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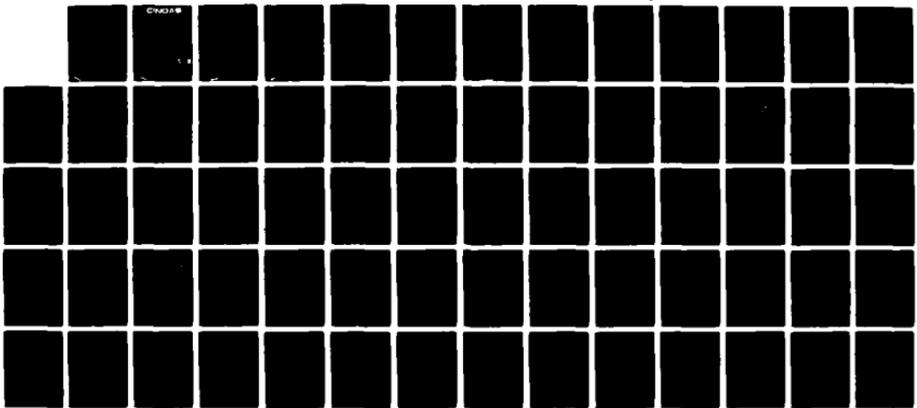
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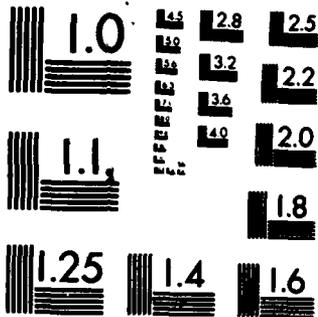
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ELECTRICAL RESISTIVITY OF VANADIUM AND ZIRCONIUM

By

P. D. Desai, H. M. James, and C. Y. Ho

CINDAS Report 63

December 1982

Prepared for

OFFICE OF STANDARD REFERENCE DATA
National Bureau of Standards
U.S. Department of Commerce
Washington, D.C. 20234

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PREFACE

This technical report was prepared by the Center for Information and Numerical Data Analysis and Synthesis (CINDAS), Purdue University, West Lafayette, Indiana, under the auspices of the Office of Standard Reference Data of the National Bureau of Standards (NBS), Department of Commerce, Washington, D.C.

This report represents the most exhaustive compilation and critical evaluation of the recorded world knowledge on the electrical resistivity of vanadium and zirconium, and is one of a series of technical reports on the electrical resistivity of selected elements. The literature search and data compilation have been done in a most extensive and detailed manner, making it possible for all users of the subject to have access to the original data without having to duplicate the laborious and costly process of literature search and data extraction. Also, for the active researchers in the field, a detailed discussion is presented for each material, reviewing the available data and information, giving details of data analysis and synthesis, and discussing the considerations involved in arriving at the final recommended values.

It is hoped that this work will prove useful not only to the engineers and scientists in the field but also to other engineering research and development programs and for industrial applications, as it provides a wealth of knowledge heretofore unknown or inaccessible to many. In particular, it is thought that the critical evaluation, analysis and synthesis, and reference data generation constitute a unique aspect of this work.

Although this report is primarily the result of financial support and interest of the NBS Office of Standard Reference Data, the extensive documentary activity essential to this work was supported by the Defense Logistics Agency of the Department of Defense. Thanks are due Dr. H. J. White, Jr., of the NBS Office of Standard Reference Data for his guidance, cooperation, and sympathetic understanding during the course of this work.

ABSTRACT

This work compiles, reviews, and discusses the available data and information on the electrical resistivity of vanadium and zirconium and presents the recommended values resulting from critical evaluation, correlation, analysis, and synthesis of the available data and information. The recommended values presented are uncorrected and also corrected for the thermal expansion of the material and cover the temperature range from 1 K to above the melting point into the molten state. The estimated uncertainties in most of the recommended values are about $\pm 2\%$ to $\pm 5\%$.

Key Words: vanadium; zirconium; conductivity; critical evaluation; data analysis; data compilation; data synthesis; electrical conductivity; electrical resistivity; elements; metals; recommended values; resistivity.

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*Figures include the recommended values.

NOMENCLATURE

A	Constant in eqs (3b) and (8)
c	Impurity concentration
C	Constant in eq (3a)
e	Base of natural logarithm
h	Planck constant divided by 2π
k	Boltzmann constant
L	Length of specimen at T
L_0	Length of specimen at T_0
AL	$AL = L - L_0$
M	Atomic weight
RER	Residual resistivity ratio
T	Temperature
T_0	Reference temperature
x	$x = h\nu/kT$
a	Constant in eqs (7) and (8)
A	Deviation from the Matthiessen's rule
θ_D	Debye temperature
θ_R	Characteristic temperature for intrinsic electrical resistivity
ρ	Electrical resistivity
ρ_0	Residual electrical resistivity
ρ_e	Electrical resistivity due to electron-electron scattering
ρ_i	Intrinsic electrical resistivity
ω	Phonon angular frequency

1. INTRODUCTION

The principal objective of this project was to exhaustively compile, critically evaluate, analyze, and synthesize all the available data and information on the electrical resistivity of a large number of selected elements and to generate recommended values over a full range of temperature from 1 K to the melting point and beyond. The results on the electrical resistivity of vanadium and zirconium are presented in this work, which is one in a series of similar works on the electrical resistivity of selected elements, some published [1-3]¹. The comprehensive study of the electrical resistivity of the elements at the Center for Information and Numerical Data Analysis and Synthesis (CINDAS) has been a continuation of a similar extensive work on the thermal conductivity of the elements [4].

The general background information on this work is given in Section 2, which includes a brief introduction to the theory of the electrical resistivity of metals and a detailed explanation of the specifics and conventions used in the presentation of the data and information.

The experimental data and information and the recommended values for the electrical resistivity of the two elements are presented in Section 3. In the discussion of the electrical resistivity of each element, individual pieces of available data and information are reviewed, details of data analysis and synthesis are given, the considerations involved in arriving at the final assessment and recommendation are discussed, the recommended values and the experimental data are compared, and the uncertainties in the recommended values are stated. The recommended values uncorrected and corrected for the thermal expansion of the material are both presented in this section. The values cover the temperature range from 1 K to above the melting point.

The last three sections are for acknowledgments, appendices, and references. There are two appendices given. The first appendix presents a logical organization of the methods for the measurement of electrical resistivity. The methods are designated with respective code letters and the same code letters are used in the 'Method Used' column of the Table of Measurement Information

¹Numbers in brackets indicate literature references listed in Section 6.

for indicating the experimental methods used by the various authors. The second appendix presents conversion factors for the units of electrical resistivity, which may be used to convert easily the electrical resistivity values in the SI units given in this work to values in any of the several other units listed.

2. GENERAL BACKGROUND

2.1. Theoretical Background

It was found experimentally by Matthiessen [5,6] that the increase in the electrical resistivity of a metal due to the presence of a small amount of another metal in solid solution is independent of the temperature. According to this Matthiessen's rule, the total electrical resistivity of an impure metal may therefore be separated into two additive contributions and written in the form

$$\rho(c,T) = \rho_0(c) + \rho_i(T) \quad (1)$$

where ρ_0 is the residual resistivity caused by the scattering of electrons by impurity atoms and lattice defects and is temperature-independent but dependent on the impurity concentration, c , and ρ_i is the temperature-dependent intrinsic resistivity arising from the scattering of electrons by lattice waves or phonons.

In reality, however, deviations from Matthiessen's rule do occur. Thus, in general the electrical resistivity of an impure metal is given by

$$\rho(c,T) = \rho_0(c) + \rho_i(T) + \Delta(c,T), \quad (2)$$

where Δ is the deviation from the Matthiessen's rule.

The intrinsic electrical resistivity which is due to scattering of electrons by phonons may be approximated by the Bloch-Grüneisen formula [7,8]:

$$\rho_i = \frac{C}{M\theta_R} \left(\frac{T}{\theta_R} \right)^5 \int_0^{\theta_R/T} \frac{x^5 e^{-x}}{(e^x - 1)^2} dx \quad (3a)$$

$$= A \left(\frac{T}{\theta_R} \right)^5 \int_0^{\theta_R/T} \frac{x^5 e^{-x}}{(e^x - 1)^2} dx, \quad (3b)$$

where C is a constant characteristic of the metal and proportional to the square of the electron-phonon interaction constant, M is the atomic weight, θ_R is a characteristic temperature of the metal which characterizes its intrinsic electrical resistivity in the same way as the Debye temperature, θ_D , characterizes its lattice specific heat, and $A \equiv C/M\theta_R$. The dimensionless variable of integration $x = \hbar\omega/kT$, where \hbar is the Planck constant divided by 2π , ω is the

phonon angular frequency, and k is the Boltzmann constant. The derivation of eq (3) is based on the simplifying assumptions that the Fermi surface is spherical, that the conduction electrons can be treated as free in the first approximation, that the spectrum of lattice vibrations is that of the Debye model, that the phonon distribution is essentially undisturbed by the scattering processes, and that electron-phonon Umklapp processes can be ignored. Consequently, it is perhaps most reasonable to expect the Bloch-Grüneisen formula to agree with experiment in the case of monovalent metals. Nevertheless, the intrinsic resistivity of many metals can be well represented by eq (3) over a wide temperature range by a suitable choice of θ_R and C , though no single values of θ_R can fit the data at all temperatures.

At low temperatures ($T \leq \theta_R/20$), eq (3a) reduces to

$$\rho_i = \frac{124.4C}{M\theta_R} \left(\frac{T}{\theta_R} \right)^5, \quad (4)$$

while at high temperatures ($T > \theta_R$), to a good approximation, it reduces to

$$\rho_i \approx \frac{C}{4M\theta_R} \left(\frac{T}{\theta_R} \right). \quad (5)$$

Thus it agrees with the experimental facts that at very low temperatures the intrinsic or ideal electrical resistivity (after subtracting ρ_0 from ρ) of most metallic elements is proportional to T^5 which is attributed to electron-phonon intraband scattering, and at high temperatures the resistivity of most metals increases approximately linearly with temperature.

In separating the electrical resistivity into its components, the temperature dependent part sometimes includes the electrical resistivity due to electron-electron scattering, ρ_e ; indeed, this is thought to be the dominant temperature-dependent term in transition metals at low temperatures. That is,

$$\rho = \rho_0 + \rho_e + \rho_i(T). \quad (6)$$

As in the case of the scattering of electrons by phonons, electron-electron collisions are of two types: normal processes in which the total wave vector is conserved, and Umklapp processes in which the total wave vectors before and after the collision differ by a reciprocal lattice vector. On the other hand, unlike electron-phonon Umklapp processes which are frozen out at

low temperatures if the Fermi surface is everywhere clear of the zone boundary, electron-electron Umklapp processes are not frozen out at low temperatures. Normal processes, involving the collision between two s-band conduction electrons, do not contribute directly to the electrical resistivity because they do not change the total momentum and thus have no effect on the current. Normal processes involving the scattering of an s-band conduction electron by a non-conducting d-band electron do contribute to the electrical resistivity, and are thought to be the dominant temperature-dependent resistive processes in transition elements and their alloys at very low temperatures, since their resistivities show the T^2 temperature dependence expected for electron-electron scattering rather than the T^5 temperature dependence expected for the intrinsic resistivity. This temperature dependence of the electrical resistivity due to electron-electron scattering:

$$\rho_0 = \alpha T^2 \quad (7)$$

comes about through the double application of the exclusion principle in the scattering processes; it applies to both the initial states and final states. In eq (7), α is a constant.

Umklapp processes between two conduction electrons do contribute to the electrical resistivity. Because these processes involve a reciprocal lattice vector, the wave functions of the electrons involved cannot be regarded as simple plane waves, but must be treated as true Bloch functions having the periodicity of the lattice. The results of this are to introduce into the expression for the resistivity the square of an interference factor. Apparently this factor is quite small, as the low temperature electrical resistivity of most ordinary metals does not show the T^2 temperature dependence expected for such a resistive mechanism.

Substituting eqs (7) and (3b) into eq (6) yields

$$\rho = \rho_0 + \alpha T^2 + A \left(\frac{T}{\theta_R} \right)^5 \int_0^{\theta_R/T} \frac{x^5 e^{-x} dx}{(e^x - 1)^2} \quad (8)$$

Equation (8) has been used frequently in analyzing the experimental data and in generating the recommended values for the electrical resistivity at low temperatures.

2.2. Presentation of Data and Information

In each of the subsections in Section 3, electrical resistivity data and information for each element are presented in the following order:

- (1) A discussion text.
- (2) A table of recommended values.
- (3) A figure presenting experimental data as a function of temperature in a log-log scale.
- (4) A figure presenting recommended values and selected experimental data (on which the recommendations were based) as a function of temperature in a log-log scale.
- (5) A figure presenting recommended values and selected experimental data (on which the recommendations were based) as a function of temperature in a linear scale.
- (6) A table giving measurement information on the experimental data presented in the figures, and
- (7) A table of experimental data for all the data sets listed in item 6 above.

In the discussion text on the electrical resistivity of each element, individual pieces of available data and information are reviewed, details of data analysis and synthesis are given, the considerations involved in arriving at the final assessment and recommendation are discussed, the recommended values and the experimental data are compared, and the uncertainties of the recommended values are stated.

The recommended values are for well-annealed high-purity specimens of the respective elements; however, those values for low temperatures are applicable only to the particular specimens having residual electrical resistivities as given at 1 K in the tables.

The recommended values uncorrected and corrected for the thermal expansion of the element are both given in the table. The uncorrected and corrected values are related by the following equation:

$$\rho_{\text{corrected}}(T) = \left[1 + \frac{\Delta L(T)}{L_0} \right] \rho_{\text{uncorrected}}(T). \quad (9)$$

where $\Delta L = L - L_0$ and L and L_0 are the lengths of the specimen at any temperature T and at a reference temperature T_0 , respectively. The thermal expansion correction amounts roughly to about -0.2% at low temperatures, zero at room temperature, about 0.3% near 500 K, and about 1.5% to 2.5% near the melting point of the element.

The recommended values in some cases are given with more significant figures than warranted, which is merely for tabular smoothness or for the convenience of internal comparison. Hence, the number of significant figures given in the table has no bearing on the degree of accuracy or uncertainty in the values; the uncertainty in the values is always explicitly stated.

In the figures, a data set consisting of a single data point is denoted by a number enclosed by a square, and a curve that connects a set of two or more data points is denoted by a ringed number. These data set numbers correspond to those listed in the accompanying tables providing measurement information and tabulating numerical data for each of the data sets. When several sets of data are too close together to be distinguishable, some of the data sets, though listed and tabulated in the tables, are omitted from the figure for the sake of clarity. The data set numbers of those data sets omitted from the figure are asterisked in both tables providing the measurement information and tabulating the experimental data.

The tables providing the measurement information contain for each set of experimental data the following information: data set number, reference number, author(s), year of publication, experimental method used for the measurement, temperature range covered by the data, name and specimen designation, specimen composition, specification and characterization, and information on measurement conditions, which are contained in the original paper. The experimental methods used for the measurement of the electrical resistivity are indicated in the column headed 'Method Used' in the table by the following code letters:

- A Direct-current potentiometer method
- B Direct-current bridge method
- C Alternating-current potentiometer method
- K Direct heating method
- R Rotating magnetic field method
- T Transient (subsecond) method
- V Voltmeter and ammeter direct reading method
- This symbol means either that the method described by the author is not sufficient for assigning a specific code letter or that the use of a code letter would not convey enough of the information reported in the research document, and therefore the method used is described briefly in the last column of the table.

Details of these and other methods for the measurement of electrical resistivity may be found in the literature references given in Appendix 5.1, which presents a complete scheme for the classification and organization of the methods.

In the tables tabulating the experimental data, all the original data reported in different units have been converted to have the same units: the SI units $10^{-8} \Omega \text{ m}$. The recommended values generated are also given in the same units. Conversion factors for the units of electrical resistivity, which may be used to convert the electrical resistivity values in the SI units given in this work to values in other units, are given in Appendix 5.2.

3. ELECTRICAL RESISTIVITY DATA AND INFORMATION

3.1. Vanadium

There are 69 sets of experimental data available for the electrical resistivity of undoped vanadium as a function of temperature. The residual resistivity of the purest sample reported in this investigation is $0.01008 \times 10^{-8} \Omega \text{ m}$. Information on the specimen characterization and measurement condition for each of the data sets is given in table 2. The data are tabulated in table 3 and shown partially in figure 1.

In the absence of a magnetic field, vanadium is a superconductor below its superconducting transition temperature (5.46 K). The superconducting transition temperature is very sensitive to the magnetic field intensity: the higher the magnetic field intensity, the lower is the superconducting transition temperature. Aleksandrov et al. [19] found that the superconducting transition temperature of vanadium would be lowered to 4.5 K in a magnetic field of ~ 0.5 kOe. Furthermore, their measurements for the nonsuperconducting state of a high purity vanadium specimen at ~ 5.4 K in a magnetic field of ~ 2.2 kOe showed an increase of about 0.45% in the electrical resistivity; thus the influence of the magnetic field on the electrical resistivity of very pure vanadium could be neglected.

The electrical resistivity measurements below room temperature have received considerable attention. This is evident in the extent of the measurements of Pan et al. [13] (data sets 6,7), Courtney [14] (data sets 8-11), Chakal'skii et al. [15] (data set 15), Jung et al. [16-18] (data sets 13-16), Aleksandrov [19] (data sets 17,18), Azhazha et al. [20] (data sets 19,20), Westlake and Alfred [37,38] (data sets 37,38), Amitin et al. [40] (data sets 41,42), Taylor and Smith [45] (data sets 52-55), and White and Woods [48,49] (data set 59). Very recent studies have been made by Gautron et al. [53] (data set 64) on a sample with the highest purity (i.e., lowest $\rho_0 = 0.01 \times 10^{-8} \Omega \text{ m}$) and by Tsai et al. [54] (data sets 65,66) on a sample with $\rho_0 = 0.0109 \times 10^{-8} \Omega \text{ m}$.

The temperature dependent part of the electrical resistivity below 21 K was reported to be proportional to T^3 by White and Woods [48,49]. This was confirmed later by results of Chakalskii et al. [15], Jung et al. [16-18], and

by Aleksandrov et al. [19]. The presence of the cubic term is evidently connected with s-d interband scattering. However, studies of Tsai et al. [55] on the sample with $\rho_0 = 0.0109 \times 10^{-8} \Omega \text{ m}$ found an additional T^2 term which they attributed to electron-electron scattering (ρ_e). In order to verify these results, Gautron et al. [53] carried out electrical resistivity measurements on an even purer specimen with $\rho_0 = 0.01 \times 10^{-8} \Omega \text{ m}$, and obtained a value of $(1.6 \pm 0.2) \times 10^{-11} \Omega \text{ cm/K}^2$ for ρ_e that was compatible with the value of $(1.3 \pm 0.2) \times 10^{-11} \Omega \text{ cm/K}^2$ obtained by Tsai et al. [54]. Gautron et al. [53] pointed out that the temperature dependent electrical resistivity above 10 K is dominated by electron-phonon interactions. Below 10 K, the electron-electron term makes a significant contribution, and it begins to dominate below 5 K. Failure to detect the ρ_e term in earlier studies [e.g., 15-18, 48-50] was attributed to the fact that these studies did not involve measurements to low enough temperatures, and also to the fact that below 10 K the electron-electron contribution is of the order of or less than ρ_0 , even for relatively pure specimens.

An anomalous behavior of the electrical resistivity between 180 and 300 K has been observed by Burger and Taylor [46], Suzuki et al. [74], Smirnov and Finkel [67], and by Rostoker and Yamamoto [73]. However, Westlake [38] found that hydrogen absorbed in the specimen affects the resistivity anomalously near 180 K and that hydrogen-free vanadium did not show such anomalous behavior.

Comparison of the electrical resistivity data below room temperature indicates that the electrical resistivity of vanadium deviates from Matthiessen's rule. The deviations are dependent not only on the concentration of impurities, but also on their type. The deviations are larger for the less pure specimens.

With the discussion given above in mind, the recommended values are based on the data of Courtney [14] (data set 11), Jung et al. [16-18] (data sets 13-16), Gautron et al. [54] (data set 64), and of Tsai [55] (data set 65), who all measured specimens with RRR > 1500. Special weight was given to the data of Gautron et al. [54] on a specimen with RRR = 1970 and $\rho_0 = 0.01 \times 10^{-8} \Omega \text{ m}$. The deviation of the data from the recommended values for somewhat less pure specimens [13,19,20,37,38,40,45,48,49] are shown in figure 1.

At the highest temperatures there is general agreement on the temperature dependence of the electrical resistivity. There are little good data from 300

to 1200 K. The recommended values in this temperature range are based on the data of Neimark et al. [28] (data set 26) and of Taylor and Groot [55] (data set 67). However, Neimark et al. have indicated rather high maximum error for their measurements, and $RRR = 400$ is reported by Taylor and Groot [55] for their sample. The recommended values from 1200 to the melting point are based on the data of Gathers et al. [10] (data set 2), Cezairliyan et al. [22-24] (data set 22), and of Peletskii et al. [32] (data set 30) [57] (data set 69). A compromise has been made between their somewhat divergent results. The scatter of the data from other investigations reported in table 2 [12,28-30,33,34,36,43,44,56] is of the order of $\pm 10\%$. The recommended values above 2202 K, in the liquid region, are based on the compromise between the only two data sets available, due to Seydel and Fucke [9] (data set 1) and to Gathers et al. [10] (data set 2). At 4000 K the divergent in their values approaches 9%. The data of Gathers et al. [10] indicate a lower melting point than the generally accepted value of 2202 K, presumably because their data were taken under pressure.

The recommended values of the electrical resistivity given in table 1 and shown in figures 2 and 3 along with the experimental data, which were used to generate these values, are for vanadium of 99.99% purity or higher, but those below 100 K are applicable specifically to vanadium having a residual resistivity of $0.0100 \times 10^{-8} \Omega \text{ m}$. The table gives both values uncorrected and corrected for thermal expansion, while the figures show only the uncorrected recommended values and mostly uncorrected experimental data. The values for the thermal expansion were taken from ref. [121]. The uncertainty in the recommended values is estimated to be within $\pm 10\%$ from 7 to 20 K and $\pm 5\%$ at lower and higher temperatures.

Vanadium is a transition element and its low-temperature electrical resistivity depends on the type as well as on the concentration of impurities. The electrical resistivity of lower-purity vanadium is, therefore, difficult to estimate, especially at low temperatures ($< 250 \text{ K}$). However, judging from the data reported by Jung et al. [16-18], it appears that for specimens having residual resistivities less than $0.5 \times 10^{-8} \Omega \text{ m}$ only small uncertainties ($< 0.01 \times 10^{-8} \Omega \text{ m}$ at 20 K, and $\sim 0.3 \times 10^{-8} \Omega \text{ m}$ at 100 K) are introduced by the application of Matthiessen's rule. The data from refs. [16-18] (data set 13) and from ref. [38] (data set 38) with sample residual resistivity of $0.52 \times 10^{-8} \Omega \text{ m}$ and $0.81 \times 10^{-8} \Omega \text{ m}$, respectively, are also shown in one of the figures for illustration.

