

2

AD-A250 611



**CHEMICAL
RESEARCH,
-DEVELOPMENT &
ENGINEERING
CENTER**

CRDEC-TR-334

**FENTANYL SYNTHETIC METHODOLOGY:
A COMPARATIVE STUDY**

**Fu-Lian Hsu
Harold D. Banks**

RESEARCH DIRECTORATE

**DTIC
ELECTE
MAY 19 1992**

March 1992

Approved for public release; distribution is unlimited.



**U.S. ARMY
ARMAMENT
MUNITIONS
CHEMICAL COMMAND**

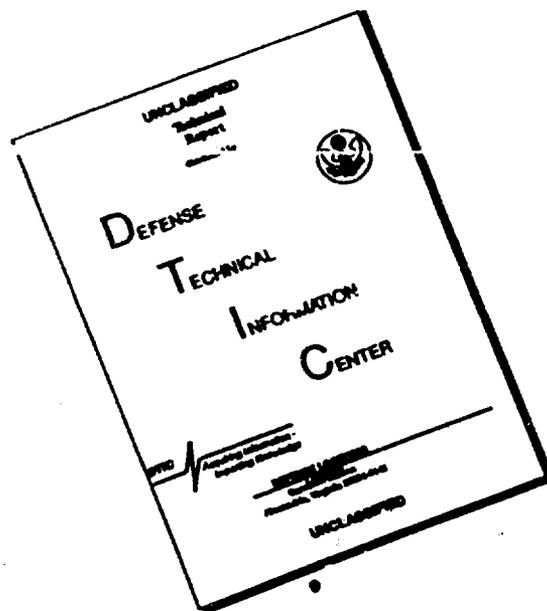
Aberdeen Proving Ground, Maryland 21010-5423

92-13322



92 5 18 139

DISCLAIMER NOTICE



THIS DOCUMENT IS BEST QUALITY AVAILABLE. THE COPY FURNISHED TO DTIC CONTAINED A SIGNIFICANT NUMBER OF PAGES WHICH DO NOT REPRODUCE LEGIBLY.

Disclaimer

The findings in this report are not to be construed as an official Department of the Army position unless so designated by other authorizing documents.

REPORT DOCUMENTATION PAGE

Form Approved
OMB No. 0704-0188

Public reporting burden for this collection of information is estimated to average 1 hour per response, including the time for reviewing instructions, searching existing data sources, gathering and maintaining the data needed, and completing and reviewing the collection of information. Send comments regarding this burden estimate or any other aspect of this collection of information, including suggestions for reducing this burden, to Washington Headquarters Services, Directorate for Information Operations and Reports, 1215 Jefferson Davis Highway, Suite 1204, Arlington, VA 22202-4302, and to the Office of Management and Budget, Paperwork Reduction Project (0704-0188), Washington, DC 20503

1. AGENCY USE ONLY (Leave blank)	2. REPORT DATE 1992 March	3. REPORT TYPE AND DATES COVERED Final, 88 Sep - 91 Dec	
4. TITLE AND SUBTITLE Fentanyl Synthetic Methodology: A Comparative Study		5. FUNDING NUMBERS PR-1C161102A71A	
6. AUTHOR(S) Hsu, Fu-Lian, and Banks, Harold D.			
7. PERFORMING ORGANIZATION NAME(S) AND ADDRESS(ES) CDR, CRDEC, ATTN: SMCCR-RSC-O, APG, MD 21010-5423		8. PERFORMING ORGANIZATION REPORT NUMBER CRDEC-TR-334	
9. SPONSORING / MONITORING AGENCY NAME(S) AND ADDRESS(ES)		10. SPONSORING / MONITORING AGENCY REPORT NUMBER	
11. SUPPLEMENTARY NOTES			
12a. DISTRIBUTION / AVAILABILITY STATEMENT Approved for public release; distribution is unlimited.		12b. DISTRIBUTION CODE	
13. ABSTRACT (Maximum 200 words) A literature review has identified six promising methods for the synthesis of fentanyl. Those methods have been compared for overall efficiency. The best method to date is identified.			
14. SUBJECT TERMS Fentanyl Anilidopiperidine Synthesis			15. NUMBER OF PAGES 18
			16. PRICE CODE
17. SECURITY CLASSIFICATION OF REPORT UNCLASSIFIED	18. SECURITY CLASSIFICATION OF THIS PAGE UNCLASSIFIED	19. SECURITY CLASSIFICATION OF ABSTRACT UNCLASSIFIED	20. LIMITATION OF ABSTRACT UL

Blank

PREFACE

The work described in this report was authorized under Project No. 1C161102A71A, Research in CW/CB Defense. This work was started in September 1988 and completed in December 1991.

