



On the Influence of Moisture on Dielectric Properties of Polyetheretherketone (PEEK) Carbon-Fiber Composites

by Bruce K. Fink, Roy L. McCullough,
and John W. Gillespie Jr.

ARL-TR-2236

June 2000

Approved for public release; distribution is unlimited.

20000627 116

The findings in this report are not to be construed as an official Department of the Army position unless so designated by other authorized documents.

Citation of manufacturer's or trade names does not constitute an official endorsement or approval of the use thereof.

Destroy this report when it is no longer needed. Do not return it to the originator.

Abstract

An analysis of local and global mechanisms of heat generation and distribution in carbon-fiber-based composites subjected to an alternating magnetic field has shown that heating is dependent upon the dielectric properties of the polymer matrix. These properties were investigated as functions of temperature, frequency, and moisture content. The results indicate little dependence of the dielectric constant on temperature or frequency, while the loss tangent exhibited a substantial dependence on both frequency and temperature. A substantial dependence of loss tangent on moisture content in polyetheretherketone (PEEK) was found.

Table of Contents

	<u>Page</u>
List of Figures.....	v
List of Tables.....	vii
1. Introduction	1
2. Review of Dielectric Properties.....	3
3. Characterization	9
4. Summary	16
5. References.....	19
Distribution List.....	21
Report Documentation Page.....	41

INTENTIONALLY LEFT BLANK.

List of Figures

<u>Figure</u>		<u>Page</u>
1.	Results of Dried-Laminate Induction-Heating Test.....	2
2.	General Relationship Between Real and Imaginary Parts of Complex Dielectric Constant in Dielectric Materials.....	6
3.	Room Temperature Dielectric Constant Results for PEEK	10
4.	Room Temperature Dielectric Loss Tangent Results for PEEK.....	11
5.	Room Temperature Results for Resistivity of PEEK.....	11
6.	Percent Weight Loss of PEEK Specimens vs. Drying Time.....	13
7.	Average Experimental Loss Tangent Data for Undried PEEK Specimens	14
8.	Average Experimental Loss Tangent Data for Three PEEK Specimens Dried at 120 °C for 24 hr.....	14
9.	Average Experimental Loss Tangent Data for Three PEEK Specimens Dried at 120 °C for 48 hr.....	15
10.	Average Experimental Loss Tangent Data for Three PEEK Specimens Dried at 120 °C for 72 hr.....	15
11.	Experimental and ICI-Reported Loss Tangent Data for PEEK at 100 kHz.....	16

INTENTIONALLY LEFT BLANK.

List of Tables

<u>Table</u>	<u>Page</u>
1. Averaged Data From Three Electrical Property Measurements on PEEK at Room Temperature	10
2. Weight Change Data for PEEK's Dielectric Loss Tangent vs. Moisture Content Testing.....	12

INTENTIONALLY LEFT BLANK.

1. Introduction

An investigation of induction heating in carbon-fiber-based composites [1] revealed that the heating resulting from an applied alternating magnetic field is due to dielectric losses in the polymer material that separates crossing fibers in a cross-ply or angle-ply laminate, as long as dielectric breakdown does not occur. The analysis included therein assumed that constant values of the fundamental dielectric properties were known. Heating in a dielectric material subjected to an alternating electric field was shown to be linearly dependent upon these material properties—dielectric constant and dissipation factor (or loss tangent). A global model has been developed [2] that established the electromagnetic response of cross-ply laminated composites to alternating magnetic fields. Three fundamentally separate submodels that consider the in-plane response, the through-thickness response, and the global generation of heat and its quantification as the surface temperature profile were developed. One of the necessary inputs to these models is the complex dielectric constant of the polyetheretherketone (PEEK) polymer. PEEK is the matrix material used in the composite prepreg product APC-2 used in the experimental work and theoretical analysis described by Fink et al. [1].

In a preliminary study, a carbon-fiber-reinforced PEEK cross-ply ($[0/90]_S$) laminate was characterized at various drying times. Ten thermocouples were used to measure the transient temperature profile on the surface of the flat laminated plate, which was heated by magnetic induction using a 10.2-cm-diameter (4 in) Helmholtz coil centered on a 10.2-cm-square (8 in) specimen. The magnetic field was applied to the specimen in five different states of drying: (1) “as-is” (ambient moisture content), (2) 24 hr, (3) 48 hr, (4) 72 hr, and (5) 96 hr at 120 °C and 28-inHg vacuum. Since the same specimen was used for all tests, the drying times were started over after each previous test. Despite PEEK’s low reported moisture content of 0.2 to 0.5% [3–5], the average equilibrium temperature exhibited significant decreases with increased drying time, as shown in Figure 1. This apparent dependence of heat production on small changes in moisture content motivated a closer examination of the effects of moisture content on the properties of the material that affect heating under an applied alternating magnetic field.

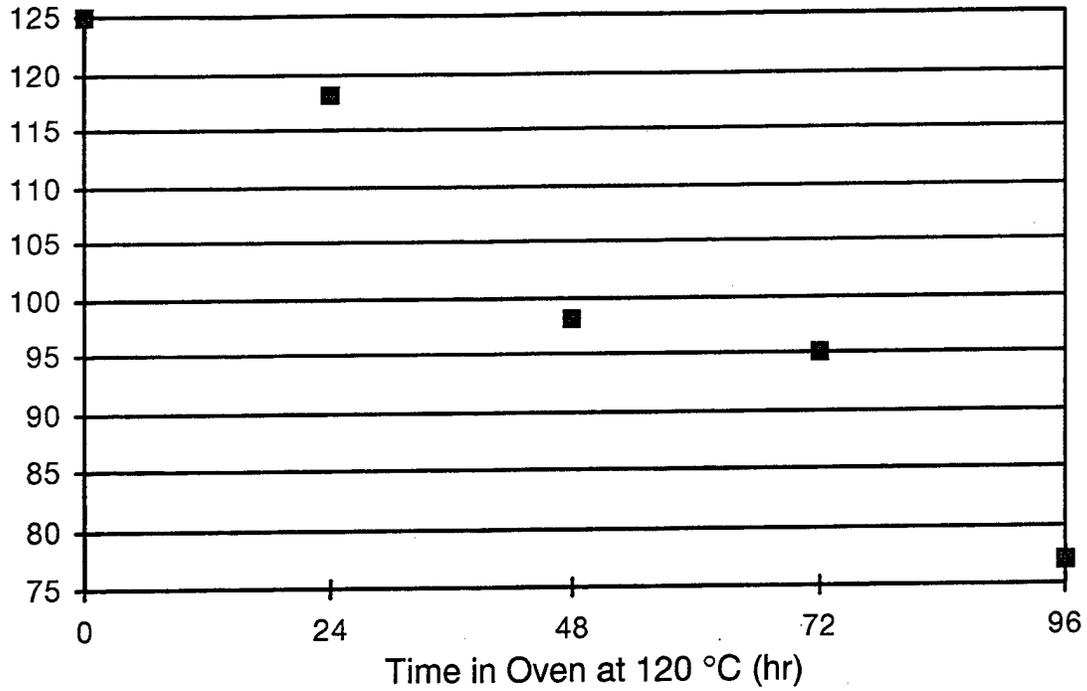


Figure 1. Results of Dried-Laminate Induction-Heating Test.

An equivalent form of an equation for heat generation in our induction heating model [1, 2] can be written as

$$W_j = \frac{\beta_j V_j^2}{h}, \quad (1)$$

where W_j is the heat generation at some node j in the plane of the laminate; V_j is the potential difference between the fibers at node j ; h is the distance through which the electric field created by V acts; and β_j is a function of several material, environmental and microstructural properties at node j .

$$\beta_j = \omega \epsilon_0 \kappa \tan \delta, \quad (2)$$

where ω is the angular frequency, ϵ_0 is the permittivity of vacuum, κ is the relative dielectric constant of the polymer, and $\tan \delta$ represents the imaginary part of the complex dielectric

constant of the polymer. Of the components of equations (1) and (2), only the real and imaginary parts of the complex dielectric constant, κ and $\tan\delta$, are potentially functions of moisture content in the composite's matrix material.

In general, the complex dielectric constant can be significantly affected by the frequency, temperature, and moisture content, as well as by the presence of mobile ions in polymeric systems [6]. Values for these properties are typically reported in the literature at room temperature and line frequency without mention of moisture content or the presence of other ionic species. Reported values are not always applicable at the frequencies often used in induction heating of PEEK-based composites (10^3 to 10^6 Hz range) and at temperatures ranging from room temperature to beyond the melting temperature of PEEK (>340 °C).

Compilations of dielectric constants and dissipation factors [7, 8] have been published for various polymers but are difficult to keep current. Manufacturers often supply data for specific materials at specific temperatures and frequencies; however, it is often necessary to characterize these properties for the particular process parameters of interest. This paper discusses the relation of structure to dielectric properties, describes the methods of experimental measurement performed, and reports the results with respect to temperature, frequency, and moisture content for PEEK.

2. Review of Dielectric Properties

Polymeric materials can be divided into two groups: (1) nonpolar and (2) polar. Nonpolar materials exhibit high resistivity, a low dielectric constant, and a negligible loss tangent; they are, for the most part, unaffected by temperature, frequency, and humidity. Polar polymers generally have lower resistivities and higher dielectric constants and loss tangents.

The dielectric constant (unit capacitance, permittivity) and dissipation factor are very different for nonpolar and polar materials as functions of temperature and frequency. The values of both properties for nonpolar materials are nearly constant over a wide frequency range and for

temperatures below the softening point. For polar materials (often called lossy materials), the dielectric constant decreases with an increase in frequency, whereas the dissipation factor increases and decreases cyclically.

Under the influence of an alternating electric field, electric dipoles are oriented in the material such that the magnitude of the induced charges increases as the polarization of the dielectric increases. There are several possible mechanisms of polarization in a dielectric material:

- (1) Electric Polarization—a shift in the center of charge of the electron cloud relative to the positive center of charge on the nucleus when an electric field is applied.
- (2) Ionic Polarization—displacement of positive and negative ions in relation with one another in respect to their equilibrium positions.
- (3) Orientation Polarization—the alignment of permanent dipoles that exist in a material in the absence of an applied field.
- (4) Space Charge Polarization—the alignment of mobile charges in the material whose movements are restricted by interfaces or other barriers in the material.

The total polarization can be expressed as the sum of these four mechanisms; however, the contribution of each source is dependent on frequency, temperature, and applied voltage. Of the four mechanisms of polarization, orientation polarization has the greatest effect on the total dissipation of energy at the frequencies considered in this work (at least for a “clean” polymer that is free of excessive foreign ionic species).

Frequency effects on polarization are related to the time dependency of the polarization mechanisms with respect to their response to an applied field. For example, electronic polarization would be expected to occur rapidly, while permanent dipole orientation would be

slow in comparison. This is usually explained by the difference in the mass of the electrons vs. the molecules and also on the number of conformations and free volume available for motion in polymers. In thermoplastics, orientation polarization relates to the conformational motion of polymeric chains and ionic polarization to the existence of ionic species or other groups either left over from the processing of the polymer or, possibly, deposited by treated fibers during resin impregnation in composite materials.

Molecular movements take a finite time, and complete orientation as induced by an alternating current may or may not be possible, depending on the frequency of the alternating electric field. Thus, at zero frequency, the dielectric constant will be at a maximum and will remain approximately constant until the dipole orientation time is of the same order as the reciprocal of the frequency. Dipole movement will now be limited, and the dipole polarization effect and the dielectric constant will be reduced. As the frequency further increases, the dipole polarization effect will tend to zero and the dielectric constant will tend to be dependent only on the electronic polarization. When there are two dipole species differing in ease of orientation, there will be two points of inflection in the dielectric constant vs. frequency curve. The dielectric constant of unsymmetrical molecules containing dipoles (polar molecules) will be dependent on the internal viscosity of the dielectric.