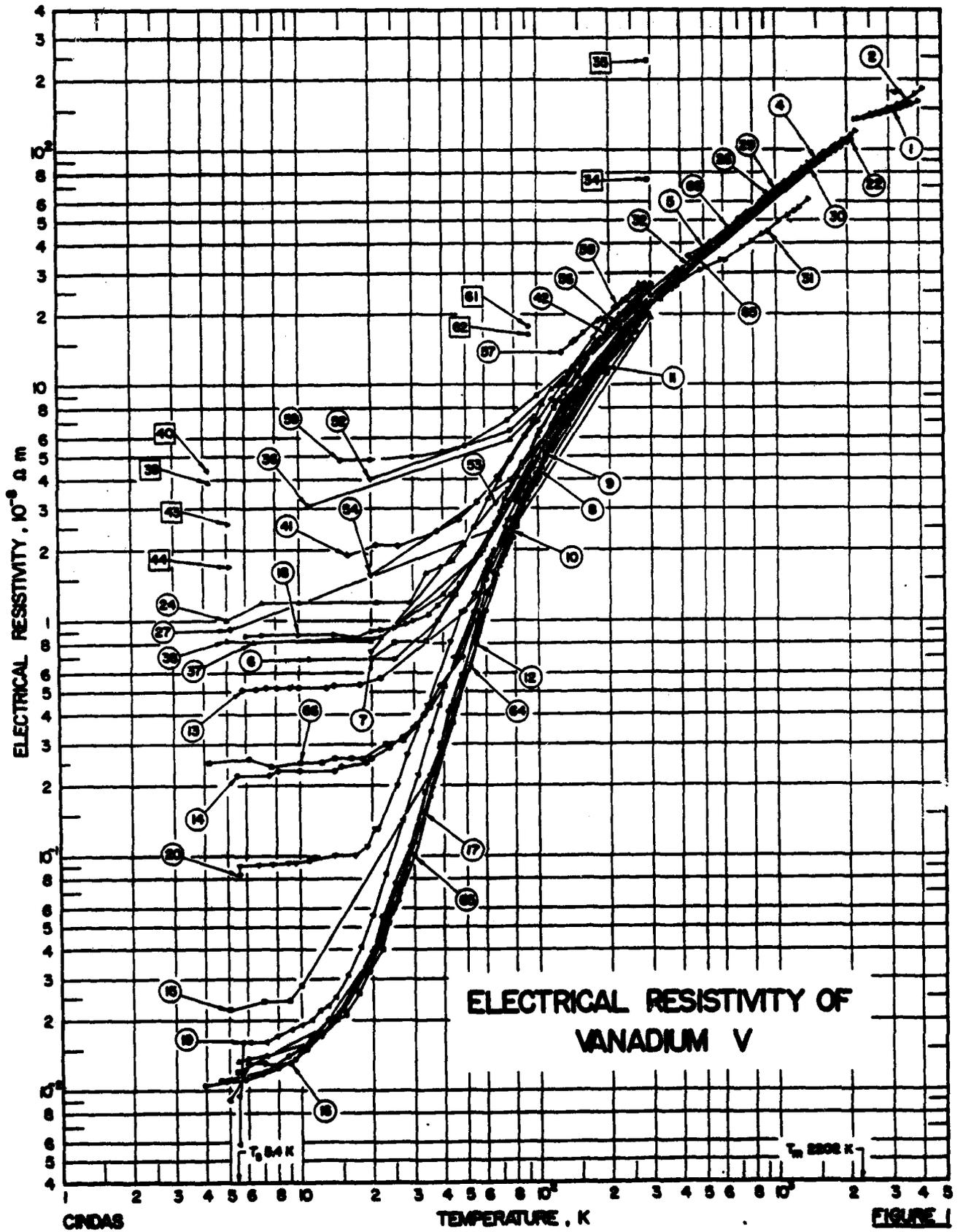
Additional information on the electrical resistivity is reported in refs. [58-95]. Data of Hensler et al. [35] (data sets 33,34), Gurp [41] (data sets 43,44), and of Wruk and Wert [51] (data sets 61,62) are for films/foils; readers are directed to refs. [96-115] for additional information/data on films. The data of Courtney [14] (data sets 8-10) are hydrogen-doped vanadium and additional information/data on various doped-vanadium samples are reported in refs. [65,72,102,116-119]. Effects of irradiation are discussed in refs. [71,72,120], of annealing temperature in refs. [66,112,116,120], and of pressure in refs. [73,74,122].

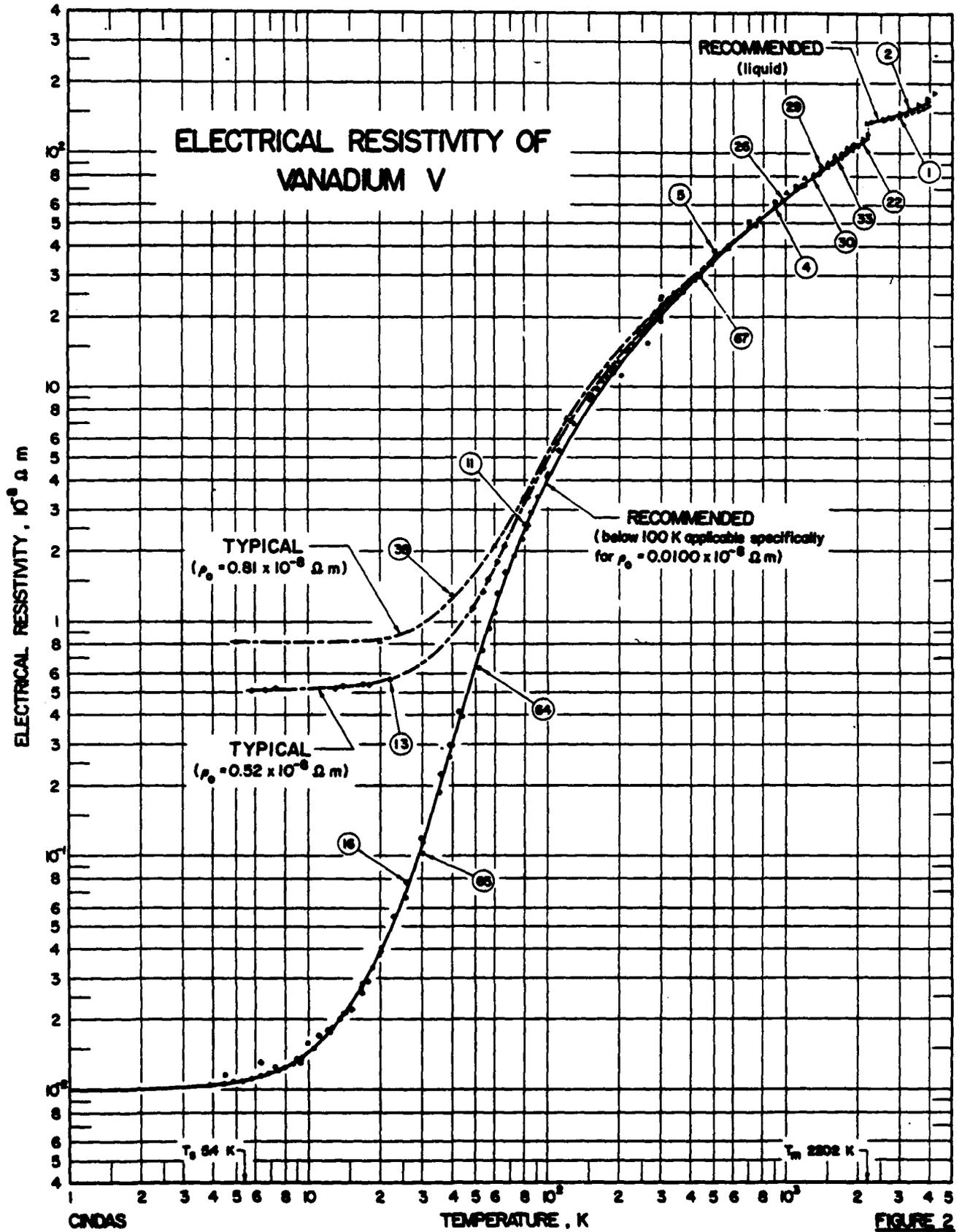
TABLE 1. RECOMMENDED VALUES FOR THE ELECTRICAL RESISTIVITY OF VANADIUM^a[Temperature, T, K; Electrical Resistivity, ρ , $10^{-8} \Omega \text{ m}$]

T	ρ		T	ρ	
	uncorrected	corrected		uncorrected	corrected
1	0.0100 ^(b)	0.0100	700	47.2	47.4
4	0.0105	0.0105	800	53.1	53.4
7	0.0117	0.0117	900	58.7	59.1
10	0.0145	0.0145	1000	64.1	64.6
15	0.0232	0.0232	1100	69.1	69.7
20	0.0391	0.0391	1200	73.8	74.5
25	0.0661	0.0660	1300	78.5	79.4
30	0.112	0.112	1400	83.2	84.2
40	0.304	0.304	1500	87.8	89.0
50	0.649	0.648	1600	92.3	93.7
60	1.114	1.112	1700	96.7	98.3
70	1.706	1.703	1800	100.9	102.7
80	2.413	2.409	1900	104.9	107.0
90	3.196	3.191	2000	108.7	111.0
100	4.01	4.00	2100	112.2	114.8
150	8.22	8.21	2202	115.6(s)	118.5(s)
200	12.43	12.42	2202		135.1(l)
250	16.37	16.36	2400		137.6
273	18.14	18.14	2600		140.4
293	19.68	19.68	2800		143.3
300	20.21	20.21	3000		146.4
350	24.2	24.2	3200		149.7
400	28.0	28.0	3400		153.3
500	34.8	34.9	3600		157.5
600	41.1	41.2	3800		162.0
			4000		166.8

^aThe values are for vanadium of purity 99.99% or higher, but those below 100 K are applicable specifically to vanadium having a residual resistivity of $0.0100 \times 10^{-8} \Omega \text{ m}$. The columns headed uncorrected and corrected refer to values uncorrected and corrected for thermal expansion, respectively. Solid line separating tabular values indicates solid to liquid state transformation.

^bAssuming superconductivity suppressed by magnetic field.





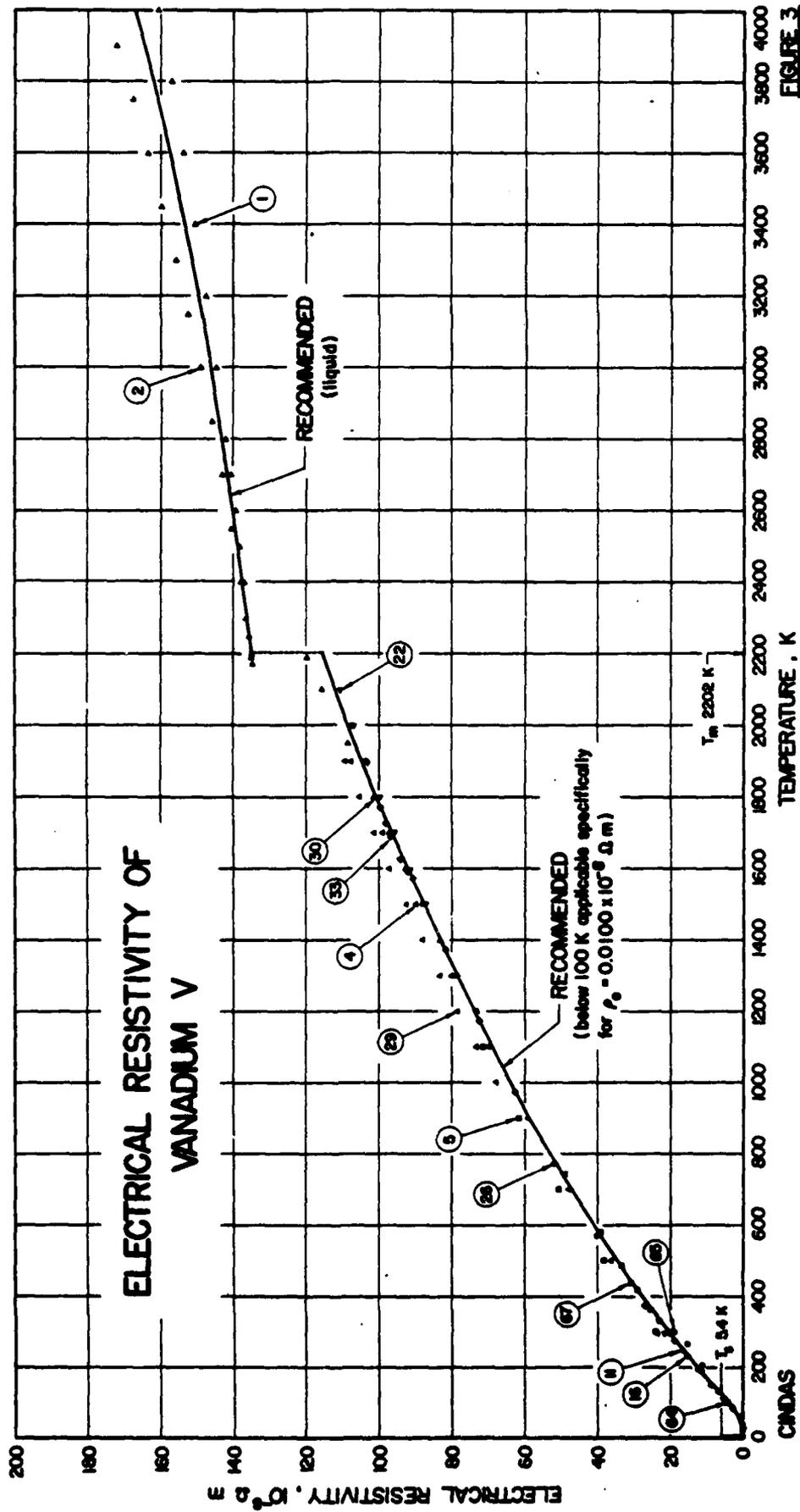


FIGURE 3

TABLE 2. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF VANADIUM V

Data Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
1	9	Seydel, U. and Fuchs, W.	1980	T	2173-4000		99.9 V; temperature measurements taken on foil samples, length 4.4 cm, cross sections 5×10^{-4} cm ² ; heated by means of a capacitor discharge with a heating rate of 10^{11} K s ⁻¹ ; for the range of T _c (melting temperature) = 2175 K \leq T \leq 6600 K, ρ ($\mu\Omega$ cm) = $1.3486 + 1.0219 \times 10^{-4}(T-T_m) + 2.1803 \times 10^{-6}(T-T_m)^2$; error in ρ stated as 5-8%.
2	10	Gachera, G.R., Shamer, J.W., Hinson, R.S., and Young, D.A.	1979		1800-4200		Wire sample 1.0 mm diameter, 25 mm long; phase change from solid to liquid occurs at 2190 K; resistivity values measured at 0.3 GPa; for the solid, ρ_0 ($\mu\Omega$ m) = $0.1077 + 5.3699 \times 10^{-4}T - 1.7255 \times 10^{-6}T^2$, 1800 K \leq T \leq 2190 K; least squares fit of data; smoothed values listed.
3*	11	Vedernikov, M.V., Dremithin, V.G., and Zhemgulov, A.	1978	A	4.2-293		No details given.
4	12	Felotskii, V.R., Ananovich, E.S., Kostomovskii, A.V., Zaretckii, E.B., Sobol, Ya.G., and Shur, B.A.	1977	A	300-1900	V1	99.8 V, 0.01 C, 0.09 Os, 0.02 Si, 0.02 Al, 0.02 Fe; density 6.1 g cm ⁻³ ; crystal orientation [100]; data not corrected for thermal expansion; error does not exceed $\pm 1.5\%$ from 300 to 1600 K and $\pm 2-5\%$ from 1600 to 2000 K; data extracted from smooth tabulated values.
5	12	Felotskii, V.R., et al.	1977	A	300-2000	V2	99.9 V, 0.06 C, 0.02 Os, 0.01 Si, 0.01 Zr, 0.01 Al; density 6.097 g cm ⁻³ ; crystal orientation 3° [001]; other specifications are same as above.
6	13	Fan, V.M., Frokhov, V.G., Shevchenko, A.D., and Dorygopol, V.P.	1977	A	11-300		Single crystal specimens; measurements taken with two directions of current flow <100> and <110>; critical temperature for superconductive transition 5.22 K; $\rho_{300}/\rho_0 = 43$, temperature coefficient of resistivity at 300 K 4.1×10^{-4} K ⁻¹ ; application of magnetic field of 40 kOe did not change the temperature dependence of ρ or shift the position of T _c ; data extracted from figure reported for measurements in zero magnetic field; values reported at 6 K are 0.5×10^{-8} Ω m and 21.5×10^{-9} Ω m at 300 K.
7	13	Fan, V.M., et al.	1977	A	20-300		Same as above except magnetic field H = 40 kOe.
8	14	Courtney, D.R.	1977	A	95-288	VR330	Electro-transported rods electropolished in a 94-6% methanol-perchloric acid, then subjected up to 10^{-7} torr in a vacuum furnace and heated to 1000°C for 1 1/4 hr and at 800°C in H ₂ for 2 hr; specimen length 4.3 cm and 0.23 cm diam.; 330 ppm H, 140 ppm O, 10 ppm N, 15 ppm C, and 165 ppm OHH+C; data from figure.
9	14	Courtney, D.R.	1977	A	76-296	VR260	Same as above except 260 ppm H, 60 ppm O, 3 ppm N, 18 ppm C, and 81 ppm OHH+C; specimen length 3.92 cm and 0.244 cm diameter.
10	14	Courtney, D.R.	1977	A	79-283	VR54	Same as above except 54 ppm H, 27 ppm O, 1 ppm N, 11 ppm C, and 39 ppm OHH+C; specimen length 2.9 cm and 0.242 cm diameter.

*Not shown in figure.

TABLE 2. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF VANADIUM V (continued)

Data Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Mass and Specimen Designation	Composition (weight percent), Specifications and Remarks
11	14	Courtney, D.R.	1977	A	81-295	VN1	Same as above except <1 ppm H and 15 ppm (O+H+C); specimen length 3.65 cm and 0.205 cm diam.; data of Jung [16,17].
12	15	Chahal'akhid, B.K., Akhachev, V.M., Rod'ko, B.A., and Shalyt, S.S.	1976	A	5-155		No details given; specimen same as that reported in data set 17.
13	16	Jung, W.D.	1975	A	6-273	Sample 1	Specimen prepared by Schmidt of the Ames Laboratory using electro-transport technique from the polycrystalline double-electrorefined vanadium supplied by the U.S. Bureau of Mines; total impurities 100 atm ppm consist of 30 atm ppm Cl, 23 atm ppm W, 22 atm ppm Cu, 10 atm ppm Fe, 5 atm ppm Mo, 4 atm ppm Mg, and 3 atm ppm Si (separate source mass-spectrometry and neglecting 1230 atm ppm O+H); $\rho_{11}/\rho_{33} = 37.6$; $\rho_{11} = 19.61 \times 10^{-8} \Omega \cdot m$; specimen diameter 0.263 cm diameter and 2.5 cm length; data extracted from figure.
14	16	Jung, W.D.	1975	A	6-265	Sample 2	Same as above except 570 atm ppm (O+H); $\rho_{11}/\rho_{33} = 81.5$ and $\rho_{11} = 18.72 \times 10^{-8} \Omega \cdot m$; specimen diameter 0.260 cm diameter and 3.47 cm length; data extracted from figure.
15	16	Jung, W.D.	1975	A	5-283	Sample 3	Same as above except 55 atm ppm O+H, 100 atm ppm Cr+V, 12 atm ppm W, 13 atm ppm Fe, 14 atm ppm Cl, and 8 atm ppm Mg; no evidence of an impurity gradient; large concentration of Cr+V likely due to surface hydrocarbon contamination not representative of sample; $\rho_{11}/\rho_{33} = 785$ and $\rho_{11} = 18.69 \times 10^{-8} \Omega \cdot m$; specimen diameter 0.205 cm diameter and 3.65 cm length; data extracted from figure.
16	16	Jung, W.D.	1975	A	5-276	Sample 4	Same as above except 28 atm ppm O+H; $\rho_{11}/\rho_{33} = 1524$ and $\rho_{11} = 18.90 \times 10^{-8} \Omega \cdot m$; specimen diameter 0.241 cm diameter and 4.3 cm length; data extracted from figure.
17	19	Aleksandrov, B.H., Semenova, E.B., Petrova, O.I., Cheryi, B.P., and Akhachev, V.M.	1975	A	5-300	Specimen No. 1	Polycrystalline; purest sample they studied is 1.4 mm diameter and 25-60 mm length; $\rho_0 = 0.0129 \times 10^{-8} \Omega \cdot m$; data extracted from figure.
18	19	Aleksandrov, B.H., et al.	1975	A	6-67	Specimen No. 4	Similar to above except $\rho_0 = 0.067 \times 10^{-8} \Omega \cdot m$; least pure sample which studied; data extracted from figure.
19	20	Akhachev, V.M., Volkenshchkin, B.V., Startsev, V.Ye., Fisbal, V.A., Cheremny, V.I., and Cheryi, B.P.	1976	A	5-270	V4	High purity sample of $\rho_{100}/\rho_0 = 1520$ prepared by complex method includes refining by vacuum electron beam melting and electron transfer; total impurities $< 3 \times 10^{-3}$ %, gas impurities 1%, and <1% hydrogen; superconducting transition temperature $T_c = 5.39$ K; error of the measurement 0.5% for $T < 15$ K and 0.01% for $T > 70$ K; anomaly near 183 K was observed; resistivity contains contribution proportional to fourth power of the temperature; these peculiarities are intensified as the purity of sample increases; data extracted from figure.