The use of trade names or manufacturers' names in this report does not constitute an official endorsement of any commercial products. This report may not be cited for purposes of advertisement.

Reproduction of this document in whole or in part is prohibited except with permission of the Commander, U.S. Army Chemical Research, Development and Engineering Center, ATTN: SMCCR-SPS-T, Aberdeen Proving Ground, MD 21010-5423. However, the Defense Technical Information Center and the National Technical Information Service are authorized to reproduce the document for U.S. Government purposes.

This report has been approved for release to the public.

3



Accession For	
NTIS GRA&I	<input checked="" type="checkbox"/>
DTIC TAB	<input type="checkbox"/>
Unannounced	<input type="checkbox"/>
Justification	
By _____	
Distribution/	
Availability Codes	
Dist	Avail and/or Special
A-1	

Blank

CONTENTS

	Page
1. INTRODUCTION	7
2. SYNTHETIC METHODS	9
3. CONCLUSION	15
LITERATURE CITED	17

Blank

FENTANYL SYNTHETIC METHODOLOGY: A COMPARATIVE STUDY

1. INTRODUCTION

The introduction of fentanyl in the 1960's by Janssen Pharmaceuticals was a momentous milestone in man's search for the ideal analgesic; i.e., a drug that provides relief from the excruciating pain of deep wounds or burns without unacceptable side effects. This search has been an endeavor of man from time immemorial. It has been known for at least five thousand years that the dried juice from the unripe seed pod of the opium poppy (*papaver somniferum*) is an extremely effective analgesic. Over the centuries, it was discovered that by carefully drying the juice from the unripe seed pod a useful preparation was produced that could be chewed directly or added to food. However, due to variation in potency depending on growing conditions, harvesting techniques and extraction procedures, the use of this natural product as a medicine in the modern sense of the term did not begin until the most active component was extracted by the German pharmacist, Serturner, in the first years of the nineteenth century¹. He named this pure compound morphine for its dream-producing properties. It should be noted that the French scientist, Derosne, came extremely close to isolating this compound even earlier², and that the experiments of these workers took place more than two decades before the birth of organic chemistry with Wohler's synthesis of urea in 1828.

Morphine was the first alkaloid to be isolated; alkaloids are nitrogen-containing bases derived from plants that often have profound physiological effects. Due to morphine's importance, vigorous attempts were made to elucidate its structure so that it or an even more effective drug could be prepared (hopefully inexpensively) in the laboratory. It soon became clear that unraveling the structure of morphine would be no simple task. Its molecular formula was determined to be $C_{17}H_{19}NO_3$, making for thousands of possible structures.

Before the days of modern spectroscopic techniques, structural assignments were made largely on the basis of degradation. Important structural features of morphine emerged, as did the relationships between morphine and other alkaloids such as thebaine and codeine. After an arduous struggle, spanning more than one century following its isolation, the

correct structure of morphine was deduced by Gulland and Robinson in 1923³. By the 1950's this structure was confirmed by total synthesis⁴ and X-ray crystallography. Morphine is agonizingly complex, with five rings and five centers of chirality. There are a total of sixteen possible stereoisomers. (Five chiral centers gives rise to a maximum of 2^5 stereoisomers; however, sixteen of these have impossible ring junctions.) In devising an efficient synthesis, even if structural and regioisomerism are successfully addressed, synthetic methodology providing the correct diastereo- and enantioselectivity must be employed. In spite of this formidable challenge, Kenner Rice at the NIH accomplished a total synthesis of the active enantiomer of morphine from commercially available 3-methoxyphenethylamine in 20 - 25% yield⁵.

