

**“Synthesis, Processing, and Properties of  
TaC-TaB<sub>2</sub>-C Ceramics”**

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NSWCCD**

**Sub-Micron and Nanostructured Ceramics  
Colorado Springs, CO  
June 9, 2009**

# Report Documentation Page

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- SEAP student Esther Showalter for participation in the experiments



## Background

- **TaC and TaB<sub>2</sub> with melting temperatures of 3980°C and 3037°C, respectively, are candidates for ultra-high temperature applications, such as propulsion systems and hypersonic vehicles. There are very limited data on the preparation and properties of ceramics containing both TaC and TaB<sub>2</sub>.**
- **TaB<sub>2</sub>-TaC system has a eutectic at 2730°C at 34 wt. % TaC and no component solubility below 2100°C.**
- **Intermediate composition ceramics in the TaC-TaB<sub>2</sub> system had higher hardness and lower wear at 800 - 900 °C than the end members.**
- **Densification of TaC requires temperatures up to 2400°C and TaB<sub>2</sub> – 2100-2200 °C.**
- **Densification temperature of TaC was decreased to 2100-2200°C by the addition of 10 wt. % fine TaB<sub>2</sub> and B<sub>4</sub>C/C mixture (0.43/0.13 wt. %).**
- **The reactions of transition metals, carbides, and oxides with B<sub>4</sub>C are well known and used for synthesis of diboride-based ceramics.**
- **Pressureless sintering of fine B<sub>4</sub>C to relative density above 95% at 2100-2250°C was accomplished by the addition of Group IV-VI transition metal carbides.**
- **Diborides and active carbon formed during processing accelerated diffusion in the bulk and along grain boundaries. MeB<sub>2</sub> had a grain-growth-inhibiting effect.**



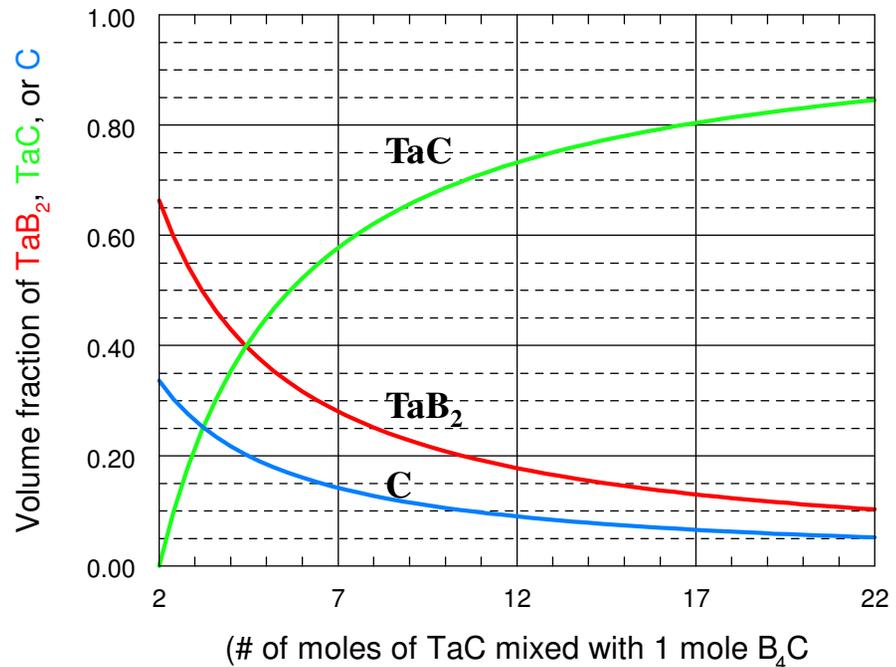
## Objective:

- Synthesize ceramic materials in the system  $\text{TaC} - \text{TaB}_2 - \text{C}$  using displacement reactions between sub-micron  $\text{TaC}$  and  $\text{B}_4\text{C}$
- Develop processing procedure for materials densification
- Characterize the microstructure and properties of ceramics

## Reactions in TaC+B<sub>4</sub>C Mixtures and Composition of the Reaction Products (in volume fractions)



Volume Fraction of Reaction Products from Heating  
x moles TaC + B<sub>4</sub>C Mixtures



## Composition of Starting and Reaction Product Components

TaC/B <sub>4</sub> C Ratio	Wt% B <sub>4</sub> C in raw materials	Calculated Volume% of Reaction Products			Calculated Wt% of Reaction Products		
		TaC	TaB <sub>2</sub>	C	TaC	TaB <sub>2</sub>	C
2	12.5	0	67	33	0	91.8	8.2
4	6.65	35	42.5	22.5	46.6	49.0	4.4
8	3.45	62	25	13	72.4	25.3	2.2
12	2.3	73.8	17.6	8.6	81.4	17.1	1.5
16	1.75	79.9	13.4	6.7	85.9	12.9	1.15

Actual carbon content in ceramics was lower. It was consumed by oxygen (3.02 wt.%) present in starting TaC during firing.



## Starting Materials

TaC - 99.9% pure, 100 nm, **made in 1976** by Atomergic Chemetals Corp., which **does not make it any more**.

Stoichiometry -  $\text{TaC}_{0.96}$  (Chemical and XRD analysis). Oxygen content – 3.02 wt. % (Laboratory Testing Inc., Hatfield, PA).

Surface area - 11.74  $\text{m}^2/\text{g}$  (NSWCCD by BET).

W was not detected by EDS or XPS. The presence of Cl and H-C bonds was determined by XPS.

**It can be assumed that the powder was prepared by gas phase deposition using Ta chloride/hydrocarbon mixtures.**

TaC - 99.5% pure; -325 mesh, CERAC Inc. Surface area - 1.02  $\text{m}^2/\text{g}$

B<sub>4</sub>C - 99.4% pure (metals basis), 1-7 Micron powder, Alfa Aesar  
Surface area - 4.84  $\text{m}^2/\text{g}$ .

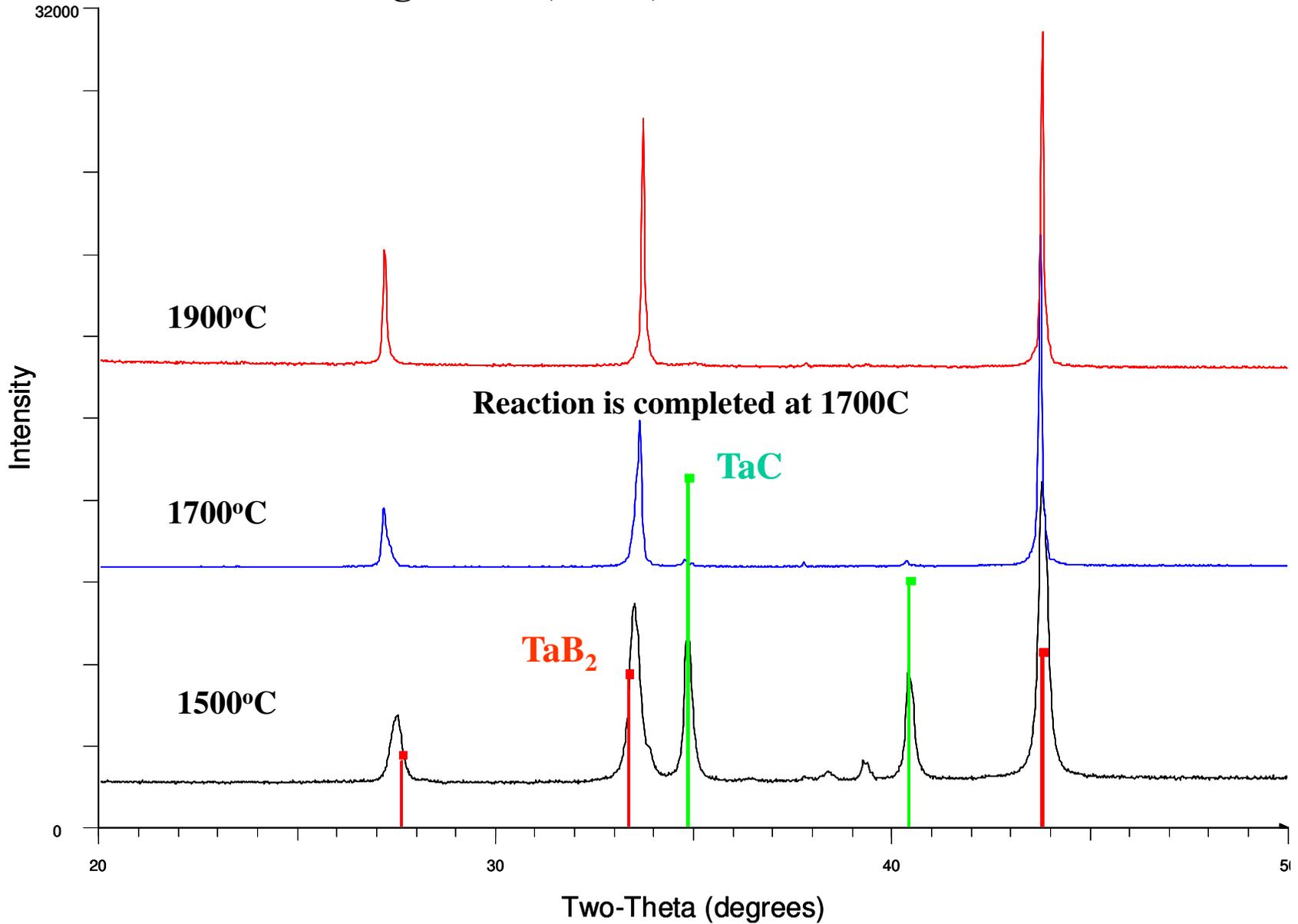
C black – surface area about 25  $\text{m}^2/\text{g}$



