

AD-A135 907

LASER-INDUCED NEUTRALIZATION AND NEGATIVE-ION FORMATION  
IN SURFACE SCATTERING(U) ROCHESTER UNIV NY DEPT OF  
CHEMISTRY K S LAM ET AL. DEC 83 UROCHESTER/DC/83/TR-44

1/1

UNCLASSIFIED

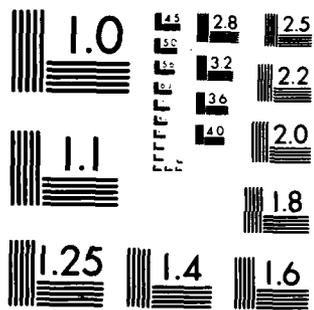
N00014-80-C-0472

F/G 20/5

NL



END  
DATE  
FILMED  
1 - 84  
DTIC



MICROCOPY RESOLUTION TEST CHART  
NATIONAL BUREAU OF STANDARDS-1963-A

(12)

AD-A135 907

OFFICE OF NAVAL RESEARCH

Contract N00014-80-C-0472

Task No. NR 056-749

TECHNICAL REPORT No. 44

Laser-Induced Neutralization and  
Negative-Ion Formation in Surface Scattering

by

Kai-Shue Lam, K. C. Liu and Thomas F. George

Prepared for Publication

in

Proceedings of the Society of Photo-Optical  
Instrumentation Engineers, Volume 459

Department of Chemistry  
University of Rochester  
Rochester, New York 14627

DTIC  
ELECTE  
DEC 16 1983  
S B

December 1983

Reproduction in whole or in part is permitted for any  
purpose of the United States Government.

This document has been approved for public release  
and sale; its distribution is unlimited.

83 12 16 019

DTIC FILE COPY

Unclassified

SECURITY CLASSIFICATION OF THIS PAGE (When Data Entered)

REPORT DOCUMENTATION PAGE		READ INSTRUCTIONS BEFORE COMPLETING FORM
1. REPORT NUMBER UROCHESTER/DC/83/TR-44	2. GOVT ACCESSION NO. 40 4/35 907	3. RECIPIENT'S CATALOG NUMBER
4. TITLE (and Subtitle) Laser-Induced Neutralization and Negative-Ion Formation in Surface Scattering		5. TYPE OF REPORT & PERIOD COVERED
		6. PERFORMING ORG. REPORT NUMBER
7. AUTHOR(s) Kai-Shue Lam, K. C. Liu and <u>Thomas F. George</u>		8. CONTRACT OR GRANT NUMBER(s) N00014-80-C-0472
9. PERFORMING ORGANIZATION NAME AND ADDRESS Department of Chemistry University of Rochester Rochester, New York 14627		10. PROGRAM ELEMENT, PROJECT, TASK AREA & WORK UNIT NUMBERS NR 056-749
11. CONTROLLING OFFICE NAME AND ADDRESS Office of Naval Research Chemistry Program Code 472 Arlington, Virginia 22217		12. REPORT DATE December 1983
		13. NUMBER OF PAGES 5
14. MONITORING AGENCY NAME & ADDRESS (if different from Controlling Office)		15. SECURITY CLASS. (of this report) Unclassified
		15a. DECLASSIFICATION/DOWNGRADING SCHEDULE
16. DISTRIBUTION STATEMENT (of this Report)  This document has been approved for public release and sale; its distribution is unlimited.		
17. DISTRIBUTION STATEMENT (of the abstract entered in Block 20, if different from Report)		
18. SUPPLEMENTARY NOTES  Prepared for publication in Proceedings of the Society of Photo-Optical Instrumentation Engineers, Volume 459		
19. KEY WORDS (Continue on reverse side if necessary and identify by block number) ION-SURFACE SCATTERING METAL SURFACES LASER-INDUCED ELECTRON TRANSFER RESONANT AND NEAR-RESONANT TRANSFER NEUTRALIZATION TWO-ELECTRON TRANSFER NEGATIVE-ION FORMATION ANDERSON CORRELATION ENERGY LASER-INDUCED METASTABLE STATES		
20. ABSTRACT (Continue on reverse side if necessary and identify by block number) In neutralization and negative-ion formation from positive ions scattering from a solid surface, a laser can be used to control the nature of resonant, near-resonant, and even non-resonant transfer of electrons from the conduction band. These spectral characteristics can be achieved by variation of only the laser frequency and intensity.		

