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DMIC Memorandum 201

COMPATIBILITY OF MATERIALS WITH ROCKET  
PROPELLANTS AND OXIDIZERS

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AD-613 553

COMPATIBILITY OF MATERIALS WITH  
ROCKET PROPELLANTS AND OXIDIZERS

W. K. Boyd, et al

Battelle Memorial Institute  
Columbus, Ohio

January 1965

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<b>13. ABSTRACT</b> Compatibility data are summarized for the storage and handling of metals and nonmetals in the following rocket propellants and oxidizers: ammonia, HiCal-3, pentaborane, trialkyl boranes, fluorine, TiOX, oxygen difluoride, ozone difluoride, chlorine trifluoride, bromine trifluoride, bromine pentafluoride, iodine pentafluoride, perchloryl fluoride, halogenated hydrocarbons, hydrazine, monomethyl hydrazine, unsymmetrical dimethyl hydrazine, Aerozine-50 hydrogen, hydrogen peroxide, methylene chloride, red fuming nitric acid, white fuming nitric acid, concentrated acid, nitrogen tetroxide, oxygen, ozone, nitronium perchlorate, and solid propellant ANP-2639 AF. (Author)			

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Fuels Oxidizers Compatibility Light metals Refractory metals High-strength steels Stainless steels Superalloys Other specific metals Metals (in general) Nonmetallics						

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 ROCKET PROPELLANTS AND GRIDIZERS

**ERRATA**

In Table 6 on page 12, several entries are in error. The corrections are listed below. In table 29, a rating for Monel is supplied that was inadvertently omitted during copying from the worksheets.

ERRATA FOR TABLE 6

Material	Temperature, F								References
	Gas				Liquid				
	Class 1	Class 2	Class 3	Class 4	Class 1	Class 2	Class 3	Class 4	
309 Stainless Steel	500			570					7,207
309 Cb Stainless Steel	500			570					7,207
310 Stainless Steel	500			660					7,207
347 Stainless Steel	390			500	-310		-320		7,120,207,274
430 Stainless Steel		400	390	600					7,207,211
Azmco Iron	390	500	167						7,159,160,207
Iron (0.004 Si)		200	400	390					82,159,160
Iron (0.79 Si)		100	300	390					82,159,160
Sheet Steel	390	660		500					7,207
SAE 1010	100	200	400	>400					82,160
SAE 1011				570					7,207
SAE 1020				390					7,207
SAE 1030	660		390	500					7,207
Music Wire				570					7,207
A-Nickel		1000	750	>1200					7,73,82,120,143,153,159,160,211,274
Monel		1000	750	>1200					7,73,82,120,143,153,159,160,211,274
Inconel		1000		<750					7,207,211
Deoxidized copper				<400					7,207
Brass 70-30		200		400					82,159,160
Magnesium MA (1.2% Mn)	140	200							82,159,160
Magnesium FS-1A	140	200							82,159,160

ERRATUM FOR TABLE 29

Monel	75	125	100	75	211
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# COMPATIBILITY OF MATERIALS WITH ROCKET PROPELLANTS AND OXIDIZERS

W. K. Boyd, W. E. Berry, and E. L. White\*

## GENERAL CONSIDERATIONS

An important consideration in rocket technology is the compatibility of each propellant with the container material. Serious problems arise because many propellants are extremely reactive and their containment is possible only with a few materials of construction. The resistance of many alloys to fuels and oxidizers is dependent entirely on the formation of an inert, corrosion-resistant film or barrier coating. In addition to corrosion problems, the presence of some metals tends to promote decomposition of the propellant. Also, certain metal/oxidizer combinations may ignite if subjected to impact.

Four years ago DMIC Memorandum 65 was issued and included all information available at that time on compatibility of materials of construction with rocket propellants and oxidizers. Since that time additional information on compatibility has been generated for new, as well as the more established, fuels and oxidizers. This report contains these new data combined with the information presented in DMIC Memorandum 65. In order to expand the usefulness of the report, the source of the data is referenced for each material.