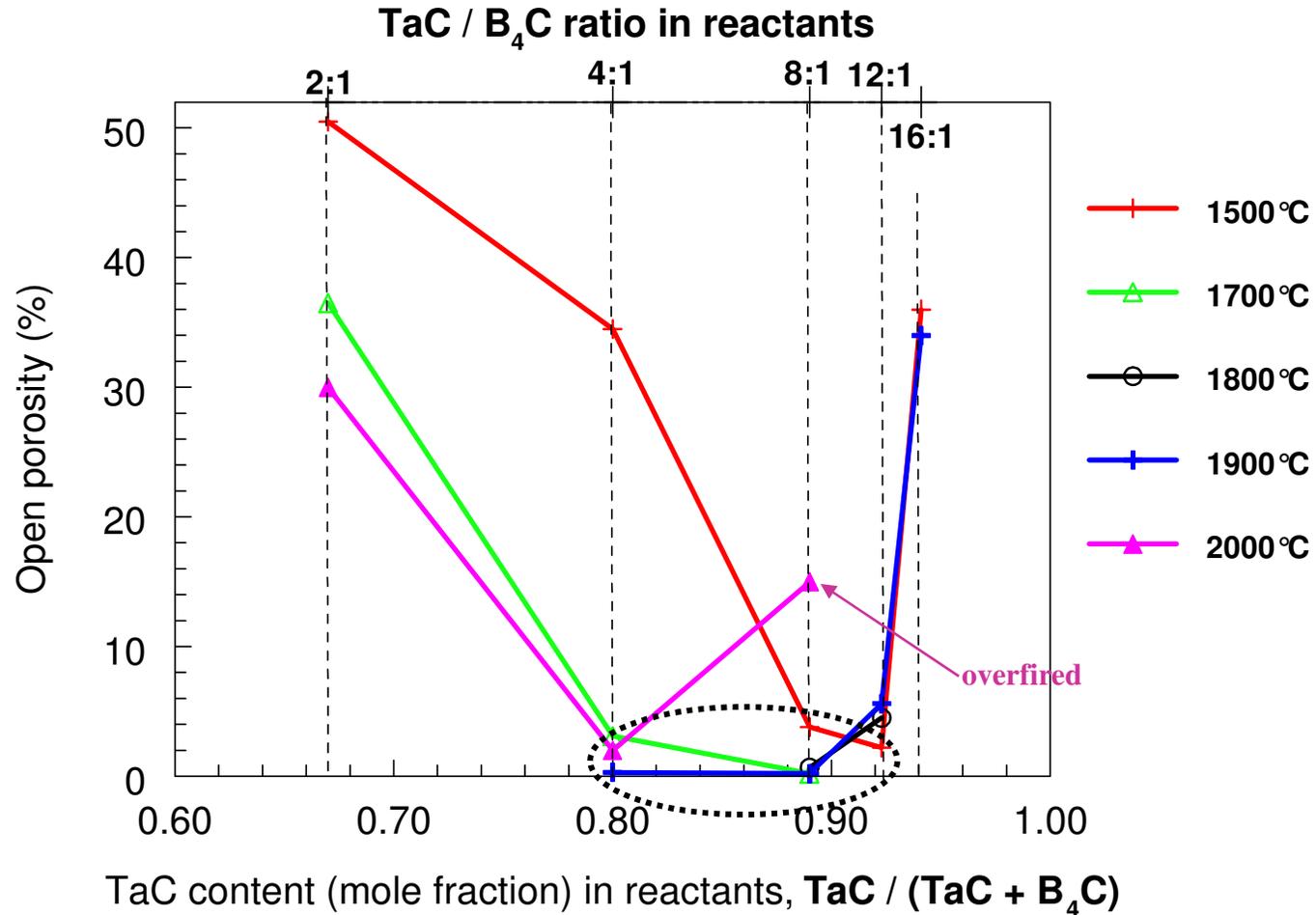
# Experimental Procedure

- Mixing the components in selected molar ratios.
- Sample preparation by cold pressing and following CIPing.
- Heating in the furnace with graphite heaters at 1500 –2100°C for 2 hours in Ar.
- Hot Pressing at 1700 and 1900°C and 20 MPa for 1hr in He.
- Characterization:
  - Phase composition by XRD
  - Microstructure by SEM
  - Vickers hardness (Load - 1kg and 10kg for 15 seconds)
  - Flexural strength (3-point)

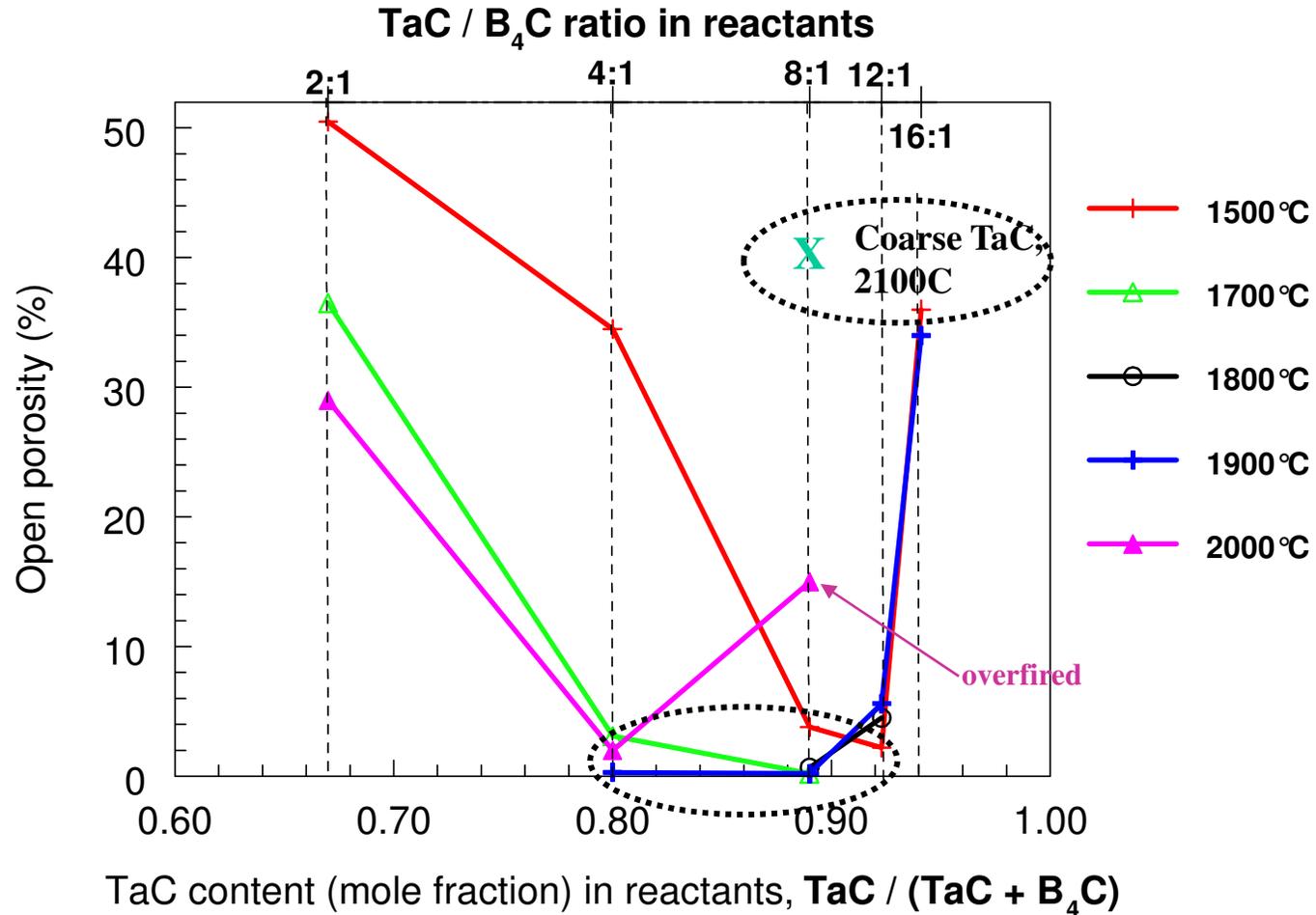
# XRD of Ceramics Prepared from $2\text{TaC} + \text{B}_4\text{C}$ Mixture by Pressureless Sintering at 1500, 1700, and 1900°C for 2 hours in He



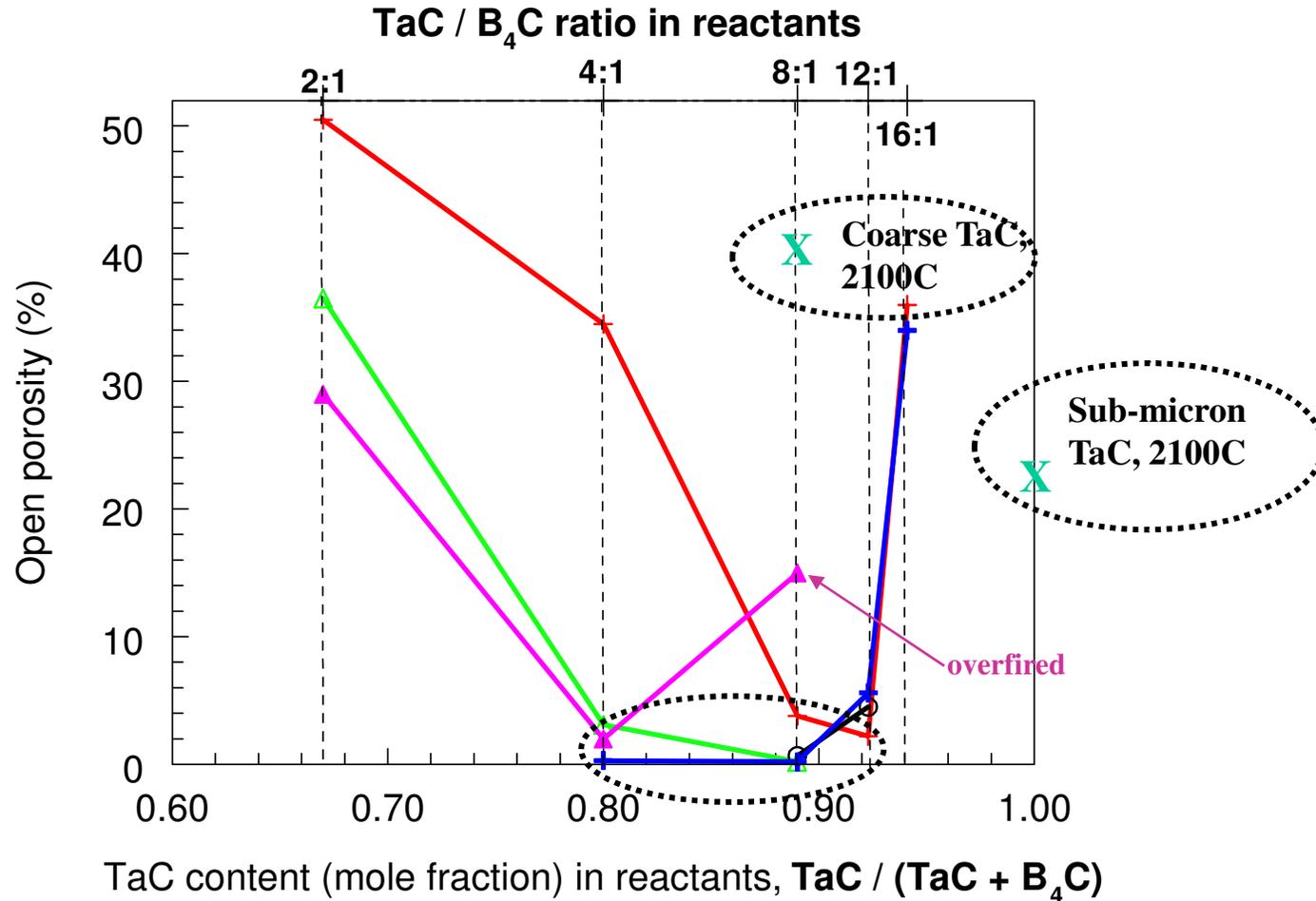
## Open Porosity of Ceramics Prepared from TaC/B<sub>4</sub>C Mixtures by Pressureless Sintering as a Function of Composition and Temperature



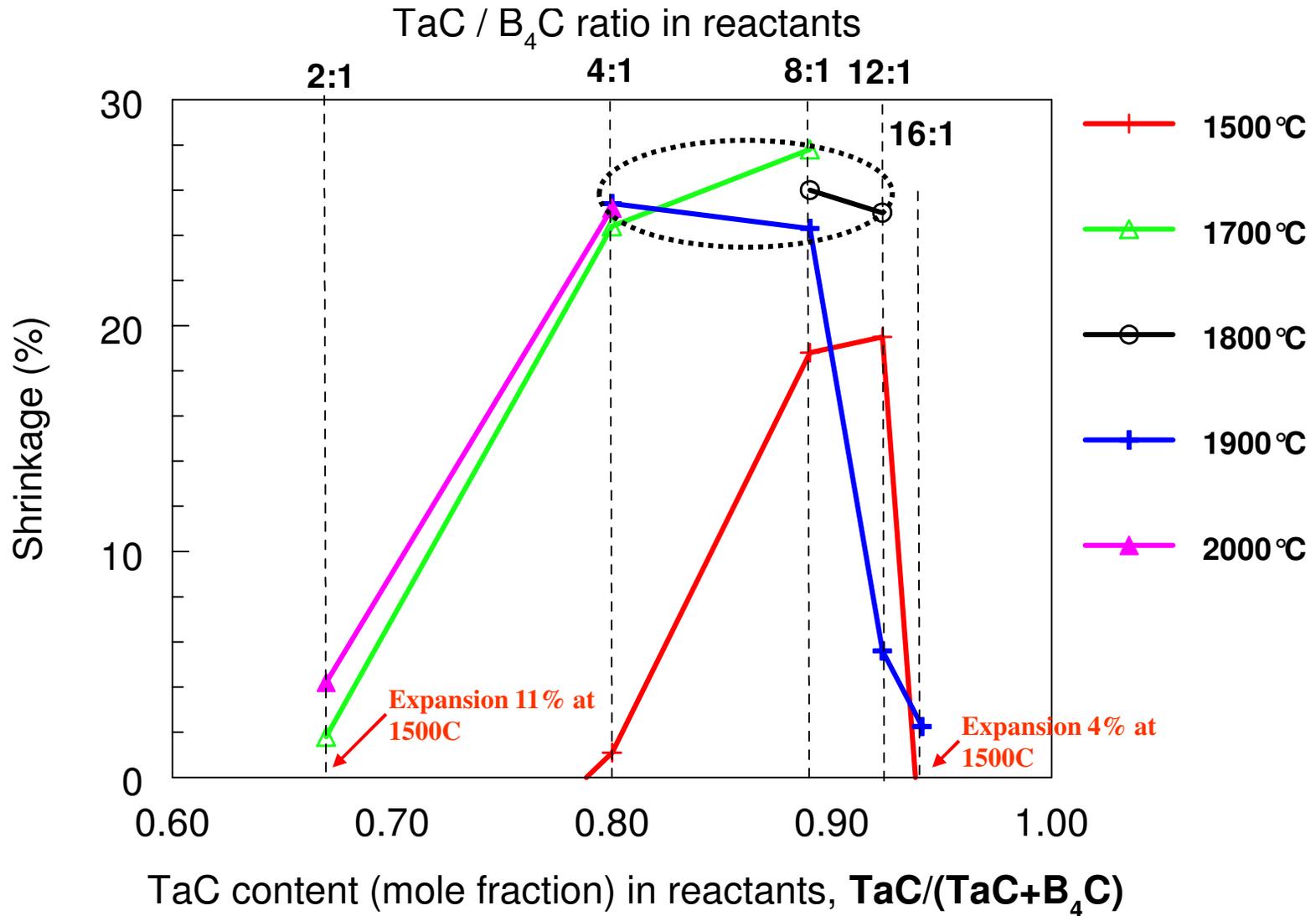
# Open Porosity of Ceramics Prepared from TaC/B<sub>4</sub>C Mixtures by Pressureless Sintering as a Function of Composition and Temperature



