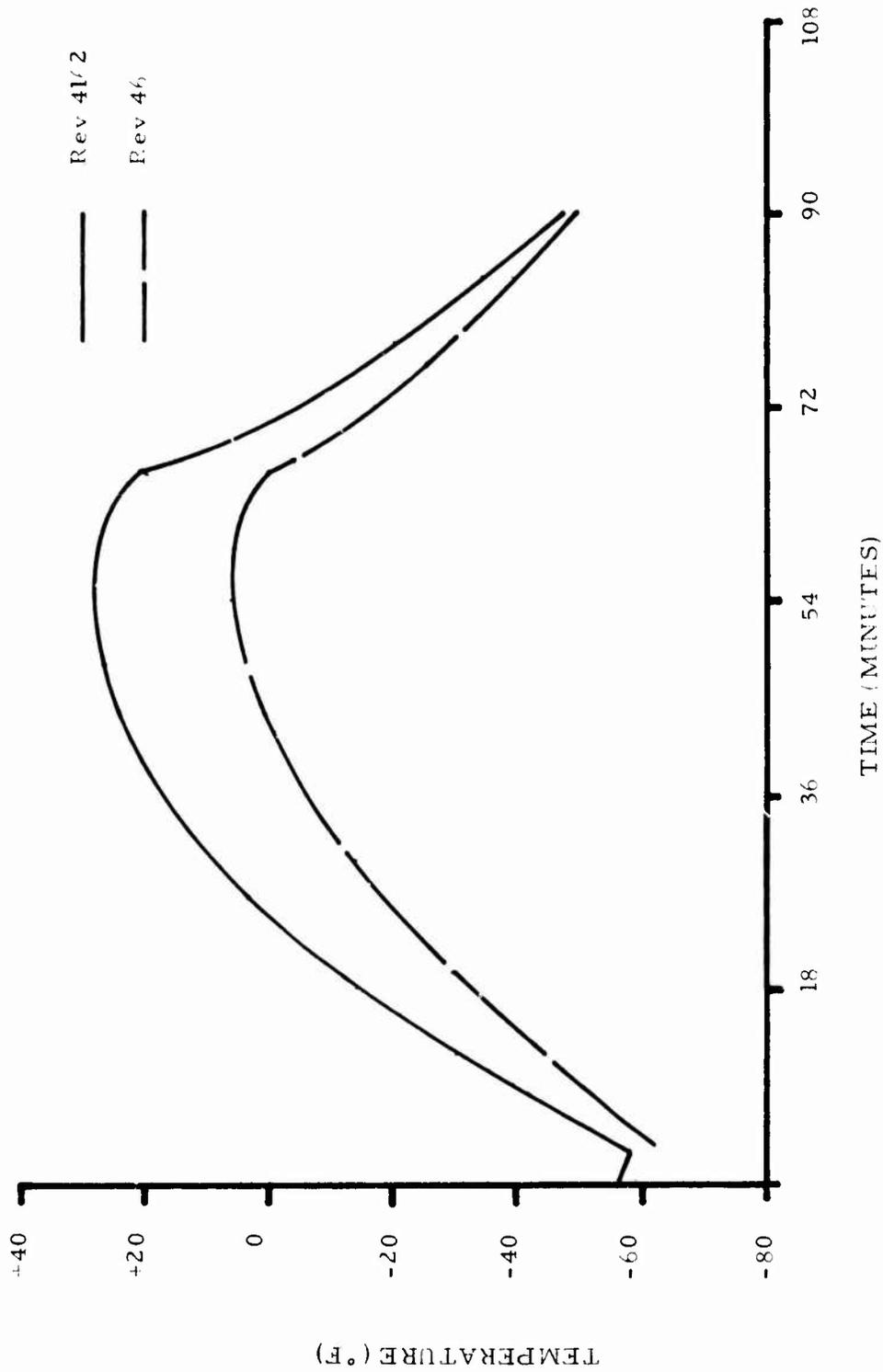


Figure 20. Sensor Temperature vs Time for PV-100 During Revolutions 46 and 4162

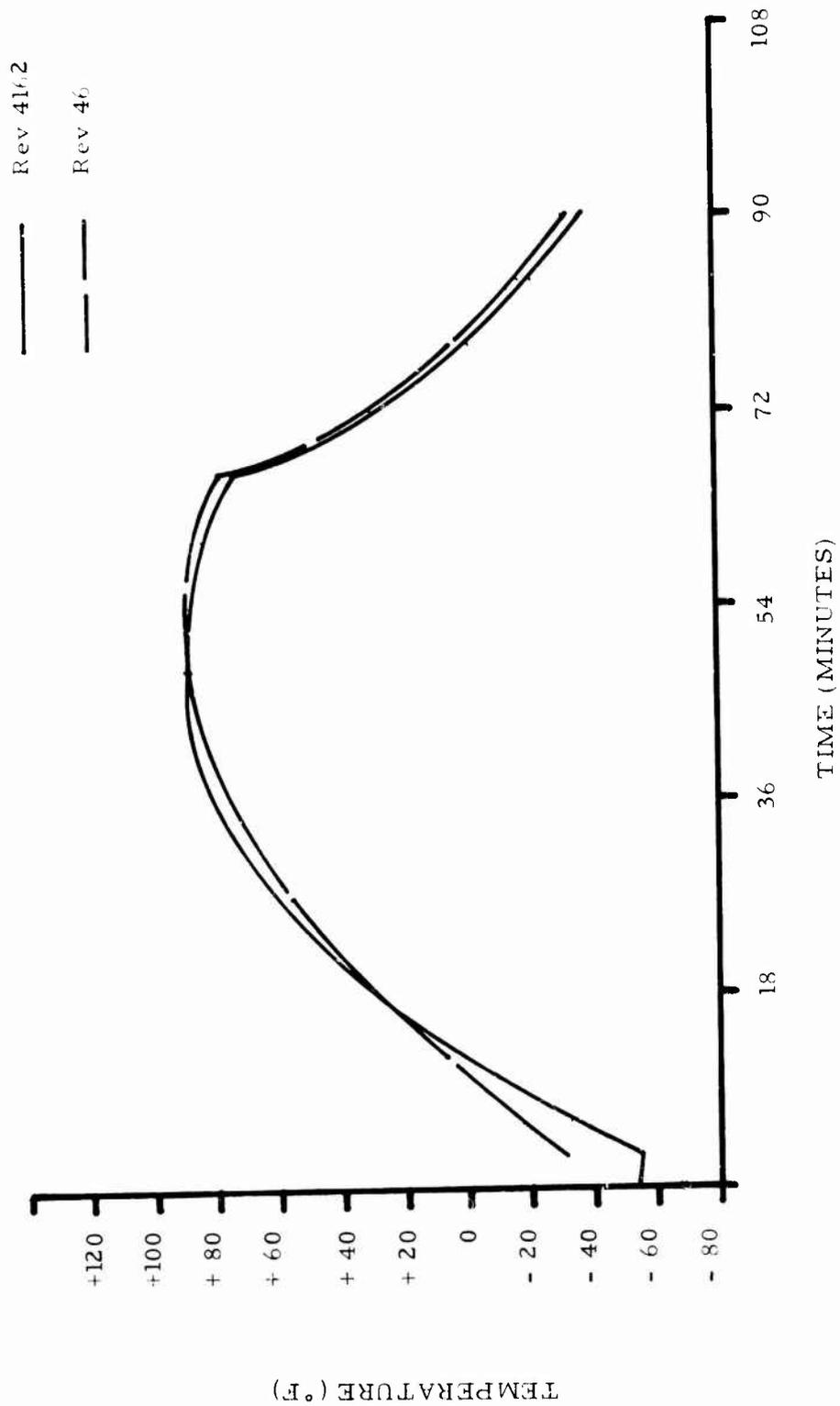


Figure 21. Sensor Temperature vs Time for 3M Black Velvet During Revolutions 46 and 4162