While the structure of morphine was being elucidated, a significant discovery was made that resulted in the introduction of first synthetic opioid (not prepared from morphine). Meperidine, a 4-phenylpiperidine, appeared on the market in 1939.⁶ It would be satisfying to report that it was prepared after an ingenious intuition concerning those structural features of morphine that produce analgesic activity. In point of fact, it was synthesized as a possible antispasmodic, having an atropine-like structure. Its analgesic activity was discovered later, by accident. Once again serendipity propels science in the right direction. Once again the value of basic, as opposed to goal-directed research, is apparent.

Meperidine, 4-phenyl-4-ethoxycarbonyl-1-methylpiperidine is much simpler structurally than morphine. Prepared in reasonably good yield starting from nitrogen mustard, it has about 60% of morphine's analgesic potency in animal studies. The reversed ester of meperidine, i.e., the propionate ester of 4-phenyl-4-hydroxyl-1-methylpiperidine, was found to have about twice the potency of morphine in pharmacological tests with animals. This discovery spurred further modification of the 4-phenylpiperidine structure, culminating in the 1960's with the discovery of fentanyl, described as a "chemical congener" of the reversed ester.⁷ This appears to be a rather broad use of the term congener: Conversion of the 4-position of the reversed ester into that of fentanyl requires replacement of oxygen by nitrogen, followed by transposition of the phenyl group from C₄ to nitrogen. In any event, fentanyl, 4-[1-(2-phenylethyl)piperidinyl]-propanamide, is an extremely potent analgesic, having a potency of about 300 times that of morphine in animal tests. Although it is not the "ideal analgesic," retaining many of the side effects that compromise the use of morphine, it does have significantly lower cardiac depressive effects, and has found extensive use in cardiac surgical procedures.

With the recent expiration of the patent on fentanyl, the following timely report schematically presents six useful routes for the synthesis of fentanyl.

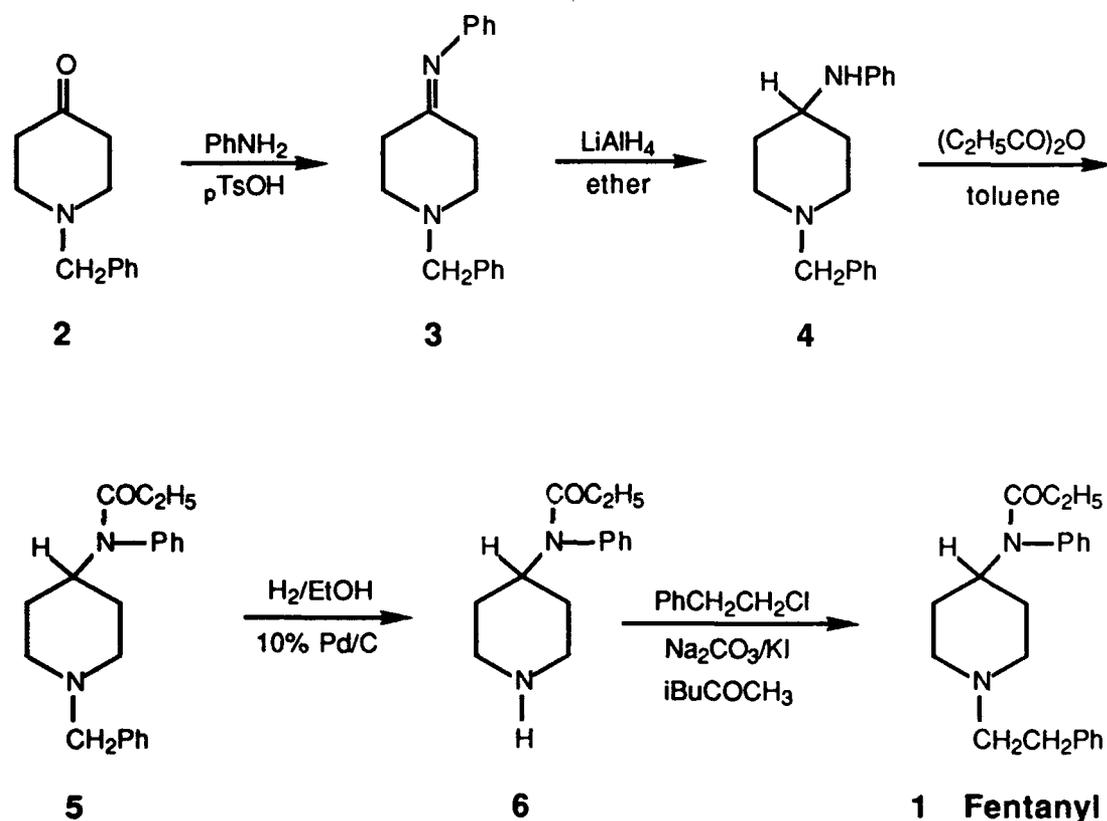
Where appropriate, starting material costs, reaction yields and comments concerning advantages and disadvantages of particular approaches are offered to guide the reader in his selection of the appropriate synthesis or to catalyze new syntheses.