TABLE 2. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF VAMADIN V (continued)

Data Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Mass and Specimen Designation	Composition (weight percent), Specifications and Remarks
20	20	Ashaba, V.M., et al.	1976	A	5-272	V6	Same as above except $D_{212}/D_0 = 220$ and $T_c = 5.52$ K.
21*	21	Alexeevskii, H.E., Mits, A.V., and Matveeva, H.M.	1975	→	300		99.9 V; resistance measured using electronic amplifier with x-y recorder.
22	22	Cesariyev, A., Righini, F., and McClure, J.L.	1974	Y	293-2100		99.9 V; polycrystalline; from Materials Research Corp.; 120 ppm C, 20 ppm Fe, 60 ppm Nb, 10 ppm W, 15 ppm O, 15 ppm P, 50 ppm Si, 70 ppm Ti, 10 ppm Ta, 30 ppm V, 15 ppm Zr, other total less than 50 ppm; tube made from rod by electro-erosion, 6.3 mm diameter (outside), 76.26 mm long; density 6.1 g cm^{-3} ; heat treated by pulse heating -30 pulses to 1900 K; 0.5% estimated total error in measurement; experimental vacuum $\sim 10^{-8}$ torr.
23*	23	Beckett, C.W.	1974				
23*	25	Humpal, K. and Ohtsuka, T.	1974	A	300		99.95 V from Material Research Corp. (V-P grade); method is electron beam furnace at pressures below 10^{-5} torr to outgas sample; $T_c = 5.20$ K; $D_{100}/D_{0.1} = 20.0$
24	26	Prehal, A.F., Buseckhin, V.A., and Polunskaya, N.V.	1974	A	5-267		No details are given; data extracted from figure.
25*	27	Lung, E. and Broecker, J.	1975	A	77,293	V811	Single crystals of [491] orientation; <10 ppm O_2 , <5 ppm of other interstitials and substitutionals; prepared by electron beam melting under UHV conditions, annealed at 1373 K; $D_{100}/D_{0.1} = 8.59$; ideal resistivity ratio 0.116; results of oxygen doping of V crystals indicated a linear increase of resistivity with increasing O_2 content.
26	28	Reimark, B.E., Belyubova, P.E., Bredikhin, B.H., Vorontsov, L.K., Korytina, S.P., and Mikhailov, A.H.	1973	→	293-1773	V8L2	99.82 V, 0.05 Al, 0.02 Ni, 0.01 Fe, 0.026 C, 0.003 Si, 0.07 O; specimen of V fused by electron beam in vacuum from pressed powder; annealed at 900°C in vacuum of 10^{-6} mm Hg and at 1540°C of 10^{-4} mm Hg; resistivity in the range 20-1100°C measured by Jaeger-Dieselhorst method and in the range 900-1400°C by Bode method; agreement between these two measurements is 15% within minimum error of measurements; resistivity value at 293 K increased from $21.3 \times 10^{-8} \Omega \text{ m}$ to $27.3 \times 10^{-8} \Omega \text{ m}$ after heating the specimen to 1100°C; data extracted from smooth tabulated values.
27	29	Chernoplev, H.A., Pomova, G.M., Samuilov, B.N., and Shibov, A.A.	1973		5-1032		Pure V (no purity or source mentioned); sample rod 60 mm long with cross section 0.7×0.7 mm; values extracted from smooth values from small figure.
28*	30	Arutyunov, A.V., Mikharukho, I.H., and Filippov, L.F.	1972		1000-1900		95.72 V, 0.13 Al, 0.09 Si, 0.05 Fe, 0.04 C, 0.055 O, 0.001 H, and 0.01 W; annealed in vacuum at 1600 K for 2 hr; sample 12 mm diameter and 90 mm length; the data reported here appeared to be same as in data set 25.

*Not shown in figure.

TABLE 2. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF VANADIUM V (continued)

Data Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
29	31	Filippov, L.P. and Yurchak, R.P.	1971	A	1000-1900		99.72 V, 0.13 Al, 0.09 Si, 0.05 Fe, 0.005 O, 0.04 C, 0.01 Ni; polycrystalline; solid and hollow rod; 90 mm length and 12 mm diameter; data extracted from smooth tabulated values; error is 2%.
30	32	Polatskii, V.E., Drenskina, V.P., and Sobol, Ya.G.	1971	A	293-1800		99.94 V, <0.001 Al, <0.001 Ni, <0.001 Fe, <0.046 O, <0.01 H, <0.001 Si; polycrystalline; density 6.099 g cm^{-3} ; specimen machined from a rod produced by electron beam melting in vacuum; specimen dimensions 10 mm diameter \times 60 mm length; measurements in vacuum of 10^{-4} torr; measurements error 1.8-2.0%; data extracted from smooth tabulated values.
31	33	L'vov, S.B., Mal'ko, P.I., and Munchenko, V.P.	1971		341-1381		99.9 V.
32	34	Voronin, L.K., Merkulov, A.B., and Reimark, B.E.	1970	A	283-1548	VEL2	99.82 V, 0.01 Fe, 0.02 Ni, 0.05 Al, 0.003 Si, 0.07 O, 0.001 H, <0.001 N, 0.024 C; electron beam melting of pressed powder; annealed at 1173 K; 1×10^{-4} mm Hg for 1 hr before measurements; sample size 150 mm \times 6 mm diameter; measurements made by Jeaget-Discalhorst method.
33	34	Voronin, L.K., et al.	1970	A	1591-1727	VEL2	Similar to the above except sample size 70 mm \times 2 mm diameter; measurements made at 2×10^{-4} mm Hg by Bode method.
34	35	Rensler, D.H., Ross, A.R., and Falls, E.H.	1970	A	293		Film deposited on sapphire substrate by sputtering from V cathode; substrate held at 673 K during sputtering and for 30 minutes post deposition annealing in vacuum and cooled slowly over several hours; thickness of film 1970 Å; temperature of measurements not reported but assumed to be 293 K.
35	35	Rensler, D.H., et al.	1970	A	293		Film deposited on sapphire substrate by sputtering from V cathode in oxygen 10^{-4} torr; thickness of film 1950 Å; other specifications are same as above.
36	36	Buehler, U.	1969		11-1090		Pure V, 0.08 O, 0.046 H, and 0.044 C; fused by electron beam; sample 80 mm long and 5 mm diameter; data extracted from figure.
37	37	Westlake, B.G. and Alfred, L.C.R.	1968	A	6-350		No details are given.
38	38	Westlake, B.G.	1967	A	5-338		Crystals of electrolytic vanadium from U.S. Bureau of Mines; 230 ppm metallic impurities, 20 ppm C, 100 ppm H, 290 ppm O; crystals electron-beam melted into ingot, rolled to 0.64 mm strips, 60 mm long \times 4.2 mm wide cut from sheet, and both rolled surfaces were ground on wet 600-grit SiC paper to produce specimen 0.4 mm thick; specimens were wrapped in Mo foil, vacuum encapsulated in quartz, annealed 4 hr at 1273 K; annealed further in dynamic vacuum 2×10^{-4} torr for 30 minutes at 1073 K for dehydrogenation; data extracted from figure.

TABLE 2. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF VANADIUM V (continued)

Data Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
39	39	Verbeiner, H.R. and Gilchrist, J.G.	1967	R	4.2	V1	Specimen from Isply Kulmann; 0.3% total impurity; 87% cold drawn.
40	39	Verbeiner, H.R. and Gilchrist, J.G.	1967	R	4.2	V2	Same as above except 97% cold drawn.
41	40	Amitin, E.B., Kovalovskaya, Yu.A., and Kovdya, Yu.Z.	1967		16-299	Sample 1	99.63 V; polycrystalline; 13.1 x 3.7 x 0.8 mm plate prepared by cutting with corundum disk under emulsion layer subjected to 10 ⁵ atm pressure at 293 K to suppress possible porosity; sample annealed in 10 ⁻⁶ mm Hg at 1123 K for 5 hr; density 6.2 g cm ⁻³ ; $\rho_{273}/\rho_0 = 11.5$; data obtained from ρ_T/ρ_{273} from figure and $\rho_{273} = 24.1 \times 10^{-9} \Omega \text{ m}$ reported by authors.
42	40	Amitin, E.B., et al.	1967		131-277	Sample 2	Sample supplied by Metal Physics Institute of Academy of Sciences of the USSR; $\rho_{273}/\rho_0 = 15$; data obtained from ρ_T/ρ_{273} from figure and $\rho_{273} = 23.6 \times 10^{-9} \Omega \text{ m}$ from Mathieson's rule.
43	41	Van Gorp, G.J.	1967	R	5.1		99.9 V, 0.05 Si, 0.03 Fe, 0.04 Ti, 0.1 O, 0.06 H; specimen from A. C. Mackay Ltd.; in the form of sheet that was some melted and cold rolled to 30 μ thickness; resistance measured by Keithly D.C. Amplifier amplifying voltage output of sample due to varying magnetic field; $\rho_{273}/\rho_{10} = 10$.
44	41	Van Gorp, G.J.	1967	R	5.2		Same as above except annealed at 10 ⁻⁶ torr at 1600°C; $\rho_{273}/\rho_{10} = 15$.
45 ^a	42	Brushlins, J.P., Vladimirovskaya, T.M., and Praktevskikh, A.A.	1966	A	293		0.01-0.05 C, 0.03-0.05 O ₂ , 0.000-0.01 H ₂ , 0.2-0.22 Si, 0.27-0.65 Fe, 0.03-0.16 Al; 22 mm x 0.42 mm diameter rod forged from ingots at 1173-1323 K; specimen heated in He atmosphere prior to forging; samples annealed at 1273 K for 30 minutes; measurements in vacuum; measurement temperature not reported, however assumed to be 293 K.
46 ^a	42	Brushlins, J.P., et al.	1966	A	293		Same as above except specimen cold-hardened.
47 ^a	42	Brushlins, J.P., et al.	1966	A	293		Same as above except diameter 0.96 mm; annealed specimen.
48 ^a	42	Brushlins, J.P., et al.	1966	A	293		Same as above except specimen cold-hardened.
49 ^a	42	Brushlins, J.P., et al.	1966	A	293		Same as above except diameter 1.33 mm; annealed specimen.
50 ^a	42	Brushlins, J.P., et al.	1966	A	293		Same as above except specimen cold-hardened.
51 ^a	43	Mrs. G. Gebhardt, E., and Bürschel, W.	1965	K	273-1762		0.06 O ₂ , 0.01 H ₂ , 0.04 H ₂ ; fused by electron beam; 0.5 mm diameter wire 16 cm long; annealed at 1500°C for 15 minutes at 1.5 x 10 ⁻⁶ torr.
44		Mrs. G.	1946				Not shown in figure.

TABLE 2. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF VANADIUM V (continued)

Data Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Specimen Designation	Composition (weight percent), Specifications and Remarks
52	45	Taylor, M.A. and Smith, C.H.L.	1962	A	20-273	V(JM)	99.63 V; ingot from Johnson Matthey Co.; specimen cut to about 10 x 1 x 1 mm; degassed in alcohol; electrolytically polished in dilute H ₂ SO ₄ , rinsed, annealed at 1073 K for 5 hr in vacuum at 10 ⁻⁶ mm Hg, cooled and process repeated again; this was done to remove strains; accurate to $\pm 1\%$; error due to irregular cross sectional area.
53	45	Taylor, M.A. and Smith, C.H.L.	1962	A	20-273	V(BMI)	99.92 V; specimen from Battelle Memorial Institute; other specifications same as above.
54	45	Taylor, M.A. and Smith, C.H.L.	1962	A	20-273	V1(USNM)	99.85 V; specimen from U.S. Bureau of Mines; other specifications same as above.
55 ^a	45	Taylor, M.A. and Smith, C.H.L.	1962	A	20-273	V2(USNM)	Similar to the above.
56	46	Burger, J. and Taylor, M.A.	1961	A	224-246		99.9 V from Battelle Memorial Institute, Columbus, OH; 0.005 C, 0.001 Si, 0.001 Cr, 0.04 Fe, 0.005 Al, 0.001 Cu, 0.001 Ni, 0.008 W, 0.0020 O; $\rho_{100} = 23 \pm 1 \times 10^{-9} \Omega$; data extracted from figure.
57	47	Brown, J.A. and Wayman, C.H.	1960	A	126-282		99.7 V, Cu reduced; annealed at 950°C; degassed at 1500°C; 0.025 in. diameter, 8 cm long; heating cycle; no indication of sudden discontinuity but deviation from linearity at 200 K; data extracted from figure.
58	47	Brown, J.A. and Wayman, C.H.	1960	A	140-288		Same as above except cooling cycle; data extracted from figure.
59	48 49	White, G.K. and Wood, S.R.	1959	A	15-390	V4	99.9 V obtained from Electrometallurgical Co.; specimen diameter 3.55 mm; annealed in vacuum at 1573 K; residual resistivity $\rho_0 = 4.83 \times 10^{-9} \Omega$ m.
60 ^a	50	Semenov, G.V.	1957	V	295		Unspecified sample of V; thermal coefficient of electrical resistivity $+0.28\%/degree$.
61	51	Urbak, D. and Vert, C.	1955	-	93	V1	Polycrystalline; 0.14 C, 0.12 O, 0.11 Ni; bec structure; foil 0.2 cm wide, 0.008 cm thick, and 4 cm long; IR drop method.
62	51	Urbak, D. and Vert, C.	1955	-	93	V2	Same as above.
63 ^a	52	Potter, H.H.	1941	A	273		Irregular pellets; specimen dimensions of 0.6 mm square and 6 mm in length.
64	53	Genstrom, G.J., Zaslowski, J.E., Williams, T.V., Weinstock, H., and Schmidt, F.A.	1961	C	3.92-298.0		Sample prepared using electrotransport technique; annealing time 800 hr, cross section was reduced to 0.85 mm square from cylinder 1.6 cm long and 2 mm diam; this was done to remedy too low signal to noise ratio; $\rho_{100}/\rho_0 = 1970$ and $\rho_{100}/\rho_{300} = 1770$, $\rho_0 = 0.01 \times 10^{-9} \Omega$ m; superconducting transition temperature, $T_c = 3.46 \pm 0.02$ K which was suppressed by 0.6T field produced by superconducting solenoid;

^aSee above in figure.

TABLE 2. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF VANADIUM V (continued)

Data Set No.	Author(s)	Year	Method Used	Temp. Range, K	Mass and Specimen Designation	Composition (weight percent), Specifications and Remarks
64 53 (cont.)	Guetron, G.J., et al.	1961	C	3.92-298.0		additionally electron-electron scattering ($\rho_{ee} = 1.6 \pm 0.2 \times 10^{-11} \Omega \text{ m K}^{-2}$), electron-phonon interband scattering ($\rho_{id} = (2.6 \pm 0.3) \times 10^{-11} \Omega \text{ m K}^{-2}$), and electron-phonon intraband scattering $\rho_{ps} = (7.3 \pm 1.1) \times 10^{-10} \Omega \text{ m K}^{-2}$.
65 54	Tsai, C.L., Fogaly, R.L., Weinstock, H., and Schmidt, F.A.	1961	C	4.5-298.1	Sample I	Sample purified using electrotransport technique; RRR = 1760; $\rho_0 = 0.0109 \times 10^{-10} \Omega \text{ m}$; superconducting transition temperature $5.43 \pm 0.03 \text{ K}$; data extracted from figure.
66 54	Tsai, C.L., et al.	1961	C	4.6-90.5	Sample II	Similar to above except less pure and $\rho_0 = 0.261 \times 10^{-10} \Omega \text{ m}$; superconducting transition temperature 5.77 K ; data extracted from figure.
67 55	Zylov, B.R. and Groot, H.	1961	K	298.9-745.0		Sample (RRR ~ 400) received from Dr. J. Cook of National Research Council, Canada; density 6.095 g cm^{-3} .
68 56	L'vov, S.H. and Rumbachev, V.P.	1965	A	292-1670		99.98 V iodide vanadium; measurement in vacuum furnace 2×10^{-5} to $8 \times 10^{-4} \text{ mm Hg}$; data extracted from figure.
69 57	Polatshii, V.E.	1978	-	200-2100		Recommended values for pure V; values based on 1968-IPTS and corrected for thermal expansion; confidence interval of the values varied from -2.8% near room temperature to $1.6-2.0\%$ in the region 1800-2000 K.

That shown in figure.

TABLE 3. EXPERIMENTAL DATA ON THE ELECTRICAL RESISTIVITY OF VANADIUM V
 [Temperature, T, K; Electrical Resistivity, ρ , $10^{-8} \Omega \cdot m$]

T	ρ	T	ρ	T	ρ	T	ρ	T	ρ	T	ρ	T	ρ	T	ρ	T	ρ						
DATA SET 1																							
2175	134.9	500	34.4	137.2	12.8	136.8	10.3	76.9	2.3	218	14.87												
2200	135.1	700	44.2	161.8	13.4	141.3	10.8	82.3	2.7	225	15.51												
2300	136.7	900	59.35	168.6	14.1	145.9	11.4	97.4	4.0	228	15.77												
2400	137.3	1100	70.1	170.9	14.7	150.4	11.9	103.3	4.5	239	16.60												
2500	138.4	1300	80.3	177.7	15.2	155.0	12.6	108.7	5.1	242	16.79												
2600	139.6	1500	90.1	184.5	15.9	159.5	13.3	117.3	5.9	246	17.13												
2700	140.8	1700	99.2	182.2	16.2	164.0	13.5	126.5	6.7	254	17.73												
2800	142.1	1900	107.9	191.3	16.6	175.4	14.9	134.5	7.4	257	17.99												
3000	144.8	DATA SET 2																					
3200	147.6	300	24.22	200.4	17.7	184.5	16.2	156.1	9.4	262	18.40												
3400	150.6	500	34.3	209.6	18.4	191.4	16.3	174.4	11.0	266	18.82												
3600	153.9	700	38.3	214.1	18.9	193.6	17.4	184.6	11.9	269	19.01												
3800	157.2	900	50.7	220.9	19.7	202.7	18.11	195.9	13.0	279	20.76												
4000	160.8	1100	61.7	236.9	21.2	214.1	19.4	199.2	13.2	283	20.02												
DATA SET 3																							
1800	101.7	1100	71.5	243.7	21.2	230.0	20.9	213.2	14.4	DATA SET 11													
1950	108.7	1300	80.2	248.2	22.2	239.1	21.6	223.4	15.4	81	2.60												
2100	115.7	1500	88.2	252.8	23.1	252.8	22.5	230.4	15.9	100	4.26												
2190(1)	119.9	1700	96.0	259.6	23.4	252.8	22.8	246.0	17.6	130	6.96												
2190(2)	135.2	1900	103.8	266.7	24.3	261.9	24.0	248.7	17.8	154	9.19												
2250	136.2	2000	107.7	275.6	25.2	268.7	24.7	251.4	18.2	170	10.82												
2400	140.6	DATA SET 4																					
2550	146.6	11.4	0.69	282.4	25.8	277.8	25.5	257.9	18.7	191	12.35												
2700	143.2	25.1	0.69	291.5	26.8	286.9	26.2	260.5	18.9	199	12.91												
3000	149.0	38.9	1.04	300.6	27.7	300.6	27.4	270.2	20.0	215	14.34												
3150	152.3	52.6	1.74	DATA SET 5																			
3300	155.8	57.1	1.98	20.5	0.69	DATA SET 6																	
3450	159.5	66.2	2.67	23.1	0.81	DATA SET 7																	
3600	163.5	73.1	3.37	34.3	0.81	DATA SET 8																	
3750	167.7	82.2	4.18	38.3	0.81	95.8	3.9	101.2	4.4	DATA SET 12													
3900	172.2	86.7	4.76	45.7	1.40	101.2	4.4	112.5	5.4	5.4	0.012												
4050	176.9	82.2	4.18	59.4	2.09	112.5	5.4	116.2	5.8	84	2.90												
4200	181.9	86.7	4.76	70.8	2.90	132.4	7.2	141.0	8.1	84	2.90												
DATA SET 6																							
2550	140.6	11.4	0.69	75.3	3.60	141.0	8.1	177.1	11.2	84	2.90												
2700	143.2	25.1	0.69	84.5	4.53	179.8	11.4	179.8	11.4	86	3.05												
3000	149.0	38.9	1.04	86.7	4.53	188.9	12.2	188.9	12.2	91	3.54												
3150	152.3	52.6	1.74	91.3	5.11	193.8	12.7	193.8	12.7	113	5.49												
3300	155.8	66.2	2.67	98.1	5.80	203.5	13.6	203.5	13.6	124	6.59												
3450	159.5	73.1	3.37	102.7	6.50	229.3	16.1	229.3	16.1	136	7.60												
3600	163.5	82.2	4.18	109.5	7.19	237.4	16.9	237.4	16.9	157	9.56												
3750	167.7	86.7	4.76	118.6	8.00	257.3	19.5	257.3	19.5	164	10.16												
3900	172.2	86.7	4.76	123.1	8.93	263.2	19.4	263.2	19.4	178	11.78												
4050	176.9	107.2	6.85	130.0	9.40	271.9	20.4	271.9	20.4	200	14.08												
4200	181.9	114.0	7.66	132.2	9.63	276.2	20.7	276.2	20.7	215	14.72												
DATA SET 7																							
4.2	1.75	130.0	9.40	132.2	9.63	288.5	21.8	288.5	21.8														
293	27.8	132.2	9.63																				
DATA SET 8																							
300	23.9	132.2	9.63																				

Not shown in figure.