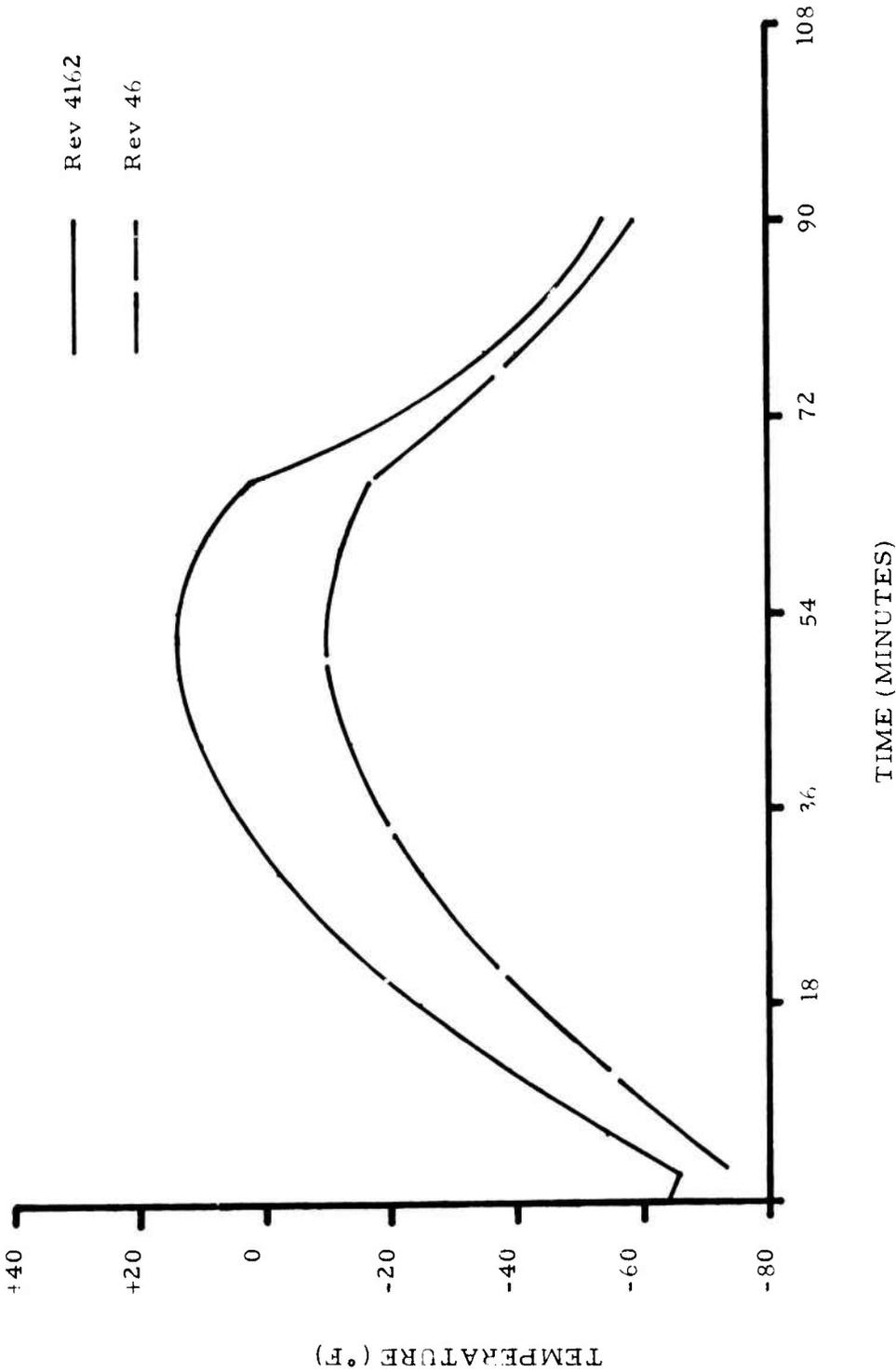


Figure 22. Sensor Temperature vs Time for TiO₂/Silicone During Revolutions 46 and 4162

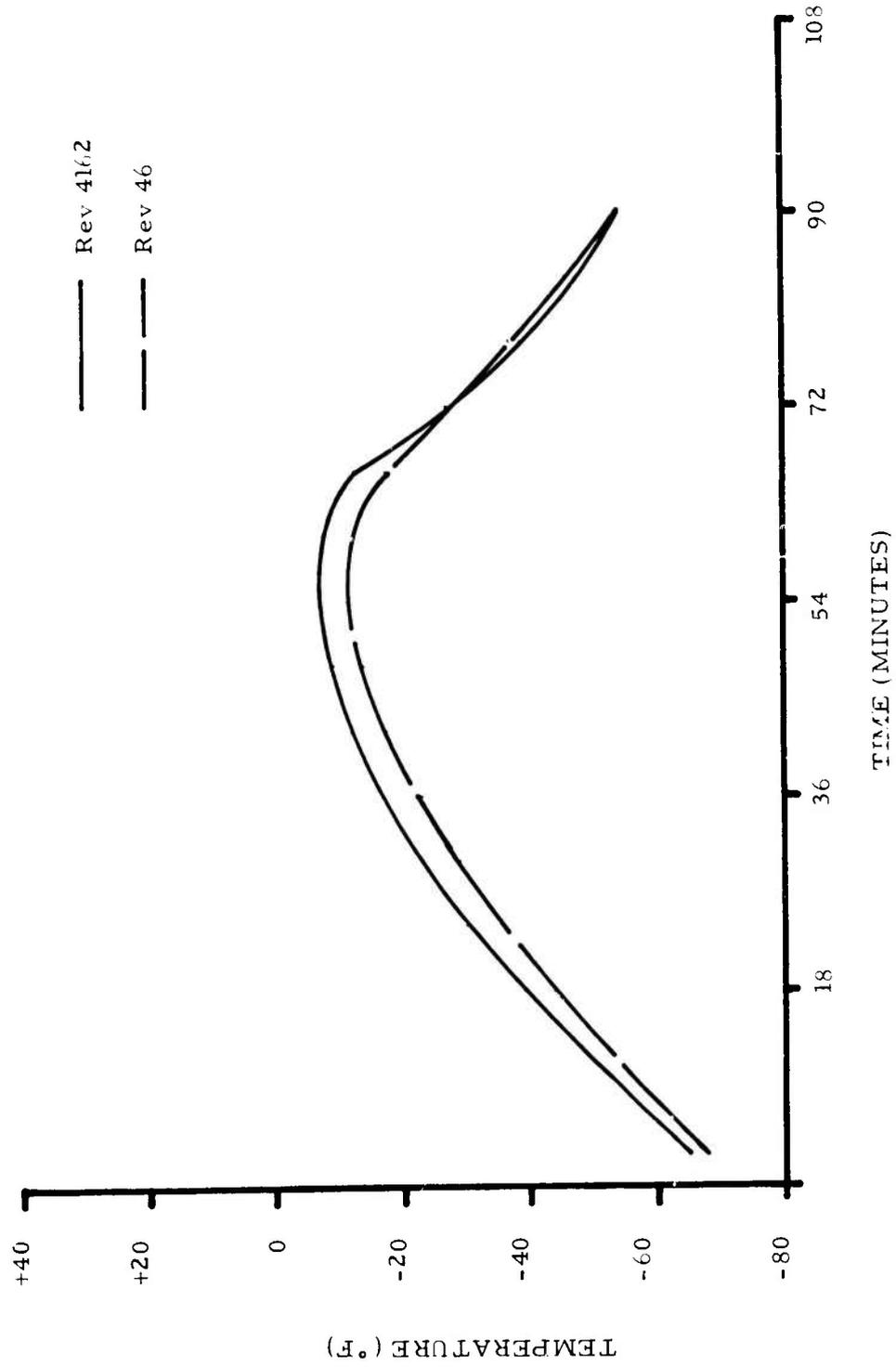


Figure 23. Sensor Temperature vs Time for OSR (0.008) During Revolutions 46 and 4162

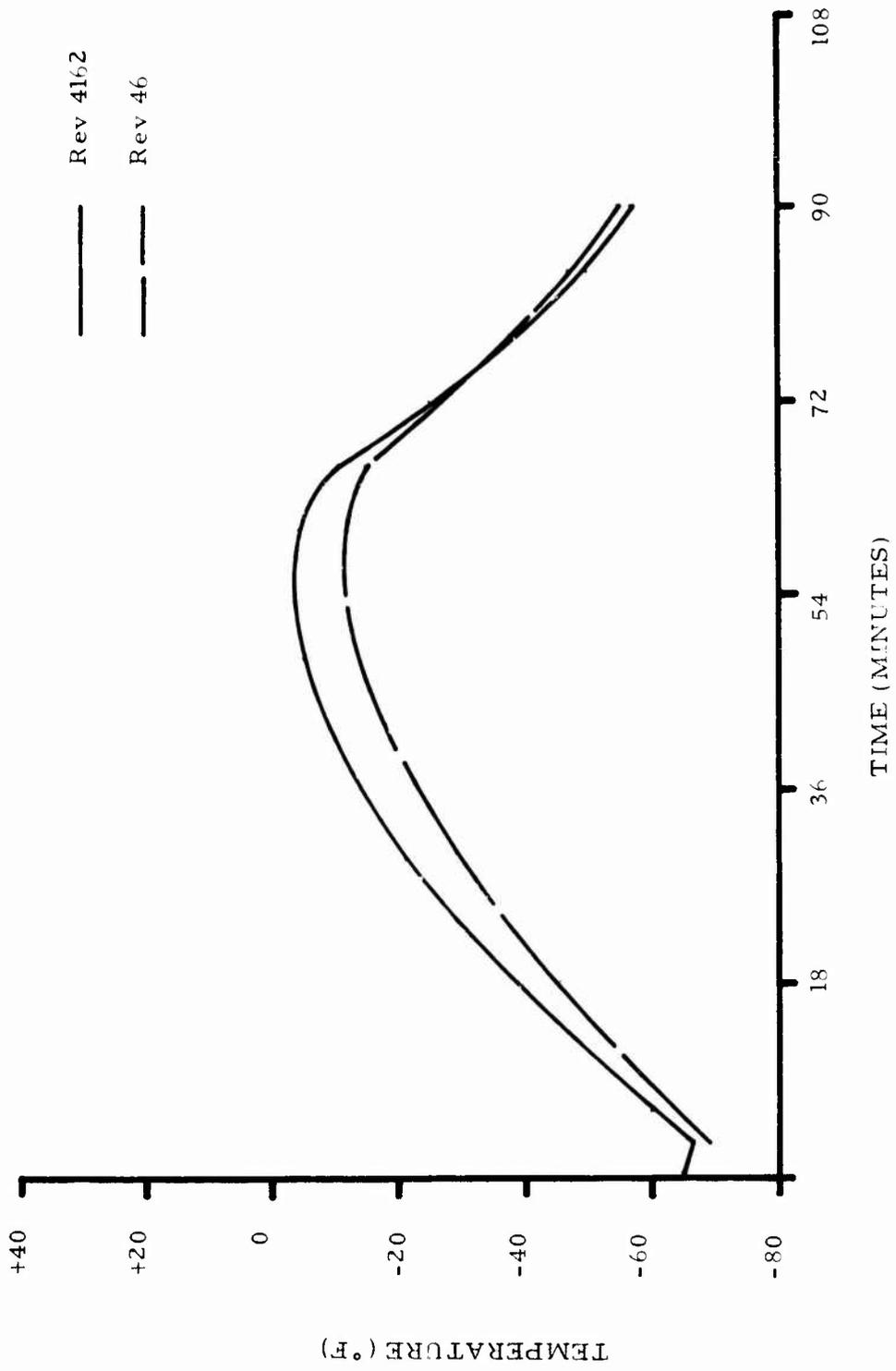


Figure 24. Sensor Temperature vs Time for OSR (0.050) During Revolutions 46 and 4162