2. SYNTHETIC METHODS

Method A. 8-10

Fentanyl (1), one of the most potent synthetic narcotic analgesic, was first synthesized in 1964 by Janssen and coworkers. Scheme 1 describes their synthetic route.

Scheme 1



Condensation of commercially available 1-benzyl-4-piperidone (**2**) (100 g/\$53.50, Aldrich) and aniline in the presence of acid catalyst followed by *in situ* lithium aluminum hydride (LiAlH₄) reduction of the intermediate Schiff base, **3**, in ether gave compound **4**. Subsequent acylation with propionyl anhydride in toluene followed by hydrogenolysis and alkylation provided the target molecule, **fentanyl** (**1**). However, no yields were given in the patent literature.

Comments:

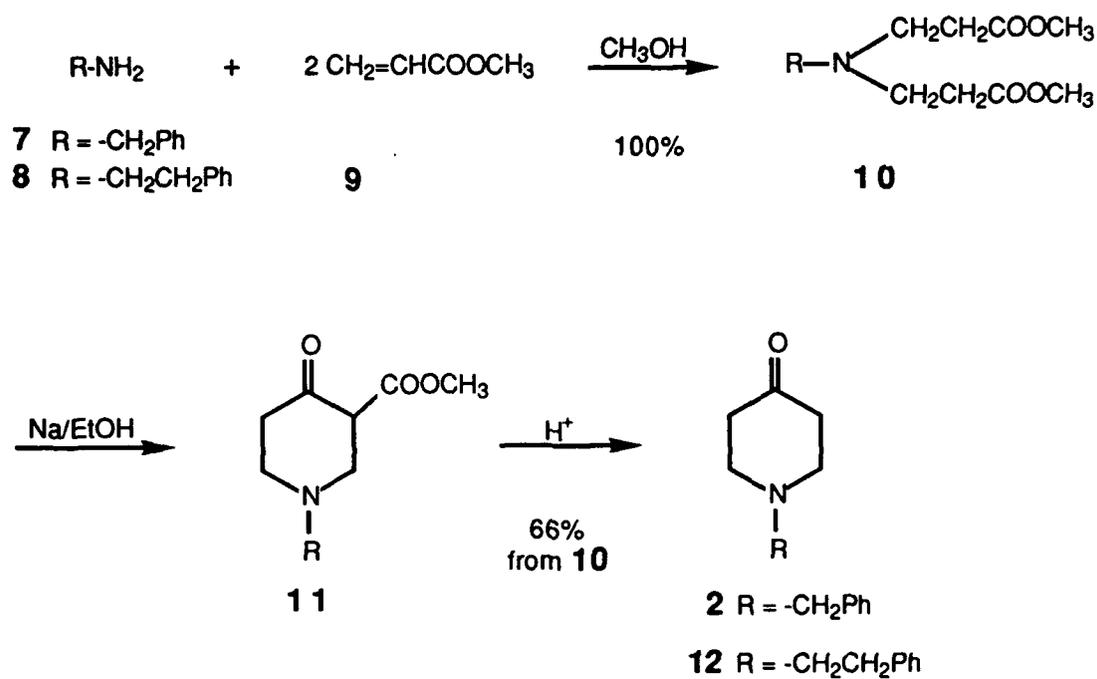
1. Although 1-benzyl-4-piperidone is cheaper, the benzyl group is used as the protecting group, thus, the deprotection and alkylation are required to complete the synthesis, necessarily decreasing yield and convenience.
2. The reducing agent LiAlH₄ and the catalyst (10% Pd/C) used in the hydrogenolysis are relatively expensive.
3. The yield of the final alkylation step in carfentanil is low, thus, the same reaction used in fentanyl synthesis may be poor.