Besides a dielectric constant dependence on temperature and frequency, polar molecules exhibit relatively high dielectric power losses at certain frequencies; the maximum power losses corresponding to the point of inflection in the dielectric constant vs. frequency curve (Figure 2). At very low frequencies, the dipole movements are able to keep in phase with changes in the electric field and power losses are low. As the frequency is increased, the point is reached when the dipole orientation cannot be completed in the time available and the dipole becomes out of phase. Measures of the fraction of energy absorbed per cycle by the dielectric from the field are the "power factor" and "dissipation factor." The delay between the changes in the field and the change in polarization leads to a current in a capacitor leading the voltage across it when a dielectric is present. The angle of lead is known as the phase angle and given the symbol θ . The value $90-\theta$ is known as the loss angle and is given the symbol δ . The power factor is defined as

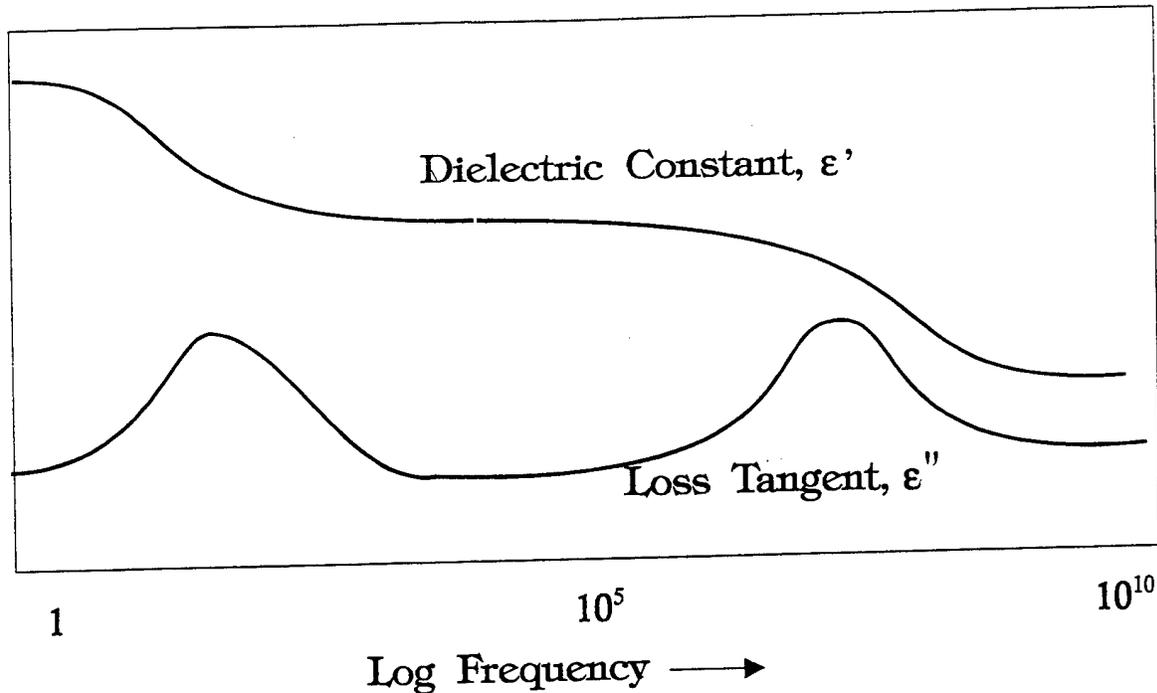


Figure 2. General Relationship Between Real and Imaginary Parts of Complex Dielectric Constant in Dielectric Materials.

$\cos(\theta)$ and the dissipation factor as $\tan(\delta)$. When δ is small, the two are equivalent. The “loss factor” is the product of the dissipation factor and the dielectric constant.

At low frequencies, when power losses are low, these values are also low, but they increase when such frequencies are reached that the dipoles cannot keep in phase. After passing through a peak at some characteristic frequency, they fall in value as the frequency further increases. This is because, at such high frequencies, there is no time for substantial dipole movement and the power losses are reduced. Because of the dependence of the dipole movement on the internal viscosity, the power factor, like the dielectric constant, is strongly dependent on temperature.

In polar polymers, the situation is more complex since there are many dipoles attached to one chain. Either these dipoles may be attached to the main chain, or the polar groups may not be directly attached to the main chain and the dipoles may, to some extent, rotate independently.

In the case with dipoles integral with the main chain, in the absence of an electric field, the dipoles will be randomly disposed but will be fixed by the disposition of the main chain atoms. On application of an electric field, complete dipole orientation is not possible because of spatial requirements imposed by the chain structure. Furthermore, in the polymeric system, the different molecules are coiled in different ways and the time for orientation will be dependent on the particular disposition. Thus, whereas simple polar molecules have a sharply defined power loss maxima, the power loss vs. frequency curve of polar polymers is broad, due to the dispersion of orientation times.

When dipoles are directly attached to the chain, their movement will obviously depend on the ability of chain segments to move. Thus, the dipole polarization effect will be less significant below the glass transition temperature than above it. PEEK is such a polymer; the offset oxygen on the ketone segment provides the polarizability of the polymer. It is expected that the loss tangent will have a significant increase directly after the glass transition temperature, which is about 143 °C.

Ionic polarizability is the inducement of dipoles by relative movement of positive and negative ions in a partial ionic solid. Ionic polarization generally has a predominate influence on the loss tangent at higher frequencies than for dipole polarization. Since the imaginary part of the complex dielectric constant (or dielectric loss factor) is large whenever the frequency reaches the upper limit for any dielectric response mechanism, such as ionic polarization, a second peak in the loss tangent vs. frequency curve is expected at some frequency higher than the dipole polarization peak.

These are generalizations that may not be the case with thermoplastic polymers with embedded fiber reinforcement as in the PEEK/AS4 carbon-fiber system being investigated. The peaks for dipole and ionic polarization may fall within the same elevated frequency range and be difficult to distinguish. Therefore, dielectric properties may be very different for pure polymers and for polymers with fibrous reinforcement. Processing reagents and surface treatment sizings

used on fibers to enhance fiber-polymer bonding can contribute ionic functional groups that would serve to increase the ionic polarizability of the polymeric regions.

Takahagi and Ishitani [9] attempted to characterize carbon-fiber surface chemistry. Their research showed a predominance of $-OH$ and $C=O$ functional groups, even without surface treatment. If the loss tangents of any associated molecules caused peaks at or near the same frequency as the dipole polarization inherent in the PEEK polymer, an effective increase in the loss tangent at that frequency would be seen.

Measuring this expected difference in dielectric properties between the neat PEEK polymer and the PEEK polymer with fiber reinforcement is not possible without first removing the fibers from the polymer due to the extreme difference in electrical properties between the polymer and the conductive carbon fibers. Our study of induction heating includes only the dielectric data obtained for virgin PEEK.

Both the dielectric constant and the dissipation factor are related to the presence of moisture. Moisture, absorbed in a filler at interfaces between different components or in laminations, can often be detected by changes in dissipation factor and capacitance [6]. Water molecules are highly polar and highly mobile. At frequencies of 10^5 – 10^6 Hz, the losses in water are ionic and increase with increases in dissolved ionizable salts. Hartshorn et al. [10] provide loss tangent vs. frequency data for several polymers in wet and dry states. The wet specimens, as a rule, gave higher loss tangent data at all frequencies. Potthoff [11] showed that increasing the temperature also increased the total effect of the presence of moisture in the system.

Moisture content studies on 150G PEEK [3] and PEEK and polyetherimide (PEI)-based composites [4, 5] indicate relatively low moisture contents in these advanced thermoplastic systems; saturated moisture absorption in PEEK-based APC-2 averaged about 0.2 weight-percent at room temperature and 100% relative humidity. Comparatively, the amorphous PES-based RADEL-C/T300 absorbed about 2.5% at equilibrium. The lower moisture absorption of PEEK is due to its semi-crystalline nature while the other resins (PEI and PES) are amorphous.

Buchman and Isayev [5] reported an equilibrium moisture content of about 0.6% for PEI-based CYPAC 7005/G30-500, while Demonet [4] reported 0.56%. The increase in moisture content from PEEK to PEI to PES may correspond to increases in loss tangent.

The next two sections describe the methods used to determine dielectric properties of PEEK over applicable ranges of frequency, temperature, and moisture pertinent to the present study and the results of those tests.

3. Characterization

The real and imaginary parts of the complex dielectric constant were measured at frequencies of 10^3 , 10^4 , 10^5 , and 10^6 Hz using a cell built for similar characterization of filled polymers by Sturman [12, 13]. The physics of the cell and the specimen preparation is described by Sturman [12].

Specimens were cut from a semi-crystalline PEEK plate; the diameter of the machined disks was 4.01 cm and the thickness was 0.32 cm. A metal sleeve connected to a variac power supply surrounded the specimen. The power was incrementally increased and held for 15 min as current caused joule heating in the sleeve which conductively transmitted the energy to the polymer specimen. A thermocouple placed on the inside of the sleeve recorded the equilibrium temperature. Fifteen minutes between cycles allowed the aluminum sleeve, the cell electrodes, and the specimen to reach thermal equilibrium with the temperature increasing in increments of 25 °C from ambient to 250 °C.

The first set of measurements was performed at room temperature and at full moisture content (no drying). Results from three measurements were averaged, tabulated in Table 1 and shown in Figures 3 through 5 for the dielectric constant, loss tangent and resistivity versus frequency respectively. Figure 3 shows that the dielectric constant drops off rapidly between 100 kHz and 1 MHz. Figure 4 shows the expected increase in loss tangent with frequency in this

Table 1. Averaged Data From Three Electrical Property Measurements on PEEK at Room Temperature

Frequency (kHz)	Capacitance (pF)	Dielectric Constant	Loss Tangent	Conductance (1/ohm)	Resistance (ohm)	Resistivity (ohm-cm)
1	13.31	3.744	0.0014	1.00E - 10	1.00E + 10	4.02E + 11
10	13.282	3.736	0.0034	2.90E - 09	3.45E + 08	1.39E + 10
100	13.209	3.716	0.0160	1.33E - 07	7.52E + 06	3.02E + 08
1,000	12.392	3.486	0.0421	7.12E - 06	1.40E + 05	5.63E + 06

Note: The capacitance in vacuo is 3.555 pF for a surface area to diameter ratio of 0.402.

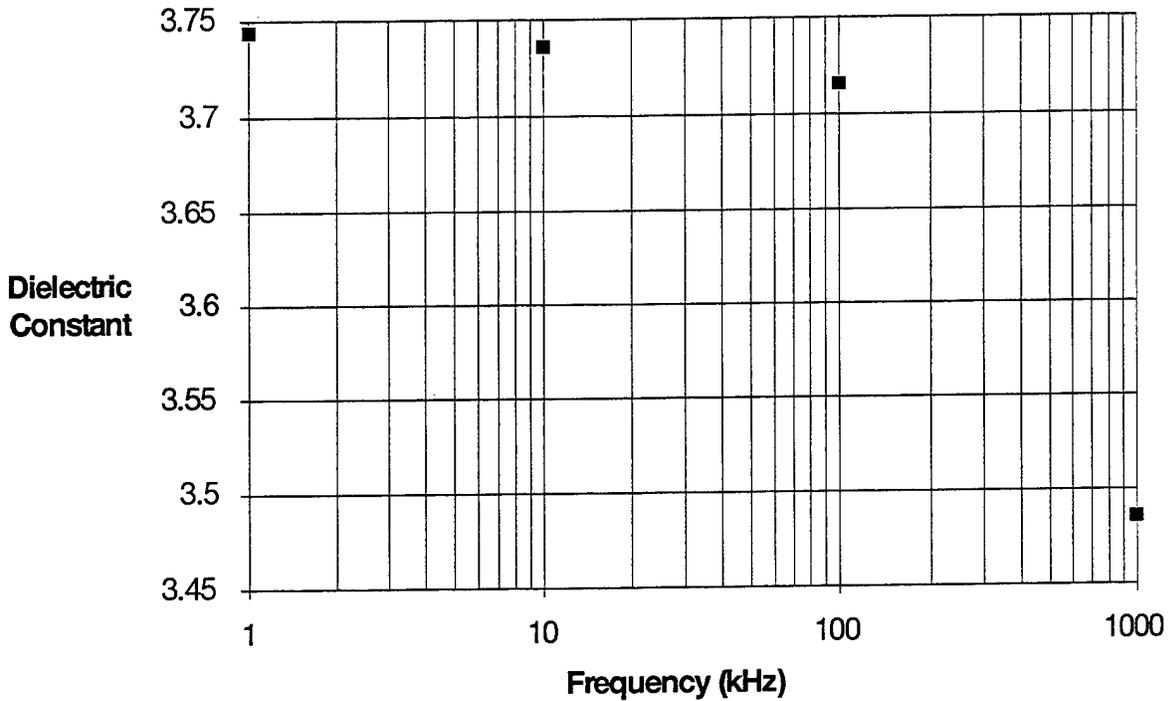


Figure 3. Room Temperature Dielectric Constant Results for PEEK.

range. The resistivity drops off linearly on the log-log plot of Figure 5 but is still of significant value at 1-MHz frequency.