DD FORM 1 JAN 79 1473

Unclassified

SECURITY CLASSIFICATION OF THIS PAGE (When Data Entered)

Laser-induced neutralization and negative-ion  
formation in surface scattering

Kai-Shue Lam

Department of Physics, California State Polytechnic University  
3801 West Temple Avenue, Pomona, California 91768

and

K. C. Liu and Thomas F. George

Department of Chemistry, University of Rochester  
Rochester, New York 14627

Abstract

In neutralization and negative-ion formation from positive ions scattering from a solid surface, a laser can be used to control the nature of resonant, near-resonant, and even non-resonant transfer of electrons from the conduction band. These spectral characteristics can be achieved by variation of only the laser frequency and intensity.

Introduction

Positive ions impacting on metal surfaces may pick up electrons resonantly or near-resonantly from the conduction band and thus become neutral atoms or negatively-charged ions on recoil.<sup>1-6</sup> These processes usually require that the empty valence level  $\epsilon_d$  (Figure 1) of the impact-ion be roughly degenerate with the electrons in the conduction band when the ion is in the impact region. If this condition is not fulfilled, neutralization or negative-ion formation can only take place through the much weaker Auger processes.<sup>7</sup> With the introduction of a laser, however, the degeneracy condition ceases to be the determining factor in charge transfer. By varying the laser frequency and intensity, the radiative bound-continuum coupling can readily be manipulated so as to resonantly, near-resonantly, or even non-resonantly transfer electrons from any region in the conduction band to the discrete valence level.

Resonant and near-resonant transfers are effected through the formation of laser-induced metastable states with energies within the band. The degree of resonance is determined by the widths of these states, which, together with the energies, can be controlled by the laser characteristics. Non-resonant transfer is due to the possible formation of a

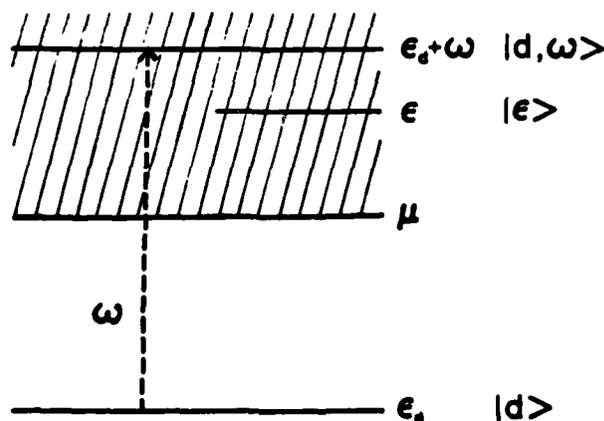


Figure 1. Schematic energy-level diagram for a model ion-surface system interacting with an external laser field near the impact region, where  $\omega$  is the photon energy. The kets denote the states associated with the designated energy levels.

laser-induced bound state with energy below the lower band edge  $\mu$ . This bound state appears when the laser intensity is raised past a certain frequency-dependent critical value, and the exact location of its energy  $\epsilon_b$  can also be controlled by the laser characteristics. Of special interest in non-resonant transfer is the onset of the threshold effect when  $\epsilon_b$  is close to  $\mu$ , in which electrons are drawn predominantly from near the lower edge of the conduction band.

In the present work, we shall focus on the system's spectral properties responsible for the above effects<sup>8</sup> rather than the dynamics, which has been treated elsewhere.<sup>3-6,9</sup>

Non-resonant transfer

We consider a model for the ion-surface system where field-free energy spectrum near the impact-region is depicted in Figure 1. A radiative bound-continuum coupling  $V_{\epsilon\omega}(t)$ , peaking around the instant of impact  $t=0$ , is responsible for the transfer of electrons from the band states  $|\epsilon\rangle$  to the valence state  $|d,\omega\rangle$ . The quantity that is ultimately of interest is the transfer probability

$$P_{\epsilon d}(t) = |\langle \epsilon | d,\omega;t \rangle|^2, \tag{1}$$

where  $|d,\omega;t\rangle$  is the time-evolved state of  $|d,\omega\rangle$ .

