

# REPORT DOCUMENTATION PAGE

*Form Approved*  
*OMB No. 0704-0188*

Public reporting burden for this collection of information is estimated to average 1 hour per response, including the time for reviewing instructions, searching existing data sources, gathering and maintaining the data needed, and completing and reviewing this collection of information. Send comments regarding this burden estimate or any other aspect of this collection of information, including suggestions for reducing this burden to Department of Defense, Washington Headquarters Services, Directorate for Information Operations and Reports (0704-0188), 1215 Jefferson Davis Highway, Suite 1204, Arlington, VA 22202-4302. Respondents should be aware that notwithstanding any other provision of law, no person shall be subject to any penalty for failing to comply with a collection of information if it does not display a currently valid OMB control number. **PLEASE DO NOT RETURN YOUR FORM TO THE ABOVE ADDRESS.**

<b>1. REPORT DATE (DD-MM-YYYY)</b> 29-11-2011		<b>2. REPORT TYPE</b> Conference Paper		<b>3. DATES COVERED (From - To)</b>	
<b>4. TITLE AND SUBTITLE</b>  <b>Functionalized Fluoroalkyl Polyhedral Oligomeric Silsesquioxane (F-POSS)</b>				<b>5a. CONTRACT NUMBER</b>	
				<b>5b. GRANT NUMBER</b>	
				<b>5c. PROGRAM ELEMENT NUMBER</b>	
<b>6. AUTHOR(S)</b> Sean M. Ramirez, Yvonne Diaz, Timothy S. Haddad and Joseph M. Mabry				<b>5d. PROJECT NUMBER</b>	
				<b>5f. WORK UNIT NUMBER</b> 23030521	
<b>7. PERFORMING ORGANIZATION NAME(S) AND ADDRESS(ES)</b> Air Force Research Laboratory (AFMC) AFRL/RZSM 9 Antares Road Edwards AFB CA 93524-7401				<b>8. PERFORMING ORGANIZATION REPORT NUMBER</b>	
<b>9. SPONSORING / MONITORING AGENCY NAME(S) AND ADDRESS(ES)</b>  Air Force Research Laboratory (AFMC) AFRL/RZS 5 Pollux Drive Edwards AFB CA 93524-7048				<b>10. SPONSOR/MONITOR'S ACRONYM(S)</b>	
				<b>11. SPONSOR/MONITOR'S NUMBER(S)</b> AFRL-RZ-ED-TP-2011-558	
<b>12. DISTRIBUTION / AVAILABILITY STATEMENT</b>  Approved for public release; distribution unlimited (PA #111052).					
<b>13. SUPPLEMENTARY NOTES</b> For presentation at the American Chemical Society National Conference, San Diego, CA 25-29 Mar 2012.					
<b>14. ABSTRACT</b> A variety of functionalized Fluorinated Polyhedral Oligomeric SilSesquioxanes (F-POSS) were synthesized and characterized. The chemical structures were confirmed using multinuclear NMR spectroscopy ( <sup>1</sup> H, <sup>13</sup> C, <sup>19</sup> F, and <sup>29</sup> Si), FT-IR, and combustion analysis. Dynamic contact angle measurements of these compounds were taken with water and hexadecane. Copolymers containing F-POSS were synthesized from monomer functionalized F-POSS. These novel structures can be used as initial building blocks for the development of low surface energy materials.					
<b>15. SUBJECT TERMS</b>					
<b>16. SECURITY CLASSIFICATION OF:</b>			<b>17. LIMITATION OF ABSTRACT</b>	<b>18. NUMBER OF PAGES</b>	<b>19a. NAME OF RESPONSIBLE PERSON</b>
<b>a. REPORT</b>	<b>b. ABSTRACT</b>	<b>c. THIS PAGE</b>			<b>19b. TELEPHONE NUMBER</b> <i>(include area code)</i>
Unclassified	Unclassified	Unclassified	SAR	4	N/A

FUNCTIONALIZED FLUORINATED POLYHEDRAL  
OLIGOMERIC SILSESQUIOXANE (F-POSS)

*Sean M. Ramirez, Yvonne Diaz, Timothy S. Haddad, and Joseph M.  
Mabry*

ABSTRACT: A variety of functionalized Fluorinated Polyhedral Oligomeric Silsesquioxanes (F-POSS) were synthesized and characterized. The chemical structures were confirmed using multinuclear NMR spectroscopy ( $^1\text{H}$ ,  $^{13}\text{C}$ ,  $^{19}\text{F}$ , and  $^{29}\text{Si}$ ), FT-IR, and combustion analysis. Dynamic contact angle measurements of these compounds were taken with water and hexadecane. Copolymers containing F-POSS were synthesized from monomer functionalized F-POSS. These novel structures can be used as initial building blocks for the development of low surface energy materials.