These data are significantly higher in value in the higher frequency range than reported by PEEK's manufacturer, ICI [14]; ICI's data for 450G PEEK indicates loss tangents below 0.004

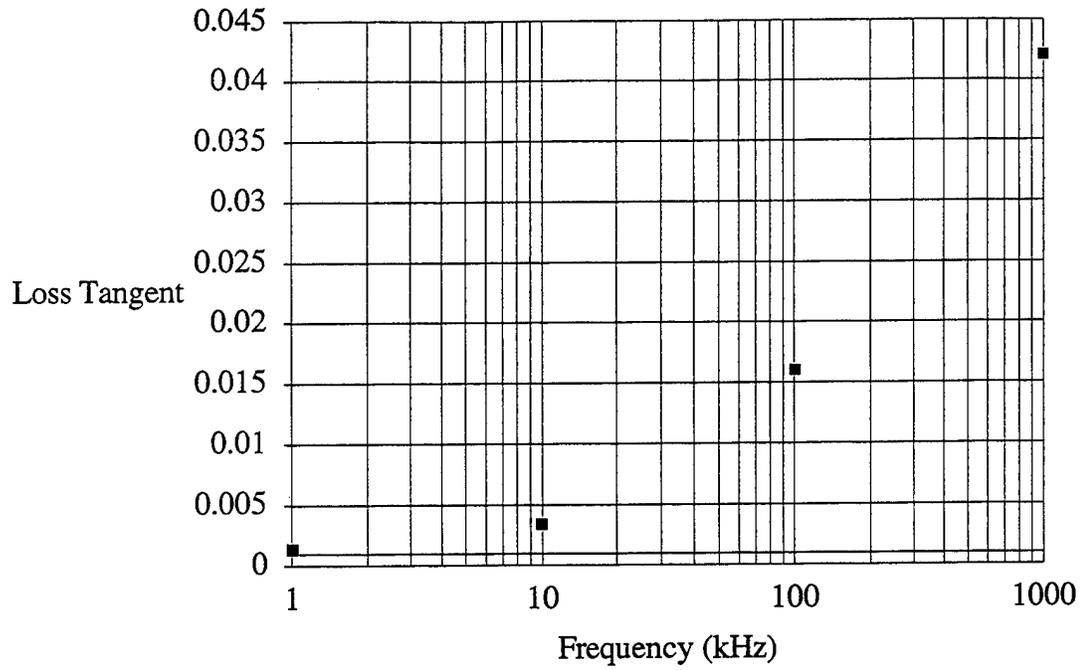


Figure 4. Room Temperature Dielectric Loss Tangent Results for PEEK.

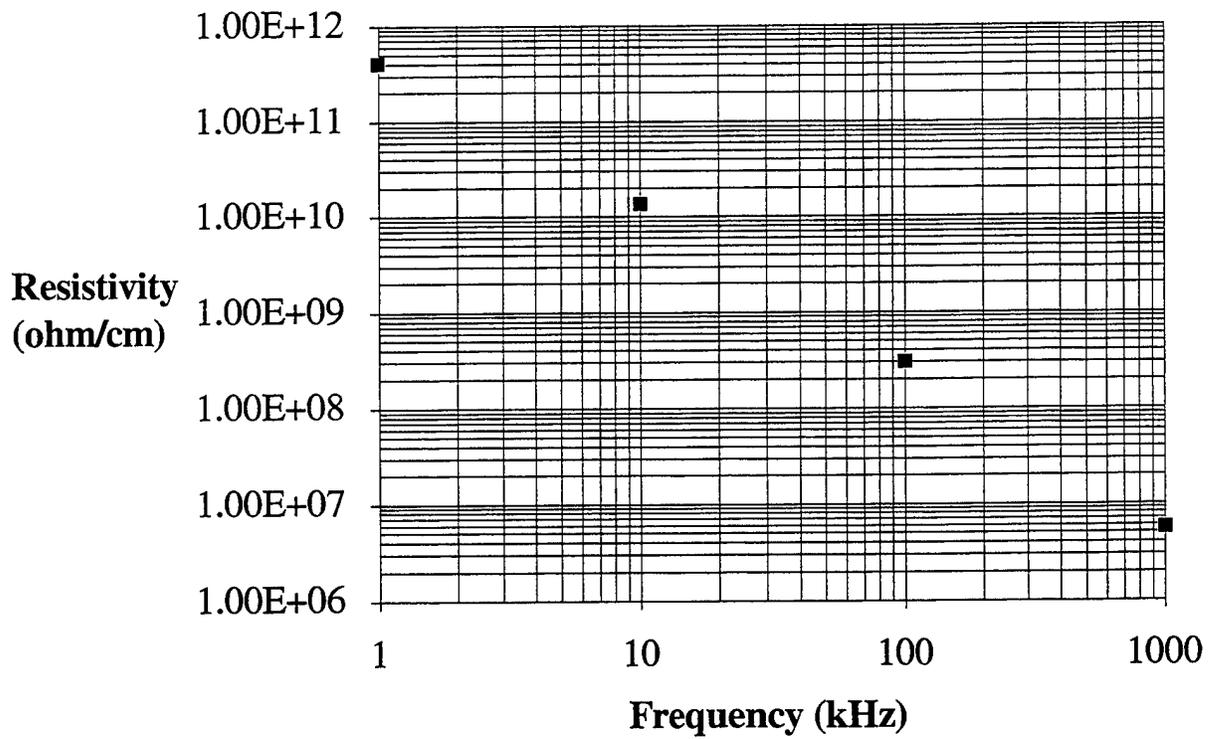


Figure 5. Room Temperature Results for Resistivity of PEEK.

at all frequencies between line frequency and 10^{10} Hz at room temperature. This discrepancy between our results and published data further precipitated an investigation of the effects of moisture content on dielectric loss tangent of PEEK.

Twelve specimens were cut from the original PEEK plate. Nine of the specimens were placed in an oven at 120 °C and 28-inHg vacuum in sets of three for periods of 24, 48, and 72 hr. The first three specimens were tested at temperatures ranging from 25 °C to 250 °C in 25 °C increments (nominally). They were assumed to have full equilibrium moisture content. The other specimens were tested as they were removed from the oven. Since each test took approximately 3 hr to run, some specimens were in the oven for 3–6 hr longer. Table 2 and Figure 6 show the dimensional and weight change data for all specimens. The solid squares represent the test data, while the line represents a spline fit to the average of the three data points at each nominal time plus 3 hr (which is the average time for each test set). From this plot, the maximum moisture content is taken to be 0.28 weight-percent; this value corresponds well to data reported in the literature.

Table 2. Weight Change Data for PEEK's Dielectric Loss Tangent vs. Moisture Content Testing

Time in Oven (hr)	Initial Weight (g)	Test Weight (g)	% Change	Average Change (%)
0	4.764	4.764	0	
0	4.812	4.812	0	0
0	4.693	4.693	0	
24	4.827	4.818	0.18	
27	4.71	4.701	0.19	0.1767
30	4.796	4.788	0.16	
48	4.774	4.762	0.25	
51	4.813	4.801	0.24	0.2533
54	4.682	4.669	0.27	
72	4.792	4.777	0.31	
75	4.845	4.832	0.27	0.28
78	4.613	4.601	0.26	

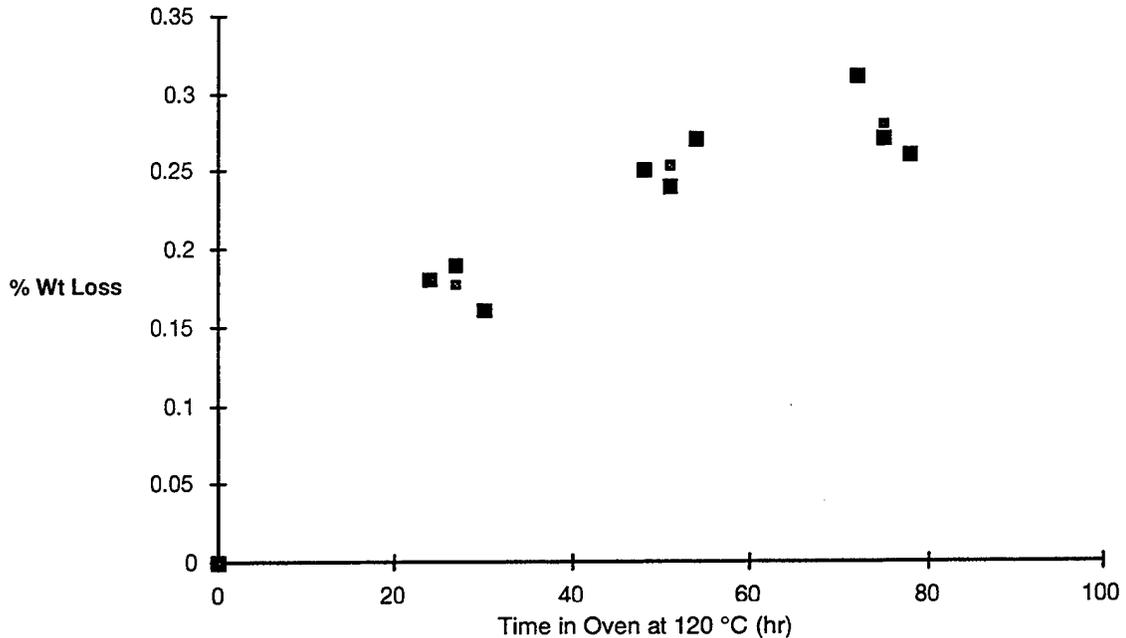


Figure 6. Percent Weight Loss of PEEK Specimens vs. Drying Time. Note That the Large Squares Represent the Three Specimens at Each Nominal Time. The Small Squares Represent the Average of the Three Data Points at Each Nominal Time.

Since maintaining the exact nominal temperatures was not possible, the results for the testing of each specimen have data points at different temperatures. However, the temperatures were generally maintained to within 5 °C of the nominal test temperature, so data has been averaged and plotted, assuming that all tests took place at the nominal test temperature. Figures 7 through 10 show the averaged loss tangent results for the 0-, 24-, 48-, and 72-hr specimens, respectively.

The influence of moisture content on the range of possible values of loss tangent at 100 kHz is shown in Figure 11. Here the 72-hr test is taken to be nearly zero moisture content and the 0-hr test is taken to be 0.28 weight-percent moisture. Also shown on Figure 11 is the value for loss tangent reported by ICI [14]. Although it was not reported as a function of moisture content, it might be concluded that their test specimens were indeed sufficiently dried prior to testing.

No attempt was made to quantify the percent moisture loss in fiber-reinforced PEEK. Furthermore, no attempt was made to increase the moisture content of the PEEK or APC-2

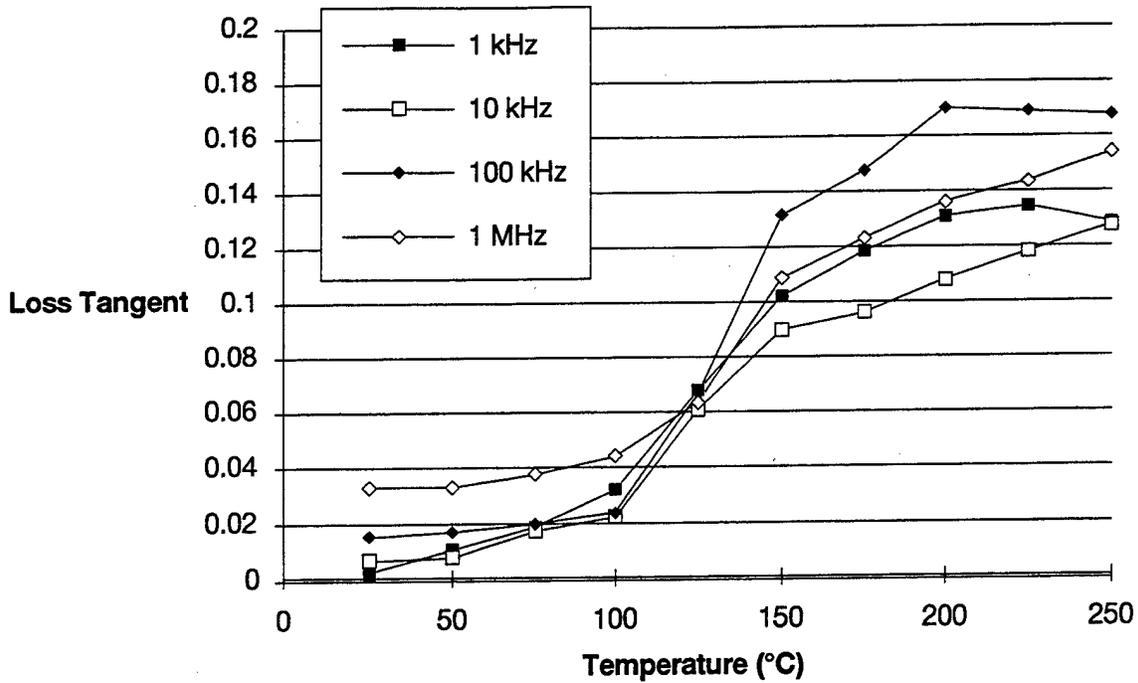


Figure 7. Average Experimental Loss Tangent Data for Three Undried PEEK Specimens.