In order to determine  $|d,\omega;t\rangle$ , it is useful to expand  $|d,\omega\rangle$  in terms of approximate eigenstates of the total Hamiltonian. We assume the impact velocity to be low enough such that the adiabatic approximation is valid: the electronic spectrum can be diagonalized independently at each  $t$ . The eigenvalue equation is then given by<sup>8</sup>

$$E(t) - [\epsilon_d(t) + \omega] + g^2 \int_{\mu}^{\infty} d\epsilon \frac{\rho(\epsilon) |V_{\epsilon\omega}(t)|^2}{\epsilon - E(t)} = 0, \quad (n=1) \tag{2}$$

where  $g^2$  is a quantity proportional to the field intensity,  $\rho(t)$  is the density of states of the band, and it has been assumed that the band is infinitely wide. This equation admits a bound-state solution  $\epsilon_b(t) < \mu$ , when

$$g^2 > g_{crit}^2 = (\omega - \mu + \epsilon_d) \int_{\mu}^{\infty} d\epsilon \frac{\rho(\epsilon) |V_{\epsilon\omega}|^2}{\epsilon - \mu}. \tag{3}$$

Furthermore, the position of this bound state shifts down as  $g^2$  is increased and/or  $\omega$  is decreased.

Since this bound state, by definition, cannot be degenerate with the band and can act as a receptor of electrons, it is the origin of a non-resonant transfer term occurring in  $P_{\epsilon d}(t)$ . An explicit expression [Eq.(8)] for this quantity will be given after we consider the source of the resonant and non-resonant transfer -- the metastable states -- in the next section.

Resonant and near-resonant transfer

When the radiative coupling is weak, a monochromatic laser gives rise to a metastable state centered at  $\epsilon_d + \omega + \Delta$  with a width  $\Gamma_0$ , where both the small shift  $\Delta$  and  $\Gamma_0$  are proportional to  $g^2$ , or the physical field intensity. The presence of this state means that electrons are transferred predominantly from an energy region in the band corresponding to its location and width, hence the designations resonant and near-resonant transfer.

When the coupling is strong, the resonance structure for  $P_{\epsilon d}(t)$  becomes much more interesting. In this case the locations of the metastable states have to be determined from the equation<sup>8</sup>

$$E(t) - [\epsilon_d(t) + \omega] + g^2 P[E(t)] = 0, \tag{4}$$

$$P[E(t)] = P \int_{\mu}^{\infty} d\epsilon \frac{\rho(\epsilon) |V_{\epsilon\omega}(t)|^2}{\epsilon - E(t)} \tag{5}$$

with  $P$  denoting the principal-value integral. The roots of Eq. (4) in general bear no simple relationship to the "unperturbed" line center  $\epsilon_{q+\omega}$ , and there may be more than one. Thus, even a monochromatic laser may induce more than one resonance region in the conduction band from which a preponderance of electron transfer takes place. Moreover, one can tune the positions of these regions through the band by varying  $\omega$  and the field intensity.

There is another interesting complication when the radiative coupling is strong. It is found that<sup>8</sup> the width of a metastable state is not strictly proportional to  $g^2$ , but to a "renormalized field intensity"

$$g_R^2 = g^2 / [1 + g^2 P'(E_m)], \tag{6}$$

where

$$P'(E_m) = \left. \frac{d}{dE} P(E) \right|_{E=E_m} \tag{7}$$

and  $E_m$  is the position of a metastable state. For a given physical field intensity and frequency, the conduction band may then be classified according to three regimes. The metastable regime [ $P'(E) > 0$ ] has the property that the physical width of a metastable state located within it is always smaller than the "bare" width, while for the unstable regime ( $-1 < g^2 P' < 0$ ) the opposite is true. Finally, there may be an unphysical regime ( $g^2 P' < -1$ ) where energies do not correspond to any physically observed states at all, since  $g_R^2$  in this regime is negative.