Since then many modifications for the synthesis of fentanyl have been reported. These modifications involve the use of either 2-phenethyl analog of **2** as the starting material, or different types of reducing agents.

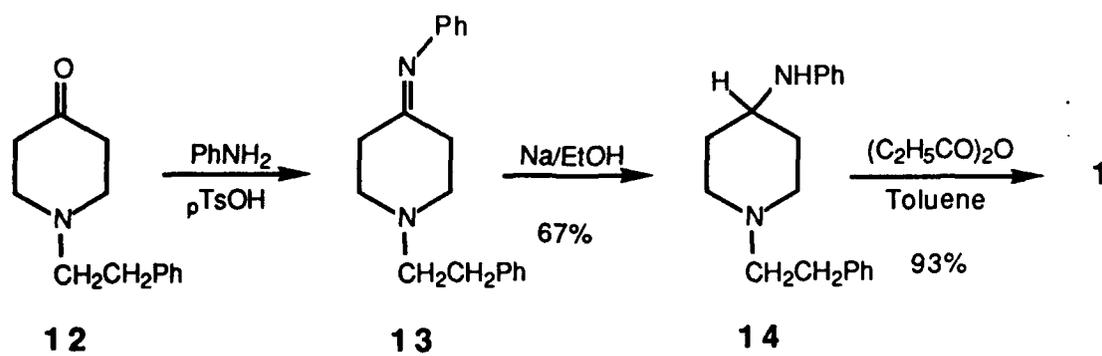
Method B.^{11,12}

Zee and his coworkers did systematic studies on each step of the synthesis in order to optimize the yield and to cut down the cost. The starting material, **12** (25 g/\$87.10, Aldrich) can be prepared from phenethylamine (**8**), and methyl acrylate (**9**) in methanol via the intermediate diester, **10**. The hydrolysis yielded the desired 4-piperidone, **12** in good yield (Scheme 2). Reductive amination of ketone **12** and aniline using Na/EtOH as the reducing agent gave **14** in 67% yield. Acylation with propionyl anhydride in toluene gave 93% of fentanyl (Scheme 3).