## FUNCTIONALIZED FLUOROALKYL POLYHEDRAL OLIGOMERIC SILSESQUIOXANE (F-POSS)

Sean M. Ramirez,<sup>1</sup> Yvonne Diaz,<sup>2</sup> Timothy S. Haddad,<sup>1</sup> and Joseph M. Mabry<sup>2</sup>

<sup>1</sup>ERC Inc., Air Force Research Laboratory, Space & Missile Propulsion Division, Edwards Air Force Base, CA 93524-7680

<sup>2</sup>Air Force Research Laboratory, Space & Missile Propulsion Division, Edwards Air Force Base, CA 93524-7680

### Introduction

Recently, Fluorinated Polyhedral Oligomeric Silsesquioxanes (F-POSS), contain a Si-O core [SiO<sub>1.5</sub>] with long-chain fluorinated alkyl group periphery, were developed for low-surface energy applications.<sup>1</sup> F-POSS was found to possess the lowest surface energy values known ( $\gamma_{sv} = 9.3$  mN/m) for a crystalline solids.<sup>2</sup> The incorporation of F-POSS in polymers has led to the development of both superhydrophobic and superoleophobic surfaces.<sup>3,4</sup> These composites were based on the physical blending of F-POSS into a polymer matrix. Currently, there are no methods to functionalize long-chain F-POSS, thereby limiting its potential in other low-surface energy applications due to poor mechanical robustness and abrasion resistance. Herein, we report the synthesis and characterization of functionalized F-POSS. The functionalized compounds can both be physically blended and covalently bound to polymer matrices. These materials possess potential applications in superhydrophobic/superoleophobic coatings and low-surface energy materials.

### Experimental

**Materials.** All dichlorosilanes were purchased from Gelest. All reactions were performed under a nitrogen atmosphere unless otherwise noted.

**Instrumentation.** <sup>1</sup>H, <sup>13</sup>C, <sup>19</sup>F, and <sup>29</sup>Si NMR spectra were obtained on a Bruker 300-MHz or 400-MHz spectrometer. A heteronuclear inverse gated decoupling pulse sequence (NONOE) with a 12 sec delay was used to acquire <sup>29</sup>Si NMR spectra. Contact angle measurements were taken on an optical contact angle system OCA (Dataphysic).

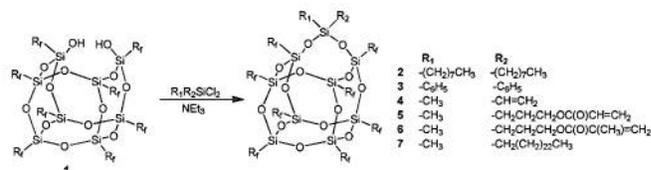
**General synthesis of functionalized F-POSS.** This synthesis will be discussed in detail in future publication. **Synthesis of compound (3).** **1** (2.90, 0.72 mmol) is dissolved in hexafluorobenzene. To this diphenyldichlorosilane (0.182 g, 0.72 mmol) is added and this was subsequently stirred for 15 min. Then triethylamine (0.19 mL, 1.44 mmol) is added slowly and the solution was stirred for an additional 12 hr. This product is subsequently precipitated from an ethyl acetate:hexafluorobenzene solvent mixture to yield compound **3**. Yield 55%. <sup>1</sup>H NMR (CDCl<sub>3</sub>; AK-225G: ppm)  $\delta$  7.69 (m, 4H), 7.46 (m, 6H), 2.12 (br m, 16H), 1.06 (m, 8H), 0.99 (m, 8H). <sup>13</sup>C NMR (C<sub>6</sub>F<sub>6</sub>, ppm)  $\delta$  133.4, 133.0, 127.07, 126.7, 123-105 (m, CF<sub>2</sub>, CF<sub>3</sub>), 24.4 (m), 2.0, 1.4, 0.9. <sup>29</sup>Si NMR (C<sub>6</sub>F<sub>6</sub>, ppm)  $\delta$  -45.0, -65.8, -68.0, -68.2 (1:2:2:4). <sup>19</sup>F NMR (CDCl<sub>3</sub>; C<sub>6</sub>F<sub>6</sub>, ppm)  $\delta$  -82.3 (3F), -116.9 (2F), -122.8 (6F), -123.7 (2F), -124.3 (2F), -127.3 (2F). IR (25 °C, KBr, cm<sup>-1</sup>) 2987, 2943, 1729, 1149, 1213, 1153, 976, 904, 814, 706, 663. m.p. = 120.2 -122.8 °C. Anal. Calcd. for C<sub>92</sub>H<sub>42</sub>F<sub>136</sub>O<sub>13</sub>Si<sub>9</sub> (found): C, 26.36 (26.22), H, 1.01 (0.91), F, 61.64 (61.38).

**General Polymerization of F-POSS monomers.** Methyl methacrylate (MMA, 1.52 g, 15.2 mmol), **7** (0.80 g, 0.002 mmol), and azobisisobutyronitrile (AIBN, 50 mg, 0.3 mmol) were dissolved in a fluorinated solvent:THF mixture (4:1). This solution was purged with N<sub>2</sub> for 25 min to remove any O<sub>2</sub> and was immediately submerged in a 65 °C oil bath for 36 hrs. The resulting solution was precipitated in hexanes, filtered and dried to yield a white powder (0.88 g, 58%).