TABLE 3. EXPERIMENTAL DATA ON THE ELECTRICAL RESISTIVITY OF VANADIUM V (continued)

T	ρ	T	ρ	T	ρ	T	ρ	T	ρ	T	ρ	
DATA SET 12 (cont.)												
44.4	0.432	5.9	0.228	5.8	0.022	90.3	2.57	69.40	1.72	20.2	0.0541	
55.6	0.796	6.2	0.228	6.7	0.024	99.9	4.22	77.30	2.31	22.9	0.0844	
76.5	2.16	7.0	0.228	8.0	0.024	112.2	5.39	300	19.6	27.2	0.143	
81.6	2.53	7.5	0.228	9.3	0.024	130.4	6.97	DATA SET 18				
87.0	3.06	8.1	0.250	10.9	0.028	150.4	8.83	6	0.861	31.4	0.223	
92.8	3.48	8.5	0.230	36.2	0.236	161.9	9.89	7	0.879	35.7	0.340	
96.9	3.84	9.2	0.230	39.2	0.309	176.4	11.1	10	0.879	38.6	0.438	
105	4.51	10.0	0.232	43.1	0.431	184.2	11.8	13	0.879	45.6	0.708	
112	5.29	10.9	0.234	48.7	0.596	197.9	12.9	14	0.879	54.1	1.13	
120	6.02	11.9	0.236	54.7	0.853	215.4	14.3	17	0.879	59.8	1.48	
136	6.85	12.9	0.239	58.4	1.07	229.5	15.5	14	0.845	65.4	1.90	
145	8.05	14.1	0.247	77.7	2.39	247.3	16.8	17	0.902	69.7	2.18	
155	8.86	15.8	0.250	82.1	2.61	260.3	17.9	20	0.902	73.9	2.39	
DATA SET 13												
5.6	0.51	19.1	0.259	90.3	3.38	261.8	18.0	22	0.914	75.3	2.74	
6.6	0.51	20.9	0.269	100.7	4.30	274.8	19.0	25	0.932	82.4	3.23	
7.3	0.52	22.8	0.280	113.3	5.50	DATA SET 17					86.6	3.65
8.1	0.52	24.9	0.296	130.8	7.08	5.49	0.0133	27	0.949	92.3	4.14	
9.2	0.52	27.9	0.322	154.5	9.27	5.95	0.0136	28	0.967	95.1	4.56	
10	0.52	30.4	0.354	170.9	10.7	7.21	0.0140	30	0.991	100.7	5.06	
11	0.52	33.8	0.407	191.3	12.5	10.10	0.0162	32	1.020	106.4	5.48	
12	0.52	36.2	0.460	217.6	14.6	13.08	0.0200	35	1.068	113.4	6.11	
13	0.52	39.4	0.533	250.7	17.1	15.99	0.0255	37	1.115	117.7	6.53	
14	0.53	41.7	0.581	283.4	19.7	19.56	0.0363	38	1.162	120.5	6.95	
15	0.53	48.3	0.778	DATA SET 16					39	1.186	129.0	7.58
16	0.54	54.3	1.06	5.4	0.009	19.56	0.0473	42	1.245	133.2	8.07	
17	0.54	73.2	2.24	5.8	0.009	22.00	0.0605	43	1.293	144.5	8.63	
18	0.54	79.9	2.72	6.3	0.013	24.34	0.0605	45	1.346	150.1	9.61	
22	0.57	86.2	3.27	7.9	0.013	26.03	0.0734	46	1.388	158.6	10.2	
49	1.14	88.1	3.38	7.9	0.013	29.99	0.112	47	1.441	174.1	11.8	
54	1.33	93.3	3.82	8.0	0.013	31.34	0.131	DATA SET 19				
57	1.51	107.0	5.03	9.3	0.013	33.35	0.154	5.5	0.00959	181.2	12.4	
62	1.80	111.1	5.36	11.2	0.017	35.98	0.198	5.5	0.00589	185.4	12.8	
66	2.10	126.7	7.84	12.1	0.018	37.00	0.226	5.5	0.00589	192.5	13.5	
96	4.51	139.3	7.99	13.6	0.020	38.01	0.244	5.7	0.0163	198.1	13.7	
109	5.72	166.8	10.2	15.2	0.022	39.23	0.272	6.1	0.0163	206.6	14.8	
125	7.18	177.5	11.2	16.2	0.026	40.09	0.294	6.5	0.0168	215.1	15.3	
148	9.27	186.8	12.9	17.9	0.029	41.92	0.346	7.2	0.0168	227.8	16.2	
182	12.1	196.3	12.8	20.1	0.039	42.88	0.370	7.9	0.0174	240.5	17.3	
242	17.1	203.5	13.3	22.9	0.055	44.00	0.400	9.1	0.0182	246.1	17.9	
245	17.2	213.9	14.2	23.8	0.077	44.90	0.431	10.7	0.0193	256.0	18.7	
273	19.5	231.4	15.6	29.7	0.119	46.12	0.471	11.9	0.0206	270.2	19.6	
DATA SET 20												
5.5	0.226	265.9	18.2	35.9	0.225	46.92	0.503	12.9	0.0223	5.65	0.083	
				39.2	0.296	47.58	0.529	13.8	0.0239	5.67	0.088	
				42.7	0.417	60.15	1.33	14.7	0.0255	5.68	0.091	
				62.1	1.33	64.20	1.40	15.8	0.0311			
				66.6	1.62	67.40	1.58	17.9	0.0412			

TABLE 3. EXPERIMENTAL DATA ON THE ELECTRICAL RESISTIVITY OF VANADIUM V (continued)

T	ρ	T	ρ	T	ρ	T	ρ	T	ρ	T	ρ		
DATA SET 28 (cont.)													
6.72	0.091	220.9	15.7	49.5	2.10	659	46.9	485.6	30.6	1177	74.0		
7.76	0.093	223.8	16.0	54.7	2.53	789	54.5	591.8	33.9	1218	75.6		
8.94	0.093	234.1	16.6	62.9	3.38	875	60.5	614.6	33.9	1275	78.1		
9.69	0.093	239.9	17.2	69.2	4.02	975	65.4	720.9	38.3	1317	80.1		
10.7	0.094	244.3	17.5	76.4	4.87	1032	69.2	796.8	40.5	1455	87.1		
11.8	0.096	251.6	18.0	96.0	7.00	DATA SET 28P							
12.4	0.096	254.5	18.4	105.4	8.50	933.4	44.9	880.4	43.8	1507	89.2		
13.6	0.098	272.0	19.6	115.7	9.78	1047.4	49.2	1047.4	49.2	1546	90.8		
14.8	0.10	DATA SET 21*											
15.8	0.10	300	23	140.5	12.8	1000	68.0	1130.9	52.5	DATA SET 33			
16.7	0.10	300	23	157.0	14.9	1100	71.4	1206.8	54.7	1591	92.3		
17.8	0.10	300	23	172.5	16.6	1200	78.5	1267.8	56.9	1626	94.5		
18.8	0.11	DATA SET 22											
19.9	0.11	293	21.72	183.9	18.3	1300	83.5	1381.2	61.3	1690	97.1		
20.6	0.13	293	21.72	195.3	19.6	1400	88.2	DATA SET 32					
21.0	0.13	1500	87.66	208.7	21.3	1500	92.8	283	20.9	293	241		
22.5	0.20	1500	89.01	228.4	23.6	1600	97.5	301	22.5	DATA SET 34			
28.4	0.27	1550	99.01	243.9	25.1	1700	105.6	320	23.5	293	75.5		
34.3	0.44	1600	91.93	251.1	26.0	1800	109.5	337	25.0	DATA SET 35			
43.1	0.81	1650	94.03	257.3	26.8	DATA SET 29							
48.9	1.1	1700	95.83	267.7	27.9	1000	68.0	359	26.4	293	241		
57.8	1.3	1750	97.06	DATA SET 25*									
60.7	1.6	1800	99.87	77	2.283	1000	73.4	402	28.9	DATA SET 36			
73.9	2.5	1850	101.69	293.2	19.62	1100	78.5	419	30.4	11	3.1		
86.9	3.7	1900	103.56	DATA SET 26									
96.9	4.7	1950	105.40	373	27.0	1200	83.5	442	31.7	77	5.9		
98.6	4.7	2000	107.28	465	33.5	1300	88.2	482	34.4	273	22.0		
101.5	5.1	2050	109.01	565	40.0	1400	92.8	500	35.8	293	23.8		
116.1	6.4	2100	110.70	600	42.1	1500	97.5	520	36.9	410	31.4		
121.9	6.8	DATA SET 23*						1600	101.5	541	468	35.1	
124.8	7.2	300	22.6	617	43.2	1700	105.6	565	40.0	575	41.4		
130.7	7.6	300	22.6	631	45.1	DATA SET 30							
136.5	8.2	300	22.6	684	46.9	293	21.02	617	43.2	680	46.6		
140.8	8.7	DATA SET 24						1200	73.3	651	45.1	773	53.5
146.6	9.2	5.2	1.07	1373	72.6	1300	78.8	684	46.9	779	52.2		
152.5	9.8	7.2	1.28	1373	81.8	1400	83.7	730	50.2	875	57.7		
159.7	10.3	10.3	1.28	1373	90.8	1500	88.5	796	53.9	925	61.4		
165.6	10.9	13.5	1.28	1773	99.8	1600	93.0	822	55.3	979	62.1		
172.9	11.5	17.6	1.27	DATA SET 27									
178.7	11.9	21.7	1.27	5.2	0.92	1700	97.4	849	56.9	DATA SET 37			
183.1	12.4	24.8	1.27	65	2.44	1800	101.6	876	58.3	6.5	0.81		
187.4	13.0	28.9	1.26	129	8.47	DATA SET 31							
194.2	13.5	34.1	1.69	229	17.16	293	25.1	907	59.9	22	0.85		
203.4	14.3	39.2	1.68	259	25.4	341.2	25.1	932	61.1	56	1.89		
212.2	14.9	44.4	1.89	337	25.4	386.9	26.2	962	62.7	59	2.05		
218.0	15.4	530	39.0	400	30.0	417.2	28.4	1051	67.6	59	2.05		
				530	39.0			1087	69.5	115.6	6.85		

*Not shown in figure.

TABLE 3. EXPERIMENTAL DATA ON THE ELECTRICAL RESISTIVITY OF VANADIUM V (cont. Inued)

T	ρ	T	ρ	T	ρ	T	ρ	T	ρ	T	ρ	T	ρ
DATA SET 37 (cont.)													
134.4	0.56	77	4.9	184	15.3	293	28.0	20	1.56	220	22.0	15	4.84
169.8	11.72	89	6.1	187	15.0	DATA SET 49*		77	3.98	232	22.9	20	4.87
191.0	13.55	100	7.1	190	16.0	DATA SET 49*		273	20.34	242	23.6	30	5.21
198.3	14.21	108	8.1	195	16.4	293	24.6	DATA SET 55*		253	24.5	40	5.57
213.5	15.48	116	8.7	198	16.6	DATA SET 50*		DATA SET 55*		261	25.0	50	7.13
233.6	17.16	133	9.5	201	16.0	293	28.6	20	1.54	282	26.7	100	9.08
242.6	17.89	136	9.8	204	17.2	DATA SET 51*		77	3.96	DATA SET 58		150	13.5
260.8	19.33	134	10.7	207	17.4	293	28.6	273	20.39	140	15.1	200	17.5
278.0	20.88	141	11.6	210	17.8	DATA SET 52*		DATA SET 56		148	15.9	250	21.5
291.0	21.69	144	11.8	214	18.2	273	20.5	224	18.4	156	16.7	273	23.1
309.6	23.10	151	12.5	216	18.2	283	22.0	225	18.5	170	17.3	288	27.1
333.6	24.90	155	12.8	222	18.8	1187	73.7	226	18.7	178	18.5	DATA SET 59	
349.6	26.13	160	13.3	223	19.0	1302	80.0	228	18.8	187	19.3	15	4.84
DATA SET 38													
5	0.82	171	14.1	225	19.1	1378	80.0	229	19.0	190	19.4	20	4.87
20	0.82	175	14.8	227	19.4	1437	86.2	230	19.1	190	19.4	30	5.21
40	1.3	179	15.2	234	19.9	1446	86.2	232	19.2	197	20.0	40	5.57
60	2.1	183	15.5	245	21.1	1493	88.7	233	19.4	207	20.9	50	7.13
80	3.4	192	16.5	249	21.5	1537	91.2	236	19.7	227	22.5	100	9.08
121	7.4	199	17.2	258	22.2	1542	90.6	237	19.8	238	23.2	200	17.5
161	11.2	206	18.0	261	22.6	1584	93.2	238	19.9	247	24.1	250	21.5
201	14.5	212	18.4	264	23.0	1643	95.4	241	20.1	256	24.8	273	23.1
241	17.8	220	19.1	266	23.0	1683	97.1	242	20.2	267	25.5	295	24.7
280	20.9	226	19.6	269	23.3	1724	98.8	243	20.3	273	25.9	390	31.4
300	22.5	229	20.0	DATA SET 43		1752	101.0	245	20.5	DATA SET 59			
320	23.8	234	20.4	5.07	2.6	1762	100.8	246	20.6				
338	25.3	239	20.8	DATA SET 44		DATA SET 32		DATA SET 57					
DATA SET 39													
4.2	3.9	244	21.3	5.15	1.7	20	4.0	126	13.8	15	4.84	20	4.87
DATA SET 40													
4.2	4.4	249	21.7	273	22.67	77	6.48	132	14.5	30	4.87	30	4.87
DATA SET 41													
16	1.9	256	22.3	DATA SET 45		273	22.67	130	14.9	40	5.21	40	5.21
21	2.1	262	22.9	293	23.6	DATA SET 46*		145	15.6	50	5.57	50	5.57
26	2.1	265	23.2	DATA SET 46*		20	0.74	151	16.2	100	9.08	100	9.08
38	2.4	275	24.0	293	28.9	77	3.18	130	14.9	150	13.5	150	13.5
47	2.7	287	25.0	DATA SET 47*		273	19.54	145	15.6	200	17.5	200	17.5
56	3.2	299	26.0	293	23.8	DATA SET 47*		151	16.2	250	21.5	250	21.5
68	4.1	131	10.1	DATA SET 42		20	0.74	167	17.5	273	23.1	273	23.1
DATA SET 42													
131	10.1	144	11.7	293	28.9	77	3.18	174	18.2	182	18.9	182	18.9
144	11.7	155	12.7	DATA SET 47*		273	19.54	182	18.9	192	19.7	192	19.7
155	12.7	161	13.3	293	23.8	DATA SET 47*		192	19.7	203	20.5	203	20.5
161	13.3	168	13.9	DATA SET 47*		DATA SET 47*		203	20.5	213	21.3	213	21.3
168	13.9	173	14.3	DATA SET 47*		DATA SET 47*		DATA SET 47*		DATA SET 47*		DATA SET 47*	
173	14.3	178	14.9	DATA SET 47*		DATA SET 47*		DATA SET 47*		DATA SET 47*		DATA SET 47*	

*not shown in figure.

TABLE 3. EXPERIMENTAL DATA ON THE ELECTRICAL RESISTIVITY OF VANADIUM V (continued)

T	ρ	T	ρ	T	ρ	T	ρ	T	ρ	T	ρ	T	ρ	T	ρ
DATA SET 60*															
293	26.0	6.348	0.0114	13.494	0.0195	8.97	0.0135	44.13	0.395	83.60	2.75	DATA SET 66 (cont.)			
DATA SET 61															
93	17.96	6.318	0.0114	13.678	0.0200	8.96	0.0147	53.89	0.0147	90.52	3.22	DATA SET 65 (cont.)			
DATA SET 62															
93	16.49	6.756	0.0116	14.306	0.0212	9.72	0.0147	57.22	0.936	DATA SET 67					
DATA SET 63*															
273	18.2	6.842	0.0117	14.395	0.0215	9.91	0.0159	81.93	2.49	298.9	21.00	DATA SET 68			
DATA SET 64 (cont.)															
3.923	0.0105	6.963	0.0117	14.550	0.0218	10.73	0.0158	85.28	2.92	330.0	23.32	DATA SET 69*			
4.145	0.0106	7.066	0.0118	14.874	0.0225	11.17	0.0178	204.4	11.2	347.1	24.10	DATA SET 70			
4.175	0.0105	7.140	0.0119	15.033	0.0228	11.62	0.0171	264.5	15.4	362.0	25.56	DATA SET 71			
4.216	0.0106	7.185	0.0118	15.188	0.0231	12.58	0.0185	298.1	19.1	368.4	25.57	DATA SET 72			
4.270	0.0106	7.203	0.0118	15.207	0.0233	12.58	0.0197	DATA SET 73							
4.289	0.0106	7.229	0.0119	15.702	0.0247	13.62	0.0197	DATA SET 74							
4.316	0.0107	7.592	0.0121	16.273	0.0260	14.17	0.0213	DATA SET 75							
4.377	0.0107	7.607	0.0122	16.350	0.0264	14.45	0.0230	DATA SET 76							
4.849	0.0109	7.723	0.0122	16.406	0.0263	14.46	0.0249	DATA SET 77							
4.873	0.0107	7.925	0.0122	16.664	0.0273	15.34	0.0239	DATA SET 78							
4.993	0.0108	7.946	0.0124	17.032	0.0282	15.65	0.0254	DATA SET 79							
5.004	0.0108	7.994	0.0124	17.155	0.0285	16.94	0.0285	DATA SET 80							
5.185	0.0109	8.208	0.0125	17.158	0.0286	16.95	0.0309	DATA SET 81							
5.282	0.0109	8.331	0.0127	17.309	0.0291	17.63	0.0328	DATA SET 82							
5.294	0.0109	8.433	0.0129	17.509	0.0315	19.06	0.0315	DATA SET 83							
5.367	0.0109	8.628	0.0139	18.128	0.0319	19.09	0.0347	DATA SET 84							
5.414	0.0110	8.776	0.0130	18.541	0.0332	19.86	0.0376	DATA SET 85							
5.467	0.0110	8.768	0.0131	18.647	0.0338	20.67	0.0406	DATA SET 86							
5.609	0.0111	9.261	0.0135	18.541	0.0332	20.67	0.0406	DATA SET 87							
5.705	0.0110	9.318	0.0136	19.577	0.0335	21.51	0.0440	DATA SET 88							
5.802	0.0112	9.441	0.0137	19.782	0.0381	22.38	0.0466	DATA SET 89							
6.199	0.0114	9.428	0.0137	20.133	0.0396	22.83	0.0494	DATA SET 90							
6.268	0.0114	9.728	0.0139	20.320	0.0406	23.30	0.0567	DATA SET 91							
6.275	0.0113	9.793	0.0137	20.024	0.0415	23.77	0.0602	DATA SET 92							
6.456	0.0114	10.066	0.0144	20.950	0.0406	23.77	0.0614	DATA SET 93							
DATA SET 94															
3.487	0.0110	10.156	0.0144	20.950	0.0406	23.77	0.0614	DATA SET 95							
5.618	0.0111	10.336	0.0147	21.10	0.0435	25.73	0.0664	DATA SET 96							
5.705	0.0110	10.336	0.0147	21.10	0.0435	25.73	0.0664	DATA SET 97							
5.802	0.0112	10.371	0.0146	21.30	0.0435	25.73	0.0664	DATA SET 98							
6.199	0.0114	10.571	0.0150	21.30	0.0435	25.73	0.0664	DATA SET 99							
6.268	0.0114	10.792	0.0153	21.30	0.0435	25.73	0.0664	DATA SET 100							
6.275	0.0113	10.828	0.0153	21.30	0.0435	25.73	0.0664	DATA SET 101							
6.456	0.0114	10.863	0.0153	21.30	0.0435	25.73	0.0664	DATA SET 102							
DATA SET 103															
3.487	0.0110	11.295	0.0160	21.30	0.0435	25.73	0.0664	DATA SET 104							
5.618	0.0111	11.585	0.0164	21.30	0.0435	25.73	0.0664	DATA SET 105							
5.705	0.0110	11.833	0.0167	21.30	0.0435	25.73	0.0664	DATA SET 106							
5.802	0.0112	11.833	0.0167	21.30	0.0435	25.73	0.0664	DATA SET 107							
6.199	0.0114	12.085	0.0170	21.30	0.0435	25.73	0.0664	DATA SET 108							
6.268	0.0114	12.216	0.0172	21.30	0.0435	25.73	0.0664	DATA SET 109							
6.275	0.0113	12.448	0.0176	21.30	0.0435	25.73	0.0664	DATA SET 110							
DATA SET 111															
3.487	0.0110	12.448	0.0181	21.30	0.0435	25.73	0.0664	DATA SET 112							
5.618	0.0111	12.725	0.0184	21.30	0.0435	25.73	0.0664	DATA SET 113							
5.705	0.0110	12.903	0.0184	21.30	0.0435	25.73	0.0664	DATA SET 114							
5.802	0.0112	12.957	0.0185	21.30	0.0435	25.73	0.0664	DATA SET 115							
6.199	0.0114	13.160	0.0189	21.30	0.0435	25.73	0.0664	DATA SET 116							
6.268	0.0114	13.416	0.0194	21.30	0.0435	25.73	0.0664	DATA SET 117							
6.275	0.0113	13.416	0.0194	21.30	0.0435	25.73	0.0664	DATA SET 118							
6.456	0.0114	13.416	0.0194	21.30	0.0435	25.73	0.0664	DATA SET 119							

*Not shown in figure.

TABLE 3. EXPERIMENTAL DATA ON THE ELECTRICAL RESISTIVITY OF VANADIUM V (continued)

T	ρ
DATA SET 69 (cont.) ^a	
1100	67.84
1200	72.76
1300	77.52
1400	82.16
1500	86.70
1600	91.17
1700	95.59
1800	99.99
1900	104.84
2000	108.84
2100	113.35

^aNot shown in figure.

3.2. Zirconium

There are 43 data sets available from 23 references [33,49,123-144] for the electrical resistivity of zirconium specimens with purity 99.8-99.99%. The temperature range covered by these data sets is from 1.7 to 2127 K. The information on specimen characterization and measurement condition for each of the data sets is given in table 5. The data sets are tabulated in table 6 and shown partially in figure 4.

From liquid-helium temperature to room temperature the only set of data for high-purity zirconium is that of White and Woods [49] (data set 27) on a specimen with RRR = 168. Above 100 K these data appear to be trustworthy, but their reliability below 100 K is not sufficient to permit reliable interpretation in terms of any low-temperature conduction mechanism. However, White and Woods pointed out a $T^{4.5}$ dependence of the temperature-dependent resistivity above 13 K as indicating rather strong electron-phonon s-s interband scattering. This and earlier work of Kemp et al. [141] (data set 31) on a specimen with RRR = 25 was supported fifteen years later by Volkenshtein et al. [131] (data set 12) using a specimen with RRR = 34. Furthermore, the data of Volkenshtein et al. [131] suggested the existence of a T^2 term below 13 K which was undoubtedly related to electron-electron scattering. T^3 dependence indicative of s-d electron-phonon scattering was neither explored nor reported by these or other low-temperature studies [131-137,140]. Careful low-temperature studies on a very pure specimen is required to detect such dependence.

The recommended values below 293 K are based on the data of White and Woods [49] (data set 27), who studied the purest specimen ($\rho_0 = 0.25 \times 10^{-8} \Omega \text{ m}$).

In the temperature range up to $T_{\alpha-\beta} = 1137 \text{ K}$ there appears to be fairly good agreement ($\pm 10\%$) among the data of Bykov et al. [127] (data set 7), L'vov et al. [33] (data set 13), Peletskii et al. [133] (data set 15), Powell and Tye [138] (data sets 22-24), Bing et al. [143] (data set 37), and of Cook et al. [144] (data set 38). The recommended values up to 800 K are based on the data of Peletskii et al. [133] (data set 15). In the temperature range from 800 to 1137 K the recommendations were guided by the data of Cezairliyan and Righini [123,124] (data set 2), Peletskii et al. [133] (data set 15) and those of Kiselev [139] (data sets 25,26). Data of Cezairliyan and Righini [123-125]

(data sets 2-5) and those of Peletskii et al. [133] (data sets 15,16) were used to generate the recommended values for β -Zr between 1137 to 2127 K. The value of $141.3 \times 10^{-8} \Omega \text{ m}$ for liquid Zr at 212 K follows the only available data of Martynyuk and Tsapkov [129] (data set 10).

The recommended values of the electrical resistivity given in table 4 and shown in figures 5 and 6 are for zirconium of 99.95% purity or higher, but those below 100 K are applicable specifically to samples with $\rho_0 = 0.250 \times 10^{-8} \Omega \text{ m}$. The table gives both values uncorrected and corrected for thermal expansion, while figures 5 and 6 show only the uncorrected values along with experimental data which were used to generate these values. Thermal expansion values needed to carry out thermal expansion correction were taken from ref. [190]. The uncertainty in the recommended values is estimated to be within $\pm 2\%$ below 1137 K, $\pm 3\%$ up to the melting point, and $\pm 4\%$ for the liquid value at 2127 K.

Zirconium is a transition element, and its low-temperature electrical resistivity depends upon the type as well as on the concentration of impurities. The low-temperature electrical resistivity of low purity zirconium is rather difficult to estimate. Data so far available does not permit one to establish the upper limit of ρ_0 for which Matthiessen's rule can be applied to estimate electrical resistivity.

The data available in the literature for the temperature dependence of a bulk sample is reviewed in this report. However, additional information on the electrical resistivity is available in refs. [50,52,82,90,145-183]. Attention is directed to refs. [163,179,184-186] for data on irradiated samples, refs. [106,111,187,188] for data on films, ref. [188] for data on doped zirconium and ref. [189] for data on pressure dependence of resistivity.