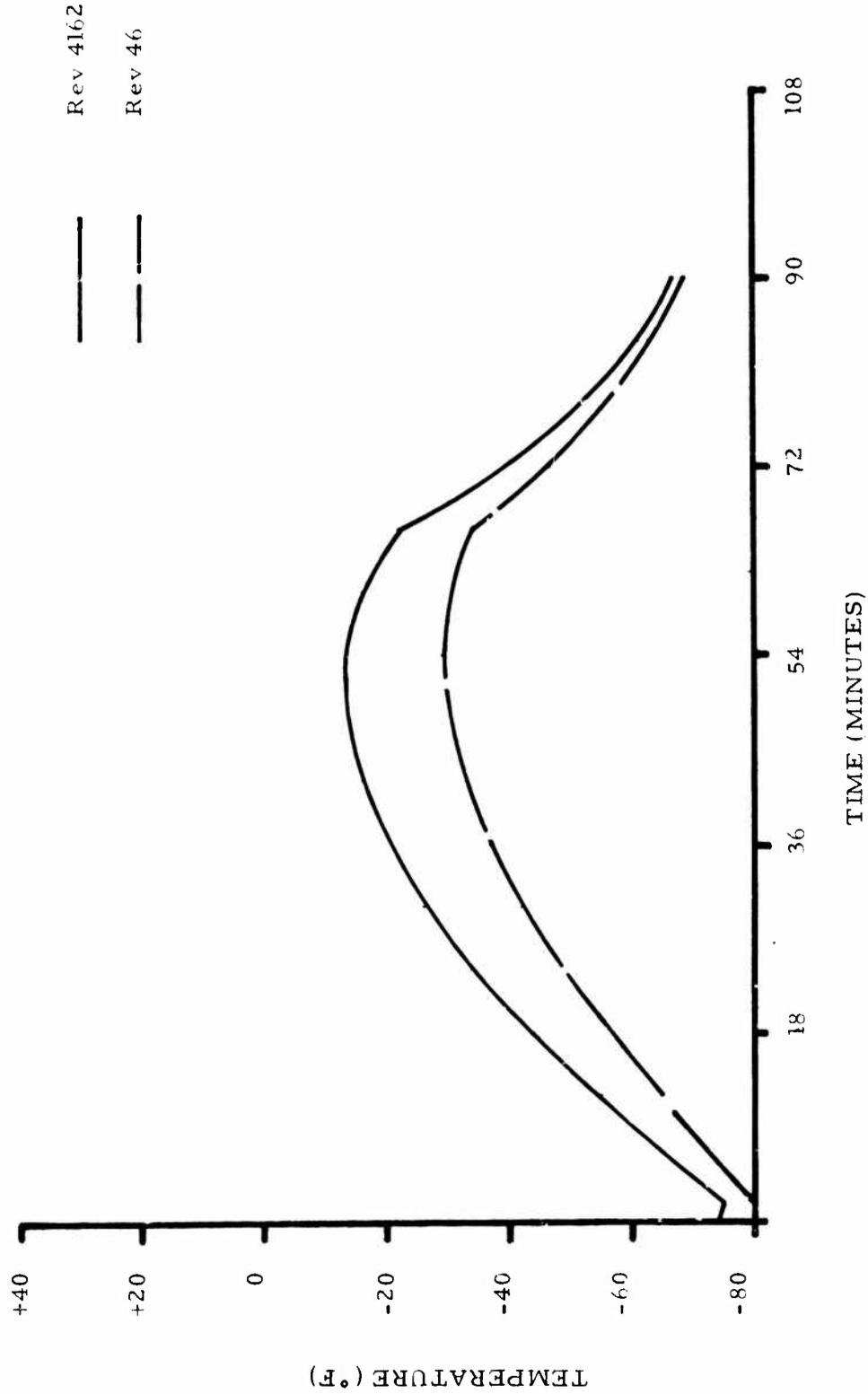


Figure 25. Sensor Temperature vs Time for $\text{Eu}_2\text{O}_3/\text{Silicone}$ During Revolutions 46 and 4162

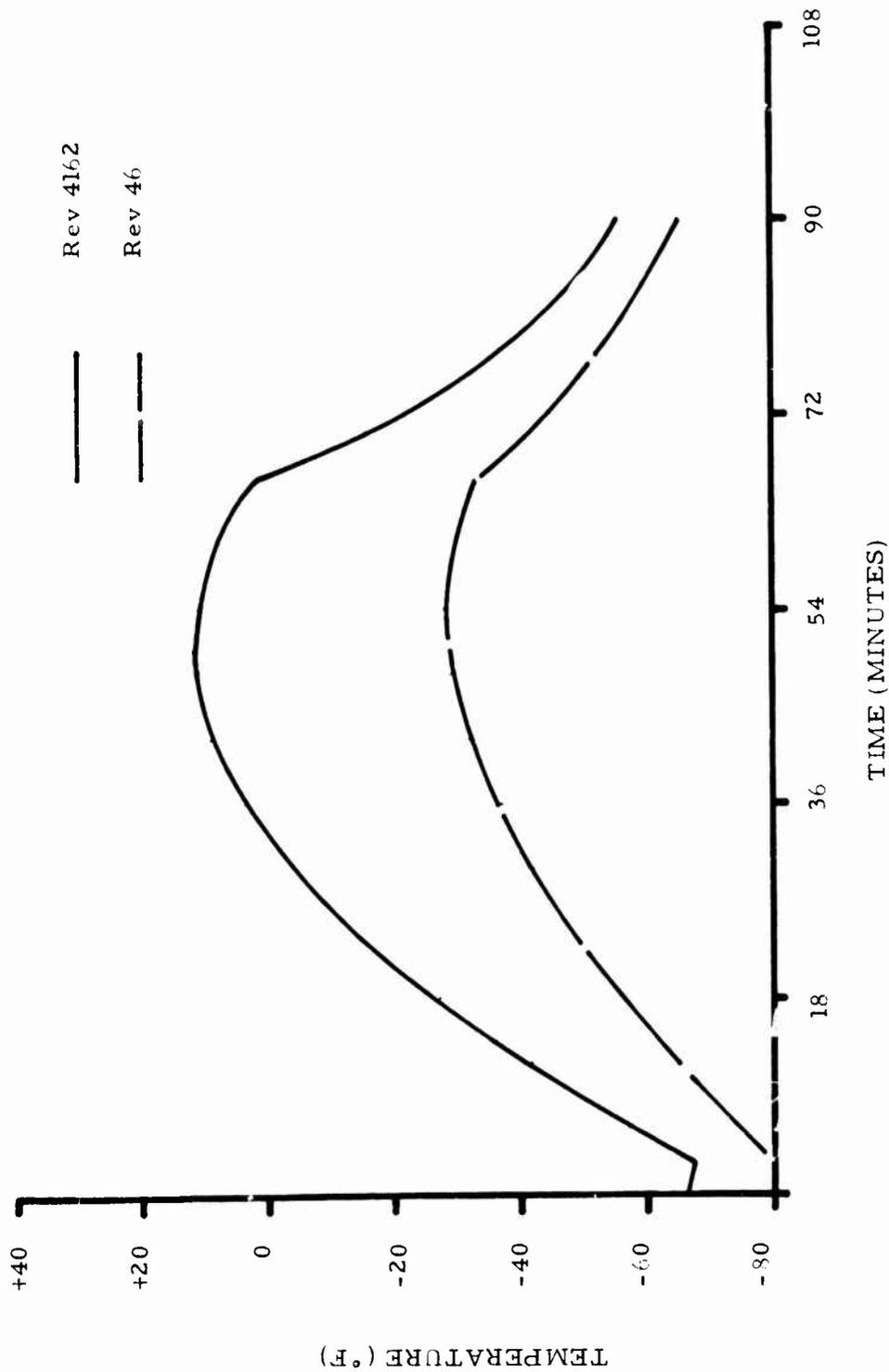


Figure 26. Sensor Temperature vs Time for α -Al₂O₃/Silicone During Revolutions 46 and 4162