This memorandum summarizes the available information on the compatibility of liquid rocket propellants with prominent materials of construction. It is pointed out that compatibility data for materials not ordinarily covered by the Defense Metals Information Center are included. These data were found during the search for information on materials that are within the scope of the DMIC, and are included for convenience. Fuels and oxidizers of current interest are discussed. The corrosion data which are presented will apply to storing, handling, and control equipment outside of missiles and to missile components excluding combustion chamber. The compatibility of materials with reaction products in combustion chambers, nozzles, etc., has not been considered. Included in the summary are data for many nonmetallic materials. These data were collected in conjunction with those obtained for metals but no concerted effort was made to secure compatibility data for nonmetals.

The memorandum is subdivided into sections according to the propellant. Each material of construction is rated for a given medium as belonging to one of four classes, based primarily upon corrosion resistance. Consideration also is given to such factors as catalytic decomposition and sensitivity to impact.

## CLASSIFICATION OF MATERIALS OF CONSTRUCTION

### Metals

#### Class 1

The Class 1 materials are those which exhibit a corrosion rate of less than 1 mil per year. The material does not promote decomposition of the propellant or oxidizer and is free from impact sensitivity.

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#### Class 2

The materials falling in Class 2 are similar to those in Class 1, except that the corrosion rate may be as great as 5 mils per year.

#### Class 3

A material in Class 3 shows only fair corrosion resistance; rates of attack may be of the order of 5 to 50 mils per year. The material may also cause a moderate breakdown of the propellant, but it is not shock sensitive under conditions likely to be encountered in service.

#### Class 4

Materials in this class are not considered usable for containing the propellant; they have corrosion rates greater than 50 mils per year, cause extensive decomposition of the propellant, cause spontaneous ignition, or react on impact.

These classifications are summarized in Table 1.

TABLE 1. COMPATIBILITY CLASSIFICATIONS FOR METALS<sup>(a)</sup>

Class	Rating	Corrosion Resistance		Decomposition of Propellant	Shock Sensitivity
		Penetration Rate, mils/year			
1	Excellent	<1		No	No
2	Good	<5		No	No
3	Fair	5 to 50		Some	No
4	Poor	>50		Extensive	Yes

(a) The classification of a material is based on the lowest rating of any one of the three properties.

### Nonmetals

Ratings for nonmetals are also somewhat arbitrary but wherever possible they follow those described in the Titan II Storable Propellant Handbook, (169). The classifications are summarized in Table 2.

TABLE 2. COMPATIBILITY CLASSIFICATIONS FOR NONMETALS

	Class			
	1	2	3	4
Volume Change, percent	0 to +25	-10 to +25	-10 to +25	<-10 or >+25
Diameter Reading Change	±3	±10	±10	<-10 or >+10
Effect on Propellant	None	Slight change	Moderate change	Severe
Visual Examination	No change	Slight change	Moderate change	Severely blistered, or cracked, dissolved
General Usage	Satisfactory, general use	Satisfactory for repeated short term use	Satisfactory for short time use	Unsatisfactory

The compatibility data have been tabulated according to the maximum temperature permissible for a given material in Classes 1, 2, and 3. The minimum temperature at which a given material becomes Class 4 (noncompatible) also is listed. For example, a notation of RT under Class 1 means that the material would fall into this classification up to room temperature. It will be noted, in many instances, that no temperature is listed for one of the more resistant classifications. This does not necessarily mean that the material does not fall in Class 1 or 2 at some temperature, but rather that insufficient data are available to assign a temperature limit.

Occasional entries indicate that a material has a higher rating at higher temperatures, e.g., Class 1 at 160 F and Class 2 at 80 F. These entries reflect a conflict in reported data. In many of these cases, it is recommended that the original references be consulted where possible, to determine which results were obtained under conditions most nearly simulating the application in question.