TABLE 4. RECOMMENDED VALUES FOR THE ELECTRICAL RESISTIVITY OF ZIRCONIUM^a[Temperature, T, K; Electrical Resistivity, ρ , $10^{-8} \Omega \text{ m}$]

T	ρ		T	ρ	
	uncorrected	corrected		uncorrected	corrected
1	0.250	0.250	700	104.2	104.5
4	0.250	0.250	800	114.9	115.3
7	0.250	0.250	900	123.1	123.6
10	0.253	0.253	1000	128.8	129.4
15	0.283	0.283	1100	132.0	132.8
20	0.357	0.357	1137	132.6(a)	133.4(a)
25	0.491	0.490	1137	110.8(β)	111.3(β)
30	0.712	0.711	1150	111.1	111.7
40	1.443	1.441	1200	112.2	112.8
50	2.495	2.492	1300	114.5	115.2
60	3.75	3.75	1400	116.5	117.3
70	5.15	5.14	1500	118.6	119.6
80	6.64	6.63	1600	120.4	121.5
90	8.18	8.17	1700	122.3	123.5
100	9.79	9.78	1800	124.0	125.4
150	17.85	17.84	1900	125.8	127.4
200	26.35	26.33	2000	127.5	129.3
250	34.9	34.9	2100	129.1	131.0
273	38.8	38.8	2127	129.5(s)	131.4(s)
293	42.1	42.1	2127		141.3(l)
300	43.3	43.3			
350	51.9	51.9			
400	60.3	60.3			
500	76.5	76.6			
600	91.5	91.7			

^aThe values are for polycrystalline zirconium of purity 99.95% or higher, but those below 200 K are applicable specifically to zirconium having a residual resistivity of $0.250 \times 10^{-8} \Omega \text{ m}$. The columns headed uncorrected and corrected refer to values uncorrected and corrected for thermal expansion, respectively. Solid line separating tabular values indicates solid to liquid state transformation, while dotted line indicates solid phase transition.

a: cph; β : bcc.

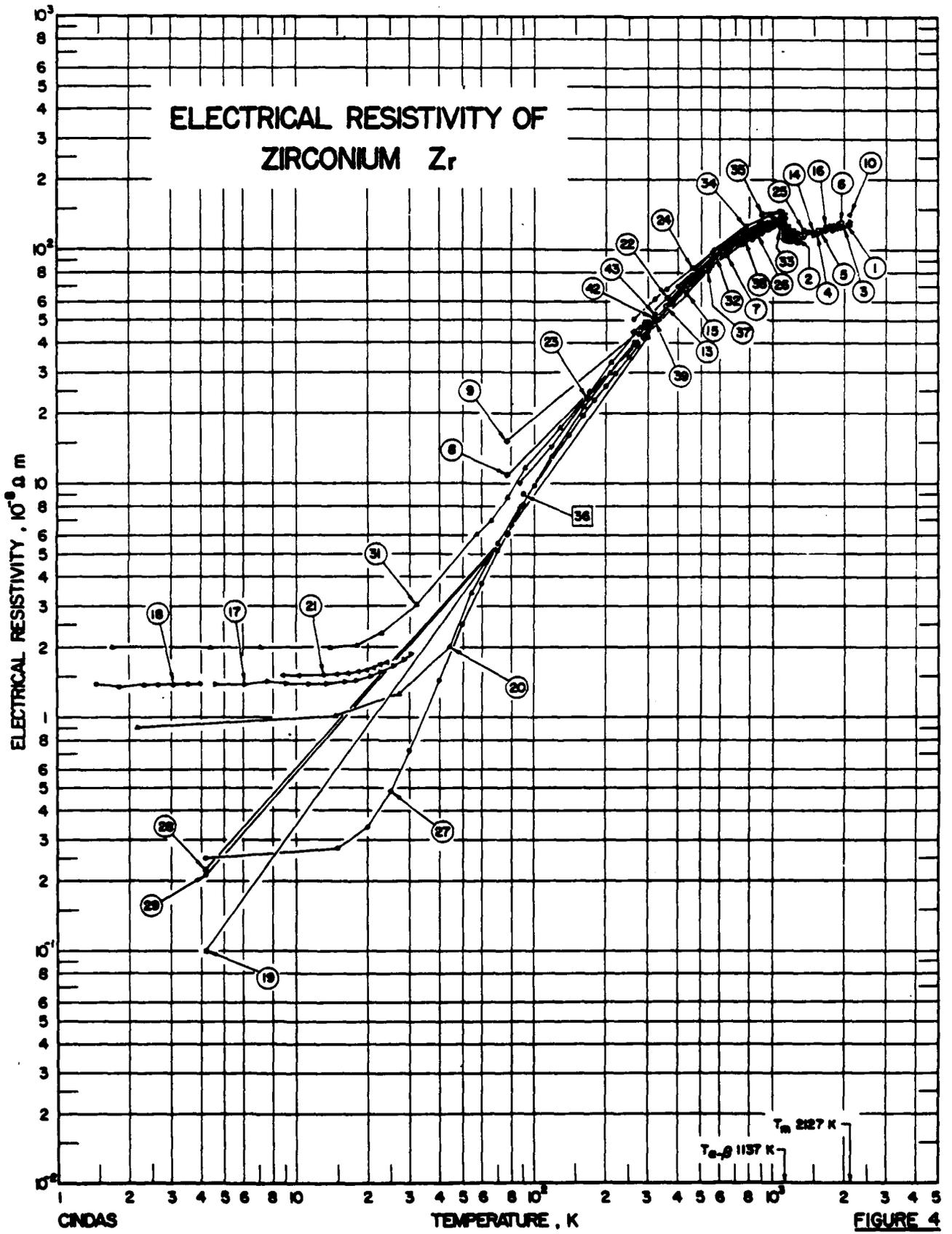
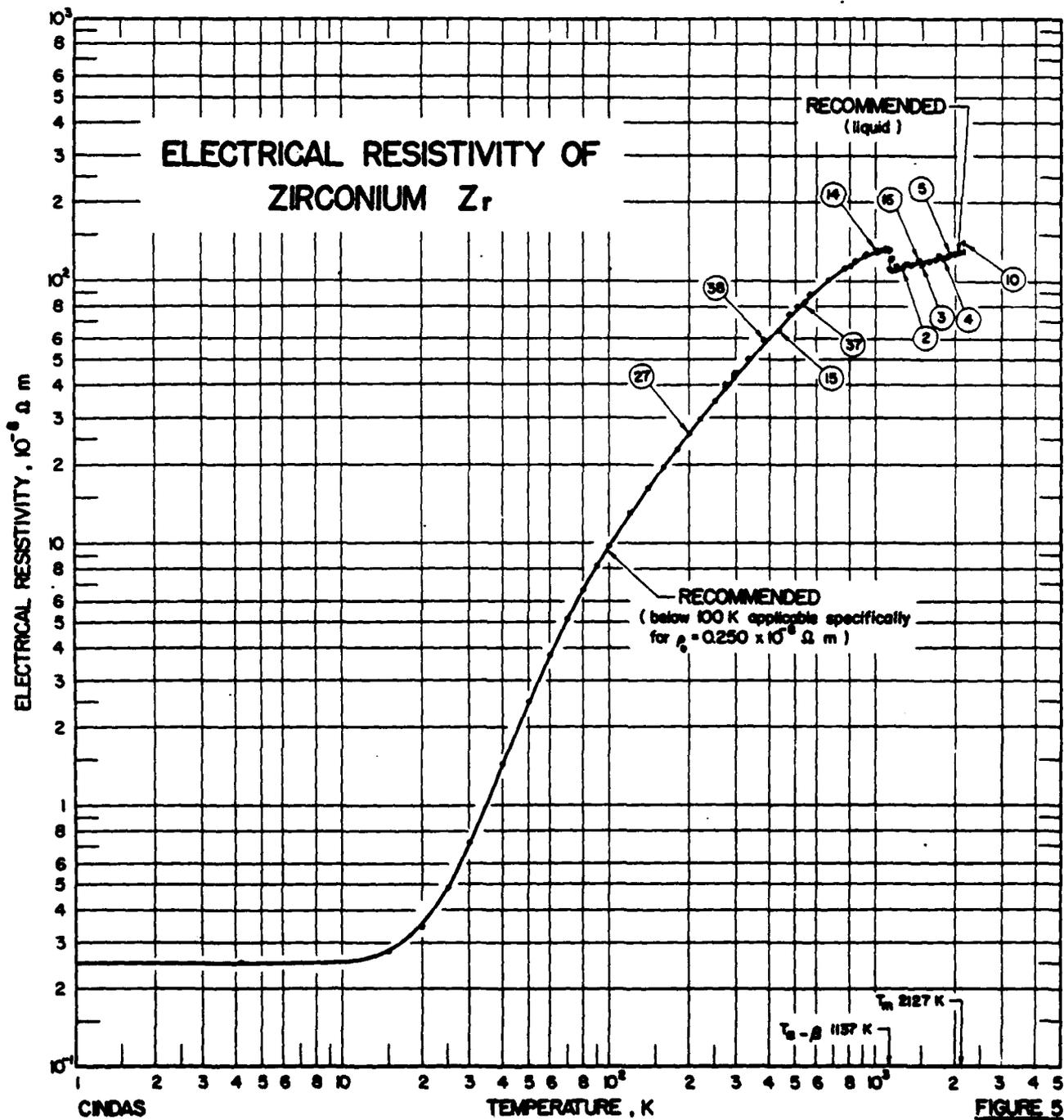


FIGURE 4



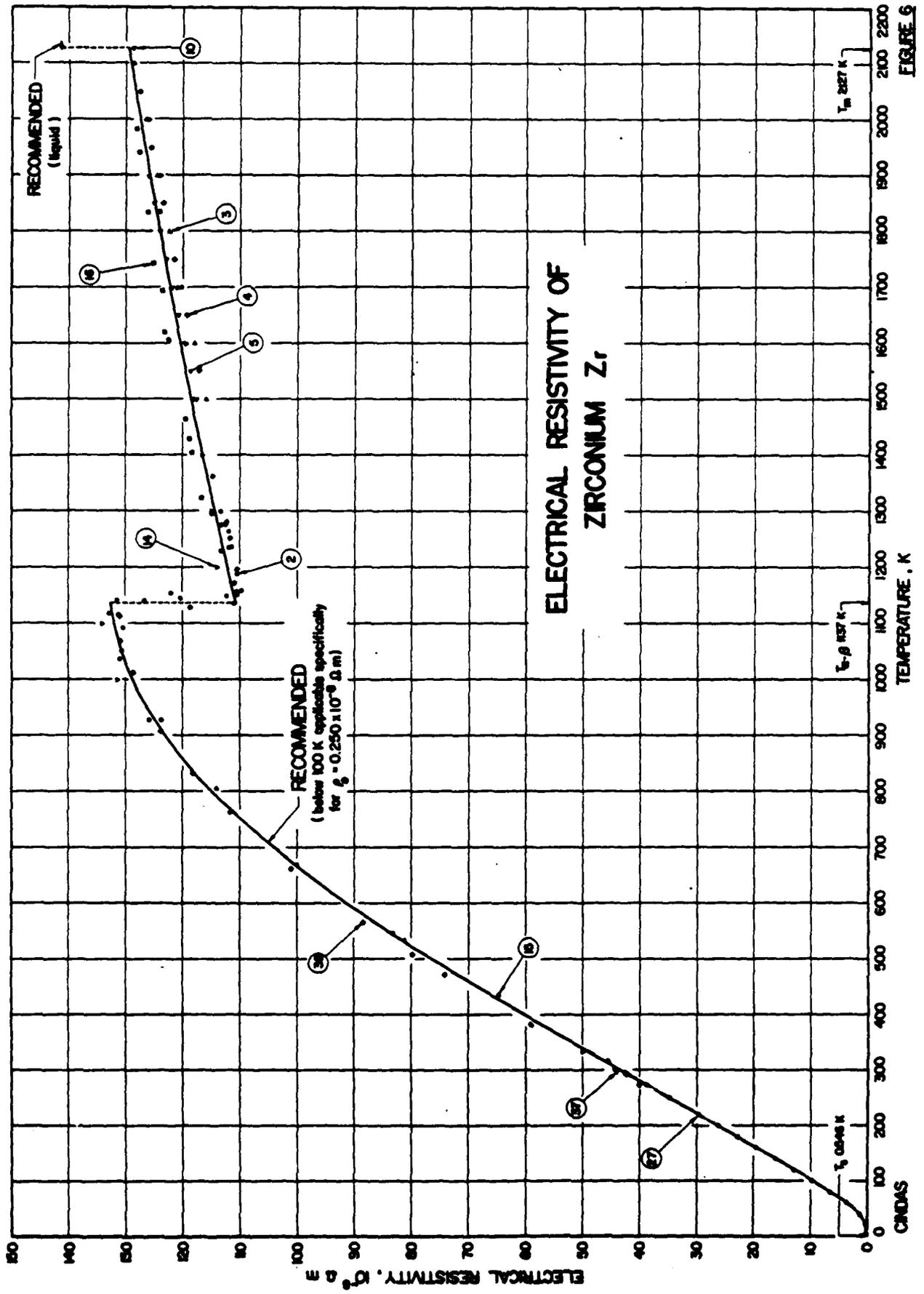


FIGURE 6

TABLE 5. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF ZIRCONIUM Zr

Ref. Set No.	Author(s)	Year	Method Used	Temp. Range, K	Specimen Designation	Composition (weight percent), Specifications and Remarks
1 123	Gasitriyev, A. and Righini, P.	1974	T	2097-2128	Specimen 11	99.98 Zr, 125 ppm O, 40 ppm Hf, 30 ppm Fe, 6 ppm Cr, 3.3 ppm Ni, 3 ppm Al, 2.1 ppm W, 1.5 ppm Mn, 1.5 ppm Si, 1.0 ppm Ti, less than 6 ppm other elements; specimen 76.2 mm long, 6.3 mm O.D., 0.25 mm thickness; small rectangular hole (0.5 x 1 mm) fabricated in the wall at middle of the specimen; approximated blackbody conditions; $T_0 = 2128$ K; data extracted from figure; estimated inaccuracy in the measurement is 13% (imprecision 10.05%).
2 123, 124	Gasitriyev, A. and Righini, P.	1974	T	1092-1265	Specimen 3	99.98 Zr, 125 ppm O, 40 ppm Hf, 30 ppm Fe, 6 ppm Cr, 3.3 ppm Ni, 3 ppm Al, 2.1 ppm W, 1.5 ppm Mn, 1.5 ppm Si, and 1.0 ppm Ti; specimen tube fabricated from rods by removing center portion using an electro-erosion technique; nominal dimensions of specimen were 76.2 mm long, 6.3 mm O.D., and wall thickness 0.5 mm; outer surfaces of the specimen were polished to reduce heat loss due to thermal radiation; α - β transformation temperature 1147 \pm 10 K; data extracted from figure; estimated inaccuracy of the measurement is 12%.
3 125	Gasitriyev, A. and Righini, P.	1974	T	1500-2100	Specimen 1	99.98 Zr, polycrystalline from Materials Research Corp., 6 ppm Cr, 3.3 ppm Ni, 125 ppm O, 2.1 ppm W, 3.0 ppm Al, 30 ppm Fe, 40 ppm Hf, 1.5 ppm Mn, 1.5 ppm Si, 1.0 ppm Ti; nominal dimensions are 76.2 mm length, 25.4 mm (effective length), 6.3 mm O.D., 0.5 mm wall thickness, and 0.5 x 1 mm rectangular blackbody hole; inaccuracy in measured value is 12%.
4 125	Gasitriyev, A. and Righini, P.	1974	T	1500-2100	Specimen 2	Similar to the above except different specimen.
5 125	Gasitriyev, A. and Righini, P.	1974	T	1500-1900	Specimen 3	Similar to the above except different specimen.
6 126	Mira, G., Hummel, H., and Kambeck, H.	1974	B	1173-1973	β -Zr	Drawn Zr wire of 0.5 mm diameter of Margrade (produced by electron beam zone melting) from Materials Research Corp., Orangeburg, NY; <10 ppm O, 40 ppm Cr, 15 ppm Al, 50 ppm Fe, 100 ppm Hf, and <75 ppm other; surface impurities were removed by polishing mechanically and electrolytically; wire was heated by D.C. for 30 minutes at 1650 C in high vacuum of 5×10^{-6} torr for recrystallization; data extracted from figure.
7 127	Dyckov, V.F., Rudnev, I.L., and Solov'ev, V.A.	1972	A	288-1282	Iodide Zirconium	0.056 Fe, <0.001 V, 0.0065 Mo, 0.0074 Nb, 0.012 Cu, 0.0041 Cr, 0.0041 Ni; measurements in 10^{-4} mm Hg vacuum; data extracted from figure.
8 128	Dyckov, V.F., Libshansk, Ye. B., and Mal'tsev, V.A.	1973	A	77,295	α -Zr	99.8 Zr (iodide); remelted in arc furnace.
9 128	Dyckov, V.F., et al.	1973	A	77,295	ω -Zr	Similar to above except subjected to hydrostatic pressure of 100 kbars at room temperature to get metastable ω -Zr phase.

TABLE 5. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF ZIRCONIUM Zr (continued)

Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Specimen Designation	Composition (weight percent), Specifications and Remarks
12	Marynski, M.M. and Tompkins, V.I.	1973	T	2127		99.76 Zr; values are reported for solid and for liquid at melting point; accuracy of measurements $\pm 5\%$.
11	Beale, G.	1973	T	4.2-293		Polycrystalline zirconium 100-250 Å thick vacuum deposited films onto very smooth, optically polished, square-shaped alkaline borosilicate substrates at room temp.; prior to the film condensation, the substrates had been degreased by baking in vacuum at 350°C for 6 hr and cleaned afterwards by both ultrasonic agitation at 50 kHz and ionic bombardment using a glow discharge of 5 kV; zirconium was evaporated from a copper liquid-nitrogen-cooled crucible employing a 270° beam deflection electron gun under pressure of the order of 10^{-4} torr; both the film thickness and the condensation rate were accurately controlled with a piezoelectric quartz crystal monitor maintained at the substrate temperature; the films were annealed for 3 hr at 300°C to remove frozen-in structural defects and subsequently cooled down to 4.2 K (tetragonal crystal structure characteristic of the β phase as shown by electron-diffraction analysis) using liquid helium as the refrigerant; the specimens were always kept under vacuum at the condensation pressure; to minimize the deformation arising from differential thermal expansion between metal and glass, both heating and cooling rates were lower than 1°C/sec; after the annealing process, measurements were taken; to avoid oxidation or adsorption of some other gases, all the experiments were performed in the vacuum conditions utilized for film preparation.
12	Velichko, B.V., Zverev, V.A., and Staryi, V.E.	1971	A	0.6-71.0		99.9 Zr, polycrystal; tabulated values calculated from $\rho_{\text{Zr}}/\rho_{\text{Zr}}$ values reported graphically assuming $36.8 \cdot 10^{-10}$ m for ρ_{Zr} ; $\rho_{\text{Zr}}/\rho_{\text{Zr}}$ = 34.
13	L'vov, S.H., Mal'ko, P.I., and Munchenko, V.P.	1971		309-1331		99.9 Zr; sample was prepared from bars (rods) obtained by iodide process; $\rho_{\text{Zr}}/\rho_{\text{Zr}}$ = 26; data extracted from figure.
14	Zverev, G.A.	1970		1000-2000	NETU 95-67-66	99.56 Zr, 0.23 Nb, 0.02 Fe, 0.04 Hf, 0.005 Co, 0.01 Ni, 0.03 Ti, 0.005 Mo, 0.005 Al, 0.01 Sn, iodide zirconium; density 6.59 g cm^{-3} ; rod specimen 56.6 mm length and 9.84 mm diameter; measurements in 5×10^{-3} mm Hg; greatest relative error in determination 2.8%; average values of several heating and cooling experiments.
15	Polozhki, V.E., Brushkin, V.P., and Sobol, Ya.G.	1970	A	302-1363		99.9 Zr, 0.01 Co, 0.005 Ni, 0.01 O ₂ , 0.009 Fe, 0.03 Nb, 0.002 Al, 0.005 Cu, 0.003 Ti, 0.005 Si; compact samples obtained by electron-beam sintering in vacuum; specimen dimensions are cylinder 60 mm long and 9 mm diameter; sample heated in resistance furnace with a molybdenum heater; measurements in 10^{-3} mm Hg; experimental error ± 1.5 to 2%.
16	Polozhki, V.E., et al.	1970	A	1229-1983		Same as above except sample heated by electron bombardment.

Not shown in figure.

TABLE 5. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF ZINCBIUM Zr (continued)

Data Set No.	Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Name and Specimen Designation	Composition (weight percent), Specifications and Remarks
17	134	Elliott, R.O. and Hill, H.H.	1970		4.6-30.6		105 ppm O ₂ , 8 ppm H ₂ , 33 ppm C, and 27 ppm Fe; heating cycle; data extracted from figure.
18	134	Elliott, R.O. and Hill, H.H.	1970		1.5-4.0		Same as above except cooling cycle; data extracted from figure.
19	135	Becterton, J.O. and Raston, D.S.	1968		4.2-300		No details given.
20	136	Clinard, F.W., Jr. and Kemper, G.D.	1968	A	2.1-295		Commercial specimen 95-175 ppm O ₂ , 40 ppm H ₂ , <40 ppm N ₂ , <1000 ppm HF, <1000 ppm Nb, 200 ppm Fe, 100 ppm Al, <100 ppm each Ti, V, Zr, Mo, and Pb; $\rho_0 = 0.8 \times 10^{-10}$ m; annealed condition; cylindrical specimen 0.25 in. diameter and 1 in. long; data extracted from figure; average of heating and cooling.
21	137	Cape, J.A. and Hule, R.H.	1965		8.8-24.1		Specimen cut from a button arc-cast in an inert atmosphere; finished sample was then measured as machined without annealing; specimen was $1 \times 0.1 \times 0.01$ in.; estimated absolute values of the resistivities are accurate to approximately $\pm 2\%$; values calculated from graphically reported values of ρ_T/ρ_{300} and tabulated values of 1.522×10^{-10} m for ρ_{300} .
22	138	Fuselli, R.H. and Tye, R.P.	1961	A	264-1196	No. 715	Graphite-milled Zr, 0.018 Fe, 0.043 C, 0.007 Al, 0.007 Nb, 0.0075 H ₂ , 0.1-0.6 O ₂ ; extruded; average of heating and cooling; data extracted from figure.
23	138	Fuselli, R.H. and Tye, R.P.	1961	A	87-1230	Van Arkel Zr	Van Arkel zirconium, 0.012 Fe, 0.016 C, 0.0025 H ₂ , and 0.3-0.6 O ₂ ; cold swaged; average of heating and cooling; data extracted from figure.
24	138	Fuselli, R.H. and Tye, R.P.	1961	A	264-886	No. 050	Arc melted low-carbon Zr; 0.045 Fe, 0.01 C, 0.008 H ₂ , 0.11 O ₂ ; extruded; average of heating and cooling; data extracted from figure.
25	139	Kiselev, H.A.	1961		730-1353		Specimen prepared from iodide metal; average of heating thermocouple and optical pyrometer measurements; $T_{\alpha-\beta} = 1138$ K; data extracted from figure.
26	139	Kiselev, H.A.	1961		855-1356		Same as above; average values of cooling thermocouple and optical pyrometer measurements; data extracted from figure.
27	49	White, G.E. and Woods, S.B.	1959	A	4.2-295	Zr3	99.95 Zr, 132 ppm Hf, 79 ppm C, 24 ppm Fe, 11 ppm Ni, 21-50 ppm O ₂ , 3-50 ppm H ₂ , <100 ppm Zn, 2-7 ppm each Ca, Cr, Mo, Si, H ₂ , and <10 ppm other elements; arc cast annealed 4 hr at 1100°C, swaged at room temp.; annealed for 15 min. at 1000°C and finally for 15 min. at 800°C in a vacuum $1-2 \times 10^{-6}$ mm Hg; values calculated from tabulated values of ideal resistivity (ρ_1), $\rho_{295} = 42.4 \times 10^{-10}$ m and $\rho_0/\rho_{295} = 5.96 \times 10^{-3}$.
28	140	Berlincourt, T.G.	1958		4.2-298	Zr1	Crystal bar from Westinghouse, 0.001 Ca, 0.016 Cu, 0.075 Fe, 0.002 H, 0.001 N, 0.016 O ₂ , 0.013 Si; $\rho_{295}/\rho_{300} = 170$.