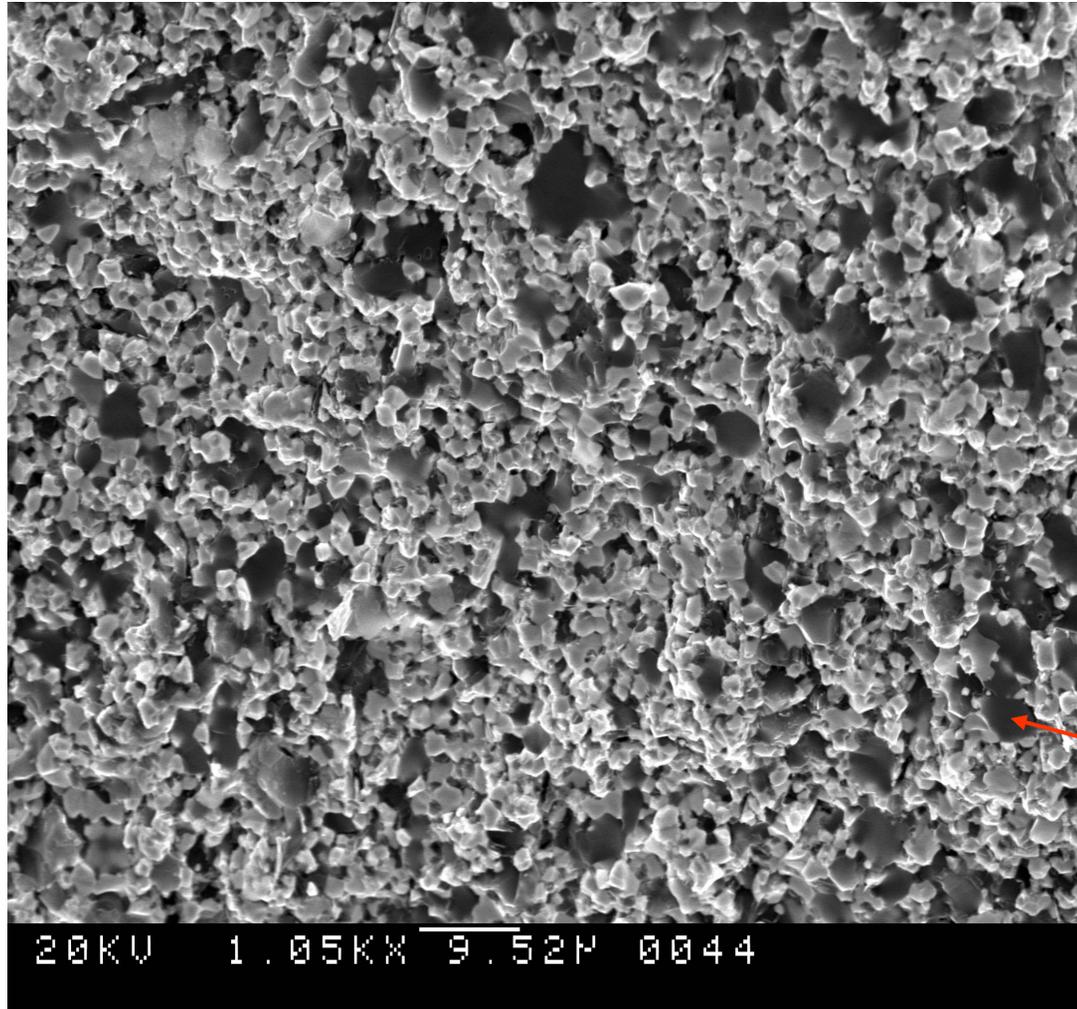
# Open Porosity of Ceramics Prepared from TaC/B<sub>4</sub>C Mixtures by Pressureless Sintering as a Function of Composition and Temperature



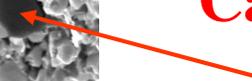
# Shrinkage of Ceramics Prepared from TaC/B<sub>4</sub>C Mixtures by Pressureless Sintering as a Function of Composition and Temperature



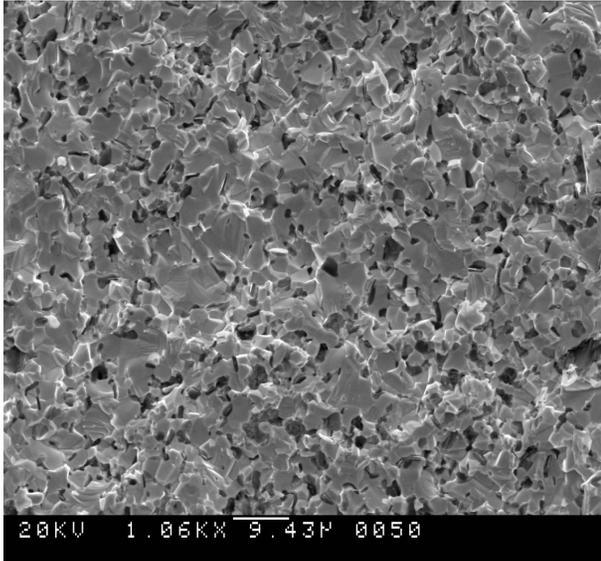
# SEM of $TaB_2 - C$ Ceramics Prepared from $TaC + B_4C$ Mixtures by Pressureless Reaction Sintering at $1900^\circ C$ for 2 hours



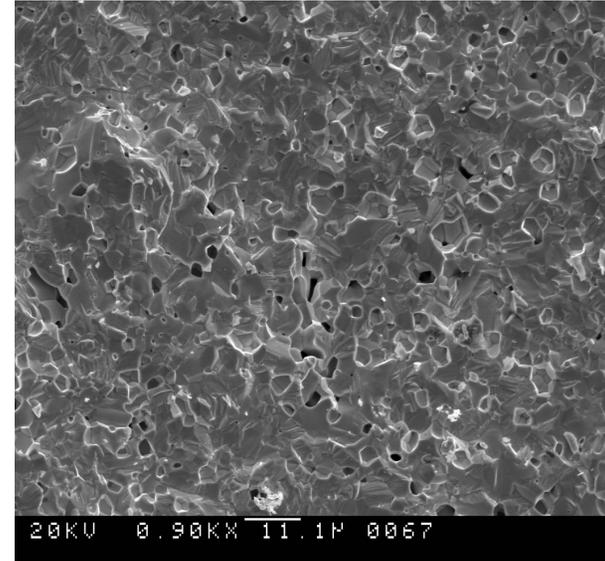
**Carbon**



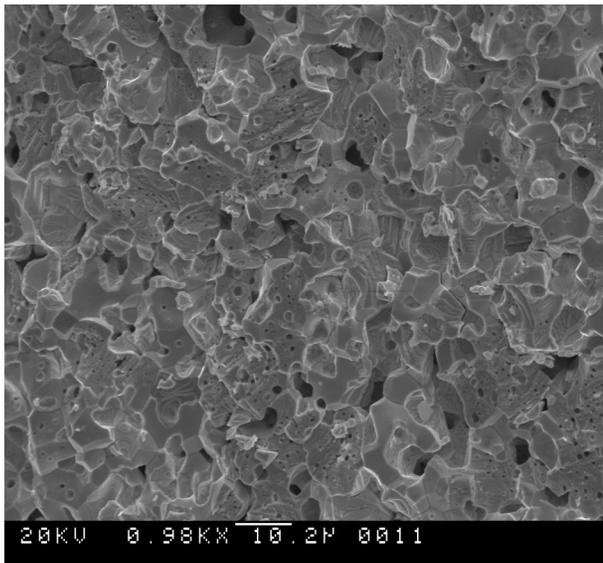
## SEM of TaC-TaB<sub>2</sub> - C Ceramics Prepared from TaC+B<sub>4</sub>C Mixtures by Pressureless Sintering at 1900°C for 2 hours



**4TaC**



**8TaC**



**12 TaC**

### **Calculated Composition in vol. %**

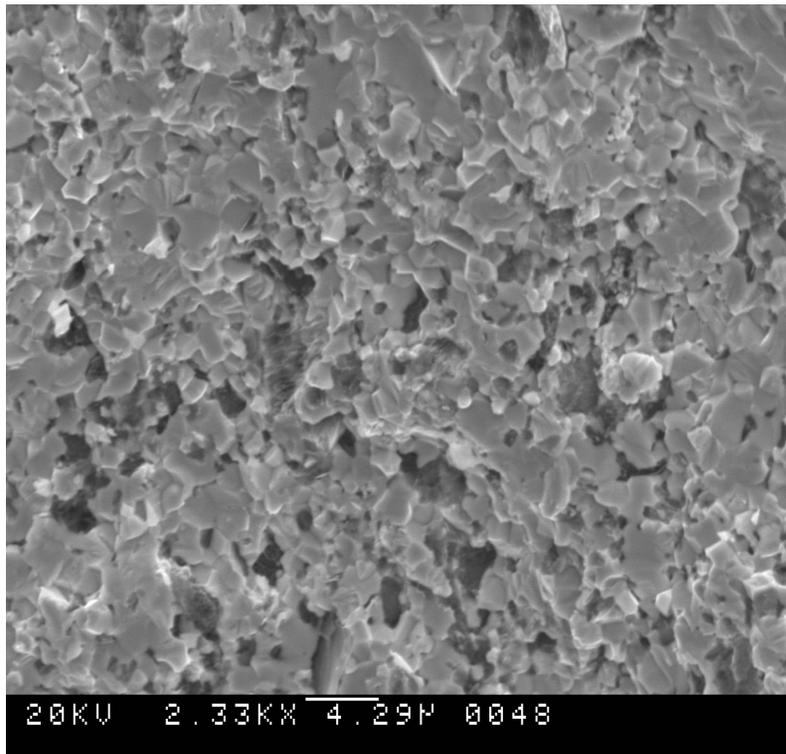
**4:1, TaC – 35, TaB<sub>2</sub> – 42.5, C – 22.5**