**Contact angle measurements.** F-POSS compounds (10 mg/mL) were dissolved in a fluorinated solvent and spun cast at a rate of 900 rpm for 30 seconds onto oxygen-plasma treated 1-inch silicon wafers.

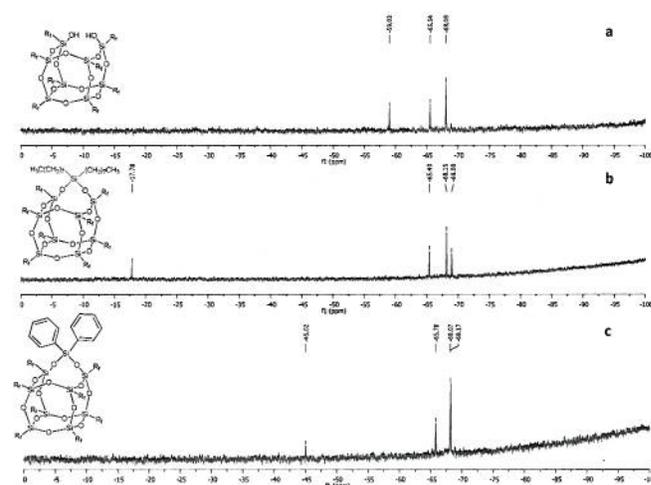
### Results and Discussion

**Synthesis of Functionalized F-POSS.** The incompletely-condensed silsesquioxane **1** can be readily reacted with a variety of dichlorosilanes (Scheme 1). For example, the reaction of **1** with diphenyldichlorosilane in the presence of triethylamine produced compound **3** (yield 55%). The main side product isolated during the reaction was closed-cage F-POSS. Multinuclear NMR (<sup>1</sup>H, <sup>29</sup>Si, <sup>19</sup>F), FT-IR, and elemental analysis were used to confirm the structure of compound **3**. The <sup>29</sup>Si peaks were observed at resonances of -45.2, -65.8, -68.0, and -68.2, with a ratio of 1:2:2:4 (Figure 1). The resonance at -45.2 ppm was attributed to the diphenyl functionalized Si.

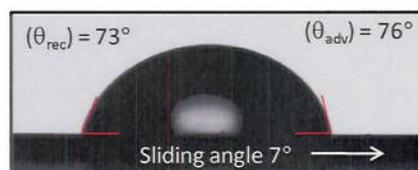


**Scheme 1.** Synthesis of F-POSS derivatives.

**Contact angle measurements of F-POSS monomers.** The wetting properties of F-POSS make it one of the lowest surface energy materials known. To demonstrate that functionalized F-POSS structures retain their impressive wetting properties, advancing ( $\theta_{adv}$ ) and receding ( $\theta_{rec}$ ) contact angle measurements were taken with water and hexadecane. Compared to unfunctionalized F-POSS and open-caged **1**, these modified F-POSS structures possess similar wetting properties to their predecessors. For example, compound **3** possess a low hysteresis, allowing for low sliding angles for hydrocarbons such as hexadecane (Figure 2). Initial observations will be explored in further detail to elucidate the impact of functionality on the wetting-properties of F-POSS.



**Figure 1.** <sup>29</sup>Si NMR spectra of a) **1** b) **2** and c) **3**.



**Figure 2.** Sliding angle of a 10  $\mu$ L drop of hexadecane rolling of Si wafer coated with compound **3**.

**Synthesis of copolymers.** Copolymerizations of methyl methacrylate (MMA) modified F-POSS (**6**) and MMA via thermally-initiated AIBN-produced PMMA-co-F-POSS copolymers. Currently, molecular weights of these polymers are being obtained to determine the impact of F-POSS in the polymerization of these monomers. The wetting behavior of these polymers is being investigated as well.

### Conclusions

Functionalized F-POSS were synthesized in a simple one-stop reaction from a variety of dichlorosilanes. Dynamic contact angle measurements of these structures were taken to determine the influence of functionalized group on surface energy. Monomer functionalized F-POSS compounds were found to copolymerize readily. These novel structures can be used in the development of new superhydrophobic and oleophobic materials.

**Acknowledgements.** The authors would like to thank the Air Force Office of Scientific Research the Air Force Research Laboratory Propulsion Directorate for financial support.

#### References

- (1) Mabry, J. M.; Vij, A.; Iacono, S. T.; Viers, b. D. *Angew. Chem., Int. Ed.* **2008**, *47*, 4137.
- (2) Chhatre, S. S.; Guardado, J. O.; Moore, B. M.; Haddad, T. S.; Mabry, J. M.; McKinley, G. H.; Cohen, R. E. *ACS Appl. Mater. Interfaces* **2010**, *2*, 3544.
- (3) Tuteja, A.; Choi, W.; Mabry, J. M.; McKinely, G. H.; Cohen, R. E. *Proc. Natl. Acad. Sci. U. S. A.* **2008**, *105*, 18200.
- (4) Tuteja, A.; Choi, W.; Ma, M.; Mabry, J. M.; Mazzella, S. A.; Rutledge, G. C.; McKinley, G. H.; Cohen, R. E. *Science* **2007**, *318*, 1618.