TABLE 5. MEASUREMENT INFORMATION ON THE ELECTRICAL RESISTIVITY OF ZIRCONIUM Zr (continued)

Ref. No.	Author(s)	Year	Method Used	Temp. Range, K	Specimen Designation	Composition (weight percent), Specifications and Remarks
29	Berlincovert, T.G.	1958		4.2-300	Zr2	Same as above except $\rho_{1773}/\rho_{4.2} = 179$.
30*	Berlincovert, T.G.	1958		4.2-300	Zr2'	Same as above except $\rho_{1773}/\rho_{4.2} = 176$.
31	Kemp, W.R.G., Klamann, P.G., and White, G.K.	1956	A	1.7-293	MS5000	99.99 Zr from Messrs. Johnson, Matthey and Co., Ltd.; 3 mm diam. rod; annealed for 5 hr. at 950°C in vacuo; data extracted from figure; $\rho_0 = 1.96 \times 10^{-10} \Omega \cdot m$.
32	Adenstedt, H.K.	1952	B	276-1213	Zr660	99.9 Zr, 0.1 Hf, 0.02 Fe, <0.005 Ti, <0.005 Al, <0.005 Si, hafnium free from Foote Mineral Co.; samples prepared from as-deposited iodide crystal bars; cold-swaged condition; Rockwell hardness A-36; first heating run; values obtained by multiplying $43.2 \times 10^{-10} \Omega \cdot m$ (resistivity at 0°C) by resistivity ratio as function of temperature reported graphically.
33	Adenstedt, H.K.	1952	B	924-1299	Zr660	Same as above except second heating run.
34	Adenstedt, H.K.	1952	B	404-1189	Zr681	Similar to the above except as deposited iodide crystal bar, 0.036 Hf, <0.005 Fe, <0.005 Ti, <0.005 Al, <0.005 Si; Rockwell hardness A-22; first heating run.
35	Adenstedt, H.K.	1952	B	902-1127	Zr681	Same as above except first cooling run.
36	Adenstedt, H.K.	1952	B	90	Zr757	Similar to the above except 0.032 Hf, 0.044 O ₂ , 0.005 Ni, 0.005 Si, and <0.003 each Al, Fe, and Ti; cold-swaged, machined and annealed at 973 K from iodide crystal bar; 0.22 in. diam. and 10 in. length; $\rho_0 = 39.6 \times 10^{-10} \Omega \cdot m$.
37	King, G., Fink, F.W., and Thompson, H.B.	1951		273-533	Hastingshouse Ingot B-216	Pure Zr, 0.04 Hf, 0.04 Fe, 0.02 Ni, 0.007 Ti, 0.003 Sn, 0.001 Al; arc-melted ingot of WM crystal bar produced from lot CB-37; ingot forged at 1650 to a 1 in. square bar; measurements made at Mettelle.
38	Cook, L.A., Coentmans, L.B., and Johnson, W.R.	1950	B	277-1277	Low-Hf	Foote crystal bar, 0.04 Hf, 0.06 Si, 0.04 Fe, 0.004 Al, 0.005 each Cu, Ca, 0.001 each Ti, Ni, Pb, Mn, 0.01 Mg, 0.003 each Ni, Cr; machined to smooth cylinder 0.358 in. diam.; annealed above recrystallization temp.; data extracted from figure.
39	Cook, G.L., et al.	1950	B	303-323	Sample A	Same as above except machined to 0.306 in. diam. cylinder.
40*	Cook, G.L., et al.	1950	B	302-315	Sample B	Same as above except swaged from 0.306 in. diam. to 0.125 in. diam. (84% reduction in area).
41*	Cook, G.L., et al.	1950	B	302-322	Sample C	Same as sample B except annealed for 1 hr. at 500°C.
42	Cook, G.L., et al.	1950	B	303-320	Sample D	Same as sample C except swaged from 0.125 in. diam. to 0.048 in. diam. (83% reduction in area).
43	Cook, G.L., et al.	1950	B	301-321	Sample E	Same as sample D except annealed for 1 hr. at 500°C.

*Not shown in figure.

TABLE 6. EXPERIMENTAL DATA ON THE ELECTRICAL RESISTIVITY OF ZIRCONIUM Zr
 [Temperature, T, K; Electrical Resistivity, ρ , 10^{-8} Ω m]

T	ρ	T	ρ	T	ρ	T	ρ	T	ρ	T	ρ	T	ρ	T	ρ
DATA SET 1															
2097.0	128.52	2115.0	128.89	2133.0	129.05	2151.0	129.21	2169.0	129.37	2187.0	129.53	2205.0	129.69	2223.0	129.85
2099.1	128.56	2115.9	128.94	2133.5	129.09	2151.5	129.25	2169.5	129.41	2187.5	129.57	2205.5	129.73	2223.5	129.89
2101.6	128.61	2116.8	129.00	2134.0	129.14	2152.0	129.30	2170.0	129.46	2188.0	129.62	2206.0	129.78	2224.0	129.94
2104.3	128.66	2117.7	129.05	2134.5	129.19	2152.5	129.35	2170.5	129.51	2188.5	129.67	2206.5	129.83	2224.5	130.00
2106.8	128.74	2118.6	129.10	2135.0	129.24	2153.0	129.40	2171.0	129.56	2189.0	129.72	2207.0	129.88	2225.0	130.06
2109.5	128.81	2119.5	129.15	2135.5	129.29	2153.5	129.45	2171.5	129.61	2189.5	129.77	2207.5	129.93	2225.5	130.12
2111.0	128.85	2120.4	129.20	2136.0	129.34	2154.0	129.50	2172.0	129.66	2190.0	129.82	2208.0	129.98	2226.0	130.18
2113.5	128.89	2121.3	129.25	2136.5	129.39	2154.5	129.55	2172.5	129.71	2190.5	129.87	2208.5	130.03	2226.5	130.23
2115.9	128.94	2122.2	129.30	2137.0	129.44	2155.0	129.60	2173.0	129.76	2191.0	129.92	2209.0	130.08	2227.0	130.28
2118.5	129.00	2123.1	129.35	2137.5	129.49	2155.5	129.65	2173.5	129.81	2191.5	129.97	2209.5	130.13	2227.5	130.33
2120.5	129.06	2124.0	129.40	2138.0	129.54	2156.0	129.70	2174.0	129.86	2192.0	130.02	2210.0	130.18	2228.0	130.38
2122.5	129.13	2124.9	129.45	2138.5	129.59	2156.5	129.75	2174.5	129.91	2192.5	130.07	2210.5	130.23	2228.5	130.43
2123.2	129.19	2125.8	129.51	2139.0	129.65	2157.0	129.81	2175.0	129.97	2193.0	130.09	2211.0	130.25	2229.0	130.45
2124.1	129.23	2126.7	129.55	2139.5	129.69	2157.5	129.85	2175.5	130.01	2193.5	130.13	2211.5	130.29	2229.5	130.49
2125.1	129.27	2127.6	129.59	2140.0	129.73	2158.0	129.89	2176.0	130.05	2194.0	130.17	2212.0	130.33	2230.0	130.53
2126.4	129.30	2128.5	129.62	2140.5	129.76	2158.5	129.92	2176.5	130.08	2194.5	130.20	2212.5	130.36	2230.5	130.56
2128.6	129.39	2129.4	129.66	2141.0	129.80	2159.0	129.96	2177.0	130.12	2195.0	130.24	2213.0	130.40	2231.0	130.60
2129.9	129.41	2130.3	129.68	2141.5	129.82	2159.5	129.98	2177.5	130.14	2195.5	130.26	2213.5	130.42	2231.5	130.62
2131.7	129.47	2131.2	129.72	2142.0	129.86	2160.0	130.02	2178.0	130.18	2196.0	130.30	2214.0	130.46	2232.0	130.66
2132.1	129.51	2132.1	129.76	2142.5	129.90	2160.5	130.06	2178.5	130.22	2196.5	130.34	2214.5	130.50	2232.5	130.70
2133.7	129.57	2133.0	129.80	2143.0	129.94	2161.0	130.10	2179.0	130.26	2197.0	130.38	2215.0	130.54	2233.0	130.74
2135.4	129.62	2133.9	129.84	2143.5	129.98	2161.5	130.14	2179.5	130.30	2197.5	130.42	2215.5	130.58	2233.5	130.78
2137.5	129.68	2134.8	129.88	2144.0	130.02	2162.0	130.18	2180.0	130.34	2198.0	130.46	2216.0	130.62	2234.0	130.82
2137.5	129.68	2135.7	129.92	2144.5	130.06	2162.5	130.22	2180.5	130.38	2198.5	130.50	2216.5	130.66	2234.5	130.86
2137.7	129.69	2136.6	129.96	2145.0	130.10	2163.0	130.26	2181.0	130.42	2199.0	130.54	2217.0	130.70	2235.0	130.90
2138.1	129.71	2137.5	129.98	2145.5	130.12	2163.5	130.28	2181.5	130.44	2199.5	130.56	2217.5	130.72	2235.5	130.92
2139.1	129.73	2138.4	129.99	2146.0	130.14	2164.0	130.30	2182.0	130.46	2200.0	130.58	2218.0	130.74	2236.0	130.94
2140.1	129.75	2139.3	130.01	2146.5	130.16	2164.5	130.32	2182.5	130.48	2200.5	130.60	2218.5	130.76	2236.5	130.96
2141.1	129.77	2140.2	130.03	2147.0	130.18	2165.0	130.34	2183.0	130.50	2201.0	130.62	2219.0	130.78	2237.0	130.98
2142.1	129.79	2141.1	130.05	2147.5	130.20	2165.5	130.36	2183.5	130.52	2201.5	130.64	2219.5	130.80	2237.5	131.00
2143.1	129.81	2142.0	130.07	2148.0	130.22	2166.0	130.38	2184.0	130.54	2202.0	130.66	2220.0	130.82	2238.0	131.02
2144.1	129.83	2142.9	130.09	2148.5	130.24	2166.5	130.40	2184.5	130.56	2202.5	130.68	2220.5	130.84	2238.5	131.04
2145.1	129.85	2143.8	130.11	2149.0	130.26	2167.0	130.42	2185.0	130.58	2203.0	130.70	2221.0	130.86	2239.0	131.06
2146.1	129.87	2144.7	130.13	2149.5	130.28	2167.5	130.44	2185.5	130.60	2203.5	130.72	2221.5	130.88	2239.5	131.08
2147.1	129.89	2145.6	130.15	2150.0	130.30	2168.0	130.46	2186.0	130.62	2204.0	130.74	2222.0	130.90	2240.0	131.10
2148.1	129.91	2146.5	130.17	2150.5	130.32	2168.5	130.48	2186.5	130.64	2204.5	130.76	2222.5	130.92	2240.5	131.12
2149.1	129.93	2147.4	130.19	2151.0	130.34	2169.0	130.50	2187.0	130.66	2205.0	130.78	2223.0	130.94	2241.0	131.14
2150.1	129.95	2148.3	130.21	2151.5	130.36	2169.5	130.52	2187.5	130.68	2205.5	130.80	2223.5	130.96	2241.5	131.16
2151.1	129.97	2149.2	130.23	2152.0	130.38	2170.0	130.54	2188.0	130.70	2206.0	130.82	2224.0	130.98	2242.0	131.18
2152.1	129.99	2150.1	130.25	2152.5	130.40	2170.5	130.56	2188.5	130.72	2206.5	130.84	2224.5	131.00	2242.5	131.20
2153.1	130.01	2151.0	130.27	2153.0	130.42	2171.0	130.58	2189.0	130.74	2207.0	130.86	2225.0	131.02	2243.0	131.22
2154.1	130.03	2151.9	130.29	2153.5	130.44	2171.5	130.60	2189.5	130.76	2207.5	130.88	2225.5	131.04	2243.5	131.24
2155.1	130.05	2152.8	130.31	2154.0	130.46	2172.0	130.62	2190.0	130.78	2208.0	130.90	2226.0	131.06	2244.0	131.26
2156.1	130.07	2153.7	130.33	2154.5	130.48	2172.5	130.64	2190.5	130.80	2208.5	130.92	2226.5	131.08	2244.5	131.28
2157.1	130.09	2154.6	130.35	2155.0	130.50	2173.0	130.66	2191.0	130.82	2209.0	130.94	2227.0	131.10	2245.0	131.30
2158.1	130.11	2155.5	130.37	2155.5	130.52	2173.5	130.68	2191.5	130.84	2209.5	130.96	2227.5	131.12	2245.5	131.32
2159.1	130.13	2156.4	130.39	2156.0	130.54	2174.0	130.70	2192.0	130.86	2210.0	130.98	2228.0	131.14	2246.0	131.34
2160.1	130.15	2157.3	130.41	2156.5	130.56	2174.5	130.72	2192.5	130.88	2210.5	131.00	2228.5	131.16	2246.5	131.36
2161.1	130.17	2158.2	130.43	2157.0	130.58	2175.0	130.74	2193.0	130.90	2211.0	131.02	2229.0	131.18	2247.0	131.38
2162.1	130.19	2159.1	130.45	2157.5	130.60	2175.5	130.76	2193.5	130.92	2211.5	131.04	2229.5	131.20	2247.5	131.40
2163.1	130.21	2160.0	130.47	2158.0	130.62	2176.0	130.78	2194.0	130.94	2212.0	131.06	2230.0	131.22	2248.0	131.42
2164.1	130.23	2160.9	130.49	2158.5	130.64	2176.5	130.80	2194.5	130.96	2212.5	131.08	2230.5	131.24	2248.5	131.44
2165.1	130.25	2161.8	130.51	2159.0	130.66	2177.0	130.82	2195.0	130.98	2213.0	131.10	2231.0	131.26	2249.0	131.46
2166.1	130.27	2162.7	130.53	2159.5	130.68	2177.5	130.84	2195.5	131.00	2213.5	131.12	2231.5	131.28	2249.5	131.48
2167.1	130.29	2163.6	130.55	2160.0	130.70	2178.0	130.86	2196.0	131.02	2214.0	131.14	2232.0	131.30	2250.0	131.50
2168.1	130.31	2164.5	130.57	2160.5	130.72	2178.5	130.88	2196.5	131.04	2214.5	131.16	2232.5	131.32	2250.5	131.52
2169.1	130.33	2165.4	130.59	2161.0	130.74	2179.0	130.90	2197.0	131.06	2215.0	131.18	2233.0	131.34	2251.0	131.54
2170.1	130.35	2166.3	130.61	2161.5	130.76	2179.5	130.92	2197.5	131.08	2215.5	131.20	2233.5	131.36	2251.5	131.56
2171.1	130.37	2167.2	130.63	2162.0	130.78	2180.0	130.94	2198.0	131.10	2216.0	131.22	2234.0	131.38	2252.0	131.58
2172.1	130.39	2168.1	130.65	2162.5	130.80	2180.5	130.96	2198.5	131.12	2216.5	131.24	2234.5	131.40	2252.5	131.60
2173.1	130.41	2169.0	130.67	2163.0	130.82	2181.0	13								

TABLE 6. EXPERIMENTAL DATA ON THE ELECTRICAL RESISTIVITY OF ZIRCONIUM Zr (continued)

T	ρ	T	ρ	T	ρ	T	ρ	T	ρ	T	ρ	T	ρ	T	ρ
DATA SET 15 (cont.)															
928.1	126.0	1.5	1.39	19.9	1.60	957	130	1138	114.6	220	29.651	DATA SET 27 (cont.)			
1037	131.1	1.8	1.36	21.3	1.64	1009	133	1159	114.8	250	34.851				
1081	136.7	2.3	1.36	22.7	1.68	1106	136	1175	115.2	273	39.851				
1118	132.9	2.6	1.39	24.1	1.73	1145	142	1197	117.2	295	42.651				
1154	122.2	3.0	1.36	DATA SET 22				1297	117.9	DATA SET 28					
1152	110.5	3.5	1.39	264	44.7	1159	139	1318	118.7						
1236	112.0	3.9	1.39	364	62.2	1179	126	1334	119.8						
1282	112.4	4.0	1.40	461	78.0	1190	120	1340	120.1						
1300	113.5	DATA SET 19				1213	122	1353	122.0						
1343	114.9	4.2	0.1	524	86.3	1230	121	DATA SET 26							
DATA SET 16				621	92.1	DATA SET 24									
1229	113.4	300	42.3	664	104	264	50.2	855	117.2						
1275	113.4	DATA SET 20				709	109	324	60.8						
1296	116.9	2.1	0.91	764	115	364	67.7	1043	133.1						
1324	116.7	24.7	1.62	812	119	421	863	1086	133.5						
1405	116.5	27.3	1.2	912	123	461	83.5	1096	133.5						
1430	118.9	44.1	2.0	957	131	521	91.2	1102	132.9						
1465	119.6	54.7	3.4	1006	134	564	97.2	1108	129.9	DATA SET 30*					
1606	122.6	70.6	5.5	1057	135	615	103	1132	120.0						
1620	123.3	87.5	7.9	1111	136	664	109	1147	118.4						
1694	123.7	105	10.8	1131	136	712	114	1172	118.0						
1743	123.2	118	13.1	1139	131	769	119	1172	117.6						
1835	126.2	132	15.5	1142	127	861	123	1234	117.2						
1853	126.3	149	18.6	1148	121	866	128	1278	117.2						
1941	127.8	163	21.4	1150	116	DATA SET 25									
1983	128.2	180	24.7	1153	116	738	111.3	1331	118.1						
DATA SET 17				1162	115	770	113.9	1340	119.6						
4.6	1.39	208	29.9	1176	115	867	121.7	1356	122.2						
4.9	1.40	246	35.2	1196	116	927	126.1	DATA SET 27							
6.1	1.39	261	37.2	DATA SET 23				4.2	0.251						
7.6	1.43	271	39.3	87	10.0	949	127.4	15	0.276						
9.1	1.40	278	40.3	118	14.3	971	128.7	20	0.341						
10.4	1.40	285	41.2	167	23.2	1002	130.4	25	0.486						
11.4	1.39	295	43.2	267	40.1	1031	131.1	30	0.721						
13.6	1.40	DATA SET 21				1049	131.8	40	1.451						
16.2	1.42	0.8	1.52	367	57.6	1068	131.8	50	2.501						
18.0	1.45	8.8	1.52	464	73.7	1099	132.2	60	3.751						
20.7	1.50	10.2	1.52	564	90.1	1102	132.2	70	5.151						
22.8	1.57	13.0	1.52	618	99.8	1115	132.2	80	6.451	DATA SET 32					
25.9	1.67	14.0	1.52	644	105	1121	129.9	90	8.151						
28.6	1.77	14.9	1.53	712	111	1126	123.4	100	9.801						
30.6	1.68	15.8	1.53	764	116	1132	118.7	120	13.051						
Not shown in figure.															

4. ACKNOWLEDGMENTS

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5. APPENDICES

5.1. Methods for the Measurement of Electrical Resistivity

At the Center for Information and Numerical Data Analysis and Synthesis (CINDAS) of Purdue University, the experimental methods for the measurement of electrical resistivity have been classified into various categories according to a similar scheme used by CINDAS for the classification of methods for the measurement of thermal conductivity [191, pp. 13a-25a]. This classification scheme of CINDAS is presented below. Note that the letters in parentheses following the respective methods are the code letter used in the 'Method Used' column of the Table of Measurement Information for indicating the experimental methods used by the various authors.

Methods for the Measurement of Electrical Resistivity

A. Steady-State Methods

1. Voltmeter and ammeter direct reading method (V) [192, p. 159; 193, pp. 244-5]
2. Direct-current potentiometer method (A) [194, pp. 151-8]
 - a. 4-probe potentiometer method
3. Direct-current bridge methods (B) [194, pp. 144-51]
 - a. Kelvin double bridge method
 - b. Mueller bridge method
 - c. Wheatstone bridge method
4. Direct-heating method (K) [195,196]

B. Non-Steady-State Methods

1. Periodic current method
 - a. Direct connection to sample
 - (1) Alternating-current potentiometer method (C) [194, pp. 161-2]
 - b. No connection to sample
 - (1) Rotating magnetic field method (R) [197]
2. Non-periodic current method
 - a. Direct connection to sample
 - (1) Transient (subsecond) method (T) [198]

5.2. Conversion Factors for the Units of Electrical Resistivity

The recommended values and experimental data for the electrical resistivity tabulated in this work are in the units: $10^{-8} \Omega \text{ m}$. Conversion factors for the units of electrical resistivity, which may be used to convert the values given in ($10^{-8} \Omega \text{ m}$) to values in other units, are given below.