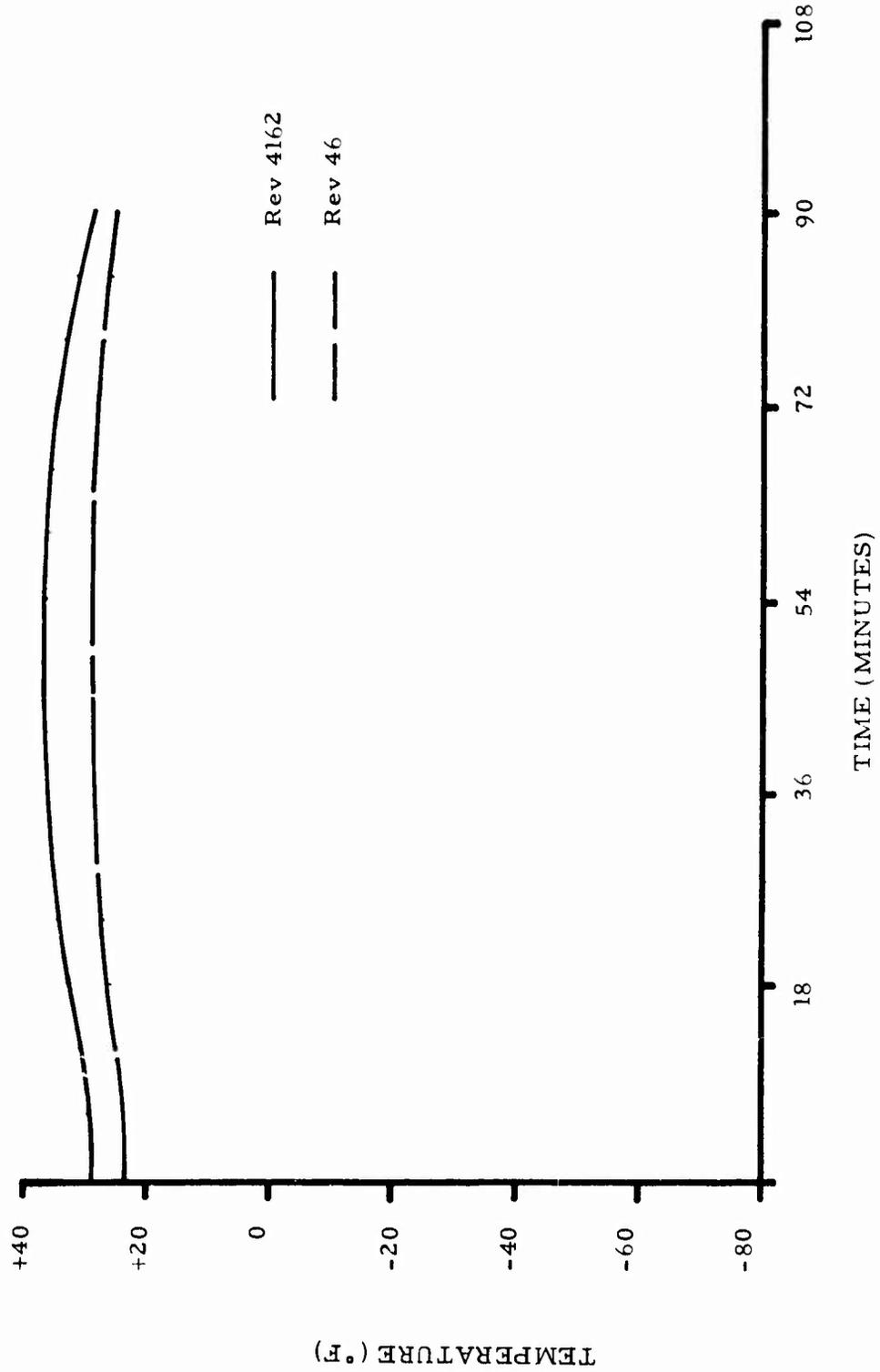


Figure 27. Sensor Temperature vs Time for Temperature Reference During Revolutions 46 and 4162

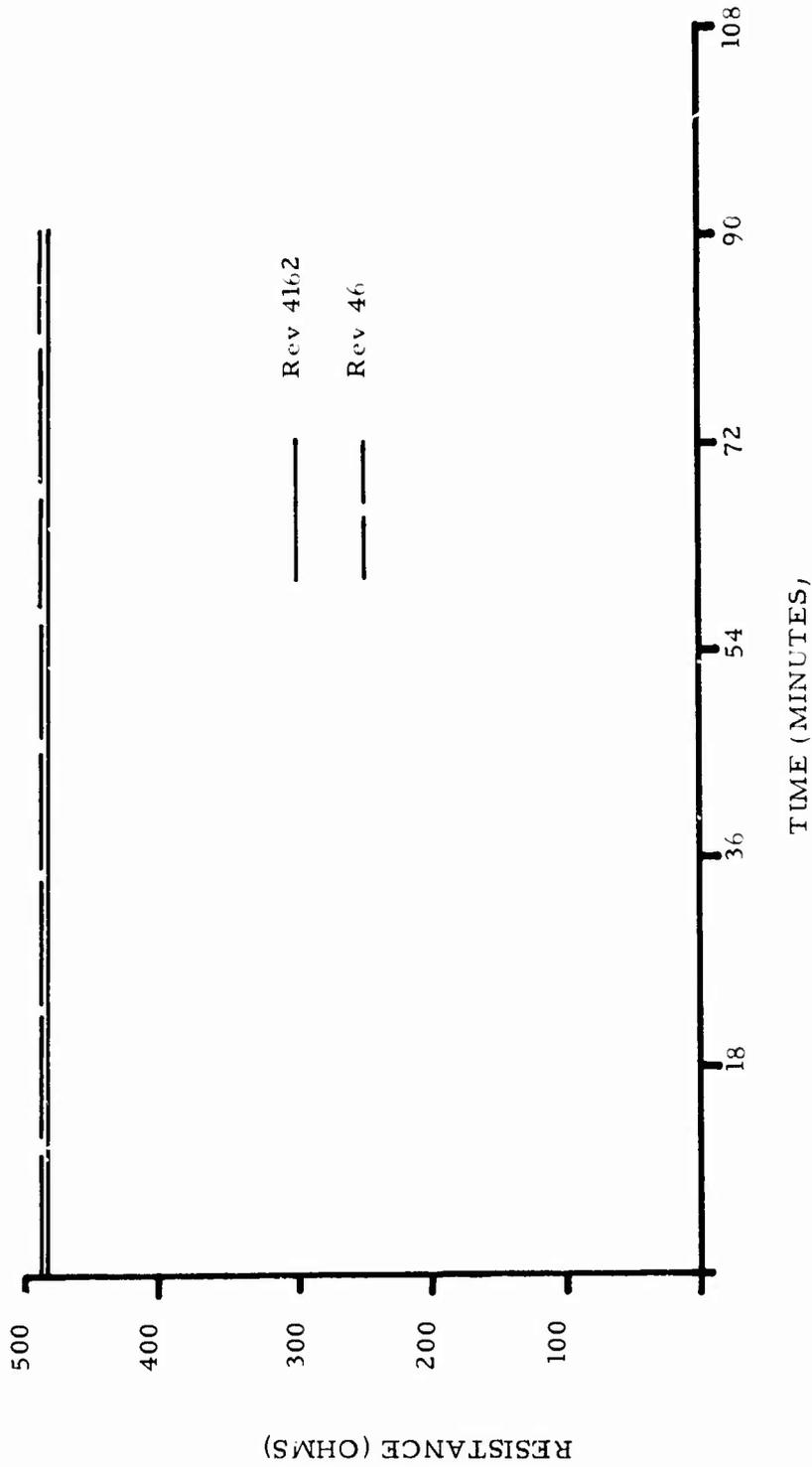


Figure 28. Resistance vs Time for Reference Resistor (499 Ohm)