Scheme 2



Scheme 3



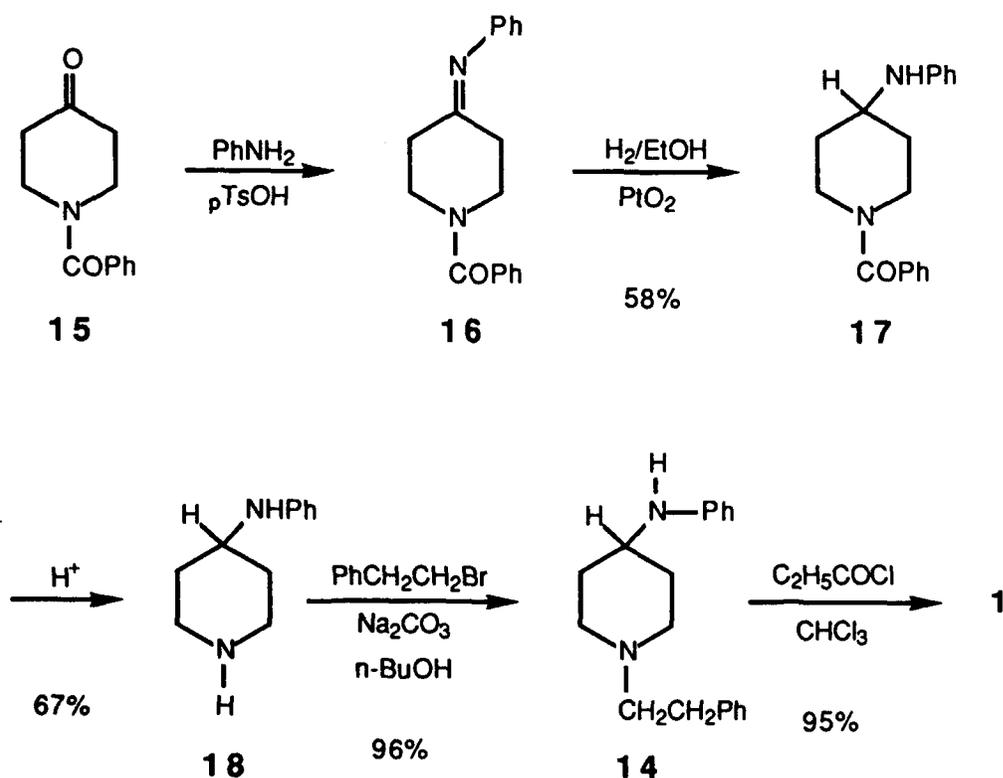
Method C.^{13,14}

The reaction sequence is similar to **Method B** except that the LiAlH_4 reduction of the Schiff base, **13** gave a 65% of the product.

Method D.¹⁵

This method utilizes the commercially available 1-benzoyl-4-piperidone (**15**) (1 g/\$15.00, Aldrich) as the starting material and the catalytic hydrogenation to reduce the Schiff base, **16**. Acid hydrolysis of **17**, followed by alkylation, gives fentanyl.

Scheme 4

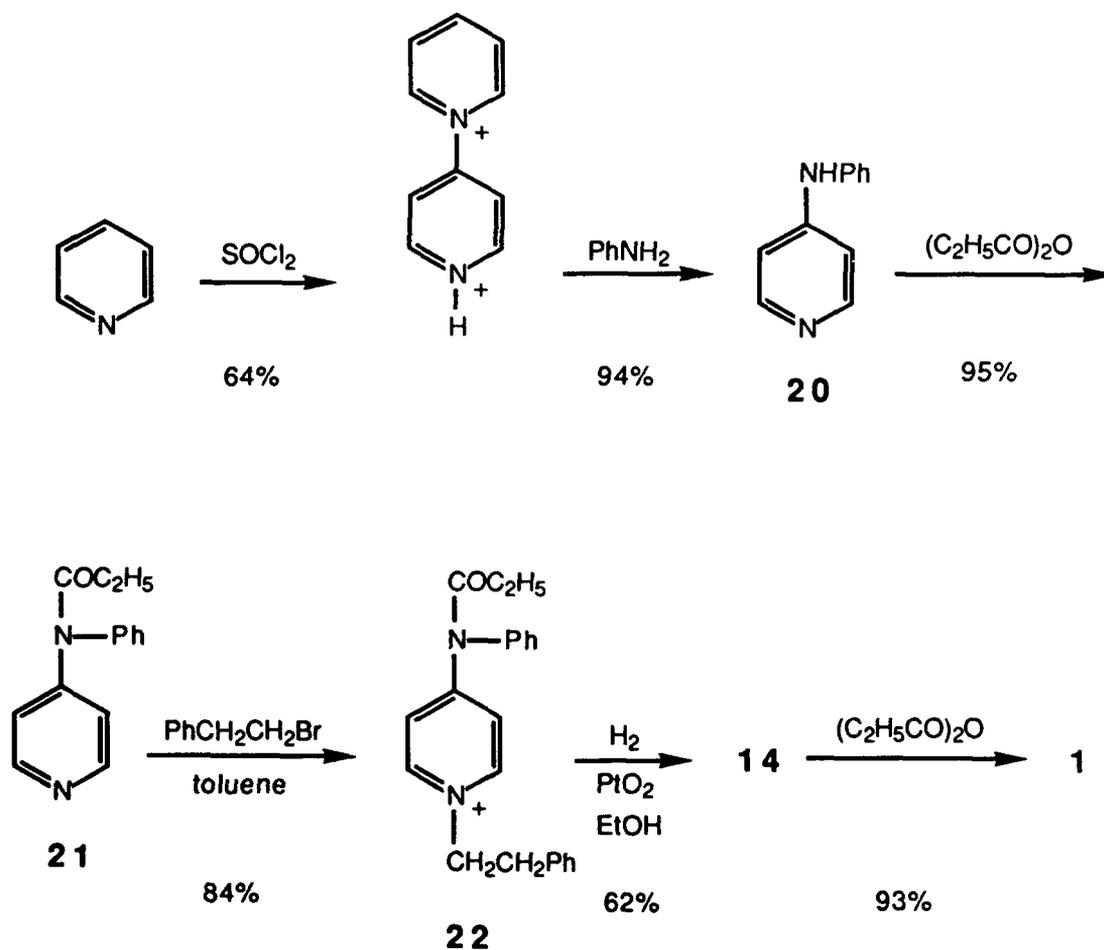


Two additional routes to fentanyl are shown in *Method E* and *F*.