Conversion Factors for the Units of Electrical Resistivity

Units to be Converted to	Multiply the Value Given in ($10^{-8} \Omega \text{ m}$) by
ohm-meter ($\Omega \text{ m}$)	1×10^{-8}
ohm-centimeter ($\Omega \text{ cm}$)	1×10^{-6}
ohm-inch ($\Omega \text{ in.}$)	3.937×10^{-7}
ohm-foot ($\Omega \text{ ft}$)	3.281×10^{-8}
microhm-centimeter ($\mu\Omega \text{ cm}$)	1
abohm-centimeter ($\text{ab}\Omega \text{ cm}$)	1×10^3
statohm-centimeter ($\text{stat}\Omega \text{ cm}$)	1.113×10^{-18}
emu (= $\text{ab}\Omega \text{ cm}$)	1×10^3
esu (= $\text{stat}\Omega \text{ cm}$)	1.113×10^{-18}
ohm-circular mil per foot ($\Omega \text{ cmil ft}^{-1}$)	6.015

Example: $1.000 \times 10^{-8} \Omega \text{ m} = 3.937 \times 10^{-7} \Omega \text{ in.}$

6. REFERENCES

1. Chi, T.C., 'Electrical Resistivity of Alkali Elements,' J. Phys. Chem. Ref. Data, 8(2), 339-438 (1979).
2. Chi, T.C., 'Electrical Resistivity of Alkaline Earth Elements,' J. Phys. Chem. Ref. Data, 8(2), 439-97 (1979).
3. Matula, R.A., 'Electrical Resistivity of Copper, Gold, Palladium, and Silver,' J. Phys. Chem. Ref. Data, 8(4), 1147-298 (1979).
4. Ho, C.Y., Powell, R.W., and Liley, P.E., 'Thermal Conductivity of the Elements: A Comprehensive Review,' J. Phys. Chem. Ref. Data, Vol. 3, Suppl. 1, 796 pp. (1974).
5. Matthiessen, A., 'Electrical Resistivity of Alloys,' Ann. Physik, 110, 190-221 (1860).
6. Matthiessen, A. and Vogt, C., 'The Influence of Temperature on the Electrical Conductivity of Alloys,' Ann. Physik, 122, 19-78 (1864).
7. Bloch, F., 'On the Quantum Mechanics of Electrons in a Crystalline Lattice,' Z. Phys., 52, 555-600 (1928).
8. Bloch, F., 'The Electrical Resistance Law at Low Temperatures,' Z. Phys., 59, 208-14 (1930).
9. Seydel, U. and Fucke, W., 'Electrical Resistivity of Liquid Titanium, Vanadium, Molybdenum, and Tungsten,' J. Phys. F, 10(8), L203-6 (1980).
10. Gathers, G.R., Shaner, J.W., Hixson, R.S., and Young, D.A., 'Very High Temperature Thermophysical Properties of Solid and Liquid Vanadium and Iridium,' High Temp.-High Pressures, 11(6), 653-68 (1979).
11. Vedernikov, M.V., Dvunitkin, V.G., and Zhumagulov, A., 'Rules Governing the Behavior of the Electrical Resistivity and Thermoelectric Power of Systems of Binary Continuous Solid Solutions of Metals,' Fiz. Tverd. Tela, 20(11), 3302-5 (1978); Engl. transl.: Sov. Phys.-Solid State, 20(11), 1904-6 (1978).
12. Peletskii, V.E. Amasovich, E.S., Kostanovskii, A.V., Zaretskii, E.B., Sobol, Ya.G., and Shur, B.A., 'Thermal Conductivity of Vanadium,' Teplofiz. Vys. Temp., 15(6), 1202-7 (1977); Engl. transl.: High Temp., 15(6), 1028-33 (1977).

13. Pan, V.M., Prokhorov, V.G., Shevchenko, A.D., and Dovgopol, V.P., 'The Physical Properties of Vanadium in the Temperature Range 4.2-300 K,' *Fiz. Nizk. Temp.*, 3(10), 1266-71 (1977); Engl. transl.: *Sov. J. Low Temp. Phys.*, 3(10), 609-12 (1977).
14. Courtney, D.R., 'Thermal Conductivity of Hydrogen Doped High Purity Vanadium,' Iowa State Univ., M.S. Thesis, 40 pp. (1977).
15. Chakal'skii, B.K., Azhazha, V.M., Red'ko, N.A., and Shalyt, S.S., 'Thermal Conductivity of Vanadium at Low Temperatures,' *Pisma Zh. Eksp. Teor. Fiz.*, 23(9), 513-15 (1976); Engl. transl.: *JETP Lett.*, 23(9), 468-70 (1976).
16. Jung, W.D., 'Thermal Conductivity of High Purity Vanadium,' Ames Lab. Rept. IS-T-693, 110 pp. (1975).
17. Jung, W.D., 'The Thermal Conductivity of High Purity Vanadium,' Iowa State Univ. of Science and Technology, Ph.D. Theses, 109 pp. (1975). [Univ. Microfilm No. 76-1850]
18. Jung, W.D., Schmidt, F.A., and Danielson, G.C., 'Thermal Conductivity of High-Purity Vanadium,' *Phys. Rev.*, 15B(2), 659-65 (1977).
19. Aleksandrov, B.N., Semenova, E.D., Petrova, O.I., Chernyi, B.P., and Azhazha, V.M., 'The Electrical Resistivity of Vanadium of Various Purities in the 4.2 to 47 K Temperature Range,' *Fiz. Nizk. Temp.*, 1(3), 388-99 (1975).
20. Azhazha, V.M., Volkenshtein, N.V., Startsev, V.Ye., Finkel, V.A., Cherepanov, V.I., and Chernyi, B.P., 'Resistivity of High Purity Vanadium. Investigation of Critical Temperature and Anomalies in the Temperature Dependence,' *Fiz. Met. Metalloved.*, 41(6), 1188-95 (1976); Engl. transl.: *Phys. Met. Metallogr.*, 41(6), 54-60 (1976).
21. Alekseevskii, N.E., Mitin, A.V., and Matveeva, N.M., 'Investigation of the Superconducting Properties of Vanadium - Aluminum and Vanadium Tin Solid Solutions,' *Zh. Eksp. Teor. Fiz.*, 69(6), 2124-31 (1975); Engl. transl.: *Sov. Phys.-JETP*, 42(6), 1080-3 (1976).
22. Cezairliyan, A., Righini, F., and McClure, J.L., 'Simultaneous Measurements of Heat Capacity, Electrical Resistivity and Hemispherical Total Emittance by a Pulse Heating Technique: Vanadium, 1500 to 2100 K,' National Bureau of Standards, Inst. for Materials Research, Rept. NBSIR-74-600, 30-46 (1974).

23. Cezairliyan, A., Righini, F., and McClure, J.L., 'Simultaneous Measurements of Heat Capacity, Electrical Resistivity, and Hemispherical Total Emittance by a Pulse Heating Technique. Vanadium, 1500 to 2100 K,' J. Res. Natl. Bur. Stand., 78A(2), 143-7 (1974).
24. Beckett, C.W., 'Thermodynamics of Chemical Species Important to Rocket Technology,' U.S. Air Force Rept. AFOSR-TR-75-0596, 200 pp. (1974). [AD A008 935]
25. Kumagai, K. and Ohtsuka, T., 'The Superconductivity of Alloys of a Transition Metal with Non-Transition Elements (and Normal State Properties),' J. Phys. Soc. Jpn., 37(2), 384-407 (1974).
26. Prekul, A.F., Rassokhin, V.A., and Volkenshtein, N.V., 'Effect of Spin Fluctuations on the Superconducting and Normal Properties of Alloys of Titanium Containing Vanadium, Niobium, or Tantalum,' Zh. Eksp. Teor. Fiz. 67(6), 2286-92 (1974); Engl. transl.: Sov. Phys.-JETP, 40(6), 1134-6. (1975.)
27. Lang, E. and Bressers, J., 'Effect of Oxygen Doping on the Electrical Resistivity of Vanadium,' Z. Metallkd., 66(10), 619-22 (1975).
28. Neimark, B.E., Belyakova, P.E., Brodskii, B.R., Voronin, L.K., Korytina, S.F., and Merkul'ev, A.N., 'Physical Properties of Vanadium,' Heat Transfer-Sov. Res., 5(2), 141-5 (1973).
29. Chernoplekov, N.A., Panova, G.Kh., Samoilov, B.N., and Shikov, A.A., 'Alteration of the Superconducting Properties of Vanadium by Introduction of Tantalum Impurity Atoms,' Zh. Eksp. Teor. Fiz., 64(1), 195-203 (1973); Engl. transl.: Sov. Phys.-JETP, 37(1), 102-6 (1973).
30. Arutyunov, A.V., Makarenko, I.N., and Filippov, L.P., 'Thermal Properties of Vanadium at High Temperatures,' Teplofiz. Svoistva Veshchestv Mater., 5, 105-8 (1972).
31. Filippov, L.P. and Yurchak, R.P., 'High Temperature Investigations of the Thermal Properties of Solids,' Inzh. Fiz. Zh., 21(3), 561-77 (1971); Engl. transl.: J. Eng. Phys., 21(3), 1209-20 (1971).
32. Peletskii, V.E., Druzhinin, V.P., and Sobol, Ya.G., 'Thermophysical Properties of Vanadium at High Temperatures,' High Temp.-High Pressures, 3(2), 153-9 (1971).

33. L'vov, S.N., Mal'ko, P.I., and Nemchenko, V.F., 'Electrical Conductivity and Thermal Conductivity of Group IV-VI Transition Metals in the 20-1200 C Range,' *Metallofizika*, **37**, 22-9 (1971).
34. Voronin, L.K., Merkul'ev, A.N., and Neimark, B.E., 'Some Physical Properties of Vanadium,' *High Temp.*, **8**(4), 737-40 (1970).
35. Hensler, D.H., Ross, A.R., and Fuls, E.N., 'Reactively Sputtered Thin Films in the Vanadium-Oxygen System Using Triode Sputtering,' *J. Electrochem. Soc.*, **116**(6), 887-9 (1969).
36. Heubner, U., 'Thermal and Electrical Conductivity of Vanadium Alloys Between 20 and 650°,' *J. Nucl. Mater.*, **32**(1), 88-100 (1969).
37. Westlake, D.G. and Alfred, L.C.R., 'Determination of the Debye Characteristic Temperature of Vanadium from the Bloch-Grueneisen Relation,' *J. Phys. Chem. Solids*, **29**(11), 1931-4 (1968).
38. Westlake, D.G., 'A Resistometric Study of Phase Equilibria at Low Temperatures in the Vanadium-Hydrogen System,' *Trans. Met. Soc. AIME*, **239**, 1341-4 (1967).
39. Wertheimer, M.R. and Gilchrist, J.G., 'Flux Jumps in Type II Superconductors,' *J. Phys. Chem. Solids*, **28**(12), 2509-24 (1967).
40. Amitin, E.B., Kovalevskaya, Yu.A., and Kovdrya, Yu.Z., 'Hall Effect and Electrical Resistivity of Vanadium at 20-300 K,' *Sov. Phys.-Solid State*, **9**(3), 704-6 (1967).
41. Van Gorp, G.J., 'The Effect of Structure on the Superconducting Properties of Vanadium and Niobium Foils,' *Philips Res. Rep.*, **22**(1), 10-35 (1967).
42. Druzhinina, I.P., Vladimirskaia, T.M., and Fraktovnikova, A.A., 'Thermoelectric Properties of Vanadium,' *Meas. Tech.*, **8**, 1032-4 (1966).
43. Hörz, G., Gebhardt, E., and Durrschnabel, W., 'Emissivity and Electrical Resistivity of Vanadium, Vanadium Nitride and Vanadium Oxide Mixed Crystals,' *Z. Metallkd.*, **56**(8), Pt. 2, 554-60 (1965).
44. Hörz, G., 'Emissivity and Electrical Resistivity of Tantalum at High Temperatures,' *Z. Metallkd.*, **57**, 871-3 (1966).

45. Taylor, M.A. and Smith, C.H.L., 'The Electrical Resistivity of Vanadium and Vanadium-Chromium Solid Solutions,' *Physica*, 28(4), 453-60 (1962).
46. Burger, J.P. and Taylor, M.A., 'Anomalies in the Magnetic Susceptibility and Electrical Resistivity of Vanadium,' *Phys. Rev. Lett.*, 6(4), 185-7 (1961).
47. Hren, J.A. and Wayman, C.M., 'Some Properties of Vanadium at Subatmospheric Temperatures,' *Trans. Met. Soc. AIME*, 218, 377-9 (1960).
48. White, G.K. and Woods, S.B., 'Low Temperature Resistivity of Transition Elements. Vanadium, Niobium, and Hafnium,' *Can. J. Phys.*, 35(3), 892-900 (1957).
49. White, G.K. and Woods, S.B., 'Electrical and Thermal Resistivity of the Transition Elements at Low Temperatures,' *Philos. Trans. R. Soc. London*, A251, 273-302 (1959).
50. Samsonov, G.V., 'The Electrical Conductivity of Certain Compounds of the Transitional Metals with Boron, Carbon and Nitrogen, and the Electrical Conductivity of Alloys of these Compounds,' *Sov. Phys.-Tech. Phys.*, 1(4), 695-701 (1957).
51. Wruck, D. and Wert, C., 'The Role of Crystal Structure on Irradiation Effects in Metals,' *Acta Metall.*, 3(2), 115-20 (1955).
52. Potter, H.H., 'Electrical Resistance and Thermoelectric Power of the Transition Metals,' *Proc. Phys. Soc. London*, 53(6), 695-705 (1941).
53. Gautron, G.J., Zablocki, J.E., Hsiang, T.Y., and Weinstock, H., 'Electron-Electron Scattering in Vanadium,' *J. Low Temp. Phys.*, 49(1/2), 185-91 (1982).
54. Tsai, C.L., Fagaly, R.L., Weinstock, H., and Schmidt, F.A., 'Electrical and Thermal Transport Coefficients of Pure Vanadium,' *Phys. Rev. B*, 23(12), 6490-40 (1981).
55. Taylor, R.E. and Groot, H., 'Thermal Conductivity of Vanadium,' *Thermophysical Properties Research Lab. Rept. TPRL 265*, 5 pp. (1982).
56. L'vov, S.N. and Nemchenko, V.F., 'Temperature Relation of Thermal Electromotive Force and Specific Electric Resistance of Titanium, Vanadium and Chromium and Their Borides, Carbides, and Nitrides,' *Vysokotemp. Neorg. Soedin.*, 100-7 (1965); Engl. transl.: Los Alamos Scientific Lab. Rept. LA-TR-67-17, 14 pp. (1967).

57. Peletskii, V.E., 'Electrical Resistivity of Vanadium in the Temperature Range 200-2100 K,' *Teplotfiz. Vys. Temp.*, 16(1), 72-6 (1978); Engl. transl.: *High Temp.*, 16(1), 57-60 (1978).
58. Altshuler, B.L. and Aronov, A.G., 'Influence of Electron-Electron Correlations on the Resistivity of Dirty Metals,' *Pisma Zh. Eksp. Teor. Fiz.*, 27(12), 700-12 (1978); Engl. transl.: *JETP Lett.*, 27(12), 662-4 (1978).
59. Roberts, B.W., 'Properties of Selected Superconductive Materials,' *Natl. Bur. Stand. Rept. NBS-TN-983*, 103 pp. (1978).
60. Kulikov, N.I., 'McMillan-Hopfield Factor and Ideal Resistivity of Transition Metals,' *J. Phys. F*, 3(6), L137-40 (1978).
61. Khanna, S.N. and Jain, A., 'Calculations on Electrical Resistivity and the Thermoelectric Power of Nonsimple Metals,' *J. Phys. F*, 7(12), 2523-30 (1977).
62. Jaffee, R.I., 'Refractory Metals,' *Proc. Int. Symp. High Temp. Technol.*, 61-75 (1960).
63. Moraga, L.A. and Vilche, A., 'The Electrical Conductivity of Inhomogeneous Thin Metallic Films,' *Thin Solid Films*, 38(2), 117-30 (1976).
64. Khanna, S.N. and Jain, A., 'Electrical Resistivity of Noble and Transition Metals Using Animalu's Model Potential,' *J. Phys. Chem. Solids*, 38, 447-50 (1977).
65. McIntire, W.R. and Cohen, J.B., 'Static Distortions and Resistivity Due to Interstitials,' *Acta Metall.*, 23(8), 953-6 (1975).
66. Grigsby, D.L., 'Vanadium Silicide. Data Sheet,' *Hughes Aircraft Co. Rept. EPIC-DS-154*, 49 pp. (1967). [AD 810 374]
67. Smirnov, Yu.M. and Finkel, V.A., 'Crystal Structure of Tantalum, Niobium, and Vanadium at 110-400 K,' *J. Exptl. Theoret. Phys., USSR*, 42, 1077-82 (1965); Engl. transl.: *Sov. Phys.-JETP*, 22(4), 750-53 (1966).
68. Grigsby, D.L., 'A Compilation on the Transition Regions of Element Superconductors,' *Hughes Aircraft Co. Rept. EPIC-IR-28*, 35 pp. (1966).
69. Biget, M., Rizk, R., Vajda, P., and Bessis, A., 'On the Spontaneous Recombination Volume of Frenkel Defects in Irradiated b.c.c. Metals,' *Solid State Commun.*, 16(7), 949-52 (1975).

70. Smith, T.F., 'Pressure Dependence of the Superconducting Transition Temperature for Vanadium,' *J. Phys. F*, 2(5), 946-56 (1972). [AD 754 662]
71. Vajda, P. and Biget, M., 'Low-Temperature Fission Neutron Damage in Vanadium and Molybdenum,' *Phys. Status Solidi*, 23(1), 251-60 (1974).
72. McLlwain, J.F., Chen, C.W., Bajaj, R., and Wechsler, M.S., 'Effect of Neutron Irradiation on Vanadium,' Iowa State Univ. of Science and Technology, Dept. of Metallurgy, Rept. IS-2833, 21 pp. (1972).
73. Rostoker, W. and Yamamoto, A.S., 'A Contribution to the Oxygen Vanadium Phase Diagram,' *Trans. Am. Soc. Met.*, 67, 1002-17 (1955).
74. Suzuki, H., Minomura, S., and Miyahara, S., 'Effect of Pressure on the Anomaly of Vanadium,' *J. Phys. Soc. Jpn.*, 21(10), 2089 (1966).
75. Carlson, O.N. and Stevens, E.R., 'Vanadium and Vanadium Alloys,' *Encycl. Chem. Technol.*, 21, 157-67 (1970).
76. Weiss, V., Jelinek, R.V., Sessler, J.G., Latorre, A.V., and Ziemer, R.D., 'Air Weapons Materials Application Handbook Metals and Alloys,' U.S. Air Force Rept. ARDC-TR-59-66 (1962). [AD 287 184]
77. Druzhinina, I.P., Fraktovnikova, A.A., and Vladimirskaia, T.M., 'Effect of Group VI Elements on Thermoelectric Properties of Vanadium,' *Teplotfiz. Vys. Temp.*, 14(1), 221-3 (1976); Engl. transl.: *High Temp.*, 14(1), 200-2 (1976).
78. Westlake, D.G. and Miller, J.F., 'Resistivity Due to Hydrogen in Transition Metal Alloys,' *J. Phys. F*, 10(5), 859-63 (1980).
79. Weinstock, H., Tsai, C.L., and Schmidt, F.A., 'Thermal and Electrical Conductivity of Pure Vanadium at Low Temperatures,' *J. Phys. (Paris), Colloq.*, 2(6), 1026-7 (1978).
80. Dunleavy, H.N. and Jones, W., 'Multiple Scattering Calculations of the Resistivity of Liquid Transition Metals,' *J. Phys. F*, 8(7), 1477-82 (1978).
81. L'vov, S.N., Mal'ko, P.I., and Nemchenko, V.F., 'Effect of Temperature on the Thermal and Electrical Conductivity and the Wiedemann-Franz Relation for D-Transition Metals,' *Metallofizika*, 64, 63-8 (1976).

82. Hirata, K., Waseda, Y., Jain, A., and Srivastava, R., 'Resistivity of Liquid Transition Metals and Their Alloys Using the T Matrix,' *J. Phys. F*, 7(3), 419-25 (1977).
83. Vasileva, E.V., Gorbova, A.S., and Prokoshkin, D.A., 'Effect of Heat Treatment on Properties of Alloy Ni25Al5,' *Metalloved. Term. Obrab. Met.*, 5, 51 (1976); Engl. transl.: *Met. Sci. Heat Treat. Met.*, 18(5-6), 453-4 (1976).
84. Kharoo, H.L., Gupta, O.P., and Hemkar, M.P., 'Phonon-Limited Electrical and Thermal Resistivities of Noble Metals,' *Phys. Rev. B*, 18(10), 5419-26 (1978).
85. Belonov, O.K. and Kornilov, I.I., 'Density of States and Some Physical Properties of Vanadium-Chromium Alloys,' *Dokl. Akad. Nauk SSSR*, 210(2), 374-6 (1973).
86. Allen, P.B., 'Superconductivity and Phonon Softening,' *Phys. Rev. Lett.*, 29(24), 1593-6 (1972).
87. Collings, E.W., 'Anomalous Electrical Resistivity, Body-Centered-Cubic Phase Stability, and Superconductivity in Titanium-Vanadium Alloys,' *Phys. Rev. B*, 2(10), 3989-99 (1974).
88. Parker, R.D. and Halloran, M.H., 'Experimental Study of the Fermi Surface of Vanadium,' *Phys. Rev. B*, 2(10), 4130-7 (1974). [AD 785 941]
89. Postnikov, V.S., Milosheko, V.E., Shunin, G.E., and Shukhalov, E.I., 'Effect of the Normal-Superconducting Transition on the Internal Friction of Vanadium,' (Savitskii, E.M., Editor), *Strukt. Svoistva Sverkhprovodnykh Mater.*, 105-8 (1974).
90. L'vov, S.N., Mal'ko, P.I., and Nemchenko, V.F., 'High Temperature Resistivity of D-Transition Metals,' *Fiz. Met. Metalloved.*, 32(3), 485-91 (1971); Engl. transl.: *Phys. Met. Metallogr.*, 32(3), 35-40 (1971).
91. Chernoplekov, N.A., Panova, G.Kh., Samoylov, B.N., Zhernov, A.P., and Shikov, A.A., 'The Resistivity of Vanadium Alloyed with Tantalum,' *Fiz. Met. Metalloved.*, 16(5), 978-82 (1973); Engl. transl.: *Phys. Met. Metallogr.*, 16(5), 72-5 (1973).
92. Kerker, G. and Bennemann, K.H., 'Theory for Superconductivity in Amorphous Transition Metals,' *Z. Phys.*, 264(1), 15-20 (1973).

93. Savitskii, E.M. and Polyakova, V.P., 'Phase Diagrams and Properties of Noble Metals and Their Alloys,' Tr. Inst. Fiz. Metal. Ural. Nauch. Tsent. Akad. Nauk SSSR, 28, 44-57 (1971).
94. Grechko, O.G., L'vov, S.N., and Bondarev, V.N., 'Physical Properties of Group IV-VI Transition Metal Germanides,' Metallidy-Str., Svoistva, Primen., 142-9 (1971).
95. Chiu, J.C.H., 'Deviations from Linear Temperature Dependence of the Electrical Resistivity of Vanadium-Chromium and Tantalum-Tungsten Alloys,' Phys. Rev. B, 13(4), 1507-14 (1976).
96. Teplov, A.A., Mikheeva, M.N., Golyanov, V.M., and Gusev, A.N., 'Superconducting Transition Temperature, Critical Magnetic Fields, and the Structure of Vanadium Films,' Zh. Eksp. Teor. Fiz., 71(3), 1122-8 (1976); Engl. transl.: Sov. Phys.-JETP, 44(3), 587-91 (1976).
97. Osipov, K.A., Orlov, A.F., and Ivanovskaya, G.F., 'The Electrical Properties of Vanadium and Tantalum Films Produced and Annealed in Super-High Vacuum,' Izv. Akad. Nauk SSSR, Met., 1, 178-81 (1974); Engl. transl.: Russ. Metall., 1, 108-10 (1974).
98. Chander, R., Howard, R.E., and Jain, S.C., 'Electrical Conductivity and Hall Effect in Thin Vanadium Films,' J. Appl. Phys., 38, 4092-3 (1967).
99. Hoffman, D.W. and Thornton, J.A., 'The Compressive Stress Transition in Aluminum, Vanadium, Zirconium, Niobium and Tungsten Metal Films Sputtered at Low Working Pressures,' Thin Solid Films, 45(2), 387-96 (1977).
100. Alekseevskii, N.E. and Sakosarenko, V.M., 'Superconducting Properties of Vanadium Films,' Phys. Status Solidi, 34A(2), 541-6 (1976).
101. Borodziuk-Kulpa, A., Stolecki, B., and Wesolowska, C., 'Electrical Properties of Vanadium Films,' Thin Solid Films, 67(1), 21-8 (1980).
102. Lang, E. and Bressers, J., 'On the Effect of Some Sample Preparation Techniques on the Electrical and Caloric Properties of Vanadium,' Z. Metallkd., 67(1), 66-71 (1976).
103. Chaplin, R.L., Sonnenberg, K., and Coltman, R.R., Jr., 'A Measurement of the Frenkel Defect Resistivity in Vanadium and Its Implications,' Radiat. Eff., 27(1-2), 119-20 (1975).