**8:1, TaC – 62, TaB<sub>2</sub> – 25, C – 13**

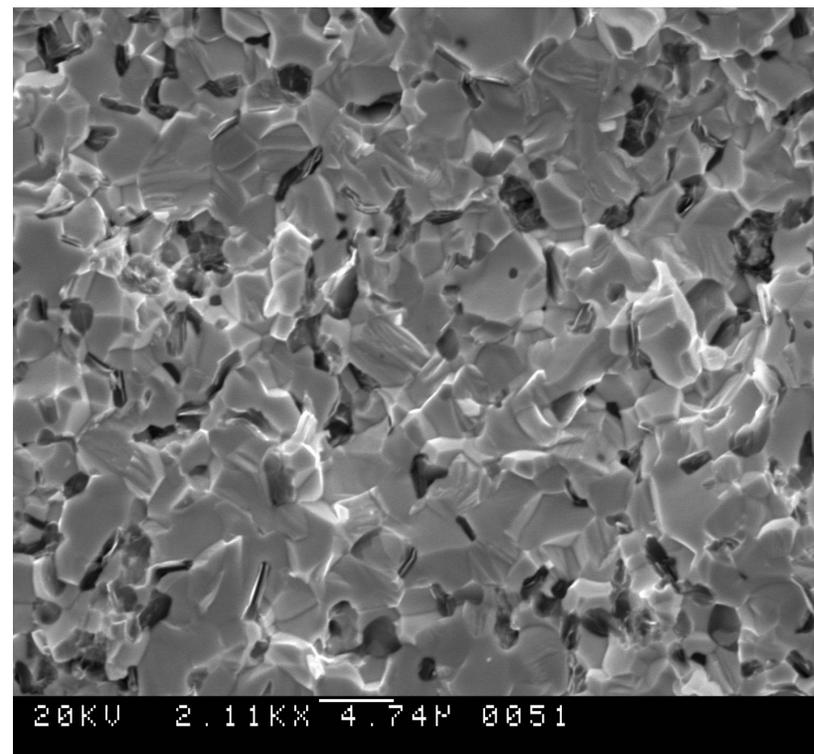
**12:1, TaC – 73.8, TaB<sub>2</sub> – 17.6, C – 8.6**

**SEM of TaC-TaB<sub>2</sub> - C Ceramics Prepared from 4TaC+B<sub>4</sub>C Mixture by Pressureless Sintering at 1700 and 1900°C for 2 hours**

Calculated composition in vol. %: TaC – 35, TaB<sub>2</sub> – 42.5, C – 22.5



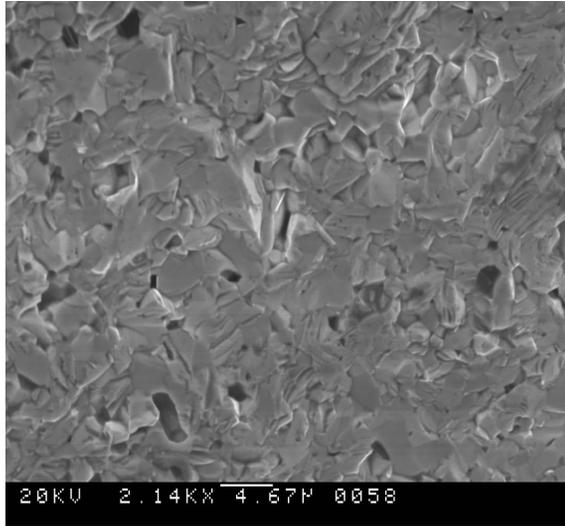
**1700°C (Open porosity 3.1%)**



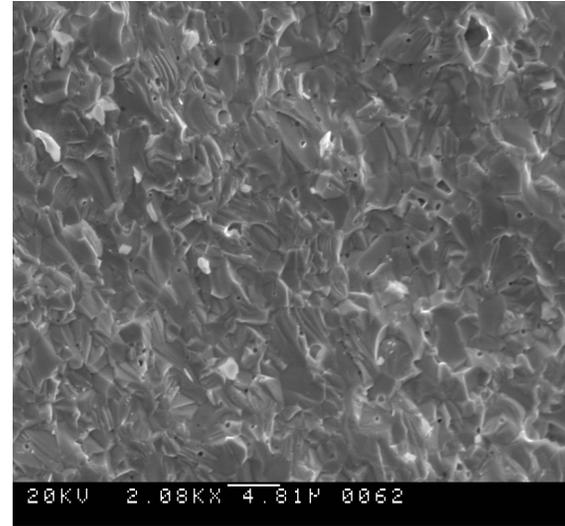
**1900°C (Open porosity 0%)**

# SEM of TaC-TaB<sub>2</sub> - C Ceramics Prepared from 8TaC+B<sub>4</sub>C Mixture by Pressureless Reaction Sintering

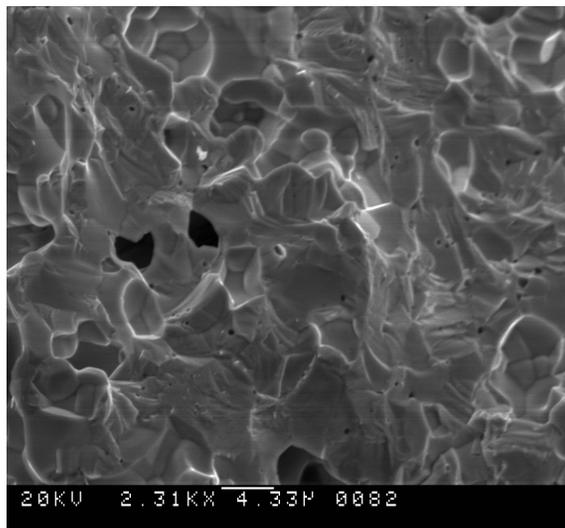
(TaC – 62, TaB<sub>2</sub> – 25, C – 13 vol. %)



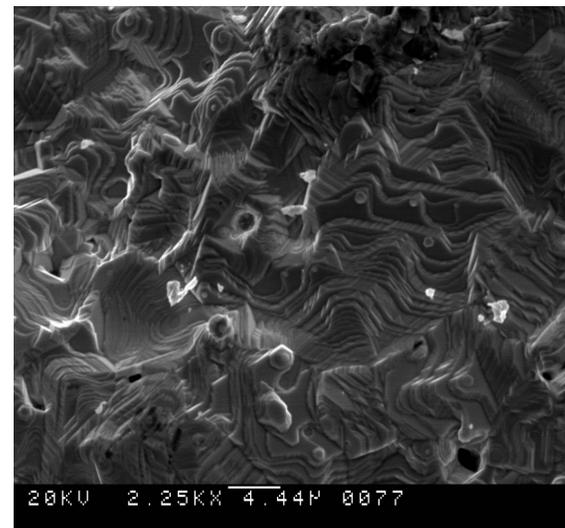
1700°C



1800°C

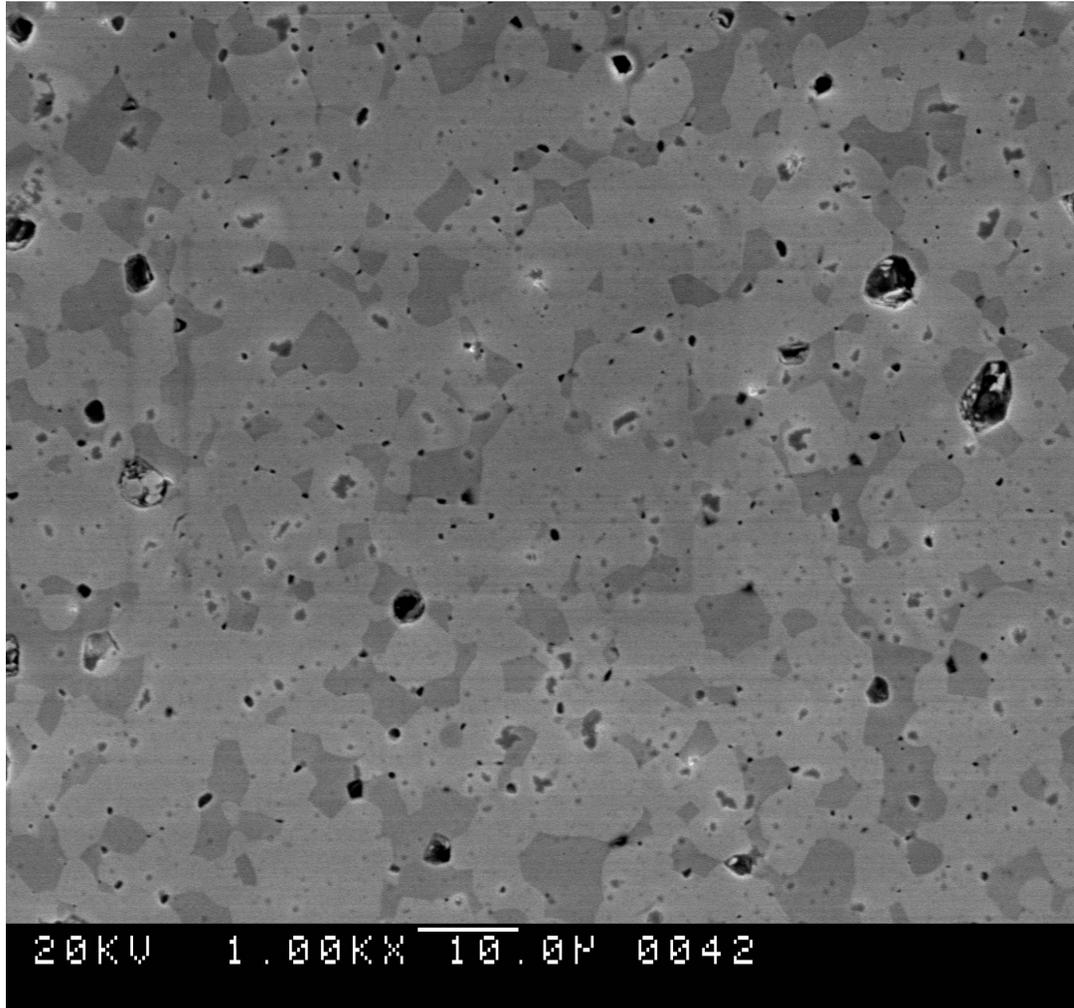


1900°C



2000°C,  
overfired

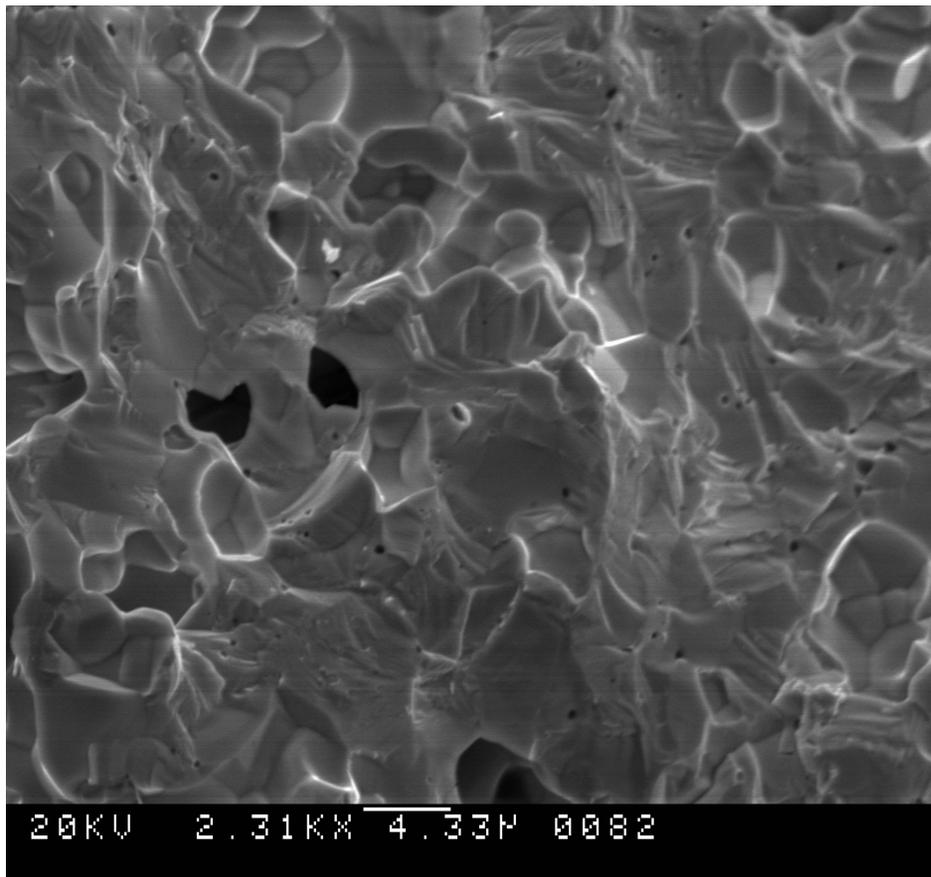
**SEM of TaC-TaB<sub>2</sub> - C Ceramics Prepared from 8TaC+B<sub>4</sub>C Mixture  
by Pressureless Reaction Sintering at 1900°C for 2 hours in He**