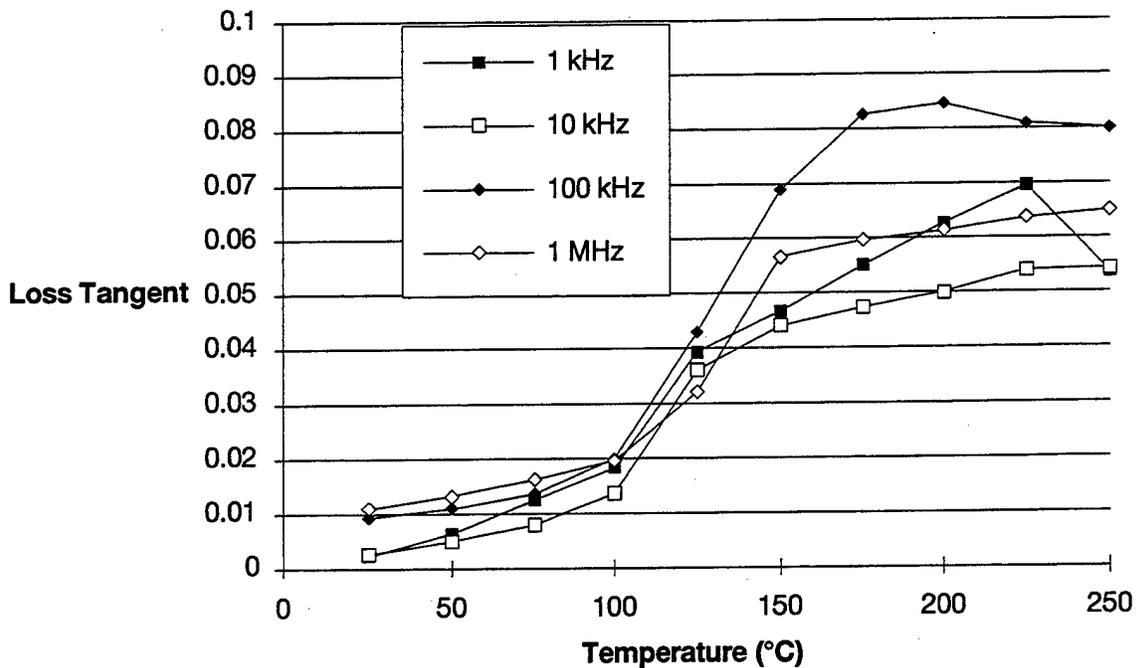


Figure 8. Average Experimental Loss Tangent Data for Three PEEK Specimens Dried at 120 °C for 24 hr.

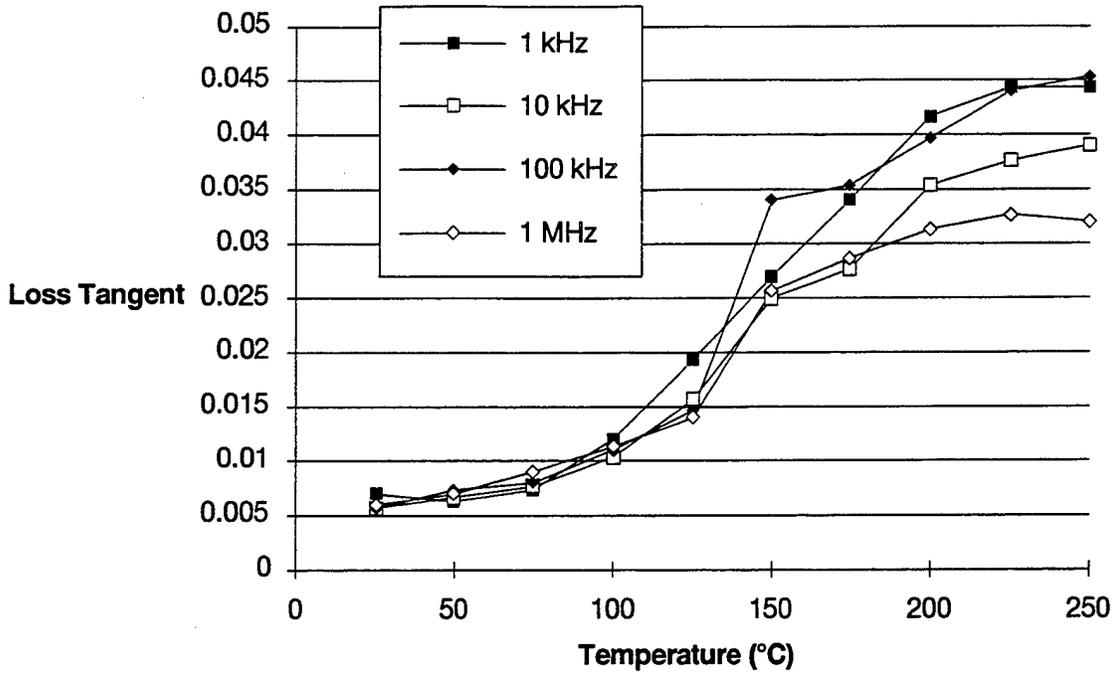


Figure 9. Average Experimental Loss Tangent Data for Three PEEK Specimens Dried at 120 °C for 48 hr.

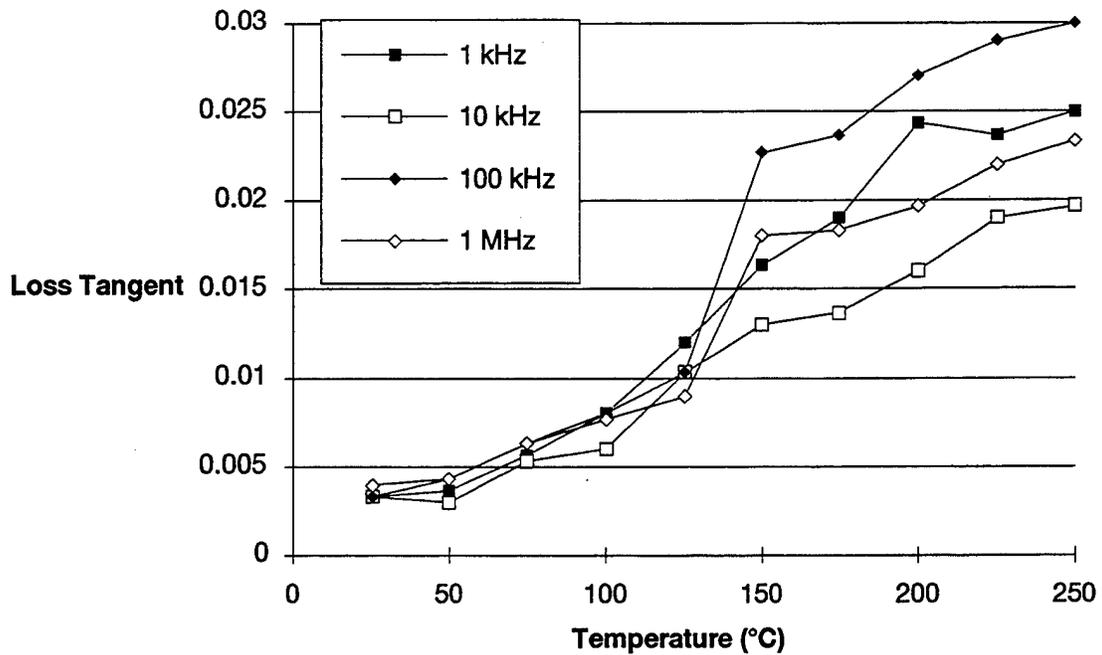


Figure 10. Average Experimental Loss Tangent Data for Three PEEK Specimens Dried at 120 °C for 72 hr.

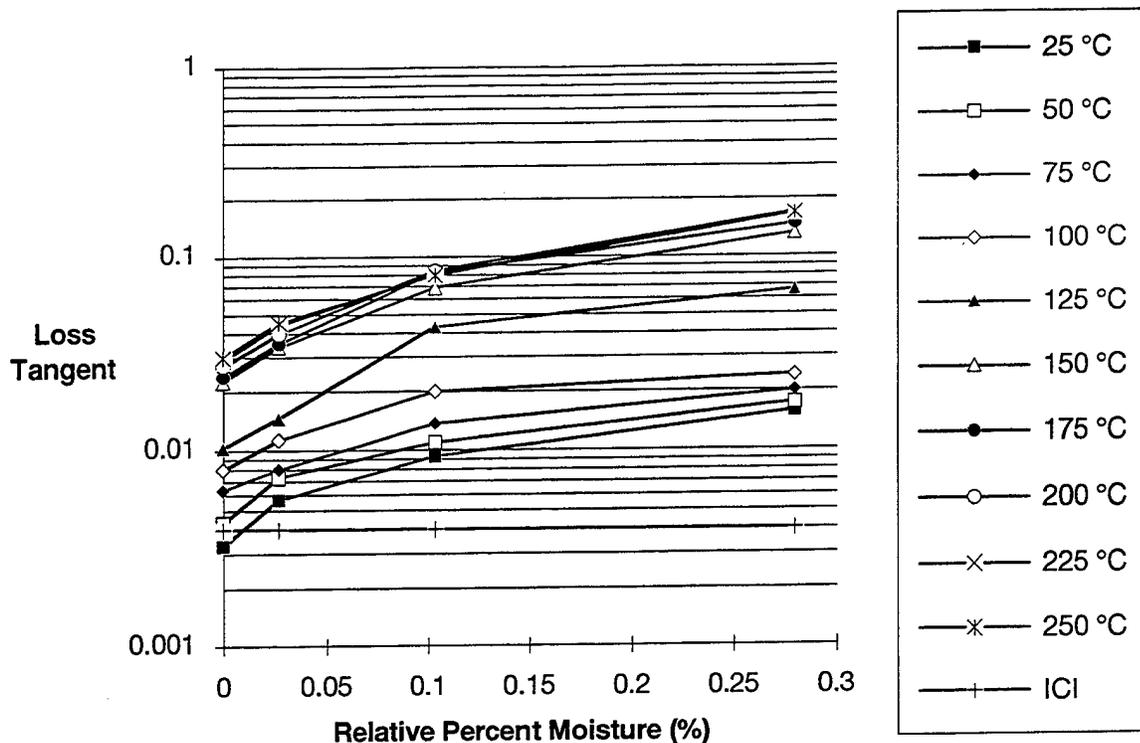


Figure 11. Experimental and ICI-Reported Loss Tangent Data for PEEK at 100 kHz. Note That the Experimental Data Approaches the ICI-Reported Data for Low Moisture Contents.

specimens, which, from the moisture dependence of loss tangent in Figure 10, might further increase heating. From this study, it appears possible to optimally localize heating at a bond interface in fusion bonding applications by increasing the moisture content of that local region; however, adding moisture in the presence of thermal gradients may nucleate undesirable voids in the composite [15].

4. Summary

Heating of continuous carbon-fiber-reinforced polymers (CFRP's) by the application of an alternating magnetic field has been shown to be due to dielectric losses in the polymer. A fundamental analysis indicated that heating in dielectric materials, such as PEEK, is linearly dependent upon the dielectric constant and the dissipation factor. Although not explicitly

characterized, induction heating in PEEK-based composites was shown to increase significantly with increases in moisture content and initial as-is loss tangent measurements showed relatively high values compared to those reported by the manufacturer. This precipitated a study of the effect of moisture content on dielectric properties. These properties were theoretically investigated as functions of temperature, frequency, ionic conductivity, and moisture content.

The imaginary part of the complex dielectric constant is an integral parameter in dielectric heating, and induction-heating thermal calculations are complicated by the loss tangent's dependence on frequency, temperature, moisture, and ionic conductivity. Fundamental tests on PEEK were performed to determine these relationships. A significant result is the extreme moisture dependence of the lossiness of PEEK. Under dry conditions, the loss data were near that reported by the manufacturer. Our immediate application of this study is to interpolate the expected loss tangent data for the frequency and temperature range expected in induction-heating experiments. This study has also established the importance of documenting the dependence of dielectric properties on frequency, temperature, and moisture.

From this study, it appears possible to optimally localize heating at a bond interface in fusion-bonding applications by increasing the moisture content of that local region; however, adding moisture in the presence of thermal gradients may nucleate undesirable voids in the composite [15].

INTENTIONALLY LEFT BLANK.