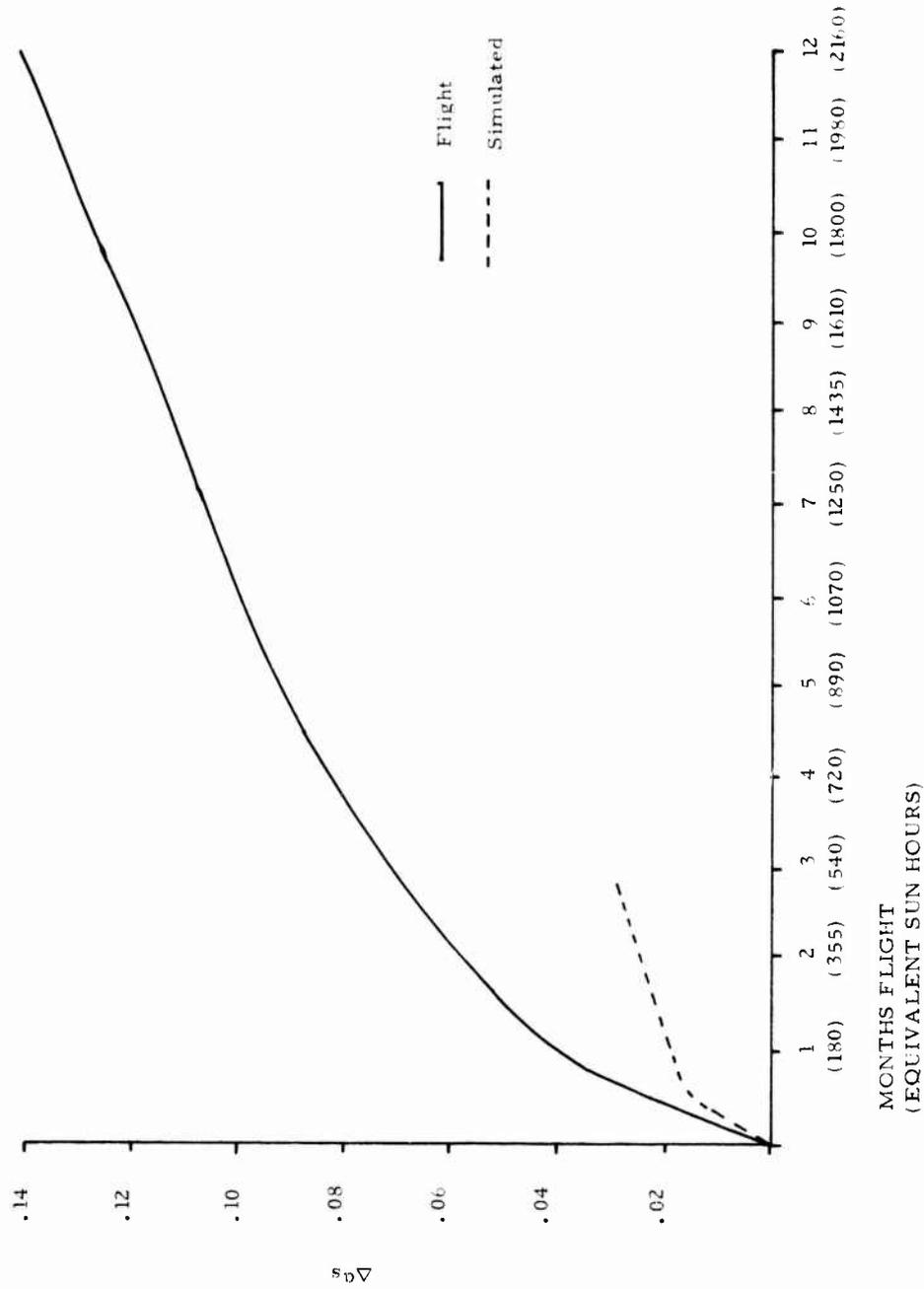


Figure 29. Comparison of Simulated and Flight Data for S-13G

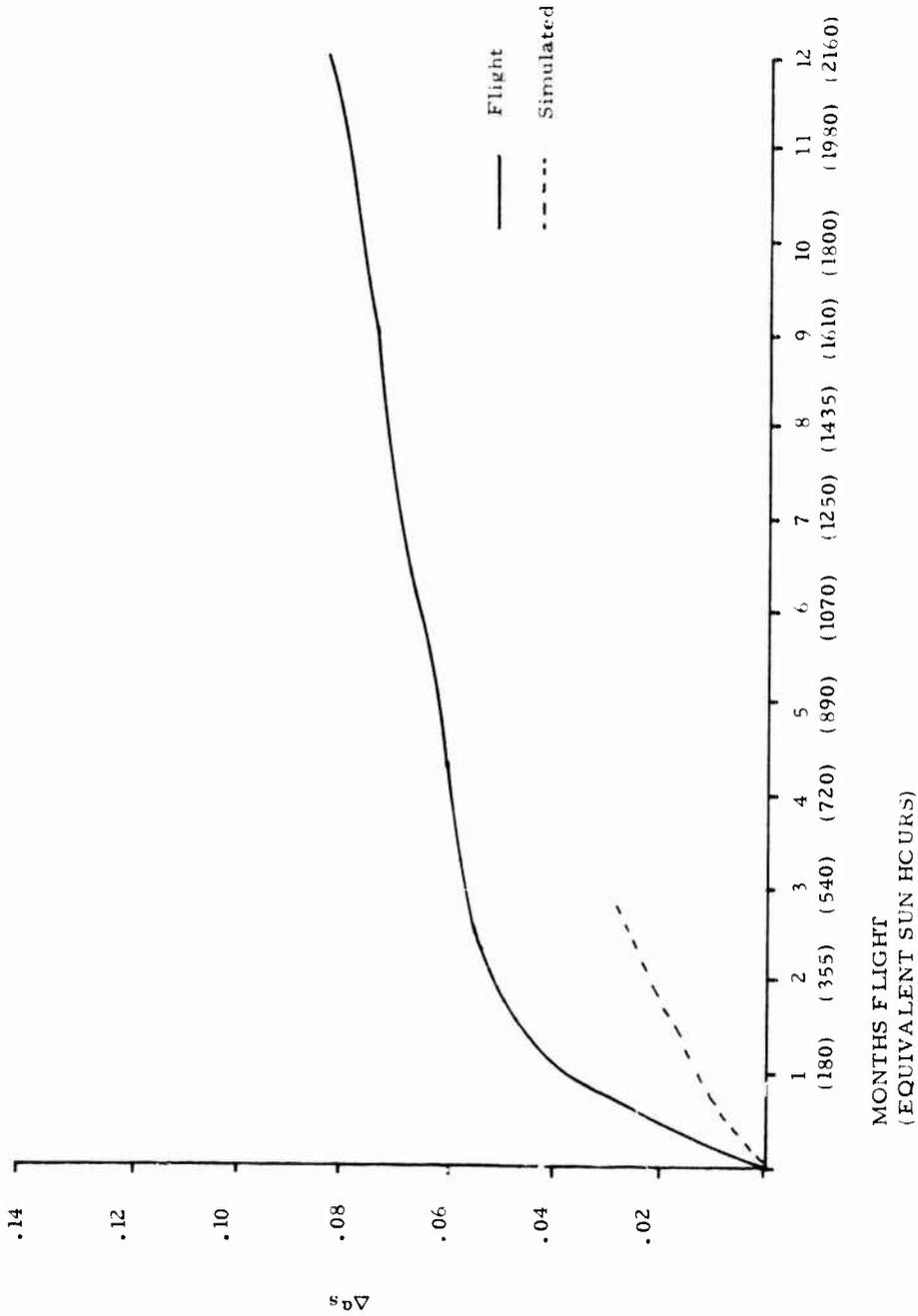


Figure 30. Comparison of Simulated and Flight Data for SiO_2 Fabric

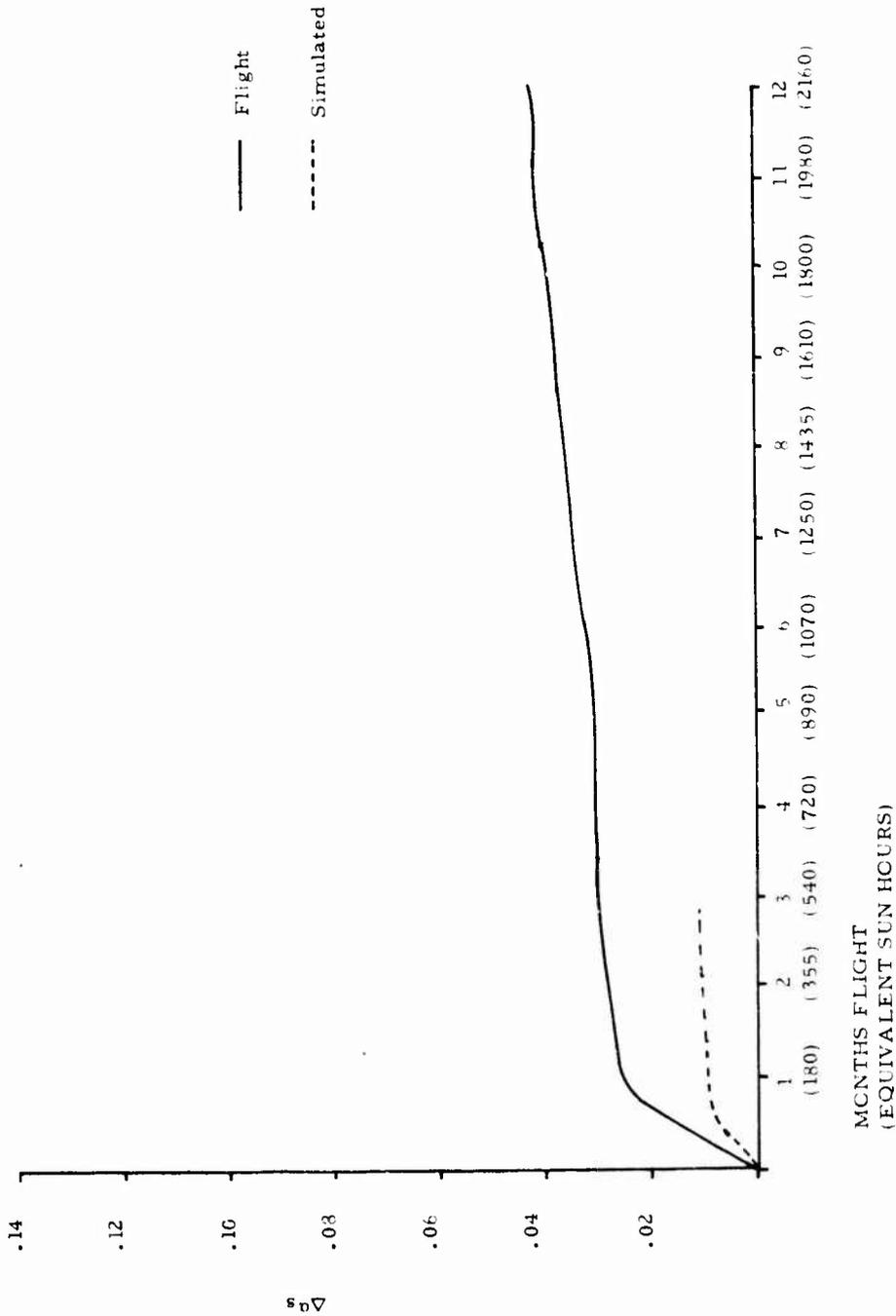