**Polished Surface**

**Composition in vol. %: TaC – 62, TaB<sub>2</sub> – 25, C – 13**

# Mechanical Properties of **8TaC/B<sub>4</sub>C** (TaC-62, TaB<sub>2</sub>-25, C – 13 vol. %) Ceramics Pressureless sintered at 1900°C



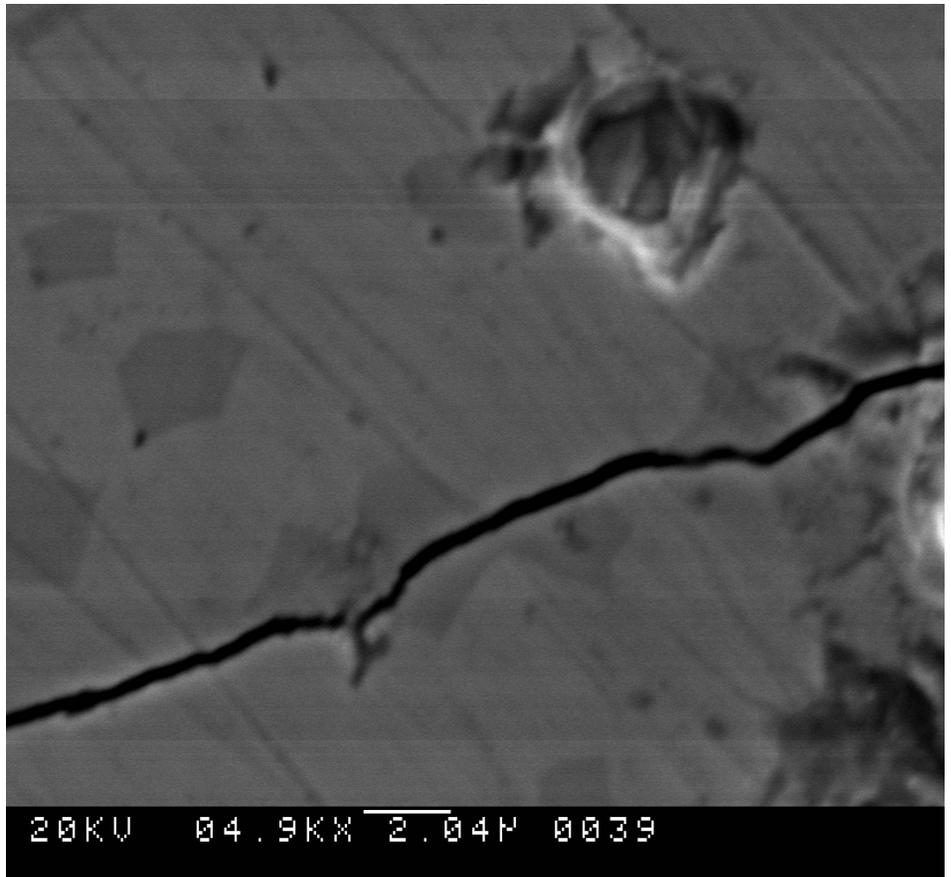
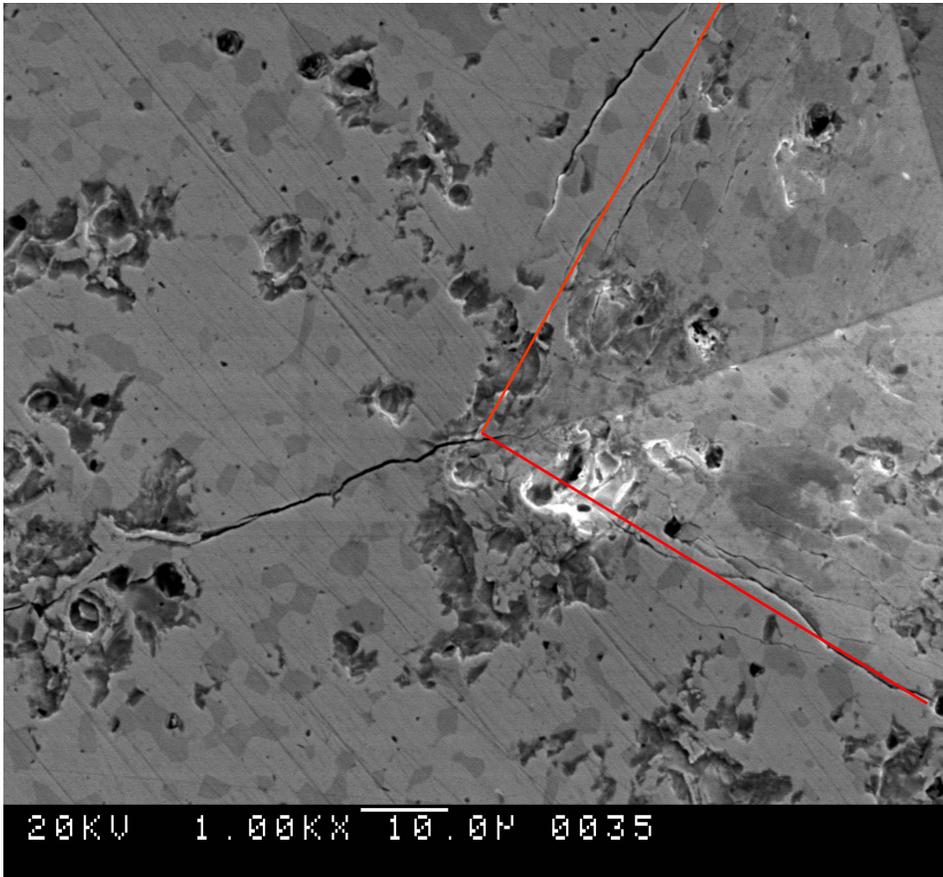
**Flexural Strength (3-point)**

**221 MPa**

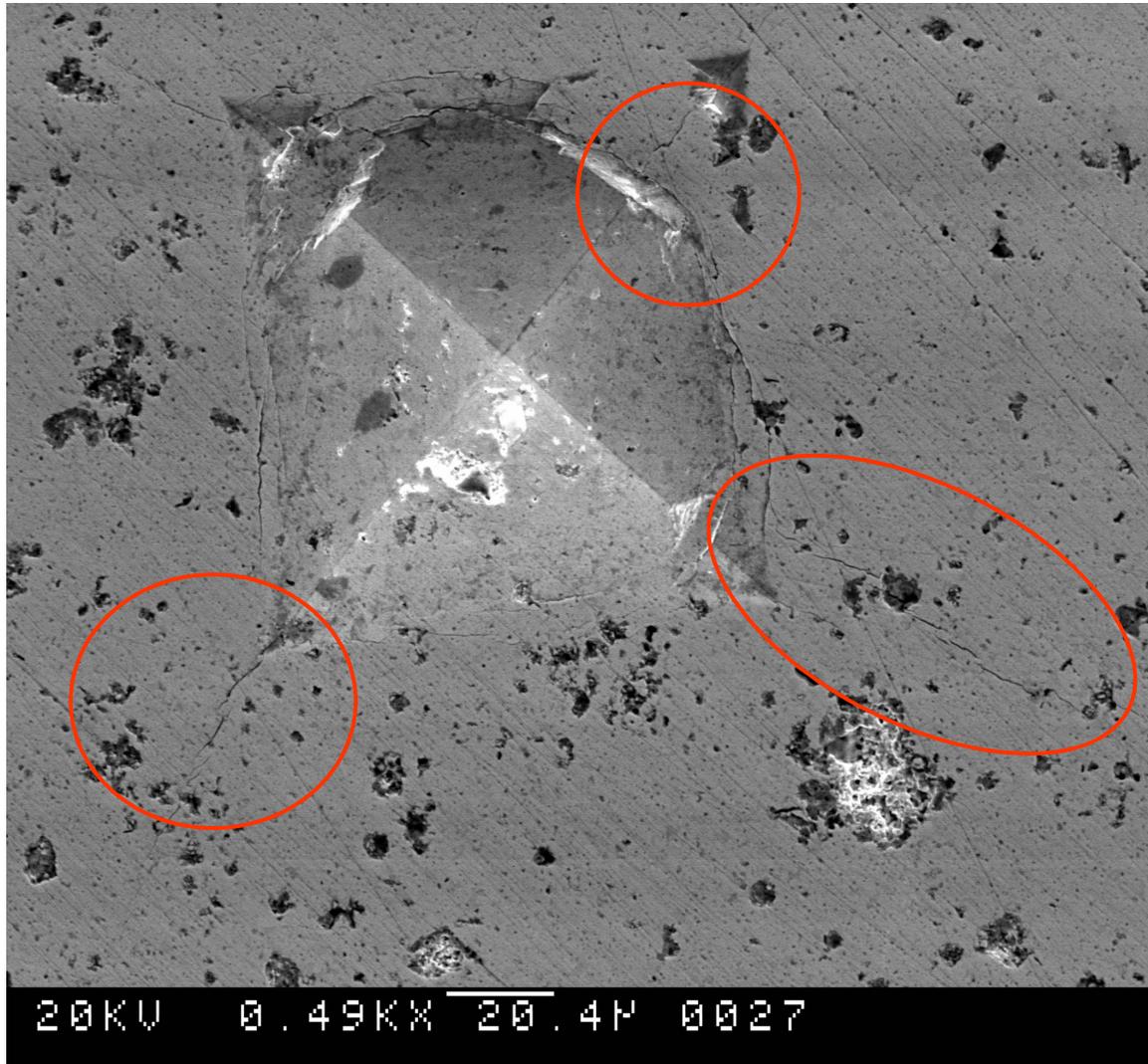
**Vickers Hardness**

**22.2 GPa**

# SEM of an Indent (10 kg load) in Ceramics from **8TaC+B<sub>4</sub>C** Mixture Pressureless Sintered at 1900 C



**SEM of an Indent (10 kg load) in Ceramics from  $12\text{TaC}+\text{B}_4\text{C}$  Mixture  
(TaC – 73.8, TaB<sub>2</sub> – 17.6, C – 8.6 vol. %) Hot Pressed at 1700 C**