5. References

1. Fink, B. K., R. L. McCullough, and J. W. Gillespie Jr. "A Local Theory of Heating in Cross-Ply Carbon Fiber Thermoplastic Composites by Magnetic Induction." *Polymer Engineering and Science*, vol. 32, p. 357, 1992.
2. Fink, B. K. Dissertation. University of Delaware, 1991.
3. Pratte, J. F., W. H. Krueger, and I. Y. Chang, "High Performance Thermoplastic Composites with Poly(etheretherketone) Matrix." *34th International SAMPE Symposium Proceedings*, 1989.
4. Demonet, C. M. "Interaction of Moisture With Resin Matrix Composites." *21st SAMPE Technical Conference Proceedings*, 1989.
5. Buchman, A., and A. I. Isayev. "Water Absorption of Some Thermoplastic Composites." *SAMPE Journal*, vol. 27, 1991.
6. Mathes, K. *Electrical and Electronic Properties of Polymers: A State-of-the-Art Compendium*. J. I. Kroschwitz, (ed.), NY: John Wiley & Sons, Inc., 1988.
7. Von Hippel, A. *Dielectric Materials and Applications*. NY: The Technology Press of M.I.T. and John Wiley & Sons, Inc., 1966.
8. Ogorkiewicz, R. M. *Engineering Properties of Thermoplastics*. 1970.
9. Takahagi, T., and A. Ishitani. "XPS Studies by Use of the Digital Difference Spectrum Technique of Functional Groups on the Surface of CF." *Carbon*, vol. 22, pp. 41-46, 1984.
10. Hartshorn, L., J. Parry, and E. Rushton. *Proc. Inst. Electr. Eng. Part 2*. Vol. 100, p. 23, 1953.
11. Potthoff, K. *Electrotech. Z.* Vol. 80, p. 688, 1959.
12. Sturman, P. C., Jr. Doctoral dissertation. University of Delaware, 1990.
13. McCullough, R. L., and J. J. Kramer. "An Investigation of the Influence of Microstructure on the Electrical Properties of Composite Materials." Center for Composite Materials Report, University of Delaware, 1991.
14. ICI Advanced Materials. "Vitrex PEEK 450G." ICI Advanced Materials, 1985.
15. Roychowdhury, S. Personal communication. 1991.

INTENTIONALLY LEFT BLANK.

<u>NO. OF COPIES</u>	<u>ORGANIZATION</u>
2	DEFENSE TECHNICAL INFORMATION CENTER DTIC DDA 8725 JOHN J KINGMAN RD STE 0944 FT BELVOIR VA 22060-6218
1	HQDA DAMO FDT 400 ARMY PENTAGON WASHINGTON DC 20310-0460
1	OSD OUSD(A&T)/ODDDR&E(R) R J TREW THE PENTAGON WASHINGTON DC 20301-7100
1	DPTY CG FOR RDA US ARMY MATERIEL CMD AMCRDA 5001 EISENHOWER AVE ALEXANDRIA VA 22333-0001
1	INST FOR ADVNCD TCHNLGY THE UNIV OF TEXAS AT AUSTIN PO BOX 202797 AUSTIN TX 78720-2797
1	DARPA B KASPAR 3701 N FAIRFAX DR ARLINGTON VA 22203-1714
1	NAVAL SURFACE WARFARE CTR CODE B07 J PENNELLA 17320 DAHLGREN RD BLDG 1470 RM 1101 DAHLGREN VA 22448-5100
1	US MILITARY ACADEMY MATH SCI CTR OF EXCELLENCE DEPT OF MATHEMATICAL SCI MADN MATH THAYER HALL WEST POINT NY 10996-1786

<u>NO. OF COPIES</u>	<u>ORGANIZATION</u>
1	DIRECTOR US ARMY RESEARCH LAB AMSRL D D R SMITH 2800 POWDER MILL RD ADELPHI MD 20783-1197
1	DIRECTOR US ARMY RESEARCH LAB AMSRL DD 2800 POWDER MILL RD ADELPHI MD 20783-1197
1	DIRECTOR US ARMY RESEARCH LAB AMSRL CS AS (RECORDS MGMT) 2800 POWDER MILL RD ADELPHI MD 20783-1145
3	DIRECTOR US ARMY RESEARCH LAB AMSRL CI LL 2800 POWDER MILL RD ADELPHI MD 20783-1145
	<u>ABERDEEN PROVING GROUND</u>
4	DIR USARL AMSRL CI LP (BLDG 305)

<u>NO. OF COPIES</u>	<u>ORGANIZATION</u>
1	DIRECTOR US ARMY RESEARCH LAB AMSRL CP CA D SNIDER 2800 POWDER MILL RD ADELPHI MD 20783-1145
1	DIRECTOR US ARMY RESEARCH LAB AMSRL OP SD TA 2800 POWDER MILL ROAD ADELPHI MD 20783-1145
3	DIRECTOR US ARMY RESEARCH LAB AMSRL OP SD TL 2800 POWDER MILL ROAD ADELPHI MD 20783-1145
1	DIRECTOR US ARMY RESEARCH LAB AMSRL OP SD TP 2800 POWDER MILL ROAD ADELPHI MD 20783-1145
2	DIRECTOR US ARMY RESEARCH LAB AMSRL OP CI AD TECH PUB BR RECORDS MGMT ADMIN 2800 POWDER MILL ROAD ADELPHI MD 20783-1197
1	HQDA DAMI FIT NOLAN BLDG WASHINGTON DC 20310-1025
1	DIRECTOR DA OASARDA SARD SO 103 ARMY PENTAGON WASHINGTON DC 20310-0103
1	DEPUTY ASST SCY FOR R&T SARD TT RM 3EA79 THE PENTAGON WASHINGTON DC 20301-7100

<u>NO. OF COPIES</u>	<u>ORGANIZATION</u>
1	COMMANDER US ARMY MATERIEL CMD AMXMI INT 5001 EISENHOWER AVE ALEXANDRIA VA 22333-0001
2	COMMANDER US ARMY ARDEC AMSTA AR AE WW E BAKER J PEARSON PICATINNY ARSENAL NJ 07806-5000
1	COMMANDER US ARMY ARDEC AMSTA AR TD C SPINELLI PICATINNY ARSENAL NJ 07806-5000
1	COMMANDER US ARMY ARDEC AMSTA AR FSE T GORA PICATINNY ARSENAL NJ
6	COMMANDER US ARMY ARDEC AMSTA AR CCH A W ANDREWS S MUSALLI R CARR M LUCIANO E LOGSDEN T LOUZEIRO PICATINNY ARSENAL NJ 07806-5000
4	COMMANDER US ARMY ARDEC AMSTA AR CC G PAYNE J GEHBAUER C BAULIEU H OPAT PICATINNY ARSENAL NJ 07806-5000

<u>NO. OF COPIES</u>	<u>ORGANIZATION</u>
1	COMMANDER US ARMY ARDEC AMSTA AR CCH P J LUTZ PICATINNY ARSENAL NJ 07806-5000
1	COMMANDER US ARMY ARDEC AMSTA AR FSF T C LIVECCHIA PICATINNY ARSENAL NJ 07806-5000
1	COMMANDER US ARMY ARDEC AMSTA AR QAC T C C PATEL PICATINNY ARSENAL NJ 07806-5000
2	COMMANDER US ARMY ARDEC AMSTA AR M D DEMELLA F DIORIO PICATINNY ARSENAL NJ 07806-5000
3	COMMANDER US ARMY ARDEC AMSTA AR FSA A WARNASH B MACHAK M CHIEFA PICATINNY ARSENAL NJ 07806-5000
2	COMMANDER US ARMY ARDEC AMSTA AR FSP G M SCHIKSNIS D CARLUCCI PICATINNY ARSENAL NJ 07806-5000

<u>NO. OF COPIES</u>	<u>ORGANIZATION</u>
1	COMMANDER US ARMY ARDEC AMSTA AR FSP A P KISATSKY PICATINNY ARSENAL NJ 07806-5000
2	COMMANDER US ARMY ARDEC AMSTA AR CCH C H CHANIN S CHICO PICATINNY ARSENAL NJ 07806-5000
9	COMMANDER US ARMY ARDEC AMSTA AR CCH B P DONADIA F DONLON P VALENTI C KNUTSON G EUSTICE S PATEL G WAGNECZ R SAYER F CHANG PICATINNY ARSENAL NJ 07806-5000
6	COMMANDER US ARMY ARDEC AMSTA AR CCL F PUZYCKI R MCHUGH D CONWAY E JAROSZEWSKI R SCHLENNER M CLUNE PICATINNY ARSENAL NJ 07806-5000
1	COMMANDER US ARMY ARDEC AMSTA AR QAC T D RIGOGLIOSO PICATINNY ARSENAL NJ 07806-5000

NO. OF
COPIES ORGANIZATION

1 COMMANDER
US ARMY ARDEC
AMSTA AR SRE
D YEE
PICATINNY ARSENAL NJ
07806-5000

1 COMMANDER
US ARMY ARDEC
AMSTA AR WET
T SACHAR
BLDG 172
PICATINNY ARSENAL NJ
07806-5000

1 COMMANDER
US ARMY ARDEC
SMCAR ASF
PICATINNY ARSENAL NJ
07806-5000

1 COMMANDER
US ARMY ARDEC
AMSTA AR WEL F
INTELLIGENCE SPECIALIST
M GUERRIERE
PICATINNY ARSENAL NJ
07806-5000

11 PROJECT MANAGER
TANK MAIN
ARMAMENT SYSTEMS
SFAE GSSC TMA
R MORRIS
C KIMKER
D GUZOWICZ
E KOPACZ
R ROESER
R DARCY
R MCDANOLDS
L D ULISSE
C ROLLER
J MCGREEN
B PATTEN
PICATINNY ARSENAL NJ
07806-5000

NO. OF
COPIES ORGANIZATION

2 PEO FIELD ARTILLERY SYSTEMS
SFAE FAS PM
H GOLDMAN
T MCWILLIAMS
PICATINNY ARSENAL NJ
07806-5000

6 PM SADARM
SFAE GCSS SD
COL B ELLIS
M DEVINE
R KOWALSKI
W DEMASSI
J PRITCHARD
S HROWNAK
PICATINNY ARSENAL NJ
07806-5000

1 COMMANDER
US ARMY ARDEC
PRODUCTION BASE
MODERN ACTY
AMSMC PBM K
PICATINNY ARSENAL NJ
07806-5000

3 COMMANDER
U S ARMY TACOM
PM TACTICAL VEHICLES
SFAE TVL
SFAE TVM
SFAE TVH
6501 ELEVEN MILE RD
WARREN MI 48397-5000

1 COMMANDER
U S ARMY TACOM
PM ABRAMS
SFAE ASM AB
6501 ELEVEN MILE RD
WARREN MI 48397-5000

1 COMMANDER
U S ARMY TACOM
PM BFVS
SFAE ASM BV
6501 ELEVEN MILE RD
WARREN MI 48397-5000

<u>NO. OF COPIES</u>	<u>ORGANIZATION</u>
1	COMMANDER U S ARMY TACOM PM AFAS SFAE ASM AF 6501 ELEVEN MILE RD WARREN MI 48397-5000
2	COMMANDER U S ARMY TACOM PM SURV SYS SFAE ASM SS T DEAN SFAE GCSS W GSI M D COCHRAN 6501 ELEVEN MILE RD WARREN MI 48397-5000
1	COMMANDER U S ARMY TACOM PM RDT&E SFAE GCSS W AB J GODELL 6501 ELEVEN MILE RD WARREN MI 48397-5000
1	COMMANDER U S ARMY TACOM PM SURVIVABLE SYSTEMS SFAE GCSS W GSI H M RYZYI 6501 ELEVEN MILE RD WARREN MI 48397-5000
1	COMMANDER U S ARMY TACOM PM BFV SFAE GCSS W BV S DAVIS 6501 ELEVEN MILE RD WARREN MI 48397-5000
1	COMMANDER U S ARMY TACOM PM LIGHT TACTICAL VEHICLES AMSTA TR S AJ J MILLS MS 209 6501 ELEVEN MILE RD WARREN MI 48397-5000

<u>NO. OF COPIES</u>	<u>ORGANIZATION</u>
1	COMMANDER U S ARMY TACOM PM GROUND SYSTEMS INTEGRATION SFAE GCSS W GSI R LABATILLE 6501 ELEVEN MILE RD WARREN MI 48397-5000
1	COMMANDER U S ARMY TACOM CHIEF ABRAMS TESTING SFAE GCSS W AB QT T KRASKIEWICZ 6501 ELEVEN MILE RD WARREN MI 48397-5000
1	COMMANDER US ARMY TACOM AMSTA SF WARREN MI 48397-5000
1	COMMANDER SMCWV QAE Q B VANINA BLDG 44 WATERVLIET ARSENAL WATERVLIET NY 12189-4050
14	COMMANDER US ARMY TACOM ASMTA TR R J CHAPIN R MCCLELLAND D THOMAS J BENNETT D HANSEN AMSTA JSK S GOODMAN J FLORENCE K IYER J THOMSON AMSTA TR D D OSTBERG L HINOJOSA B RAJU AMSTA CS SF H HUTCHINSON F SCHWARZ WARREN MI 48397-5000