Method E.¹⁶

The major difference between this method and the others is that the piperidine nucleus of fentanyl is derived from pyridine. Thus, 4-anilinopyridine (**20**) was prepared from pyridine and thionyl chloride. Treatment of **20** with propionic anhydride affords the amide **21**. Subsequent alkylation with 2-phenethyl bromide yields the pyridinium bromide intermediate, **22**. Hydrogenation of **22** over platinum oxide reduces the pyridine ring and also cleaves the propionyl group, affording **14** which can be converted into **1** according to Scheme 4.

Scheme 5

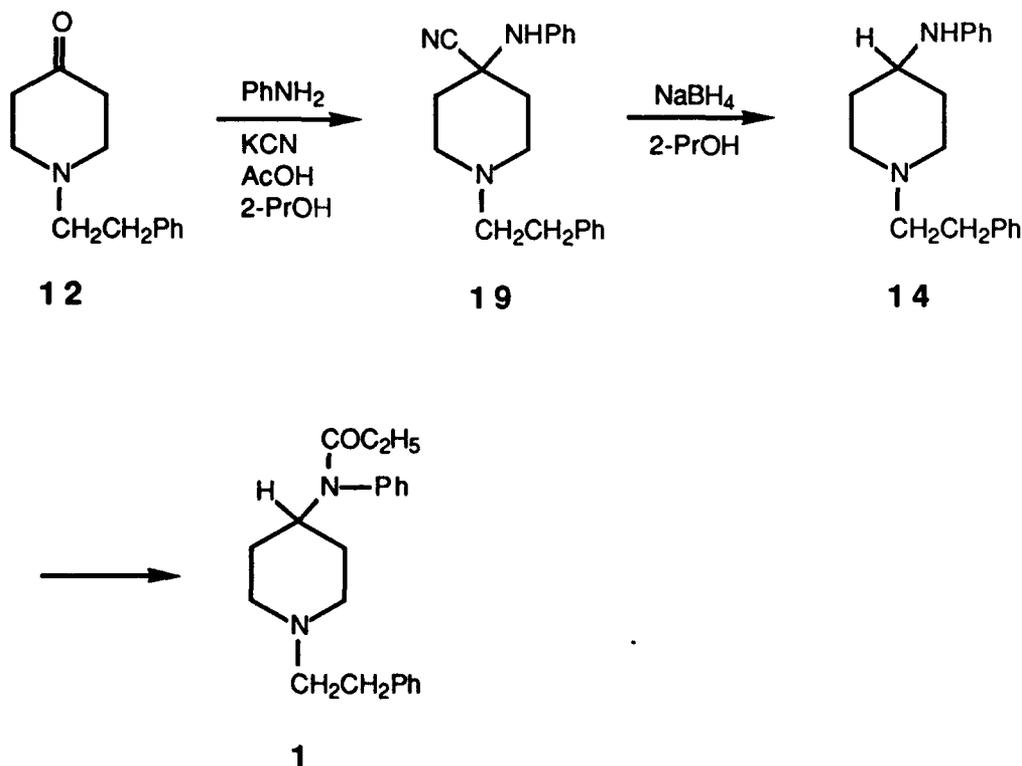


Method F.¹⁷

The best yield for the synthesis of **14** is 67% as reported by Zee *et al.* (cf. **Method B**). An alternate method to prepare **14** has been developed at CRDEC.

The Strecker synthesis of **12** with aniline, KCN, and AcOH in 2-propanol gave excellent yield of α -aminonitrile, **19**. Reductive decyanation of **19** with NaBH₄ in 2-PrOH afforded **14** in 85-90% yield. These two steps can be carried out in one pot without the isolation of **19**. Thus, this process represents a considerable improvement over the others.

Scheme 6



3. CONCLUSION

Scheme 3 represents one of the best procedures for the synthesis of **fentanyl**. Our modification of Scheme 3 has been demonstrated to give the best yield of the product. So this method as shown in Scheme 6 will be further explored at CRDEC.