The electron-transfer spectrum

Neglecting complications due to the collision dynamics, an approximate formula revealing the essential features of this spectrum can be obtained as<sup>8</sup>

$$P_{ed}(t) = \frac{g^2 |v_{\epsilon\omega}(t)|^2 \beta_b^4(t)}{[\epsilon - \epsilon_b(t)]^2} + \sum_i \frac{g_{Ri}^2 |v_{\epsilon\omega}(t)|^2}{i [\epsilon - E_i(t)]^2 + \Gamma_i^2/4} \tag{8}$$

where the sum is over the metastable states of energies  $E_i$  and  $\Gamma_i$  is the physical width given by

$$\Gamma_i = \Gamma_0(E_i) / [1 + g^2 P'(E_i)] \tag{9}$$

$$\beta_b \equiv \left[ 1 + g^2 \int_{\mu}^{\infty} d\epsilon \frac{\rho(\epsilon) |v_{\epsilon\omega}|^2}{(\epsilon - \epsilon_b)^2} \right]^{-1/2} \tag{10}$$

The first term accounts for the main contributions to non-resonant transfer. It is due entirely to the laser-induced bound state and is present only when the physical field intensity is larger than a certain critical value [Eq.(3)]. Since  $\epsilon_b$  is always less than  $\mu$ , this term is especially significant near the threshold ( $\epsilon \geq \mu$ ) region and contributes a long-range non-resonant "tail" for large  $\epsilon$ . The crucial fact is that its importance can always be enhanced by tuning  $\epsilon_b$  close to  $\mu$ . We call this the threshold effect due to non-resonant transfer.

The summation term contains contributions to the resonant and near-resonant transfer. Based on the discussion of the previous section, one can also effectively "engineer" the locations and widths of the resonance peaks. Within the metastable regime line narrowing can be enhanced, while in the unstable regime line broadening is facilitated. Moreover, the unphysical regime serves as a filter to block off certain regions of the band from electron transfer.

We stress again that the unusual range of spectral phenomena described by Eq. (8) can all be obtained by a judicious variation of the laser frequency and the physical intensity.

The Anderson correlation energy U

We have shown elsewhere<sup>9</sup> that the Anderson correlation energy  $U$ <sup>10</sup> plays an important role in the two-electron transfer process of negative-ion formation. Essentially,  $U$  arises from the Coulomb repulsion between two electrons of opposite spin in the discrete valence level, and imposes an energy barrier for the transfer of the second electron once the first electron has been transferred. This condition manifests itself in overall stricter resonance

459-03

requirements in resonant and near-resonant transfer than in the neutralization case. Hence, in light of the uncertainty principle, short interaction times, or equivalently, large impact velocities are in general favored.

The laser can again be tuned to advantage in this situation. Without arbitrarily increasing the impact velocity of the positive ion, one can either make use of the bound state to enhance the non-resonant part of the transfer, or make use of a "metastable" state in the unstable regime to relax the resonance requirements, or use a suitable combination of the two effects. If, in addition, one can "probe" the spectrum on very short-time scales (on the order of typical collision times) so that resonance requirements are lax even in the field-free case, the barrier effects of U may be largely overcome. This probing can again be best achieved by a laser -- one with pulse durations on the order of collision times.

Acknowledgments

This research was supported in part by the Air Force Office of Scientific Research (AFSC), United States Air Force, under Grant AFOSR-82-0046, and the Office of Naval Research. The United States Government is authorized to reproduce and distribute reprints for governmental purposes notwithstanding any copyright notation hereon. TFG acknowledges the Camille and Henry Dreyfus Foundation for a Teacher-Scholar Award (1975-84) and the John Simon Guggenheim Foundation for a Fellowship (1983-84).

References

1. Overborsch, E. G., Rasser, B., Tenner, A. D. and Los, J., Surf Sci., Vol. 92, p. 310. 1980.
2. Overborsch, E. G. and Los, J., Surf. Sci., Vol. 108, p. 117. 1981.
3. Bloss, W. and Hone, D., Surf. Sci., Vol. 72, p. 277. 1978.
4. Brako, R. and News, D. M., Surf. Sci., Vol. 108, p. 253. 1981.
5. Tully, J. C., Phys. Rev. B, Vol. 16, p. 4324. 1977.
6. Rasser, B., van Wunnik, J. N. M. and Los, J., Surf. Sci., Vol. 118, p. 697. 1982.
7. Hagstrum, H. D., J. Vacuum Sci. Technol., Vol. 12, p. 7. 1975.
8. The details of a general theory on bound-continuum systems applicable to the present context are in a manuscript by K. S. Lam and T. F. George recently submitted to Phys. Rev A.
9. Lam, K. S., Liu, K. C. and George, T. F., Phys. Rev. Lett., submitted.
10. Anderson, P. W., Phys. Rev., Vol. 124, p. 41. 1961.

