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The N-silylphosphoranimines \[e.g. Me_3SiN=P(OCH_2CF_3)Me_2\] undergo a variety of reactions including deprotonation/substitution of the P-Me groups, transilylation, and condensation with reactive phosphines. Some new 1,2-addition and oxidation reactions of the P-C systems, \(\text{R}_2\text{NP}=-\text{CHR}\), \(\text{R}_2\text{NP}(=\text{NR})\text{-CHR}\), and \(\text{R}_2\text{NP}=-\text{CR-CHR}\) (\(R=\text{SiMe}_3\)) are also reported.
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by


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SILYLATED PHOSPHINES AND PHOSPHORANIMINES

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Abstract The N-silylphosphoranimines [e.g., Me₃SiN=P(OCH₂CF₃)Me₂] undergo a variety of reactions including deprotonation/substitution of the P-Me groups, transilylation, and condensation with reactive phosphines. Some new 1,2-addition and oxidation reactions of the P=C systems, R₂NP=CHR, R₂NP(-NR)=CHR, and R₂NP=CR=CHR (R = SiMe₃) are also reported.

The chemistry of compounds containing Si-N-P and/or Si-C-P linkages is usually markedly different from that of the nonsilylated analogues. On the one hand, the Si-N or Si-C bonds may function as reactive sites so that processes such as transilylation and condensation reactions via silane elimination are commonly found. Alternatively, the steric and electronic properties of the silyl groups can impart both kinetic stability and interesting chemical reactivity to a variety of low-coordinate phosphorus systems. We report here several illustrative examples of these principles.

REACTIONS OF N-SILYLPHOSPHORANIMINES

Various P-functionalized N-silylphosphoranimines are useful precursors to cyclic and polymeric phosphazenes. Recently, the synthetic utility of such compounds has
been greatly expanded through two general types of reactions. First, deprotonation/silylation of the P-methyl group followed by a Peterson olefination affords a series of vinyl derivatives (eq 1). Some of these

\[ \text{Me}_3\text{SiN}=\text{POCH}_2\text{CF}_3 \xrightarrow{(1) \text{n-BuLi}} \text{Me}_3\text{SiN}=\text{POCH}_2\text{CF}_3 \]

\[ \text{Me}_3\text{SiN}=\text{POCH}_2\text{CF}_3 \xrightarrow{(2) \text{Me}_3\text{SiCl}} \text{Me}_3\text{SiN}=\text{POCH}_2\text{CF}_3 \]

Second, we have studied a broad spectrum of Si-N bond cleavage reactions. For example, transilylation processes yield a series of novel N-silylphosphoranimines (eq 2) bearing reactive substituents on silicon as well as some bis(phosphoranimino)anes and -siloxanes (eq 3). The related reactions involving cleavage of the Si-N bond in \( \text{Me}_3\text{SiN}=\text{P}(X)\text{Me}_2 \) \( [X = \text{Cl, Br, OCH}_2\text{CF}_3, \text{and N(SiMe}_3)_2] \) by reactive phosphines, e.g., \( \text{Ph}_2\text{PY} (Y = \text{Cl, OCH}_2\text{CF}_3) \), are generally more complex.

\[ \text{CF}_3\text{CH}_2\text{O} \quad \text{RSiMe}_2\text{Cl} \quad \text{CF}_3\text{CH}_2\text{O} \]

\[ \text{Me}_3\text{SiN}=\text{PMe}_2 \xrightarrow{\text{Me}_3\text{SiN}=\text{PMe}_2} \text{RMe}_2\text{SiN}=\text{PMe}_2 \]

\( R = \text{Ph, CH}=\text{CH}_2, \text{Cl, (CH}_2)_3\text{CN, CH}_2\text{CH}_2\text{C(O)Me, SiMe}_2(\text{CH}_2)_2\text{SiMe}_2\text{Cl} \)

"Monomers" undergo condensation polymerization to yield poly(phosphazenes) bearing substituted C=C side groups.

\( R_2\text{C}(O) = \text{Me}_2\text{C}(O), \text{PhC}(O)\text{H, PhC}(O)\text{Me, PhC}(O)\text{CF}_3, \text{PhC}(O)\text{CH}=\text{CH}_2, \text{Me}_2\text{C}=\text{CH}-\text{C}(O)-\text{CH}=\text{CMeo}_2 \)
These reactions occur with elimination of Me₃SiY and, depending upon the choice of reactants, lead to various open-chain ionic species (e.g., A and B).

Surprisingly, the permethylated analogue of A (X = OCH₂CF₃) is produced in high yield in the "transilylation" of Me₂SiHCl with Me₃SiN=P(OCH₂CF₃)Me₂.