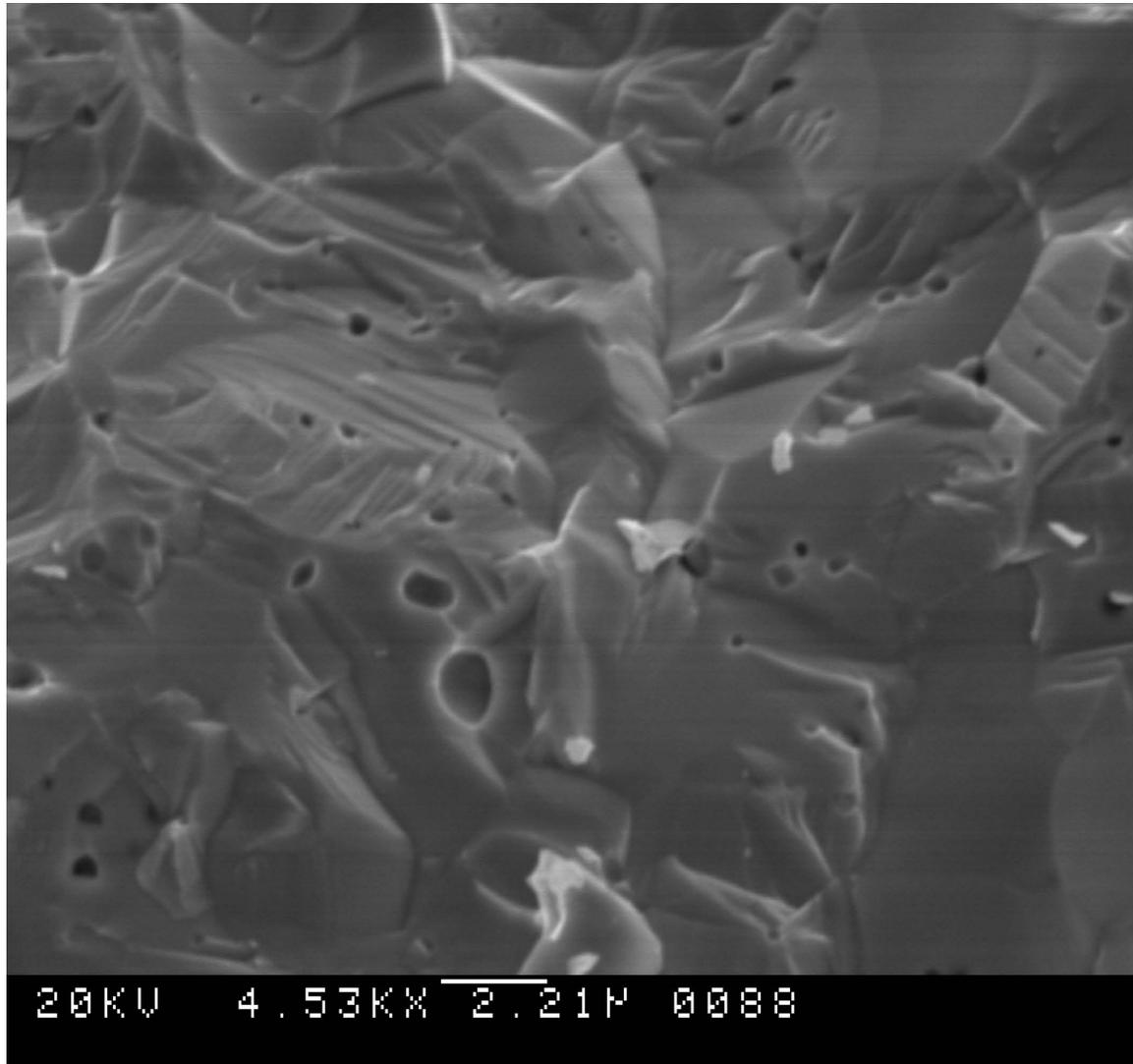
## Use of nano-TaC and nano-C black to Densify the End Members of the Reaction:



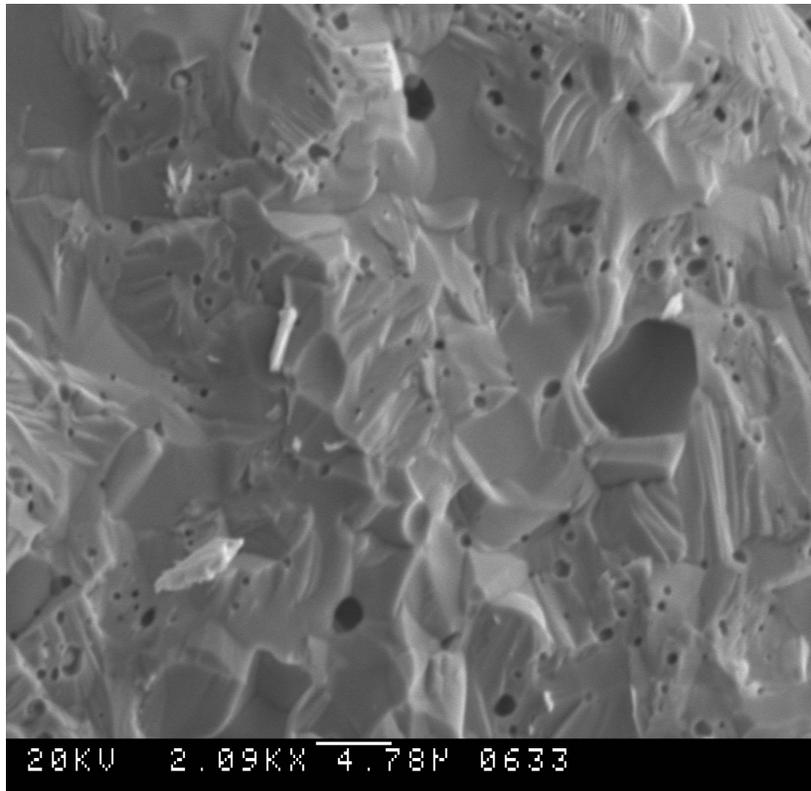
- The end-member mixture was pressureless sintered to full density at 1850°C (2 hours).
- The mixture of **nano TaC and TaB<sub>2</sub> (without carbon black)** had open porosity of 19% after firing at 1800°C for 2 hours.
- The mixture of **6TaC, 2TaB<sub>2</sub> and 3C** containing **coarse TaC (-325 mesh) and C black** had open porosity of 32.5% after pressureless firing at 2100°C

Both sub-micron TaC and carbon black are needed for pressureless densification of TaC/TaB<sub>2</sub>/C ceramics. **C black should be in the range of 2 to 8 wt. % of the TaC/Cmixture.**

**SEM of TaC-TaB<sub>2</sub> - C Ceramics Prepared from  $6\text{TaC} + 2\text{TaB}_2 + 3\text{C}$   
Mixture Pressurless Sintered at 1850°C**

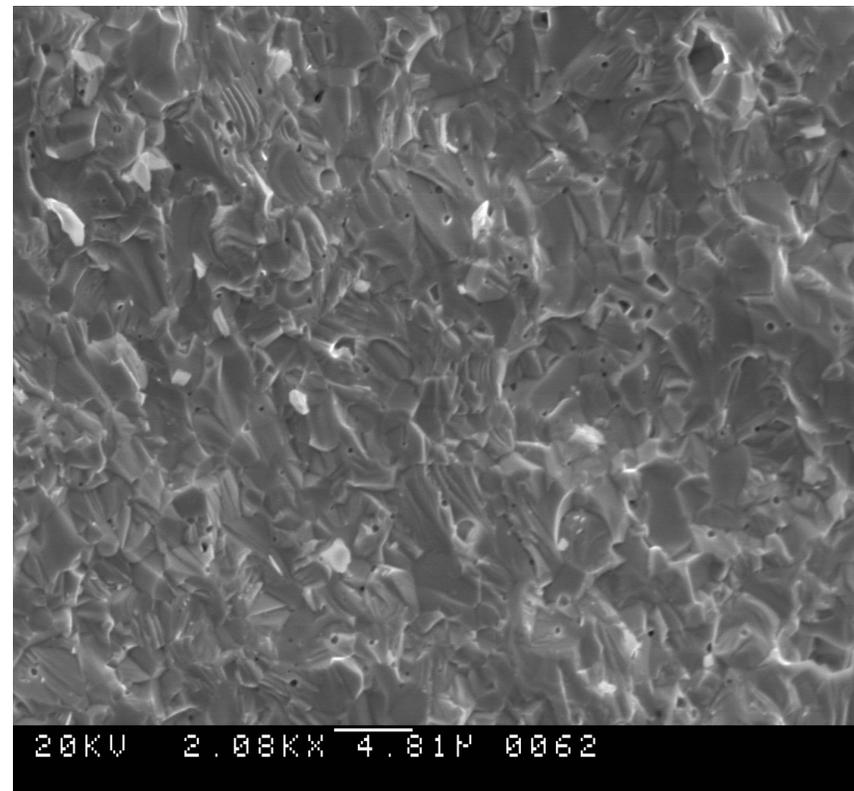


**SEM of TaC-TaB<sub>2</sub> - C Ceramics Prepared from 8TaC+B<sub>4</sub>C (reaction)  
and from 6TaC + 2TaB<sub>2</sub>+ 3C (end members) Mixtures by Pressurless  
Sintering**



**6TaC + 2TaB<sub>2</sub> + 3C, 1850°C**

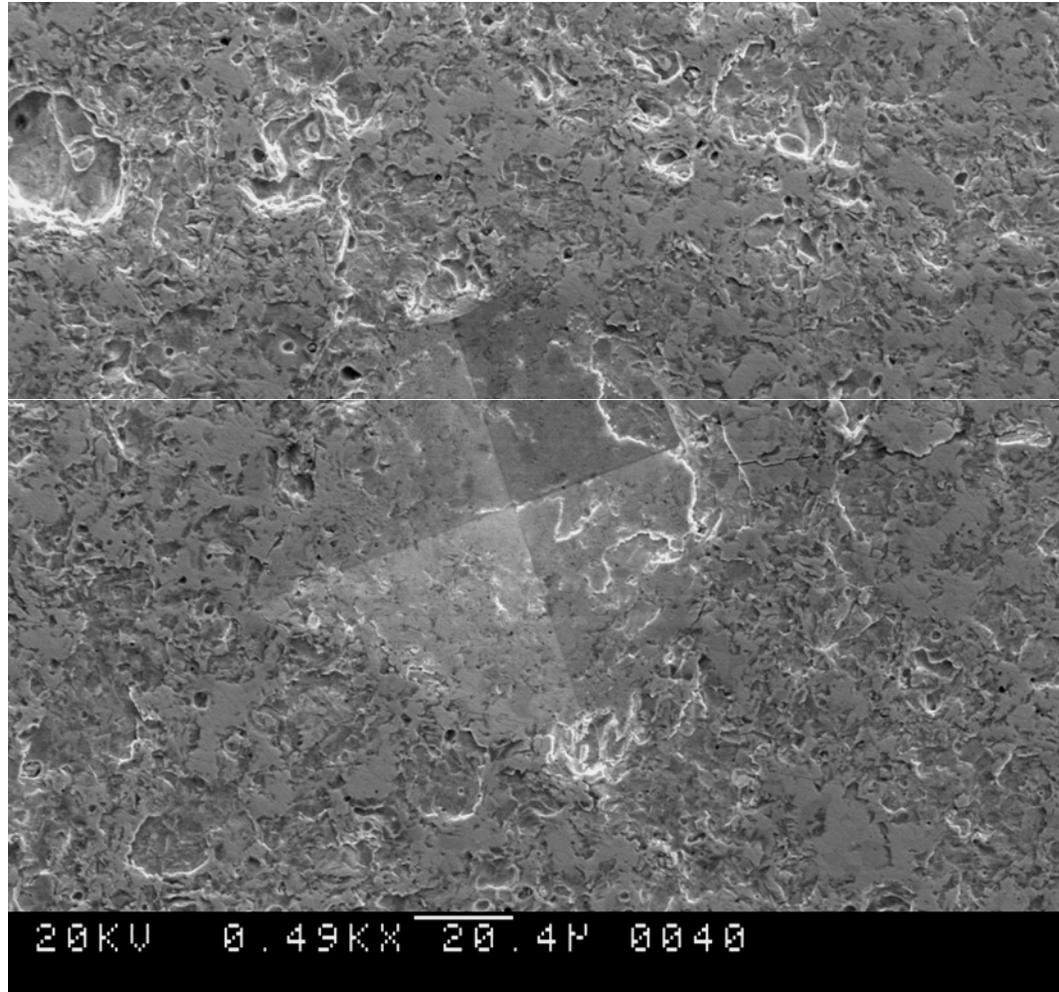
**Flexural Strength – 391 MPa**



**8TaC+B<sub>4</sub>C, 1800°C**

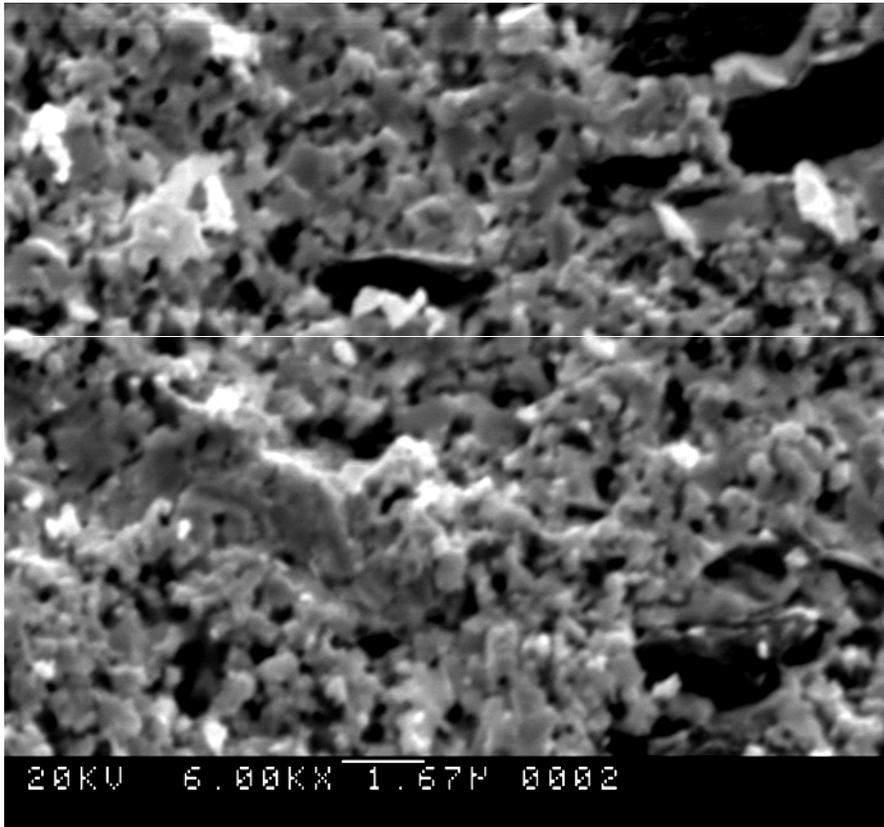
**Flexural Strength – 221 MPa**

**SEM of an Indent (10 kg load) in Ceramics from  $6\text{TaC} + 2\text{TaB}_2 + 3\text{C}$   
(end members) Mixture Pressurless Sintered at  $1850^\circ\text{C}$**