104. Lakh, K.I. and Stasyuk, Z.V., 'Effect of Adsorption on the Electrical Conductivity of Vanadium Thin Films,' *Fiz. Elektron. (Lvov)*, **5**, 102-4 (1972).
105. Reale, C., 'Determination of Charge-Transport Parameters for Metals with Cubic Symmetry by a Study of Size Effects on Thin Films,' *Phys. Status Solidi*, **58B(1)**, K5-7 (1973).
106. Schmidt, P.H., 'Superconductivity of Transition Metal Thin Films Deposited by Noble Gas Ion Beam Sputtering,' *J. Vac. Sci. Technol.*, **10(5)**, 611-15 (1973).
107. Linker, G., Meyer, O., and Gettings, M., 'Back-Scattering Energy Loss Parameter Measurements in Thin Metal Films,' *Thin Solid Films*, **19(2)**, 177-85 (1973).
108. Felsch, W., 'Effect of Surface Charge on the Superconductivity of Vanadium Films,' *Low Temp. Phys.-LT 13, Proc. Int. Conf. Low Temp. Phys., 13th (Timmerhaus, K.D., O'Sullivan, W.J., and Hammel, E.F., Editors)*, **3**, 543-6 (1974).
109. Linker, G., 'Influence of Ion Bombardment on the Superconducting Transition Temperature of Evaporated Vanadium Layers,' *J. Nucl. Mater.*, **72(1-2)**, 275-81 (1978).
110. Kreinina, G.S., 'Current-Voltage Characteristics of Palladium and Vanadium Films,' *Radiotekh. Elektron.*, **17(6)**, 1269-72 (1972).
111. Schmidt, P.H., Castellano, R.N., Barz, H., Cooper, A.S., and Spencer, E.G., 'Variation of Superconducting Transition Temperatures of Transition-Metal Thin Films Deposited with the Noble Gases,' *J. Appl. Phys.*, **44(4)**, 1833-6 (1973).
112. Korshunov, V.D. and Tarakanov, V.I., 'Action of Gamma-Rays on Metal Thin Films,' *Izv. Vyssh. Ucheb. Zaved., Fiz.*, **15(11)**, 143-5 (1972); Engl. transl.: *Sov. Phys. J.*, **15(11)**, 1674-6 (1972).
113. Korshunov, V.D. and Tarakanov, V.I., 'Influence of Gamma Rays on the Structure and Properties of Thin Films,' *Fiz. Khim. Obrab. Mater.*, **5**, 46-9 (1974).
114. Noer, R.J., 'Superconductive Tunneling in Vanadium with Gaseous Impurities,' *Phys. Rev. B*, **12(11)**, 4882-5 (1975).

115. Nestell, J.E., Jr., 'Optical Properties and Structure of Vanadium, Niobium, Tantalum, Chromium, Molybdenum, and Tungsten Films,' Dartmouth College, Ph.D. Thesis, 233 pp. (1979). [Univ. Microfilm No. 80-01173]
116. Eto, M. and Narutani, T., 'Contribution of Dissolved or Precipitated Oxygen to the Electrical Resistivity of Vanadium,' J. Mater. Sci., 9(11), 1902-4 (1974).
117. Peterson, D.T. and Jensen, C.L., 'Electromigration of Hydrogen and Deuterium in Vanadium and Niobium by a Resistance Method,' Metall. Trans., 9A(11), 1673-7 (1978).
118. Watanabe, K. and Fukai, Y., 'Electrical Resistivity Due to Interstitial Hydrogen and Deuterium in Vanadium, Niobium, Tantalum, and Palladium,' J. Phys. F, 10(8), 1795-801 (1980).
119. Druzhinina, I.P., Fraktovnikova, A.A., and Vladimirskaya, T.M., 'Influence of Rhenium on the Thermoelectric Properties of Vanadium,' Teplofiz. Vys. Temp., 14(3), 652-4 (1976); Engl. transl.: High Temp., 14(3), 579-81 (1976).
120. Stanley, J.T., Williams, J.M., Brundage, W.E., and Wechsler, M.S., 'Effect of Interstitial Impurities on the Annealing of Neutron-Irradiated Vanadium,' Acta Metall., 20(2), 191-8 (1972).
121. Touloukian, Y.S. and Ho, C.Y., Editors, Properties of Selected Ferrrous Alloying Elements, Vol. III-1 of McGraw-Hill/CINDAS Data Series on Material Properties, McGraw-Hill Book Co., New York, 269 pp. (1981).
122. Brandt, N.B. and Zarubina, O.A., 'Superconductivity of Vanadium at Pressures Up to 250 Kbar,' Fiz. Tverd. Tela, 15(11), 3423-5 (1973); Engl. transl.: Sov. Phys.-Solid State, 15(11), 2281-2 (1974).
123. Cozairliyan, A. and Righini, F., 'Measurement of Melting Point, Radiance Temperature (at Melting Point), and Electrical Resistivity (above 2100 K) of Zirconium,' Rev. Int. Hautes Temp. Refract., 12(3), 201-7 (1975).
124. Cozairliyan, A. and Righini, F., 'Thermodynamic Studies of the Alpha-Beta Phase Transformation in Zirconium Using a Subsecond Pulse Heating Technique,' J. Res. Natl. Bur. Stand., 79A(1), 81-4 (1975). [AD A013 906]

125. Cezairliyan, A. and Righini, F., 'Simultaneous Measurements of Heat Capacity, Electrical Resistivity and Hemispherical Total Emittance by a Pulse Heating Technique: Zirconium, 1500 to 2100 K,' J. Res. Natl. Bur. Stand., 78A(4), 509-14 (1974).
126. Hörz, G., Hammel, M., and Kanbach, H., 'Electrical Resistivity of Beta-Zirconium-Oxygen Solid Solutions as a Function of Oxygen Concentration and Temperature,' J. Nucl. Mater., 55(3), 291-8 (1975).
127. Bykov, V.N., Rudnev, I.I., and Solov'ev, V.A., 'Anomalies of the Physical Properties of Alpha-Titanium and Alpha-Zirconium,' Elektron. Str. Fiz. Svoistva Tverd. Tela, 1, 42-7 (1972).
128. Mal'tsev, V.A., 'Physical Properties of the Omega Phase of Zirconium,' Fiz. Met. Metalloved., 36(2), 413-14 (1973); Engl. transl.: Phys. Met. Metallogr., 36(2), 174-6 (1973).
129. Martynyuk, M.M. and Tsapkov, V.I., 'The Applicability of Mott's Formula to the Fusion of Transition Metals,' Zh. Fiz. Khim., 47(5), 1308-98 (1973); Engl. transl.: Russ. J. Phys. Chem., 47(5), 741-2 (1973).
130. Reale, C., 'Determination of Charge-Transport Parameters for the Group IV Metals,' Rev. Bras. Fis., 3(3), 431-9 (1973).
131. Volkenshtein, N.V., Novoselov, V.A., and Startsev, V.E., 'Role of Inter-electron Collisions in the Electric Resistance of Transition Metals,' Zh. Eksp. Teor. Fiz., 60(3), 1078-85 (1971); Engl. transl.: Sov. Phys.-JETP, 33(3), 584-7 (1971).
132. Zhorov, G.A., 'Emissivity of Metals of the IVb Subgroup at High Temperatures,' High Temp., 8(3), 501-4 (1970).
133. Peletskii, V.E., Druzhinin, V.P., and Sobol, Ya.G., 'Emissivity, Thermal Conductivity, and Electrical Conductivity of Remelted Zirconium at High Temperatures,' High Temp., 8(4), 732-6 (1970).
134. Elliott, R.O. and Hill, H.H., 'Resistance Minima in Zirconium-Plutonium Alloys,' J. Less-Common Met., 22(1), 123-6 (1970).
135. Betterton, J.O., Jr. and Easton, D.S., 'Electrical Resistivity of Zirconium Alloys,' USARC Rept. ORNL-4370, 23-4 (1968).

136. Clinard, F.W., Jr. and Kempter, C.P., 'Low-Temperature Electrical Properties of Some Transition Metals and Transition-Metal Carbides,' *J. Less-Common Met.*, 15(1), 59-73 (1968).
137. Cape, J.A. and Hake, R.R., 'Localized Magnetic Impurity States in Titanium, Zirconium, and Hafnium,' *Phys. Rev. A*, 139(1), 142-9 (1965).
138. Powell, R.W. and Tye, R.P., 'The Thermal and Electrical Conductivities of Zirconium and of Some Zirconium Alloys,' *J. Less-Common Met.*, 3, 202-15 (1963).
139. Kiselev, N.A., 'Device for Determining the Electrical Resistance and the Melting Point of Metals and Alloys,' *Metallurgiya, Metallovedenie, Fiziko-Khimicheskie Metody Issledovaniya*, 3, 1 (1958); Engl. transl.: OTS-60-51087, 3, 280-7 (1961).
140. Berlincourt, T.G., 'Hall Effect, Resistivity, and Magnetoresistivity of Thorium, Uranium, Zirconium, Titanium and Niobium,' *Phys. Rev.*, 114(4), 969-78 (1959).
141. Kemp, W.R.G., Klemens, P.G., and White, G.K., 'Thermal and Electrical Conductivities of Iron, Nickel, Titanium, and Zirconium at Low Temperatures,' *Aust. J. Phys.*, 2, 180-8 (1956).
142. Adenstedt, H.K., 'Physical, Thermal and Electrical Properties of Hafnium and High Purity Zirconium,' *Trans. Am. Soc. Metals*, 44, 949-73 (1952).
143. Bing, G., Fink, F.W., and Thompson, H.B., 'The Thermal and Electrical Conductivities of Zirconium and Its Alloys,' *USAEC Rept. BMI-65*, 19 pp. (1951).
144. Cook, L.A., Castleman, L.S., and Johnson, W.E., 'Preliminary Report on the Electrical Resistivity of Zirconium,' *Westinghouse Electric Corp. Rept. WAPD-25*, Dec. 20 (1950).
145. Nakagawa, M., Mansel, W., Boening, K., Rosner, P., and Vogl, G., 'Spontaneous Recombination Volumes of Frenkel Defects in Neutron-Irradiated Non-Face-Centered Cubic Metals,' *Phys. Rev. B*, 19(2), 742-8 (1979).
146. Azhazha, V.M., V'yugov, P.N., Reshetova, L.N., San'kov, A.A., and Tarasova, M.I., 'Purification of Zirconium by Vacuum Zone Refining,' *Izv. Akad. Nauk SSSR, Met.*, 1, 41-4 (1975); Engl. transl.: *Russ. Metall.*, 1, 36-9 (1975).

147. Hörz, G. and Hammel, M., 'Kinetics and Mechanism of the Reaction of Beta-Zirconium with Oxygen at Low Pressures,' *J. Nucl. Mater.*, 55(3), 284-90 (1975).
148. Baranov, I.A., Bychkov, Yu.F., Korzhov, V.P., Mal'tsev, V.A., Slavgorodskii, M.P., and Shmalevich, R.S., 'Effect of Rhodium on Superconductivity of Zirconium and Some of Its Alloys,' *Sverkhprovodyashchie Splavy Soedin.*, 140-7 (1972).
149. Fedorov, G.B., Zuev, M.T., Smirnov, E.A., and Kissil, A.E., 'Physical Properties of Uranium Zirconium Alloys at Low Temperatures,' *At. Energ.*, 34(2), 85-8 (1972).
150. Pande, B.M., Anand, M.S., and Argarwala, R.P., 'Recovery of Deformed Zirconium Above Room Temperature,' *Phys. Status Solidi*, 10A(2), K137-9 (1972).
151. Claisse, F., Cormier, M., and Frigout, C., 'The Exponential Temperature Dependence of the Electrical Resistivity of Hexagonal Close-Packed Hafnium, Titanium and Zirconium Alloys,' *High Temp.-High Pressures*, 4(4), 395-9 (1972).
152. Hall, L.A. and Germann, F.E.E., 'Survey of Electrical Resistivity Measurements on 8 Additional Pure Metals in the Temperature Range 0 to 273 K,' *Natl. Bur. Stand. Tech. Note* 365-1, 85 pp. (1970).
153. Blumenthal, W.B. and Roach, J.D., 'Zirconium Metal and Its Metal-Like Compounds,' *Encycl. Chem. Technol.*, 22, 614-29 (1970).
154. Golutvin, Yu.M. and Maslennikova, E.G., 'The Heat Capacity of Metallic Hafnium,' *Izv. Akad. Nauk SSSR, Met.*, 5, 174-83 (1970); Engl. transl.: *Russ. Metall.*, 5, 129-35 (1970).
155. Parker, E.R., Materials Data Book for Engineers and Scientists, McGraw-Hill Book Co., New York, 398 pp. (1967).
156. Bridgman, P.W., 'The Resistance of 72 Elements, Alloys and Compounds to 100,000 kg/cm,' *Proc. Amer. Acad. Arts Sci.*, 81(4), 165-251 (1952).
157. Squire, C.F. and Kaufmann, A.R., 'The Magnetic Susceptibility of Titanium and Zirconium,' *J. Chem. Phys.*, 9(9), 673-7 (1941).

158. Fast, J.D., 'The Manufacturing of the Pure Metal of the Titanium Group by Thermal Replacements of Their Iodide. IV. The Occurrence of the Lower Zirconium Iodide During the Manufacturing of Ductile Zirconium,' *Z. Anorg. Allgem. Chem.*, 239, 145-54 (1938).
159. McLennan, J.C., Howlett, L.E., and Wilhelm, J.O., 'On the Electrical Conductivity of Certain Metals at Low Temperatures,' *Trans. Roy. Soc. Can.*, 23, Sect. III, 287-306 (1929).
160. Clausing, P. and Moubis, G., 'Electrical Resistance of Titanium at Low Temperature,' *Physica*, 7, 245-50 (1927).
161. Bridgman, P.W., 'The Compressibility and Pressure Coefficient of Resistance of Zirconium and Hafnium,' *Proc. Amer. Acad. Arts Sci.*, 63, 347-50 (1928).
162. Meissner, W. and Voigt, B., 'Measurements with the Aid of Liquid Helium. XI. Resistance of Pure Metals at Low Temperatures,' *Ann. Physik*, 7(7), Pt. 5, 761-97, 892-936 (1930).
163. Akatsu, E., 'Chemistry of the Fission Products,' Japan Atomic Energy Research Inst., Tokyo, Rept. JAERI-M-7873, 78 pp. (1978).
164. Volkenshtein, N.V., Galoshina, E.V., and Panikovskaya, T.N., 'Temperature Dependence of Magnetic Susceptibility Anisotropy of Transition Metal Crystals with a Hexagonal Close Packed Structure,' *Zh. Eksp. Teor. Fiz.*, 67(4), 1468-73 (1974); Engl. transl.: *Sov. Phys.-JETP*, 40(4), 730-2 (1975).
165. Touloukian, Y.S., Editor, Thermophysical Properties of High Temperature Solid Materials. Volume 1: Elements, MacMillan Co., New York, 1152 pp. (1967).
166. Vaynshteyn, A.A., Rusinov, P.S., and Papina, N.V., 'The Heterogeneous Properties of Polycrystals with an Axial Texture,' *Fiz. Met. Metalloved.*, 42(5), 957-61 (1979); Engl. transl.: *Phys. Met. Metallogr.*, 42(5), 49-53 (1979).
167. Moers, K., 'Conductivity Measurements on High Melting Carbides, Nitrides and Borides at Room, Low and High Temperatures,' *Z. Anorg. Allgem. Chem.*, 192, 262-75 (1931).

168. Martyniuk, W.M. and Tsapkov, V.I., 'Electrical Resistance, Enthalpy, and Phase Transitions of Titanium, Zirconium, and Hafnium During Pulsed Heating,' *Izv. Akad. Nauk SSSR, Met.*, 181-8 (1974); Engl. transl.: *Russ. Metall.*, 108-12 (1974).
169. Rassmann, G. and Merz, A., 'Developments in the Field of High Temperature Materials,' *Technik*, 17(2), 74-9 (1962).
170. Grigsby, D.L., 'Electrical Resistivity of Selected Materials,' Hughes Aircraft Co. Rept. EPIC-IR-44, 15 pp. (1966).
171. Antler, M. and Krumbein, S.J., 'Contact Properties of Conductive Hardmetals and of Tin Nickel Plate,' in Electrical Contacts-1965, Proc. Eng. Seminar on Electrical Contacts, Univ. of Maine, 103-37 (1965).
172. Fuschillo, N. and Lindberg, R.A., 'Electrical Conductors at Elevated Temperatures,' U.S. Air Force Rept. ASD-TRD-62-481 (1962). [AD 299 020]
173. Dayton, R.W., 'Zirconium and Its Alloys,' in Reactor Handbook: Materials. Vol. 4, 459-504 (1955).
174. Lustman, B. and Kerze, F., Jr., The Metallurgy of Zirconium, McGraw-Hill Book Co., New York, 776 pp. (1955).
175. Rosenberg, H.M., 'The Thermal Conductivity of Metals at Low Temperatures,' *Philos. Trans. R. Soc. London*, A247, 441-97 (1955).
176. DeBoer, J.H. and Clausing, P., 'Electrical Resistance of Titanium, Zirconium, and Solid Solutions,' *Physica*, 10, 267-9 (1930).
177. DeBoer, J.H. and Fast, J.D., 'The Production of the Pure Metals of the Titanium Group by Thermal Replacement of Their Iodides. III. Hafnium,' *Z. Anorg. Allgem. Chem.*, 187, 193-208 (1930).
178. Huett, R., 'The Physical Ordering Principle of Crystals in Nature,' *Wissenschaftlicher Bericht, Gesamthochschule Siegen, Abteilung Gummersbach*, 1-2 (1978).
179. Brehm, C. and Lehr, P., 'Plastic Deformation of Titanium and Zirconium,' *Melaux*, 46(551-2), 253-71; (553), 325-34; (554), 359-81 (1971).
180. Stewart, R.B. and Johnson, V.J., 'A Compendium of the Properties of Materials at Low Temperature (Phase),' U.S. Air Force Rept. WADD-TR-60-56, Part IV, 501 pp. (1961).

181. Greig, D. and Morgan, G.J., 'The Electrical Resistivity of Transition Metals at High Temperatures,' *Philos. Mag.*, 27(8), 929-40 (1973).
182. Pav, T. and Saxl, I., 'Physical Properties of Zirconium and Its Alloys,' *Ustav Jaderneho Vyzkumu, Ceskoslovenska, Akademie Ved. Rez. Rept. UJV-2708-M*, 119 pp. (1971).
183. Degtyareva, V.F., Karimov, Yu.S., and Rabin'kin, A.G., 'Superconductivity and Magnetic Susceptibility of the Alpha and Omega Modifications of Titanium and Zirconium,' *Fiz. Tverd. Tela*, 15(11), 3436-8 (1973); Engl. transl: *Sov. Phys.-Solid State*, 15(11), 2293-4 (1974).
184. Rosenbaum, M., Bisogni, E.A., and Blewitt, T.H., 'Resistivity Changes in Neutron-Irradiated Zirconium at 77 K,' *J. Nucl. Mater.*, 48(2), 201-3 (1973).
185. Omar, A.M., Robinson, J.E., and Thompson, D.A., 'The 10-16 MeV Proton Irradiation of Iron, Zirconium and Copper: Resistivity-Dose Measurements,' *J. Nucl. Mater.*, 84(1-2), 173-82 (1979).
186. Filippov, V.F., 'Study of Some Properties of Zirconium with Additions of Transition Metals,' *Joint Publications Research Service Rept. JPRS-59873*, 75-83 (1973).
187. Reale, C., 'Results of Combined Measurements of Resistivity and Thermopower of Transition and Noble Metal Films,' *Phys. Lett. A*, 50(1), 53-4 (1974).
188. Bastl, Z., 'Influence of Interstitial Hydrogen on Hall Voltage and on Electric Resistance of Thin Zirconium Films,' *Collect. Czech. Chem. Commun.*, 40(7), 1987-96 (1975).
189. Eichler, A. and Gey, W., 'Superconductivity in Alpha- and Omega-Zirconium Under High Pressure,' *Z. Phys.*, 251(4), 321-3 (1972).
190. Touloukian, Y.S., Kirby, R.K., Taylor, R.E., and Desai, P.D., Thermal Expansion, Metallic Elements and Alloys. Vol. 12 of Thermophysical Properties of Matter - The TPRC Data Series, IFI/Plenum Data Corp., New York, 1348 pp. (1975).

191. Touloukian, Y.S., Powell, R.W., Ho, C.Y., and Kleens, P.G., Thermal Conductivity - Metallic Elements and Alloys, Vol. 1 of Thermophysical Properties of Matter - The TPRC Data Series, IFI/Plenum Data Corp., New York, 1595 pp. (1970).
192. Laws, F.A., Electrical Measurements, 2nd Edition, McGraw-Hill Book Co., Inc., New York, 739 pp. (1938).
193. Harris, F.K., Electrical Measurements, John Wiley and Sons, Inc., New York, 784 pp. (1952).
194. Meaden, G.T., Electrical Resistance of Metals, Plenum Press, New York, 218 pp. (1965).
195. Taylor, R.E. and Groot, H., 'Operating Manual for Kohlrausch Apparatus,' Thermophysical Properties Research Laboratory Rept. TPRL 291, 11 pp., 1978.
196. Taylor, R.E., 'A Description of the Thermophysical Properties Research Laboratory,' Thermophysical Properties Research Laboratory Rept. TPRL 181 (Revised), 72 pp., 1982.
197. Radenac, A., Lacoste, M., and Roux, C., 'Apparatus Meant for the Measurement of the Electrical Resistivity of Metals and Alloys by the Method of the Rotating Field Up to About 2000 K,' Rev. Int. Hautes Temp. Refract., 7(4), 389-96 (1970).
198. Cezairliyan, A., Morse, M.S., Burman, H.A., and Beckett, C.W., 'High-Speed (Subsecond) Measurement of Heat Capacity, Electrical Resistivity, and Thermal Radiative Properties of Molybdenum in the Range 1900-2800 K,' J. Res. Natl. Bur. Stand., Sect. A, 74A(1), 65-92 (1970).