In many cases, a material does not fit into a classification because (1) there is a scarcity of numerical data, or (2) the decomposition effects on the propellant are of primary concern. Hydrogen peroxide is a good example of a propellant for which it is difficult to classify construction materials. In such a case, the classification used in the table is described in detail for the material in question. Many materials are listed by trade names. Similar materials marketed under other trade names probably would be given the same classification. However, only materials for which actual data are available are listed.

#### SOURCES OF INFORMATION

The information on which this memorandum is based came from a variety of sources. A list of references is given at the end of the text. Appropriate references for a particular propellant material are listed at the beginning of each section. References are also included for each individual entry. In addition to reference material obtained from published literature, specialists in companies active in the development of rocket propellants were contacted either by letter, by telephone, or in person. The cooperating companies are listed below.

Aerogel-General Corporation  
 Food Machinery and Chemical Corporation  
 Becco Chemical Division  
 Belmont Smelting and Refining Works, Inc.  
 General Dynamics Corporation  
 Convair Astronautics Division  
 Callery Chemical Company  
 Research and Development Division  
 Celanese Corporation of America  
 Columbia-Southern Chemical Corporation  
 Commercial Solvents Corporation  
 Douglas Aircraft Company, Inc.  
 Diamond Alkali Company  
 The Dow Chemical Company  
 Texas Division  
 E. I. du Pont de Nemours and Company, Inc.  
 Foote Mineral Company  
 Allied Chemical Corporation  
 General Chemical Division  
 Hughes Tool Company  
 HEF, Inc. - Hooker Chemical Corporation and  
 Foote Mineral Company

Hercules Powder Company  
 California Institute of Technology  
 Jet Propulsion Laboratory  
 Lockheed Aircraft Corporation  
 Missile Systems Division  
 Lithium Corporation of America  
 Menasco Manufacturing Company  
 Metal Hydrides, Inc.  
 Monsanto Chemical Company  
 Research and Engineering Division  
 Chemetron Corporation  
 National Cylinder Gas Division  
 Union Carbide Corporation  
 Linde Company  
 Union Carbide Metals Company  
 National Carbon Company  
 Olin-Mathieson Chemical Corporation  
 Pennsalt Chemicals Corporation  
 Thiokol Chemical Corporation  
 Reaction Motors Division  
 North American Aviation, Inc.  
 Rocketdyne Division  
 Rohm & Haas Company  
 Stanford Research Institute  
 Solar Aircraft Company  
 Sinclair Research Laboratories, Inc.  
 Titanium Metals Corporation of America  
 Virginia-Carolina Chemical Corporation  
 Wyandotte Chemicals Corporation.

General information on compatibility and properties of large groups of propellants are contained in References 3, 23, 81, 102, 110, 151, 196, 198, 211, and 237. Information on handling, safety, and toxicity is included in References 81, 195, 291, and 302.

#### AMMONIA (NH<sub>3</sub>)\*

Ammonia is a pungent colorless gas that is alkaline in nature. It can be liquified at room temperature at pressures above 100 psia. The vapor irritates the eyes and respiratory tract. The threshold-limit value of toxicity in the atmosphere is 50 ppm.

Stainless steel, carbon steel, nickel alloys, silver, platinum, gold, and tantalum are sufficiently resistant to anhydrous ammonia to be placed in Class 1, as shown in Table 3. Inconel, gold, platinum, and tantalum are Class 1 materials in moist ammonia. Carbon steel and cast iron are also quite resistant and are normal materials of construction for ammonia service.

The copper alloys are less resistant than steel and have the disadvantage of being susceptible to cracking in ammonia atmosphere.

The upper temperature limit of many metals is related to the initiation of the nitriding process. Inconel is more resistant to nitriding than other nickel alloys, mild steel, or stainless steel.

Many organic materials are suitable for ammonia service. Plastics and elastomers usually resist attack up to their softening point.

Most inorganic construction materials are not attacked by ammonia. Graphitic materials are considered best for handling ammonia gas at very high temperatures.