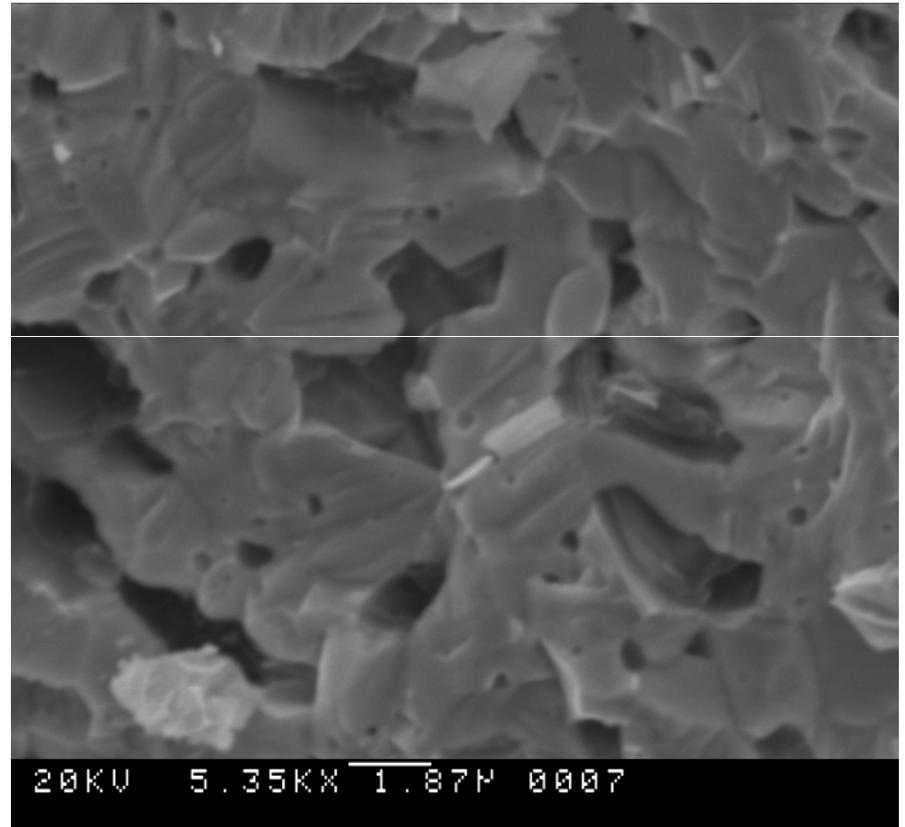


# **Hot Pressing Experiments**

**SEM of TaC-TaB<sub>2</sub> - C Ceramics Prepared from 8TaC+B<sub>4</sub>C Mixture by Reaction Hot Pressing at 1700 and 1900°C for 1 hours in Ar**



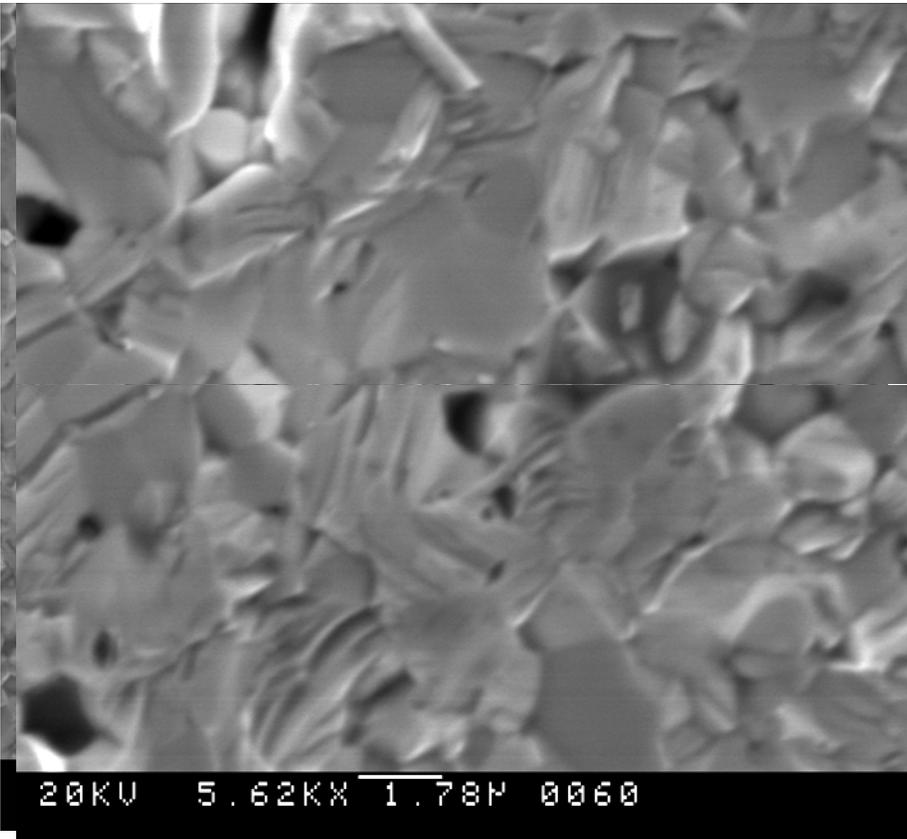
**1700°C**



**1900°C**

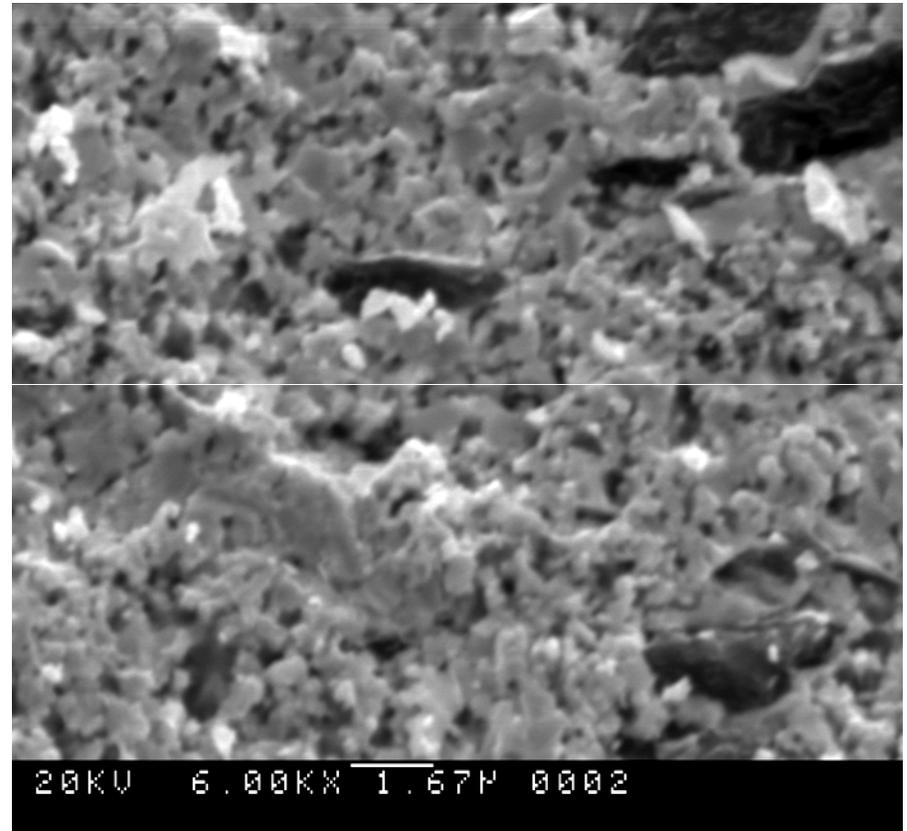
Open porosity about 12%

**SEM of TaC-TaB<sub>2</sub> - C Ceramics Prepared from 8TaC+B<sub>4</sub>C Mixture  
by Reaction Pressureless Sintering and Hot Pressing at 1700 C**