Accession For	
NTIS GRA&I	<input checked="" type="checkbox"/>
DTIC TAB	<input type="checkbox"/>
Unannounced	<input type="checkbox"/>
Justification	
By	
Distribution/	
Availability Codes	
Dist	Avail and/or Special
A-1	



TECHNICAL REPORT DISTRIBUTION LIST, GEN

	<u>No. Copies</u>		<u>No. Copies</u>
Office of Naval Research Attn: Code 413 800 N. Quincy Street Arlington, Virginia 22217	2	Naval Ocean Systems Center Attn: Technical Library San Diego, California 92152	1
ONR Pasadena Detachment Attn: Dr. R. J. Marcus 1030 East Green Street Pasadena, California 91106	1	Naval Weapons Center Attn: Dr. A. B. Amster Chemistry Division China Lake, California 93555	1
Commander, Naval Air Systems Command Attn: Code 310C (H. Rosenwasser) Washington, D.C. 20360	1	Scientific Advisor Commandant of the Marine Corps Code RD-1 Washington, D.C. 20380	1
Naval Civil Engineering Laboratory Attn: Dr. R. W. Drisko Port Hueneme, California 93401	1	Dean William Tolles Naval Postgraduate School Monterey, California 93940	1
Superintendent Chemistry Division, Code 6100 Naval Research Laboratory Washington, D.C. 20375	1	U.S. Army Research Office Attn: CRD-AA-IP P.O. Box 12211 Research Triangle Park, NC 27709	1
Defense Technical Information Center Building 5, Cameron Station Alexandria, Virginia 22314	12	Mr. Vincent Schaper DTNSRDC Code 2830 Annapolis, Maryland 21402	1
DTNSRDC Attn: Dr. G. Bosmajian Applied Chemistry Division Annapolis, Maryland 21401	1	Mr. John Boyle Materials Branch Naval Ship Engineering Center Philadelphia, Pennsylvania 19112	1
Naval Ocean Systems Center Attn: Dr. S. Yamamoto Marine Sciences Division San Diego, California 91232	1	Mr. A. M. Anzalone Administrative Librarian PLASTEC/ARRADCOM Bldg 3401 Dover, New Jersey 07801	1
Dr. David L. Nelson Chemistry Program Office of Naval Research 800 North Quincy Street Arlington, Virginia 22217	1		

TECHNICAL REPORT DISTRIBUTION LIST, 056

Dr. G. A. Somorjai  
Department of Chemistry  
University of California  
Berkeley, California 94720

Dr. J. Murday  
Naval Research Laboratory  
Surface Chemistry Division (6170)  
455 Overlook Avenue, S.W.  
Washington, D.C. 20375

Dr. J. B. Hudson  
Materials Division  
Rensselaer Polytechnic Institute  
Troy, New York 12181

Dr. Theodore E. Madey  
Surface Chemistry Section  
Department of Commerce  
National Bureau of Standards  
Washington, D.C. 20234

Dr. Chia-wei Woo  
Department of Physics  
Northwestern University  
Evanston, Illinois 60201

Dr. Robert M. Hexter  
Department of Chemistry  
University of Minnesota  
Minneapolis, Minnesota

Dr. J. E. Demuth  
IBM Corporation  
Thomas J. Watson Research Center  
P.O. Box 218  
Yorktown Heights, New York 10598

Dr. M. G. Lagally  
Department of Metallurgical  
and Mining Engineering  
University of Wisconsin  
Madison, Wisconsin 53706

Dr. Adolph B. Amster  
Chemistry Division  
Naval Weapons Center  
China Lake, California 93555

Dr. W. Kohn  
Department of Physics  
University of California, San Diego  
La Jolla, California 92037

Dr. R. L. Park  
Director, Center of Materials  
Research  
University of Maryland  
College Park, Maryland 20742

Dr. W. T. Peria  
Electrical Engineering Department  
University of Minnesota  
Minneapolis, Minnesota 55455

Dr. Keith H. Johnson  
Department of Metallurgy and  
Materials Science  
Massachusetts Institute of Technology  
Cambridge, Massachusetts 02139

Dr. J. M. White  
Department of Chemistry  
University of Texas  
Austin, Texas 78712

Dr. R. P. Van Duyne  
Chemistry Department  
Northwestern University  
Evanston, Illinois 60201

Dr. S. Sibener  
Department of Chemistry  
James Franck Institute  
5640 Ellis Avenue  
Chicago, Illinois 60637