Figure 31. Comparison of Simulated and Flight Data for FEP/A1

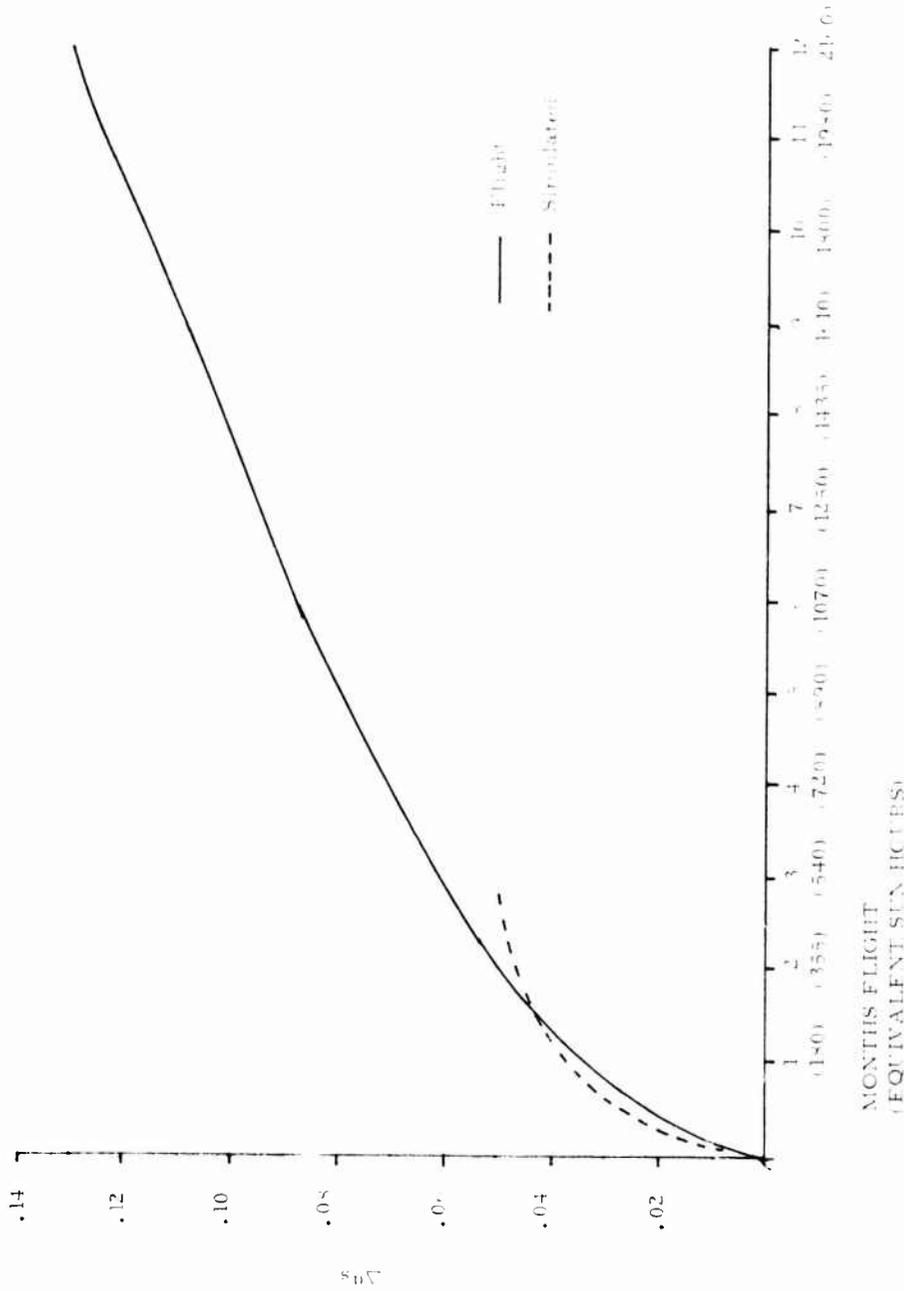


Figure 32. Comparison of Simulated and Flight Data for $Zn_2TiO_4/01650$

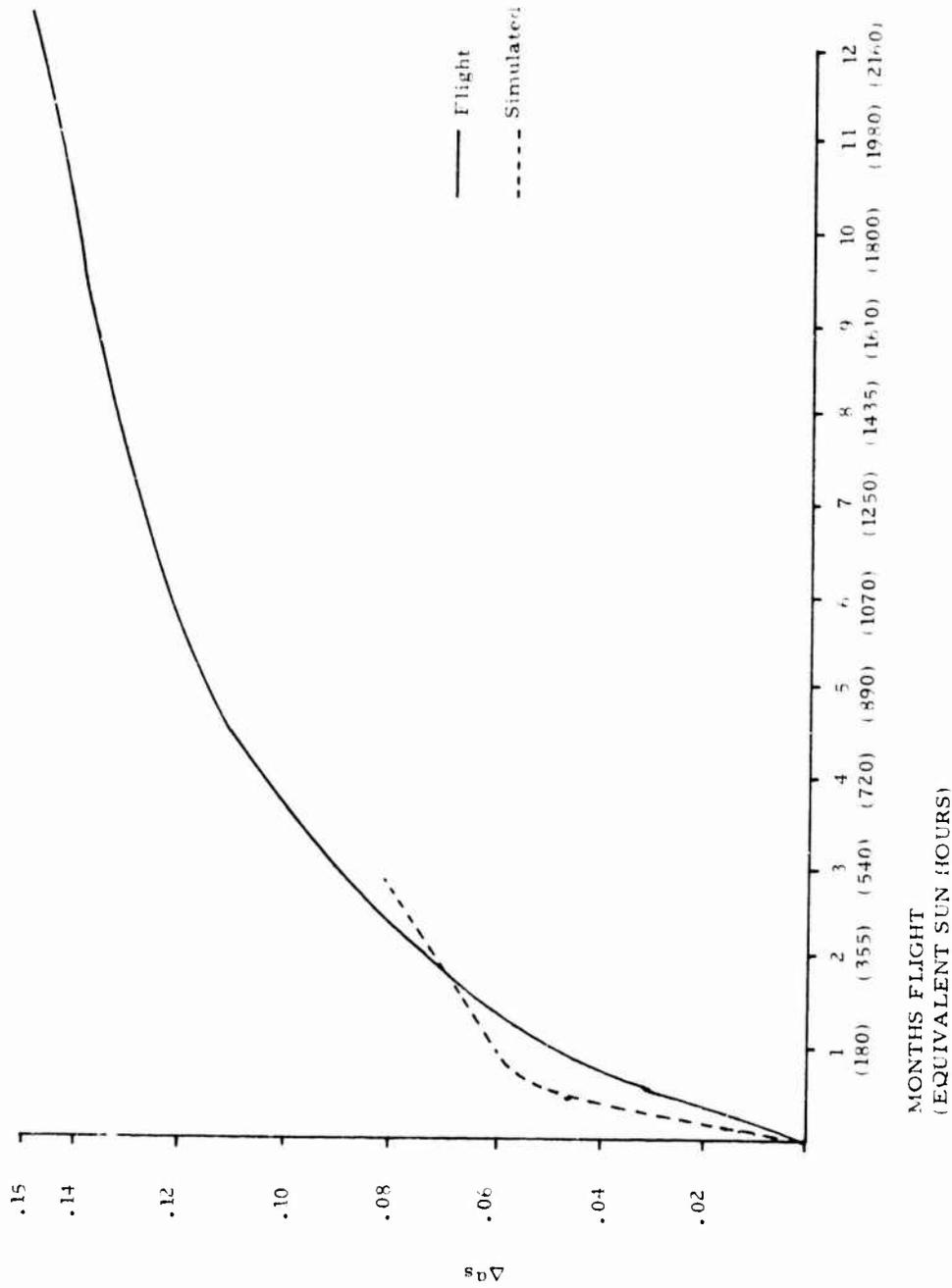


Figure 33. Comparison of Simulated and Flight Data for PV-100