BLANK

LITERATURE CITED

1. Sertürner, F.W.A. "Darstellung der reinen Mohnsaure (Opiumsäure) nebst einer chemischen Untersuchung des Opium mit vorzuglicher Hinsicht auf einen darin neu entdeckten Stoff und die dahin gehörigen Bemerkungen," J. Pharm. Arzte Apotheker Chem., Vol. 14, pp 47 - 93 (1806).
2. (a) Lockemann, G.J., "Friedrich Wilhelm Serturmer, The Discoverer of Morphine," J. Chem., Educ., Vol. 34., pp 277 - 279, (1951.) (b) Derosne, A., "Memoire Sur l'Opium," Ann. Chim., Vol. 45, pp 257 - 286 (1804).
3. Gulland, J.M., and Robinson, R, "Constitution of Codeine and Thebaine," Mem Proc. Manchester Lit. Phil. Soc., Vol. 69, p. 79 (1925).
4. Gates, M., and Tschudi, G., "The Synthesis of Morphine," J. Am. Chem. Soc., Vol. 78, pp1380 - 1393 (1956) and papers cited therein.
5. Rice, K.C., "The Development of a Practical Total Synthesis of Natural and Unnatural Codeine, Morphine and Thebaine," The Chemistry and Biology of Isoquinoline Alkaloids, Philipson, J.D., Roberts, M.F., Zenk, M.H., EDS., Springer-Verlag, New York, pp 191 - 203, 1985.
6. Eisleb, O. and Shaumann, O., "Dolantin, A New Antispasmodic and Analgesic," Dtsch. Med. Wochschr., Vol. 65, p. 967 (1939).
7. Soudijn, W., "The Pharmacology and Pharmacokinetics of Fentanyl," Int. Cong. Symp. Ser. R. Soc. Med., Vol.3, pp 3 - 10 (1978).
8. Janssen, P. A. J., "Method for Producing Analgesia," U.S. Patent 3,141,823 (1964).
9. Janssen, P. A. J., "1-Aralkyl-4-(n-Aryl-Carbonylamino)-piperidines and Related Compounds," U.S. Patent 3,164,000 (1965).
10. Casy, A. F.; Hassan, M. A.; Simmonds, A. B.; Staniforth, D. , " Structure-activity Relations in Analgesics Based on 4-Anilinopiperidine," J. Pharm. Pharmac. Vol. 21, pp 434-440 (1969).
11. Zee, S.H.; Lai, C.L.; Wu, Y.M.; Chen, G.S., "Preparation of Fentanyl From Phenethylamine and Methyl Acrylate," Natl. Sci. Council. Monthly, ROC, Vol. 9, pp 387-397 (1981).

12. Benke, B. , "N-[1-(2-Substituted-ethyl) -4-piperidyl]carboxamides," Hung. Patent 157,325 (1970); Chem. Abstr. ,Vol. 73, 25305y (1970).
13. Jonczyk, A., "N-(2-Phenylethyl)-4-N-propoinylanilinopiperidine," Pol. Patent 72,416 (1974); Chem. Abstr. Vol. 84, 43865n, (1976).
14. Jonczyk, A.; Jawdosiuik, J.; Makosza M. , "Poszukiwanie Nowej Metody Syntezy Srodka Analgetycznego "Fentanyl," Przem. Chem. Vol. 57, pp 131-134, 180 - 182 (1978).
15. Zong, R.-S.; Yin, D.-X; Ji, R.-Y., "Synthetic Studies of Potential Analgesics. II. Synthesis of Fentanyl," Yao Hsueh Hsueh Pao, " Vol. 14, pp 362-367 (1979).
16. Zee. S.-H.; Wang, W.-K., "A New Process for the Synthesis of Fentanyl," Chin. Chem. Soc. (Taipei), Vol. 27, pp 147-149 (1980).
17. Banks, H. D.; von Ostwalden, P.; Hsu, F.-L. , "An Improved Synthesis of 4-Anilinopiperidines," In Proceedings of the 1988 U. S. Army CRDEC Scientific Conference on Chemical Defense Research, 15-18 November, CRDEC-SP-013, U. S. Army Chemical Research, Development and Engineering Center, Aberdeen Proving Ground, MD, August 1989, UNCLASSIFIED Report, pp 775 - 780.