<u>NO. OF COPIES</u>	<u>ORGANIZATION</u>
1	COMMANDER SMCWV SPM T MCCLOSKEY BLDG 253 WATERVLIET ARSENAL WATERVLIET NY 12189-4050
10	BENET LABORATORIES AMSTA AR CCB R FISCELLA G D ANDREA M SCAVULO G SPENCER P WHEELER K MINER J VASILAKIS G FRIAR R HASENBEIN SMCAR CCB R S SOPOK WATERVLIET NY 12189
2	TSM ABRAMS ATZK TS S JABURG W MEINSHAUSEN FT KNOX KY 40121
3	ARMOR SCHOOL ATTN ATZK TD R BAUEN J BERG A POMEY FT KNOX KY 40121
2	HQ IOC TANK AMMO TEAM AMSIO SMT R CRAWFORD W HARRIS ROCK ISLAND IL 61299-6000
1	DIRECTOR U S ARMY AMCOM SFAE AV RAM TV D CALDWELL BUILDING 5300 REDSTONE ARSENAL AL 35898

<u>NO. OF COPIES</u>	<u>ORGANIZATION</u>
4	DIRECTOR US ARMY CECOM NIGHT VISION AND ELECTRONIC SENSORS DIRECTORATE AMSEL RD NV CM CCD R ADAMS R MCLEAN A YINGST AMSEL RD NV VISP E JACOBS 10221 BURBECK RD FT BELVOIR VA 22060-5806
2	CDR USA AMCOM AVIATION APPLIED TECH DIR J SCHUCK FT EUSTIS VA 23604-5577
1	U S ARMY CRREL P DUTTA 72 LYME RD HANOVER NH 03755
1	US ARMY CERL R LAMPO 2902 NEWMARK DR CHAMPAIGN IL 61822
2	U S ARMY CORP OF ENGINEERS CERD C T LIU CEW ET T TAN 20 MASS AVE NW WASHINGTON DC 20314
10	DIRECTOR US ARMY NATL GRND INTEL CTR D LEITER S EITELMAN M HOLTUS M WOLFE S MINGLEDORF H C ARDLEIGH J GASTON W GSTATTENBAUER R WARNER J CRIDER 220 SEVENTH STREET NE CHARLOTTESVILLE VA 22091

<u>NO. OF COPIES</u>	<u>ORGANIZATION</u>
6	US ARMY SBCCOM SOLDIER SYSTEMS CENTER BALLISTICS TEAM J WARD MARINE CORPS TEAM J MACKIEWICZ BUS AREA ADVOCACY TEAM W HASKELL SSCNC WST W NYKVIST T MERRILL S BEAUDOIN KANSAS ST NATICK MA 01760-5019
1	US ARMY COLD REGIONS RSCH & ENGRNG LAB P DUTTA 72 LYME RD HANOVER NH 03755
1	SYSTEM MANAGER ABRAMS BLDG 1002 RM 110 ATZK TS LTC J H NUNN FT KNOX KY 40121
9	US ARMY RESEARCH OFFICE A CROWSON J CHANDRA H EVERETT J PRATER R SINGLETON G ANDERSON D STEPP D KISEROW J CHANG PO BOX 12211 RESEARCH TRIANGLE PARK NC 27709-2211
1	DIRECTORATE OF CMBT DEVELOPMENT C KJORO 320 ENGINEER LOOP STE 141 FT LEONARD WOOD MO 65473-8929

<u>NO. OF COPIES</u>	<u>ORGANIZATION</u>
1	COMMANDANT U S ARMY FIELD ARTILLERY CTR AT FT SILL ATFS CD LTC BUMGARNER FT SILL OK 73503 5600
1	CHIEF USAIC LTC T J CUMMINGS ATZB COM FT BENNING GA 31905-5800
1	NAVAL AIR SYSTEMS CMD J THOMPSON 48142 SHAW RD UNIT 5 PATUXENT RIVER MD 20670
1	NAVAL SURFACE WARFARE CTR DAHLGREN DIV CODE G06 DAHLGREN VA 22448
1	NAVAL SURFACE WARFARE CTR TECH LIBRARY CODE 323 17320 DAHLGREN RD DAHLGREN VA 22448
3	NAVAL RESEARCH LAB I WOLOCK CODE 6383 R BADALIANCE CODE 6304 L GAUSE WASHINGTON DC 20375
1	NAVAL SURFACE WARFARE CTR CRANE DIVISION M JOHNSON CODE 20H4 LOUISVILLE KY 40214-5245
2	COMMANDER NAVAL SURFACE WARFARE CTR CADEROCK DIVISION R PETERSON CODE 2020 M CRITCHFIELD CODE 1730 BETHESDA MD 20084
2	NAVAL SURFACE WARFARE CTR U SORATHIA C WILLIAMS CD 6551 9500 MACARTHUR BLVD WEST BETHESDA MD 20817

<u>NO. OF COPIES</u>	<u>ORGANIZATION</u>	<u>NO. OF COPIES</u>	<u>ORGANIZATION</u>
1	DAVID TAYLOR RESEARCH CTR SHIP STRUCTURES & PROTECTION DEPARTMENT CODE 1702 J CORRADO BETHESDA MD 20084	1	NAVSEA OJRI PEO DD21 PMS500 G CAMPONESCHI 2351 JEFFERSON DAVIS HWY ARLINGTON VA 22242-5165
2	DAVID TAYLOR RESEARCH CTR R ROCKWELL W PHYLLAIER BETHESDA MD 20054-5000	1	EXPEDITIONARY WARFARE DIV N85 F SHOUP 2000 NAVY PENTAGON WASHINGTON DC 20350-2000
1	OFFICE OF NAVAL RESEARCH D SIEGEL CODE 351 800 N QUINCY ST ARLINGTON VA 22217-5660	1	AFRL MLBC 2941 P STREET RM 136 WRIGHT PATTERSON AFB OH 45433-7750
8	NAVAL SURFACE WARFARE CTR J FRANCIS CODE G30 D WILSON CODE G32 R D COOPER CODE G32 J FRAYSSE CODE G33 E ROWE CODE G33 T DURAN CODE G33 L DE SIMONE CODE G33 R HUBBARD CODE G33 DAHLGREN VA 22448	1	AFRL MLSS R THOMSON 2179 12TH STREET RM 122 WRIGHT PATTERSON AFB OH 45433-7718
1	NAVAL SEA SYSTEMS CMD D LIESE 2531 JEFFERSON DAVIS HIGHWAY ARLINGTON VA 22242-5160	2	AFRL F ABRAMS J BROWN BLDG 653 2977 P STREET STE 6 WRIGHT PATTERSON AFB OH 45433-7739
1	NAVAL SURFACE WARFARE M LACY CODE B02 17320 DAHLGREN RD DAHLGREN VA 22448	1	AFRL MLS OL 7278 4TH STREET BLDG 100 BAY D L COULTER HILL AFB UT 84056-5205
1	OFFICE OF NAVAL RES J KELLY 800 NORTH QUINCEY ST ARLINGTON VA 22217-5000	1	OSD JOINT CCD TEST FORCE OSD JCCD R WILLIAMS 3909 HALLS FERRY RD VICKSBURG MS 29180-6199
2	NAVAL SURFACE WARFARE CTR CARDEROCK DIVISION R CRANE CODE 2802 C WILLIAMS CODE 6553 3A LEGGETT CIR BETHESDA MD 20054-5000	1	DEFENSE NUCLEAR AGENCY INNOVATIVE CONCEPTS DIV R ROHR 6801 TELEGRAPH RD ALEXANDRIA VA 22310-3398

<u>NO. OF COPIES</u>	<u>ORGANIZATION</u>
1	WATERWAYS EXPERIMENT D SCOTT 3909 HALLS FERRY RD SC C VICKSBURG MS 39180
3	DARPA M VANFOSSEN S WAX L CHRISTODOULOU 3701 N FAIRFAX DR ARLINGTON VA 22203-1714
2	SERDP PROGRAM OFC PM P2 C PELLERIN B SMITH 901 N STUART ST SUITE 303 ARLINGTON VA 22203
1	FAA MIL HDBK 17 CHAIR L ILCEWICZ 1601 LIND AVE SW ANM 115N RENTON VA 98055
2	FAA TECH CENTER D OPLINGER AAR 431 P SHYPRYKEVICH AAR 431 ATLANTIC CITY NJ 08405
1	OFC OF ENVIRONMENTAL MGMT U S DEPT OF ENERGY P RITZCOVAN 19901 GERMANTOWN RD GERMANTOWN MD 20874-1928
1	LOS ALAMOS NATL LAB F ADDESSIO MS B216 PO BOX 1633 LOS ALAMOS NM 87545
1	OAK RIDGE NATL LAB R M DAVIS PO BOX 2008 OAK RIDGE TN 37831-6195

<u>NO. OF COPIES</u>	<u>ORGANIZATION</u>
5	DIRECTOR LAWRENCE LIVERMORE NATL LAB R CHRISTENSEN S DETERESA F MAGNESS M FINGER MS 313 M MURPHY L 282 PO BOX 808 LIVERMORE CA 94550
7	NIST R PARNAS J DUNKERS M VANLANDINGHAM MS 8621 J CHIN MS 8621 D HUNSTON MS 8543 J MARTIN MS 8621 D DUTHINH MS 8611 100 BUREAU DR GAITHERSBURG MD 20899
1	OAK RIDGE NATL LAB C EBERLE MS 8048 PO BOX 2009 OAK RIDGE TN 37831
1	OAK RIDGE NATL LAB C D WARREN MS 8039 PO BOX 2009 OAK RIDGE TN 37922
4	DIRECTOR SANDIA NATL LABS APPLIED MECHANICS DEPT DIVISION 8241 W KAWAHARA K PERANO D DAWSON P NIELAN PO BOX 969 LIVERMORE CA 94550-0096
1	LAWRENCE LIVERMORE NATIONAL LAB M MURPHY PO BOX 808 L 282 LIVERMORE CA 94550

<u>NO. OF COPIES</u>	<u>ORGANIZATION</u>	<u>NO. OF COPIES</u>	<u>ORGANIZATION</u>
3	NASA LANGLEY RESEARCH CTR MS 266 AMSRL VS W ELBER F BARTLETT JR G FARLEY HAMPTON VA 23681-0001	1	GRAPHITE MASTERS INC J WILLIS 3815 MEDFORD ST LOS ANGELES CA 90063-1900
1	NASA LANGLEY RESEARCH CTR T GATES MS 188E HAMPTON VA 23661-3400	1	ADVANCED GLASS FIBER YARNS T COLLINS 281 SPRING RUN LN STE A DOWNINGTON PA 19335
1	USDOT FEDERAL RAILROAD RDV 31 M FATEH WASHINGTON DC 20590	1	COMPOSITE MATERIALS INC D SHORTT 19105 63 AVE NE PO BOX 25 ARLINGTON WA 98223
1	DOT FHWA J SCALZI 400 SEVENTH ST SW 3203 HNG 32 WASHINGTON DC 20590	1	COMPOSITE MATERIALS INC R HOLLAND 11 JEWEL COURT ORINDA CA 94563
1	FHWA E MUNLEY 6300 GEORGETOWN PIKE MCLEAN VA 22101	1	COMPOSITE MATERIALS INC C RILEY 14530 S ANSON AVE SANTA FE SPRINGS CA 90670
1	CENTRAL INTELLIGENCE AGENCY OTI WDAG GT W L WALTMAN PO BOX 1925 WASHINGTON DC 20505	2	COMPOSIX D BLAKE L DIXON 120 O NEILL DR HEBRUN OHIO 43025
1	MARINE CORPS INTEL ACTY D KOSITZKE 3300 RUSSELL RD SUITE 250 QUANTICO VA 22134-5011	4	CYTEC FIBERITE R DUNNE D KOHLI M GILLIO R MAYHEW 1300 REVOLUTION ST HAVRE DE GRACE MD 21078
1	NATL GRND INTELLIGENCE CTR DIRECTOR IANG TMT 220 SEVENTH ST NE CHARLOTTESVILLE VA 22902-5396	2	SIMULA J COLTMAN R HUYETT 10016 S 51ST ST PHOENIX AZ 85044
1	DIRECTOR DEFENSE INTELLIGENCE AGENCY TA 5 K CRELLING WASHINGTON DC 20310	1	SIOUX MFG B KRIEL PO BOX 400 FT TOTTEN ND 58335