**Pressureless Sintering**

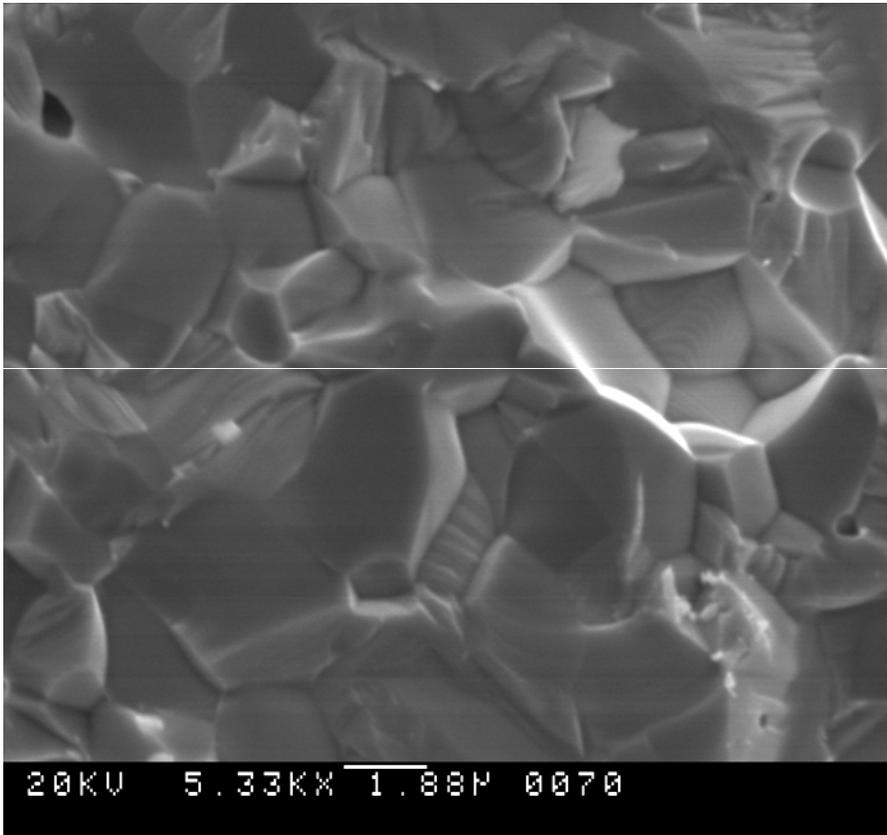
Open porosity 0.2%



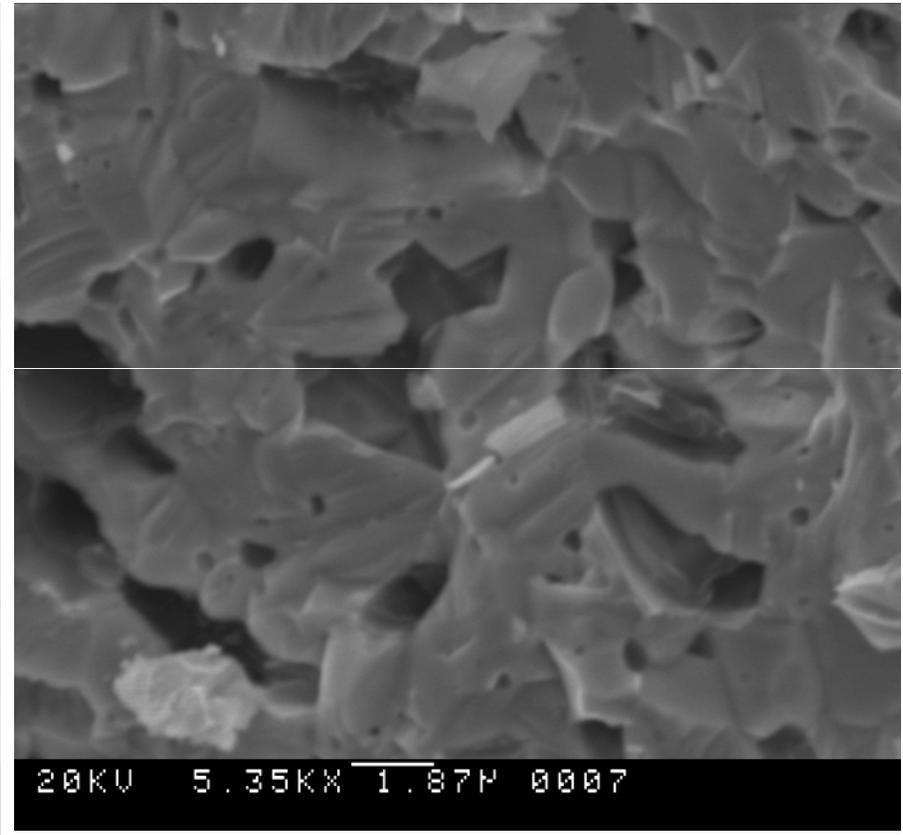
**Hot Pressing**

Open porosity 12%

# SEM of TaC-TaB<sub>2</sub> - C Ceramics Prepared from **8TaC+B<sub>4</sub>C** Mixture by Reaction Pressureless Sintering and Hot Pressing at 1900° C

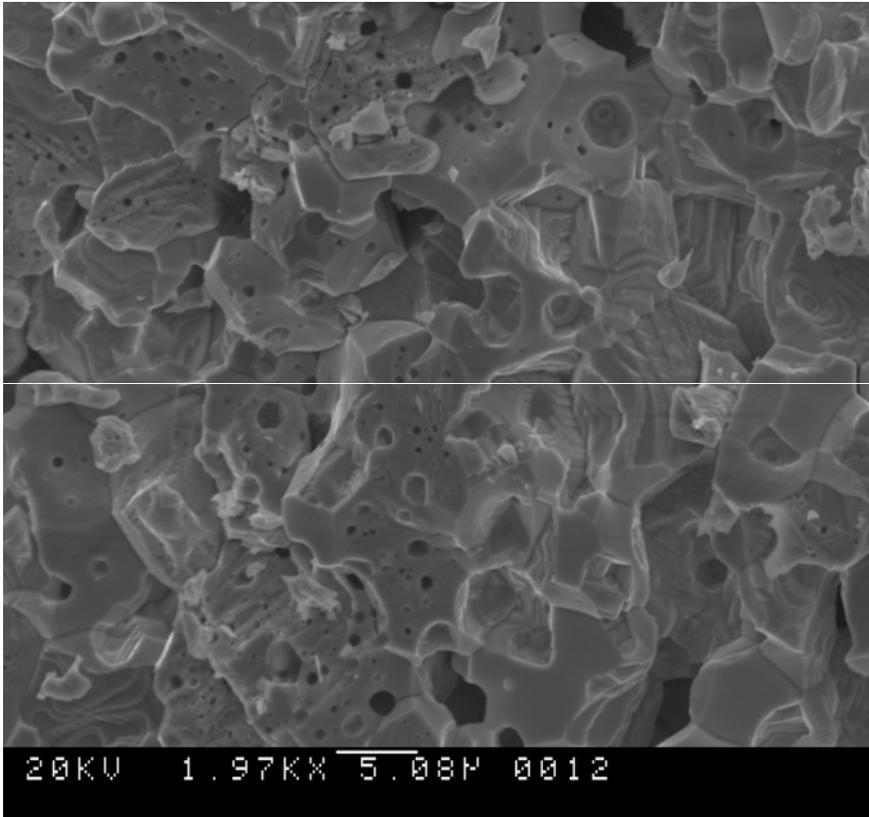


**Pressureless Sintering**  
Fully dense sample



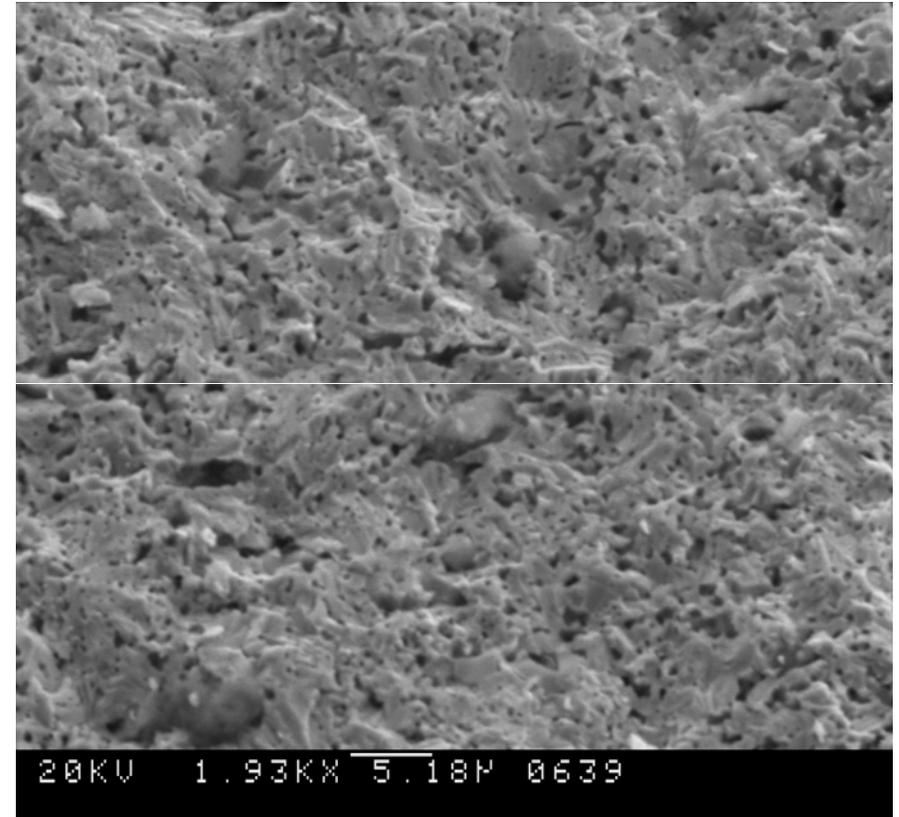
**Hot Pressing**  
Open porosity 12.5%

## SEM of TaC-TaB<sub>2</sub> - C Ceramics Prepared from 12TaC+B<sub>4</sub>C Mixture by Reaction Sintering and Hot Pressing



**Pressureless at 1800°C for 2 hours**

Open porosity 4.5%

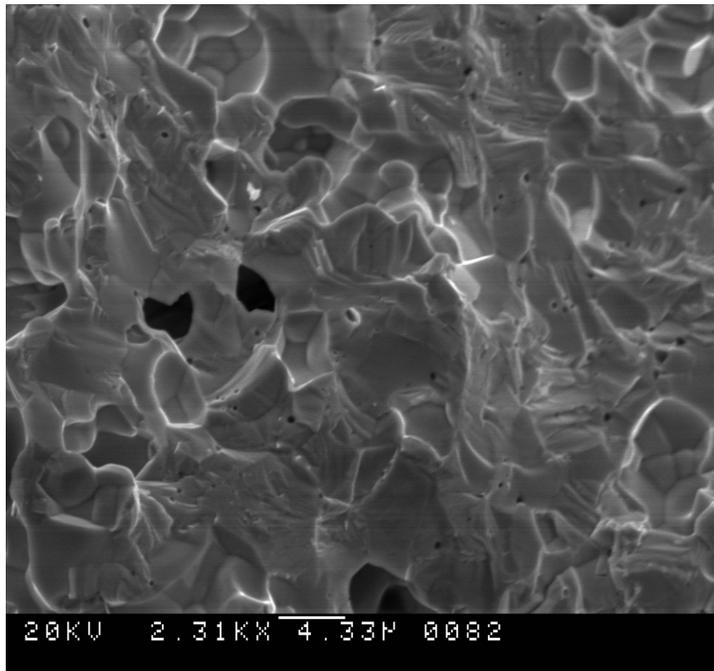


**HP at 1700°C for 1 hour**

Open porosity 12.5%

# SEM, Flexural Strength (3-point), and Vickers Hardness of Ceramics: **8TaC/B<sub>4</sub>C** (TaC – 62, TaB<sub>2</sub> – 25, C – 13 vol. %)

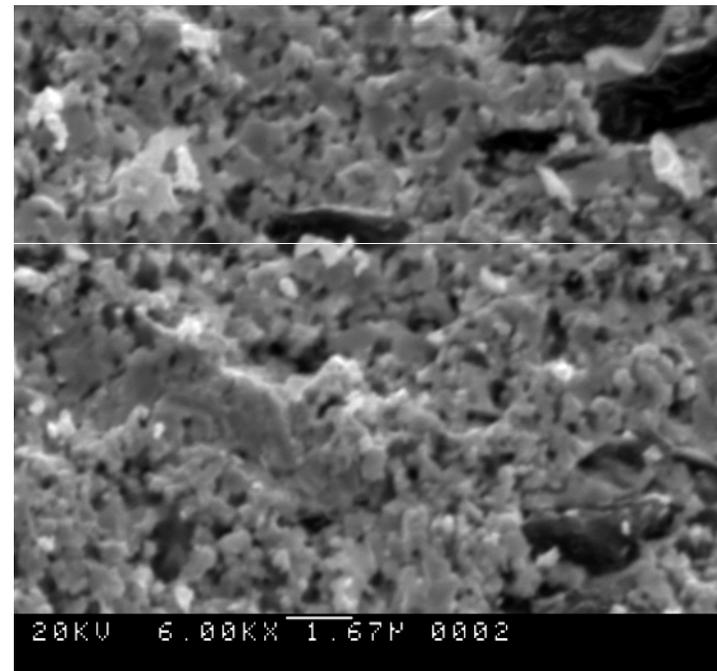
Pressureless sintered 1900°C



**Strength-221 MPa**

**Hardness-22.2 GPa**

Hot pressed 1700°C and 20MPa



**Strength-293MPa**

**Hardness-8.2 GPa**

## Summary

- Dense ceramics containing (in v.%) 35-73.8 TaC, 25-62 TaB<sub>2</sub>, and 8.6-22.5 C were prepared by reactive pressureless sintering of sub-micron TaC with B<sub>4</sub>C at 1700-1900°C.
- Reaction between TaC and B<sub>4</sub>C was completed at about 1700°C.
- The densification of multi-phase ceramics was promoted by the formation of TaB<sub>2</sub> and active carbon. Carbon eliminated oxygen from the grain boundaries additionally enhancing sintering.
- Use of sub-micron TaC and carbon black together led to pressureless densification of non-reactive mixture of **6TaC, 2TaB<sub>2</sub>, 3C** at 1850°C.
- The materials of all the tested compositions were not densified if a coarse TaC powder was used even in the presence of carbon black.
- Additional work is necessary to optimize processing parameters to remove oxygen at low temperatures, decrease shrinkage and stresses to increase the strength of ceramics.
- Low-temperature hot pressing is very promising for the development of high-strength multi-phase ceramics with a sub-micron grain size.
- Based on the results of this work and considering high potential of TaC for UHT applications, it is important to develop feasible low-cost method to produce sub-micron TaC and other carbides.