\*Ammonia: see References 80, 81, 82, 94, 102, 109, 110, 127, 128, 151, 181, 199, 211, 214, 217, 221, 287, and 295.

Hi-Cal-3\*

The composition and properties of Hi-Cal-3 are classified.

Many of the common construction materials are compatible with Hi-Cal-3. Mild steels, stainless steels, copper alloys, nickel alloys, aluminum, titanium, tantalum, and lead can all be used up to 120 F and are listed as Class 1 in Table 4. Above this temperature, no data are available. Organic materials which are compatible with Hi-Cal-3 are listed in Table 4 as Class 2.

Pentaborane (B<sub>5</sub>H<sub>9</sub>)\*\*

Pure pentaborane is a clear colorless liquid that possesses an odor similar to that of rotten pumpkin. It is pyrophoric and highly toxic. Maximum allowable exposure is less than 1.0 ppm. It has a vapor pressure of 77 mm at 77 F and boils at 137 F.

Most of the metals used in rocket construction are compatible with pentaborane, including iron, steel, stainless steel, aluminum, copper, brass, magnesium, titanium, etc. Data are presented in Table 5. Teflon, Viton, Kel-F, and fluorosilicon rubber are included among the plastics and elastomers that are compatible with pentaborane. Pentaborane forms shock-sensitive mixtures with most of the chlorinated hydrocarbons that are used as degreasers or solvents.

Trialkyl Boranes\*\*\*

Materials which have withstood 1 month's exposure in triethylborane [(C<sub>2</sub>H<sub>5</sub>)<sub>3</sub>B] (a colorless liquid that boils at 203 F) at 160 F with no apparent attack include:

<u>Metals</u>	<u>Nonmetals</u>
Mild steel	Teflon
Stainless steel	Phenolite
Aluminum	Garlock 900 packing
Brass	Garlock red rubber
Nickel	Koppers 6200
Monel	Super Dylan polyethylene
Inconel	
Lead	
Copper (pitted)	

Materials which have withstood 2 weeks' exposure in tri-n-butyl borane [(C<sub>4</sub>H<sub>9</sub>)<sub>3</sub>B] (a colorless liquid that exerts a vapor pressure of 20 mm at 228 F) at 122 F with no apparent attack include:

<u>Metals</u>	<u>Nonmetals</u>
Mild steel	Teflon gasket
Stainless steel	Palmer to gasket
Aluminum	Koroseal
Brass	Kel-F
Nickel	Hycar
Copper	Nylon gasket

\*Hi-Cal-3: see Reference 103.

\*\*Pentaborane: see References 56 and 294.

\*\*\*Trialkyl boranes: See Reference 57.

A number of fluorine compounds are being considered as oxidizers for rocket-propulsion systems. These include fluorine, chlorine trifluoride, bromine trifluoride, bromine pentafluoride, iodine pentafluoride, oxygen difluoride, oxygen difluoride-oxygen mixtures, perchloryl fluoride, perchloryl fluoride-tetrafluoro-hydrazine mixtures, and fluorine-oxygen mixtures (FLOX). All of these materials are extremely active chemically. Under the proper conditions, they will react with almost every known element; hence, they present severe corrosion and compatibility problems. These oxidizers, however, are not for the most part susceptible to thermal and catalytic breakdown and present little or no problem in this respect. Therefore, compatibility ratings are based primarily on the reaction of the medium with the material in question.

Fluorine (F<sub>2</sub>)\*

Fluorine is a yellowish gas that has a pungent odor and is irritating to the respiratory tract. The threshold limit of toxicity of fluorine in the atmosphere is 0.1 ppm. It is normally handled as a liquid at -310 to -320 F.