Dr. Arold Green  
Quantum Surface Dynamics Branch  
Code 3817  
Naval Weapons Center  
China Lake, California 93555

Dr. S. L. Bernasek  
Princeton University  
Department of Chemistry  
Princeton, New Jersey 08544

TECHNICAL REPORT DISTRIBUTION LIST, 056

Dr. F. Carter  
Code 6132  
Naval Research Laboratory  
Washington, D.C. 20375

Dr. Richard Colton  
Code 6112  
Naval Research Laboratory  
Washington, D.C. 20375

Dr. Dan Pierce  
National Bureau of Standards  
Optical Physics Division  
Washington, D.C. 20234

Professor R. Stanley Williams  
Department of Chemistry  
University of California  
Los Angeles, California 90024

Dr. R. P. Messmer  
Materials Characterization Lab.  
General Electric Company  
Schenectady, New York ~~22217~~  
12301

Dr. Robert Gomer  
Department of Chemistry  
James Franck Institute  
5640 Ellis Avenue  
Chicago, Illinois 60637

Dr. Ronald Lee  
R301  
Naval Surface Weapons Center  
White Oak  
Silver Spring, Maryland 20910

Dr. Paul Schoen  
Code 5570  
Naval Research Laboratory  
Washington, D.C. 20375

Dr. John T. Yates  
Department of Chemistry  
University of Pittsburgh  
Pittsburgh, Pennsylvania 15260

Dr. Richard Greene  
Code 5230  
Naval Research Laboratory  
Washington, D.C. 20375

Dr. L. Kesmodel  
Department of Physics  
Indiana University  
Bloomington, Indiana 47403

Dr. K. C. Janda  
California Institute of Technology  
Division of Chemistry and Chemical  
Engineering  
Pasadena, California 91125

Professor E. A. Irene  
Department of Chemistry  
University of North Carolina  
Chapel Hill, North Carolina 27514

Dr. Adam Heller  
Bell Laboratories  
Murray Hill, New Jersey 07974

Dr. Martin Fleischmann  
Department of Chemistry  
Southampton University  
Southampton SO9 5NH  
Hampshire, England

Dr. John W. Wilkins  
Cornell University  
Laboratory of Atomic and  
Solid State Physics  
Ithaca, New York 14853

Dr. Richard Smardzewski  
Code 6130  
Naval Research Laboratory  
Washington, D.C. 20375

TECHNICAL REPORT DISTRIBUTION LIST, 056

Dr. R. G. Wallis  
Department of Physics  
University of California  
Irvine, California 92664

Dr. D. Ramaker  
Chemistry Department  
George Washington University  
Washington, D.C. 20052

Dr. P. Hansma  
Physics Department  
University of California  
Santa Barbara, California 93106

Dr. J. C. Hemminger  
Chemistry Department  
University of California  
Irvine, California 92717

Professor T. F. George  
Chemistry Department  
University of Rochester  
Rochester, New York 14627

Dr. G. Rubloff  
IBM  
Thomas J. Watson Research Center  
P.O. Box 218  
Yorktown Heights, New York 10598

Professor Horia Metiu  
Chemistry Department  
University of California  
Santa Barbara, California 93106

Captain Lee Myers  
AFOSR/NC  
Bolling AFB  
Washington, D.C. 20332

Professor Roald Hoffmann  
Department of Chemistry  
Cornell University  
Ithaca, New York 14853

Dr. R. W. Plummer  
Department of Physics  
University of Pennsylvania  
Philadelphia, Pennsylvania 19104

Dr. E. Yeager  
Department of Chemistry  
Case Western Reserve University  
Cleveland, Ohio 41106

Professor D. Hercules  
University Pittsburgh  
Chemistry Department  
Pittsburgh, Pennsylvania 15260

Professor N. Winograd  
Department of Chemistry  
Pennsylvania State University  
University Park, Pennsylvania 16802

Dr. G. D. Stein  
Mechanical Engineering Department  
Northwestern University  
Evanston, Illinois 60201

Professor A. Steckl  
Department of Electrical and  
Systems Engineering  
Rensselaer Polytechnic Institute  
Troy, New York 12181

Professor G. H. Morrison  
Department of Chemistry  
Cornell University  
Ithaca, New York 14853

Dr. David Squire  
Army Research Office  
P.O. Box 12211  
Research Triangle Park, NC 27709

LMED

8