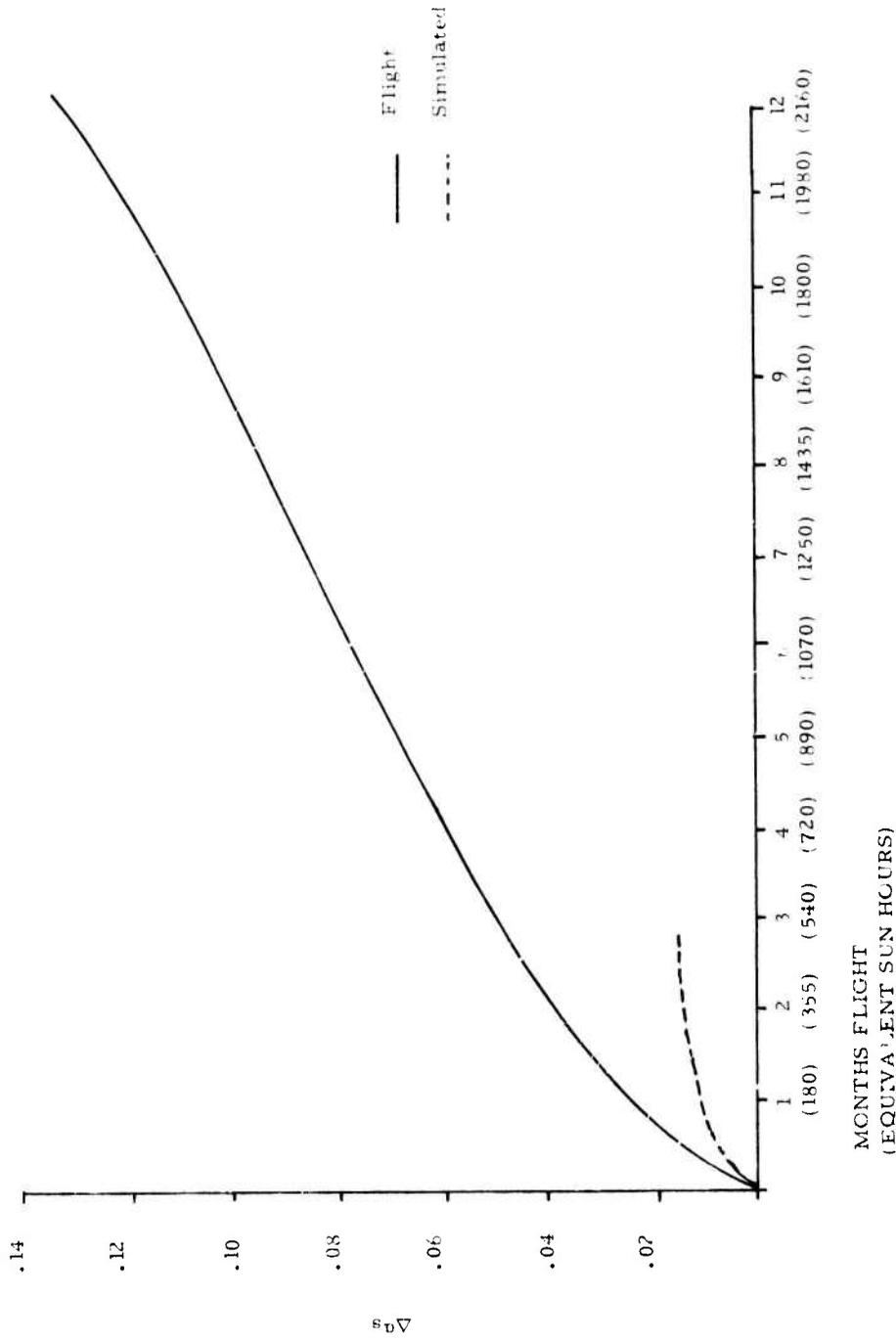


Figure 34. Comparison of Simulated and Flight Data for $TiO_2/Silicone$

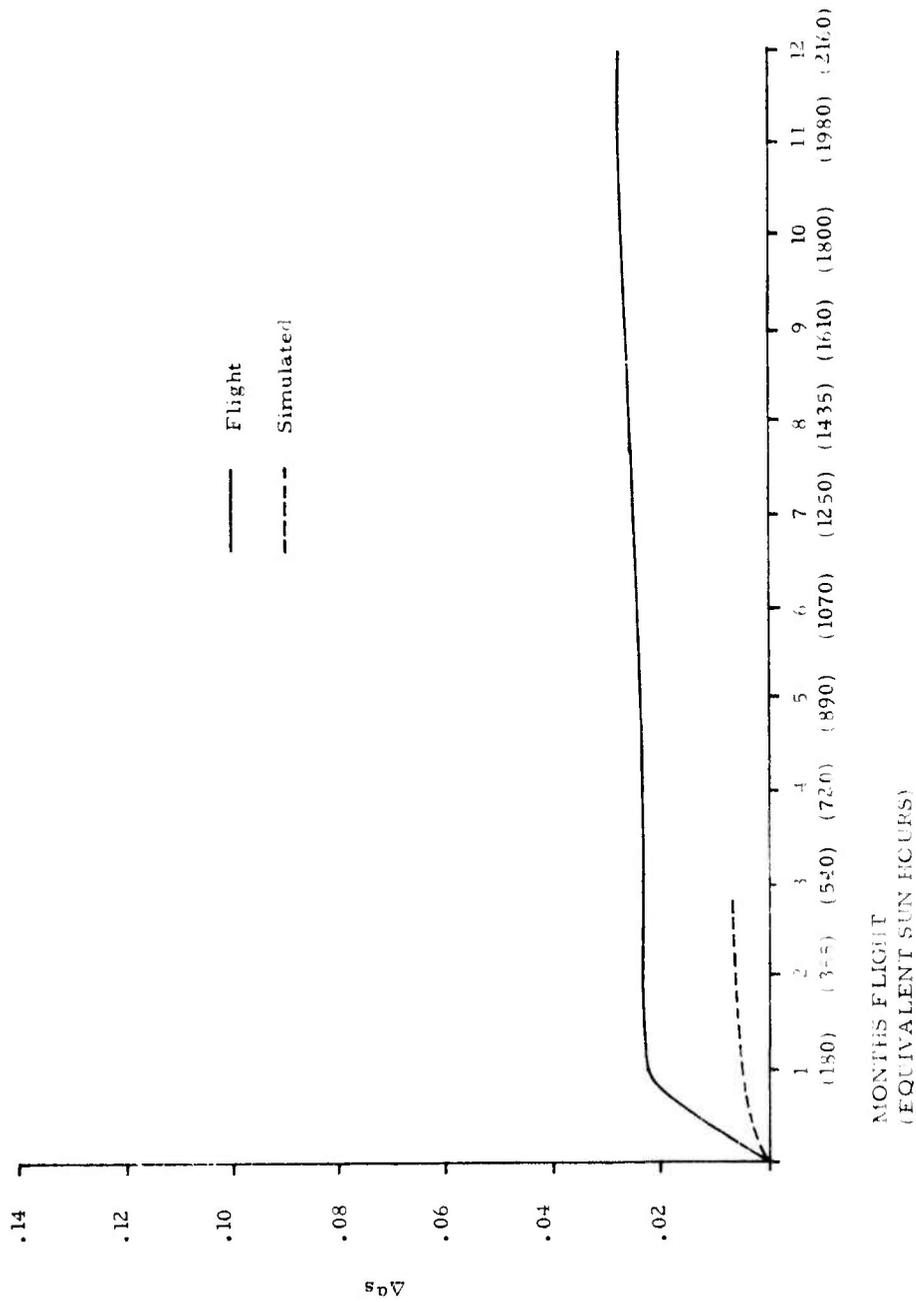


Figure 35. Comparison of Simulated and Flight Data for OSR (0.008)