<u>NO. OF COPIES</u>	<u>ORGANIZATION</u>
2	PROTECTION MATERIALS INC M MILLER F CRILLEY 14000 NW 58 CT MIAMI LAKES FL 33014
3	FOSTER MILLER J J GASSNER M ROYLANCE W ZUKAS 195 BEAR HILL RD WALTHAM MA 02354-1196
1	ROM DEVELOPMENT CORP R O MEARA 136 SWINEBURNE ROW BRICK MARKET PLACE NEWPORT RI 02840
2	TEXTRON SYSTEMS T FOLTZ M TREASURE 201 LOWELL ST WILMINGTON MA 08870-2941
1	JPS GLASS L CARTER PO BOX 260 SLATER RD SLATER SC 29683
1	O GARA HESS & EISENHARDT M GILLESPIE 9113 LESAINTE DR FAIRFIELD OH 45014
2	MILLIKEN RESEARCH CORP H KUHN M MACLEOD PO BOX 1926 SPARTANBURG SC 29303
1	CONNEAUGHT INDUSTRIES INC J SANTOS PO BOX 1425 COVENTRY RI 02816

<u>NO. OF COPIES</u>	<u>ORGANIZATION</u>
1	BATTELLE C R HARGREAVES 505 KING AVE COLUMBUS OH 43201-2681
2	BATTELLE NATICK OPERATIONS J CONNORS B HALPIN 209 W CENTRAL ST STE 302 NATICK MA 01760
1	BATTELLE NW DOE PNNL T HALL MS K231 BATTELLE BLVD RICHLAND WA 99352
3	PACIFIC NORTHWEST LAB M SMITH G VAN ARSDALE R SHIPPELL PO BOX 999 RICHLAND WA 99352
1	ARMTEC DEFENSE PRODUCTS S DYER 85 901 AVE 53 PO BOX 848 COACHELLA CA 92236
2	ADVANCED COMPOSITE MATLS CORP P HOOD J RHODES 1525 S BUNCOMBE RD GREER SC 29651-9208
2	GLCC INC J RAY M BRADLEY 103 TRADE ZONE DR STE 26C WEST COLUMBIA SC 29170
2	AMOCO PERFORMANCE PRODUCTS M MICHNO JR J BANISAUKAS 4500 MCGINNIS FERRY RD ALPHARETTA GA 30202-3944

<u>NO. OF COPIES</u>	<u>ORGANIZATION</u>
1	SAIC M PALMER 2109 AIR PARK RD S E ALBUQUERQUE NM 87106
1	SAIC ATTN G CHRYSSOMALLIS 3800 W 80TH ST STE 1090 BLOOMINGTON MN 55431
1	AAI CORPORATION DR T G STASTNY PO BOX 126 HUNT VALLEY MD 21030-0126
1	JOHN HEBERT PO BOX 1072 HUNT VALLEY MD 21030-0126
12	ALLIANT TECHSYSTEMS INC C CANDLAND C AAKHUS R BECKER B SEE N VLAHAKUS R DOHRN S HAGLUND D FISHER W WORRELL R COPENHAFER M HISSONG D KAMDAR 600 2ND ST NE HOPKINS MN 55343-8367
3	ALLIANT TECHSYSTEMS INC J CONDON E LYNAM J GERHARD WV01 16 STATE RT 956 PO BOX 210 ROCKET CENTER WV 26726-0210
1	APPLIED COMPOSITES W GRISCH 333 NORTH SIXTH ST ST CHARLES IL 60174

<u>NO. OF COPIES</u>	<u>ORGANIZATION</u>
1	PROJECTILE TECHNOLOGY INC 515 GILES ST HAVRE DE GRACE MD 21078
1	CUSTOM ANALYTICAL ENG SYS INC A ALEXANDER 13000 TENSOR LN NE FLINTSTONE MD 21530
2	LORAL VOUGHT SYSTEMS G JACKSON K COOK 1701 W MARSHALL DR GRAND PRAIRIE TX 75051
5	AEROJET GEN CORP D PILLASCH T COULTER C FLYNN D RUBAREZUL M GREINER 1100 WEST HOLLYVALE ST AZUSA CA 91702-0296
3	HEXCEL INC R BOE F POLICELLI J POESCH PO BOX 98 MAGNA UT 84044
3	HERCULES INC G KUEBELER J VERMEYCHUK B MANDERVILLE JR HERCULES PLAZA WILMINGTON DE 19894
1	BRIGS COMPANY J BACKOFEN 2668 PETERBOROUGH ST HERDON VA 22071-2443
1	ZERNOW TECHNICAL SERVICES L ZERNOW 425 W BONITA AVE STE 208 SAN DIMAS CA 91773

NO. OF
COPIES ORGANIZATION

NO. OF
COPIES ORGANIZATION

2 OLIN CORPORATION
FLINCHBAUGH DIV
E STEINER
B STEWART
PO BOX 127
RED LION PA 17356

1 BOEING
R BOHLMANN
PO BOX 516 MC 5021322
ST LOUIS MO 63166-0516

1 OLIN CORPORATION
L WHITMORE
10101 9TH ST NORTH
ST PETERSBURG FL 33702

2 BOEING DEFENSE
AND SPACE GRP
W HAMMOND
J RUSSELL
S 4X55
PO BOX 3707
SEATTLE WA 98124-2207

1 DOW UT
S TIDRICK
15 STERLING DR
WALLINGFORD CT 06492

2 BOEING ROTORCRAFT
P MINGURT
P HANDEL
800 B PUTNAM BLVD
WALLINGFORD PA 19086

5 SIKORSKY AIRCRAFT
G JACARUSO
T CARSTENSAN
B KAY
S GARBO M S S330A
J ADELMANN
6900 MAIN ST
PO BOX 9729
STRATFORD CT 06497-9729

1 BOEING
DOUGLAS PRODUCTS DIV
L J HART SMITH
3855 LAKEWOOD BLVD
D800 0019
LONG BEACH CA 90846-0001

1 PRATT & WHITNEY
D HAMBRICK
400 MAIN ST MS 114 37
EAST HARTFORD CT 06108

1 LOCKHEED MARTIN
S REEVE
8650 COBB DR
D 73 62 MZ 0648
MARIETTA GA 30063-0648

1 AEROSPACE CORP
G HAWKINS M4 945
2350 E EL SEGUNDO BLVD
EL SEGUNDO CA 90245

1 LOCKHEED MARTIN
SKUNK WORKS
D FORTNEY
1011 LOCKHEED WAY
PALMDALE CA 93599-2502

2 CYTEC FIBERITE
M LIN
W WEB
1440 N KRAEMER BLVD
ANAHEIM CA 92806

1 LOCKHEED MARTIN
R FIELDS
1195 IRWIN CT
WINTER SPRINGS FL 32708

1 HEXCEL
T BITZER
11711 DUBLIN BLVD
DUBLIN CA 94568

1 MATERIALS SCIENCES CORP
B W ROSEN
500 OFFICE CENTER DR STE 250
FORT WASHINGTON PA 19034

<u>NO. OF COPIES</u>	<u>ORGANIZATION</u>
1	NORTHROP GRUMMAN CORP ELECTRONIC SENSORS & SYSTEMS DIV E SCHOCH 1745A WEST NURSERY RD MAILSTOP V 16 LINTHICUM MD 21090
2	NORTHROP GRUMMAN ENVIRONMENTAL PROGRAMS R OSTERMAN A YEN 8900 E WASHINGTON BLVD PICO RIVERA CA 90660
1	UNITED DEFENSE LP PO BOX 359 D MARTIN SANTA CLARA CA 95052
1	UNITED DEFENSE LP PO BOX 58123 G THOMAS SANTA CLARA CA 95052
2	UNITED DEFENSE LP MAIL DROP M53 R BARRETT V HORVATICH 328 W BROKAW RD SANTA CLARA CA 95052-0359
3	UNITED DEFENSE LP GROUND SYSTEMS DIVISION M PEDRAZZI MAIL DROP N09 A LEE MAIL DROP N11 M MACLEAN MAIL DROP N06 1205 COLEMAN AVE SANTA CLARA CA 95052
4	UNITED DEFENSE LP 4800 EAST RIVER RD R BRYNSVOLD P JANKE MS170 T GIOVANETTI MS236 B VAN WYK MS389 MINNEAPOLIS MN 55421-1498

<u>NO. OF COPIES</u>	<u>ORGANIZATION</u>
2	GENERAL DYNAMICS LAND SYSTEMS D REES M PASIK PO BOX 2074 WARREN MI 48090-2074
1	GENERAL DYNAMICS LAND SYSTEMS DIVISION D BARTLE PO BOX 1901 WARREN MI 48090
1	GENERAL DYNAMICS LAND SYSTEMS MUSKEGON OPERATIONS W SOMMERS JR 76 GETTY ST MUSKEGON MI 49442
1	GENERAL DYNAMICS AMPHIBIOUS SYS SURVIVABILITY LEAD G WALKER 991 ANNAPOLIS WAY WOODBIDGE VA 22191
5	INSTITUTE FOR ADVANCED TECH T KIEHNE H FAIR P SULLIVAN W REINECKE I MCNAB 4030 2 W BRAKER LN AUSTIN TX 78759
2	CIVIL ENGR RSCH FOUNDATION H BERNSTEIN PRESIDENT R BELLE 1015 15TH ST NW STE 600 WASHINGTON DC 20005
1	ARROW TECH ASSO 1233 SHELBURNE RD STE D 8 SOUTH BURLINGTON VT 05403-7700

<u>NO. OF COPIES</u>	<u>ORGANIZATION</u>	<u>NO. OF COPIES</u>	<u>ORGANIZATION</u>
1	CONSULTANT R EICHELBERGER 409 W CATHERINE ST BEL AIR MD 21014-3613	1	UNIVERSITY OF UTAH DEPT OF MECH & INDUSTRIAL ENGR S SWANSON SALT LAKE CITY UT 84112
1	UCLA MANE DEPT ENGR IV H THOMAS HAHN LOS ANGELES CA 90024-1597	2	PENNSYLVANIA STATE UNIV R MCNITT C BAKIS 227 HAMMOND BLDG UNIVERSITY PARK PA 16802
2	U OF DAYTON RESEARCH INSTUTE RAN Y KIM AJIT K ROY 300 COLLEGE PARK AVE DAYTON OH 45469-0168	1	PENNSYLVANIA STATE UNIV RENATA S ENGEL 245 HAMMOND BLDG UNIVERSITY PARK PA 16801
1	MIT P LAGACE 77 MASS AVE CAMBRIDGE MA 01887	1	PURDUE UNIVERSITY SCHOOL OF AERO & ASTRO C T SUN W LAFAYETTE IN 47907-1282
1	IIT RESEARCH CENTER D ROSE 201 MILL ST ROME NY 13440-6916	1	STANFORD UNIVERSITY DEPARTMENT OF AERONAUTICS AND AEROBALLISTICS DURANT BUILDING S TSAI STANFORD CA 94305
1	GEORGIA TECH RESEARCH INSTITUTE GEORGIA INSTITUTE OF TECHNOLOGY P FRIEDERICH ATLANTA GA 30392	1	UNIVERSITY OF DAYTON J M WHITNEY COLLEGE PARK AVE DAYTON OH 45469-0240
1	MICHIGAN ST UNIVERSITY R AVERILL 3515 EB MSM DEPT EAST LANSING MI 48824-1226	7	UNIVERSITY OF DELAWARE CTR FOR COMPOSITE MATERIALS J GILLESPIE M SANTARE G PALMESE S YARLAGADDA S ADVANI D HEIDER D KUKICH 201 SPENCER LABORATORY NEWARK DE 19716
1	UNIVERSITY OF KENTUCKY LYNN PENN 763 ANDERSON HALL LEXINGTON KY 40506-0046		
1	UNIVERSITY OF WYOMING D ADAMS PO BOX 3295 LARAMIE WY 82071		

<u>NO. OF</u> <u>COPIES</u>	<u>ORGANIZATION</u>	<u>NO. OF</u> <u>COPIES</u>	<u>ORGANIZATION</u>
1	UNIVERSITY OF ILLINOIS AT URBANA CHAMPAIGN NATL CENTER FOR COMPOSITE MATERIALS RESEARCH 216 TALBOT LABORATORY J ECONOMY 104 S WRIGHT ST URBANA IL 61801		<u>ABERDEEN PROVING GROUND</u>
		1	COMMANDER US ARMY MATERIEL SYS ANALYSIS P DIETZ 392 HOPKINS RD AMXSY TD APG MD 21005-5071
3	THE UNIVERSITY OF TEXAS AT AUSTIN CENTER FOR ELECTROMECHANICS J PRICE A WALLS J KITZMILLER 10100 BURNET RD AUSTIN TX 78758-4497	1	DIRECTOR US ARMY RESEARCH LAB AMSRL OP AP L APG MD 21005 5066
		115	DIR USARL AMSRL CI AMSRL CI H W STUREK AMSRL CI S A MARK AMSRL CS IO FI M ADAMSON AMSRL SL B J SMITH AMSRL SL BA AMSRL SL BL D BELY R HENRY AMSRL SL BG A YOUNG AMSRL SL I AMSRL WM B A HORST E SCHMIDT AMSRL WM BA W D AMICO F BRANDON AMSRL WM BC P PLOSTINS D LYON J NEWILL S WILKERSON A ZIELINSKI AMSRL WM BD B FORCH R FIFER R PESCE RODRIGUEZ B RICE
3	VA POLYTECHNICAL INSTITUTE & STATE UNIVERSITY DEPT OF ESM M W HYER K REIFSNIDER R JONES BLACKSBURG VA 24061-0219		
1	NORTH CAROLINA STATE UNIVERSITY CIVIL ENGINEERING DEPT W RASDORF PO BOX 7908 RALEIGH NC 27696-7908		
1	UNIVERSITY OF MARYLAND DEPT OF AEROSPACE ENGINEERING ANTHONY J VIZZINI COLLEGE PARK MD 20742		
1	DREXEL UNIVERSITY ALBERT S D WANG 32ND AND CHESTNUT STREETS PHILADELPHIA PA 19104		
1	SOUTHWEST RSCH INSTITUTE ENGR & MATL SCIENCES DIV J RIEGEL 6220 CULEBRA RD PO DRAWER 28510 SAN ANTONIO TX 78228-0510		