REATIONS OF SILYLATED P-C COMPOUNDS

In this area our recent work has included a comparative study of the reactivity of the P-C bonds in the analogous P(III) and P(V) compounds, (Me₃Si)₂NP=CHSiMe₃ and (Me₃Si)₂NP(=NSiMe₃)=CHSiMe₃, respectively. In general, the 1,2-addition of polar reagents to the P=C bond occurs more readily in the P(V) system. For example, while the former does not react with PhPCl₂, the latter undergoes Si-N cleavage and ring formation (eq 4) subsequent to the 1,2-addition process. Some reagents, on the other hand, react smoothly with both substrates. Cyclopentadiene, for example, affords a Diels-Alder adduct in each case.
A stable, acyclic 1-phosphadiene has also been prepared (eq 5) and we have begun to investigate its derivative chemistry. Interestingly, simple oxidation reactions (e.g., with sulfur or Me₃SiN₃) are accompanied by isomerization to four-membered ring products (eq 6). In the sulfur reaction, a three-membered disulfide ring, \((\text{Me}_3\text{Si})_2\text{NP(S)}-\text{S-C(SiMe}_3\text{)})\text{CH=CHSiMe}_3\) is also produced.

\[
\begin{align*}
\text{Cl} & \quad \text{H} \\
\text{DBU} & \quad \text{SiMe}_3 \\
\text{Cl} & \quad \text{H} \\
\text{DBU} & \quad \text{Me}_3\text{Si} \\
\text{Cl} & \quad \text{H} \\
\text{DBU} & \quad \text{Me}_3\text{Si} \\
\text{Cl} & \quad \text{H} \\
\text{DBU} & \quad \text{Me}_3\text{Si} \\
\text{Cl} & \quad \text{H} \\
\text{DBU} & \quad \text{Me}_3\text{Si} \\
\text{Cl} & \quad \text{H} \\
\text{DBU} & \quad \text{Me}_3\text{Si} \\
\text{Cl} & \quad \text{H} \\
\text{DBU} & \quad \text{Me}_3\text{Si} \\
\text{Cl} & \quad \text{H} \\
\text{DBU} & \quad \text{Me}_3\text{Si} \\
\text{Cl} & \quad \text{H} \\
\text{DBU} & \quad \text{Me}_3\text{Si} \\
\text{Cl} & \quad \text{H} \\
\text{DBU} & \quad \text{Me}_3\text{Si} \\
\text{Cl} & \quad \text{H} \\
\text{DBU} & \quad \text{Me}_3\text{Si} \\
\text{Cl} & \quad \text{H} \\
\text{DBU} & \quad \text{Me}_3\text{Si} \\
\text{Cl} & \quad \text{H} \\
\text{DBU} & \quad \text{Me}_3\text{Si} \\
\text{Cl} & \quad \text{H} \\
\text{DBU} & \quad \text{Me}_3\text{Si} \\
\end{align*}
\]

\[
\begin{align*}
\text{Me}_3\text{Si} & \quad \text{H} \\
\text{Me}_3\text{Si} & \quad \text{H} \\
\text{Me}_3\text{Si} & \quad \text{H} \\
\text{Me}_3\text{Si} & \quad \text{H} \\
\text{Me}_3\text{Si} & \quad \text{H} \\
\text{Me}_3\text{Si} & \quad \text{H} \\
\text{Me}_3\text{Si} & \quad \text{H} \\
\text{Me}_3\text{Si} & \quad \text{H} \\
\text{Me}_3\text{Si} & \quad \text{H} \\
\text{Me}_3\text{Si} & \quad \text{H} \\
\text{Me}_3\text{Si} & \quad \text{H} \\
\text{Me}_3\text{Si} & \quad \text{H} \\
\text{Me}_3\text{Si} & \quad \text{H} \\
\text{Me}_3\text{Si} & \quad \text{H} \\
\text{Me}_3\text{Si} & \quad \text{H} \\
\text{Me}_3\text{Si} & \quad \text{H} \\
\text{Me}_3\text{Si} & \quad \text{H} \\
\text{Me}_3\text{Si} & \quad \text{H} \\
\text{Me}_3\text{Si} & \quad \text{H} \\
\text{Me}_3\text{Si} & \quad \text{H} \\
\text{Me}_3\text{Si} & \quad \text{H} \\
\text{Me}_3\text{Si} & \quad \text{H} \\
\end{align*}
\]

\[
\begin{align*}
\text{E} & \quad \text{NSiMe}_3, S \\
\text{E} & \quad \text{NSiMe}_3, S \\
\text{E} & \quad \text{NSiMe}_3, S \\
\text{E} & \quad \text{NSiMe}_3, S \\
\text{E} & \quad \text{NSiMe}_3, S \\
\text{E} & \quad \text{NSiMe}_3, S \\
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\text{E} & \quad \text{NSiMe}_3, S \\
\text{E} & \quad \text{NSiMe}_3, S \\
\text{E} & \quad \text{NSiMe}_3, S \\
\text{E} & \quad \text{NSiMe}_3, S \\
\text{E} & \quad \text{NSiMe}_3, S \\
\end{align*}
\]

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