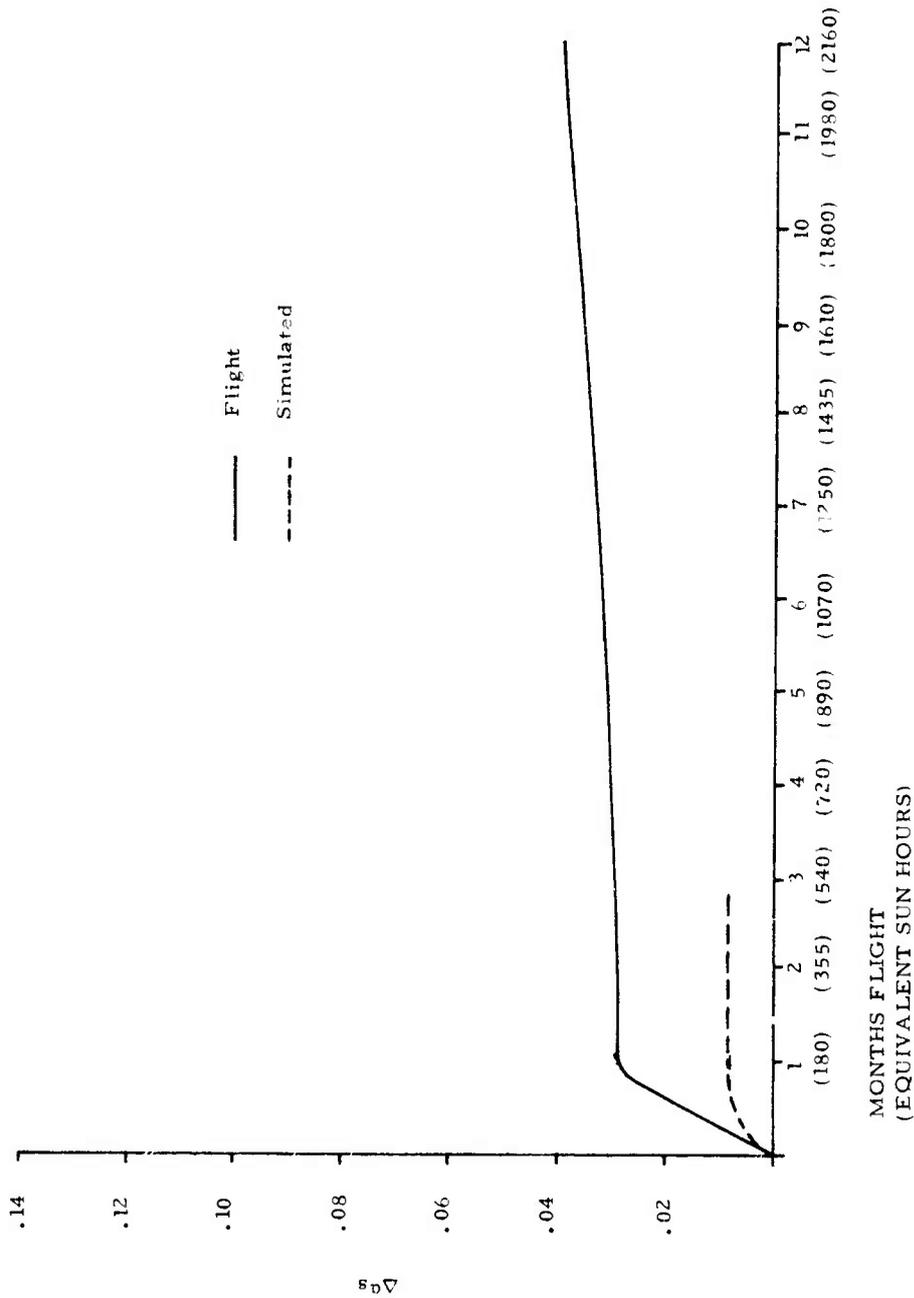


Figure 36. Comparison of Simulated and Flight Data for OSR (0.050)

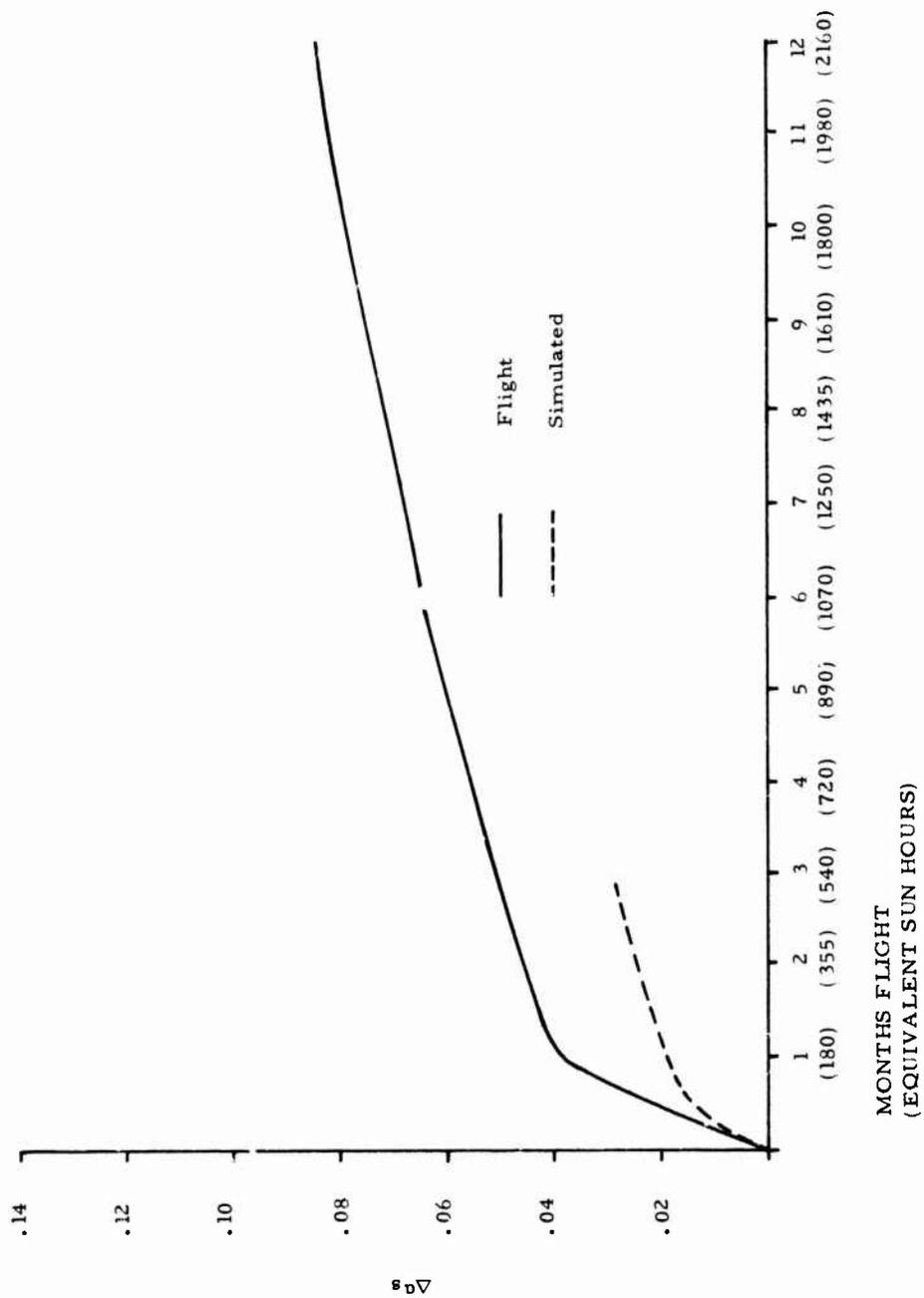


Figure 37. Comparison of Simulated and Flight Data for $\text{Eu}_2\text{O}_3/\text{Silicone}$

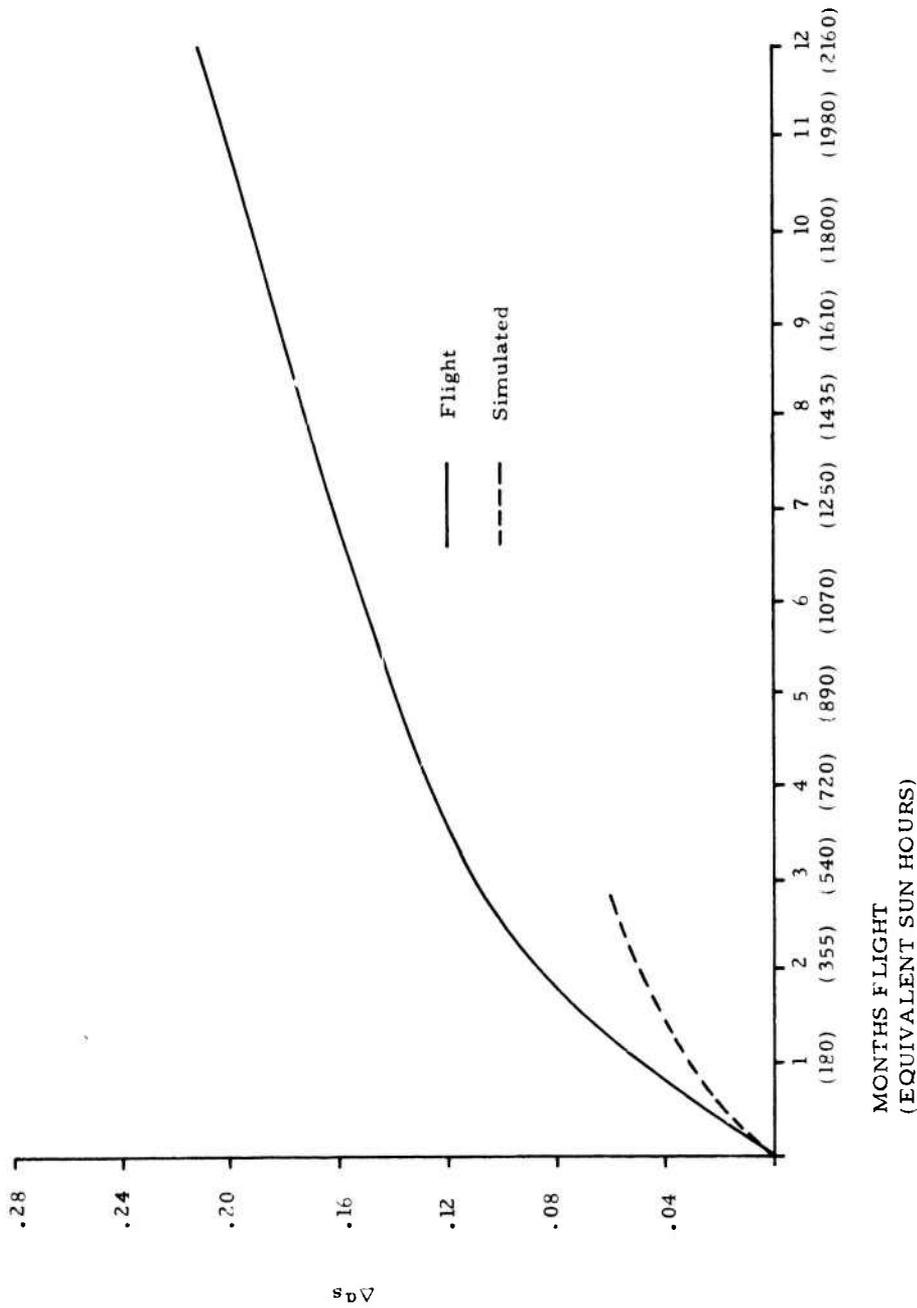


Figure 38. Comparison of Simulated and Flight Data for α -Al₂O₃/Silicone