NO. OF
COPIES ORGANIZATION

ABERDEEN PROVING GROUND (CONT)

AMSRL WM BE
G WREN
C LEVERITT
D KOOKER
AMSRL WM BR
C SHOEMAKER
J BORNSTEIN
AMSRL WM M
D VIECHNICKI
G HAGNAUER
J MCCAULEY
B TANNER
AMSRL WM MA
R SHUFORD
P TOUCHET
N BECK TAN
D FLANAGAN
L GHIORSE
D HARRIS
S MCKNIGHT
P MOY
S NGYUEN
P PATTERSON
G RODRIGUEZ
A TEETS
R YIN
AMSRL WM MB
B FINK
J BENDER
T BLANAS
T BOGETTI
R BOSSOLI
L BURTON
K BOYD
S CORNELISON
P DEHMER
R DOOLEY
W DRYSDALE
G GAZONAS
S GHIORSE
D GRANVILLE
D HOPKINS
C HOPPEL
D HENRY
R KASTE
M KLUSEWITZ
M LEADORE
R LIEB

NO. OF
COPIES ORGANIZATION

ABERDEEN PROVING GROUND (CONT)

AMSRL WM MB
E RIGAS
J SANDS
D SPAGNUOLO
W SPURGEON
J TZENG
E WETZEL
A ABRAHAMIAN
M BERMAN
A FRYDMAN
T LI
W MCINTOSH
E SZYMANSKI
AMRSL WM MC
J BEATTY
J SWAB
E CHIN
J MONTGOMERY
A WERESCZCAK
J LASALVIA
J WELLS
AMSRL WM MD
W ROY
S WALSH
AMSRL WM T
B BURNS
AMSRL WM TA
W GILLICH
T HAVEL
J RUNYEON
M BURKINS
E HORWATH
B GOOCH
W BRUCHEY
AMSRL WM TC
R COATES
AMSRL WM TD
A DAS GUPTA
T HADUCH
T MOYNIHAN
F GREGORY
A RAJENDRAN
M RAFTENBERG
M BOTELER
T WEERASOORIYA
D DANDEKAR
A DIETRICH

NO. OF
COPIES ORGANIZATION

ABERDEEN PROVING GROUND (CONT)

AMSRL WM TE
 A NIILER
 J POWELL
AMSRL SS SD
 H WALLACE
AMSRL SS SE R
 R CHASE
AMSRL SS SE DS
 R REYZER
 R ATKINSON
AMSRL SE L
 R WEINRAUB
 J DESMOND
 D WOODBURY

NO. OF
COPIES ORGANIZATION

1 R MARTIN
MERL
LTD
TAMWORTH RD
HERTFORD SG13 7DG
UK

1 PW LAY
SMC SCOTLAND
DERA ROSYTH
ROSYTH ROYAL DOCKYARD
DUNFERMLINE FIFE KY 11 2XR
UK

1 T GOTTESMAN
CIVIL AVIATION ADMINISTRATION
PO BOX 8
BEN GURION INTERNL AIRPORT
LOD 70150 ISRAEL

1 S ANDRE
AEROSPATIALE
A BTE CC RTE MD132
316 ROUTE DE BAYONNE
TOULOUSE 31060
FRANCE

1 J BAUER
DAIMLER BENZ AEROSPACE
D 81663 MUNCHEN
MUNICH
GERMANY

3 DRA FORT HALSTEAD
PETER N JONES
DAVID SCOTT
MIKE HINTON
SEVEN OAKS KENT TN 147BP
UNITED KINGDOM

1 MR FRANCOIS LESAGE
DEFENSE RESEARCH ESTAB
VALCARTIER
PO BOX 8800
COURCELETTE QUEBEC COA
IRO CANADA

NO. OF
COPIES ORGANIZATION

2 ROYAL MILITARY COLLEGE OF
SCIENCE SHRIVENHAM
D BULMAN
B LAWTON
SWINDON WILTS SN6 8LA
UNITED KINGDOM

1 SWISS FEDERAL ARMAMENTS
WKS
WALTER LANZ
ALLMENDSTRASSE 86
3602 THUN
SWITZERLAND

1 PROFESSOR SOL BODNER
ISRAEL INSTITUTE OF
TECHNOLOGY
FACULTY OF MECHANICAL ENGR
HAIFA 3200 ISRAEL

1 DSTO MATERIALS RSRCH LAB
DR NORBERT BURMAN NAVAL
PLATFORM VULNERABILITY SHIP
STRUCTURES & MATERIALS DIV
PO BOX 50
ASCOT VALE VICTORIA
AUSTRALIA 3032

1 PROFESSOR EDWARD CELENS
ECOLE ROYAL MILITAIRE
AVE DE LA RENAISSANCE 30
1040 BRUXELLE
BELGIQUE

1 DEF RES ESTABLISHMENT
VALCARTIER
ALAIN DUPUIS
2459 BOULEVARD PIE XI NORTH
VALCARTIER QUEBEC
CANADA
PO BOX 8800 COURCELETTE
GOA IRO QUEBEC CANADA

<u>NO. OF COPIES</u>	<u>ORGANIZATION</u>
1	INSTITUT FRANCO ALLEMAND DE RECHERCHES DE SAINT LOUIS DE MARC GIRAUD RUE DU GENERAL CASSAGNOU BOITE POSTALE 34 F 68301 SAINT LOUIS CEDEX FRANCE
1	J MANSON ECOLE POLYTECH DMX LTC CH 1015 LAUSANNE SWITZERLAND
1	TNO PRINS MAURITS LAB DR ROB IJSSELSTEIN LANGE KLEIWEG 137 PO BOX 45 2280 AA RIJSWIJK THE NETHERLANDS
2	FOA NAT L DEFENSE RESEARCH ESTAB DR BO JANZON R HOLMLIN DIR DEPT OF WEAPONS & PROTECTION S 172 90 STOCKHOLM SWEDEN
2	DEFENSE TECH & PROC AGENCY GRND MR I CREWETHER GENERAL HERZOG HAUS 3602 THUN SWITZERLAND
1	MINISTRY OF DEFENCE RAFAEL DR MEIR MAYSELESS ARMAMENT DEVELOPMENT AUTH PO BOX 2250 HAIFA 31021 ISRAEL
1	DR AKE PERSSON DYNAMEC RESEARCH AB PARADISGRND 7 S 151 36 SODERTALJE SWEDEN

<u>NO. OF COPIES</u>	<u>ORGANIZATION</u>
1	ERNST MACH INSTITUT EMI DIRECTOR HAUPTSTRASSE 18 79576 WEIL AM RHEIN GERMANY
1	ERNST MACH INSTITUT EMI DR ALOIS STILP ECKERSTRASSE 4 7800 FREIBURG GERMANY
1	DR IR HANS PASMAN TNO DEFENSE RESEARCH POSTBUS 6006 2600 JA DELFT THE NETHERLANDS
1	DR BITAN HIRSCH TACHKEMONY ST 6 NETAMUA 42611 ISRAEL
1	PROF DR MANFRED HELD DEUTSCHE AEROSPACE AG DYNAMICS SYSTEMS PO BOX 1340 D 86523 SCHROBENHAUSEN GERMANY

REPORT DOCUMENTATION PAGE			Form Approved OMB No. 0704-0188	
Public reporting burden for this collection of information is estimated to average 1 hour per response, including the time for reviewing instructions, searching existing data sources, gathering and maintaining the data needed, and completing and reviewing the collection of information. Send comments regarding this burden estimate or any other aspect of this collection of information, including suggestions for reducing this burden, to Washington Headquarters Services, Directorate for Information Operations and Reports, 1215 Jefferson Davis Highway, Suite 1204, Arlington, VA 22202-4302, and to the Office of Management and Budget, Paperwork Reduction Project(0704-0188), Washington, DC 20503.				
1. AGENCY USE ONLY (Leave blank)		2. REPORT DATE June 2000	3. REPORT TYPE AND DATES COVERED Final, January 1996 - December 1996	
4. TITLE AND SUBTITLE On the Influence of Moisture on Dielectric Properties of Polyetheretherketone (PEEK) Carbon-Fiber Composites			5. FUNDING NUMBERS AH42	
6. AUTHOR(S) Bruce K. Fink, Roy L. McCullough,* and John W. Gillespie Jr.*				
7. PERFORMING ORGANIZATION NAME(S) AND ADDRESS(ES) U.S. Army Research Laboratory ATTN: AMSRL-WM-MB Aberdeen Proving Ground, MD 21005-5069			8. PERFORMING ORGANIZATION REPORT NUMBER ARL-TR-2236	
9. SPONSORING/MONITORING AGENCY NAMES(S) AND ADDRESS(ES)			10. SPONSORING/MONITORING AGENCY REPORT NUMBER	
11. SUPPLEMENTARY NOTES *University of Delaware, Newark, DE 19716				
12a. DISTRIBUTION/AVAILABILITY STATEMENT Approved for public release; distribution is unlimited.			12b. DISTRIBUTION CODE	
13. ABSTRACT (Maximum 200 words) An analysis of local and global mechanisms of heat generation and distribution in carbon-fiber-based composites subjected to an alternating magnetic field has shown that heating is dependent upon the dielectric properties of the polymer matrix. These properties were investigated as functions of temperature, frequency, and moisture content. The results indicate little dependence of the dielectric constant on temperature or frequency, while the loss tangent exhibited a substantial dependence on both frequency and temperature. A substantial dependence of loss tangent on moisture content in polyetheretherketone (PEEK) was found.				
14. SUBJECT TERMS carbon, composites, dielectric PEEK, moisture induction			15. NUMBER OF PAGES 48	
			16. PRICE CODE	
17. SECURITY CLASSIFICATION OF REPORT UNCLASSIFIED	18. SECURITY CLASSIFICATION OF THIS PAGE UNCLASSIFIED	19. SECURITY CLASSIFICATION OF ABSTRACT UNCLASSIFIED	20. LIMITATION OF ABSTRACT UL	

INTENTIONALLY LEFT BLANK.

USER EVALUATION SHEET/CHANGE OF ADDRESS

This Laboratory undertakes a continuing effort to improve the quality of the reports it publishes. Your comments/answers to the items/questions below will aid us in our efforts.

1. ARL Report Number/Author ARL-TR-2236 (Fink) Date of Report June 2000

2. Date Report Received _____

3. Does this report satisfy a need? (Comment on purpose, related project, or other area of interest for which the report will be used.) _____

4. Specifically, how is the report being used? (Information source, design data, procedure, source of ideas, etc.) _____

5. Has the information in this report led to any quantitative savings as far as man-hours or dollars saved, operating costs avoided, or efficiencies achieved, etc? If so, please elaborate. _____

6. General Comments. What do you think should be changed to improve future reports? (Indicate changes to organization, technical content, format, etc.) _____

CURRENT ADDRESS

 Organization

 Name E-mail Name

 Street or P.O. Box No.

 City, State, Zip Code

7. If indicating a Change of Address or Address Correction, please provide the Current or Correct address above and the Old or Incorrect address below.

OLD ADDRESS

 Organization

 Name

 Street or P.O. Box No.

 City, State, Zip Code

(Remove this sheet, fold as indicated, tape closed, and mail.)
(DO NOT STAPLE)

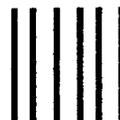
DEPARTMENT OF THE ARMY

OFFICIAL BUSINESS

BUSINESS REPLY MAIL
FIRST CLASS PERMIT NO 0001,APG,MD

POSTAGE WILL BE PAID BY ADDRESSEE

DIRECTOR
US ARMY RESEARCH LABORATORY
ATTN AMSRL WM MB
ABERDEEN PROVING GROUND MD 21005-5069



NO POSTAGE
NECESSARY
IF MAILED
IN THE
UNITED STATES