Many metals perform well in liquid and gaseous fluorine as can be seen from the data in Table 6. It is believed that a protective fluoride film which forms on the surface of most metals imparts corrosion resistance. On the other hand, some experiments have revealed no increased corrosion on specimens immersed in liquid fluorine and wire brushed to remove any film, suggesting no passivating effect from metal fluorides.(275)

Traces of water in the system react with fluorine to form hydrofluoric acid. This acid tends to attack some of the materials which are resistant to uncontaminated fluorine. Since moisture may be present in many systems, Monel is usually chosen as a construction material because, in addition to being resistant to fluorine, it also is resistant to hydrofluoric acid.

Titanium ignites\*\* in liquid fluorine if subjected to impact or rupture. However, ignition does not propagate as it does in liquid oxygen. Although ignition has occurred in at least one specimen of aluminum alloys, aluminum is not considered to be impact sensitive in liquid fluorine.(275) Dynamite cap explosions against tubes filled with liquid fluorine have failed to cause ignition in Monel, nickel, copper, brass, 304 stainless steel, 316 stainless steel, 347 stainless steel, and 1100 aluminum.(275) Tensile tests to fracture in liquid fluorine have not caused ignition of AM350, 304L stainless steel, 301 stainless steel, ASM 6434, 2014-T6 and 7075-T6 aluminum, Inconel X, and Ti-6Al-4V.(233)

\*Fluorine: see References 7, 14, 15, 16, 20, 40, 52, 58, 73, 81, 82, 83, 86, 90, 91, 93, 94, 99, 100, 102, 110, 120, 121, 122, 123, 132, 134, 143, 146, 151, 152, 153, 159, 160, 161, 162, 163, 175, 184, 191, 194, 197, 199, 201, 202, 203, 204, 207, 211, 212, 214, 215, 217, 221, 224, 225, 226, 233, 237, 243, 251, 252, 253, 259, 261, 263, 264, 265, 266, 268, 269, 270, 271, 272, 273, 274, 275, 276, 277, 278, 290, 297, and 299.

\*\*Ignition - fluorine: see References 141, 233, 274, 275, and 276.

Because of its strong oxidizing properties, fluorine reacts readily and often violently with most organic materials. Only the highly fluorinated hydrocarbons such as Teflon and Kel-F withstand continued service.

Cleanliness is essential for fluorine systems. If the metal surfaces in contact with fluorine or for that matter almost any strong oxidizer are contaminated with organic materials, such as traces of oil or grease, local hot spots may form which in turn may cause the violent burning of the encasing material.

It is recommended that equipment to handle fluorine be first thoroughly cleaned to remove all contaminants, e.g., organic matter, and then passivated with fluorine diluted with an inert gas.

#### FLOX\*

Mixtures of liquid fluorine and liquid oxygen (called FLOX) have received some consideration as oxidizers. Typical ratios are 40:60 and 20:80 fluorine:oxygen. The limited compatibility data that are available for FLOX are summarized in Table 7. In general, it appears that any material which performs well in liquid fluorine also performs well in FLOX.

#### Oxygen Difluoride (OF<sub>2</sub>)\*\*

Oxygen difluoride is a colorless gas and a brownish yellow liquid. It boils at -220 F. It is toxic and possesses about the same lethal characteristics as phosgene.

The limited data on the compatibility of materials with oxygen difluoride and oxygen difluoride-oxygen mixtures are summarized in Tables 8 and 9. In general, the materials behave as well as, or better than, in fluorine. No detonation has been observed in steel cylinders when filled with OF<sub>2</sub> liquid or gas and struck by 0.22-caliber long rifle bullets fired from 50 feet. (124)

#### Ozone Difluoride (O<sub>3</sub>F<sub>2</sub>)

Ozone difluoride is a viscous blood-red liquid at -297 F. At -250 F it decomposes into oxygen and oxygen difluoride.

There is little published information on O<sub>3</sub>F<sub>2</sub>. The corrosive effect of 0.05 percent O<sub>3</sub>F<sub>2</sub>-LOX on stainless steel is reported to be about the same order of magnitude as that of fluorine. (83) Compatibility data are summarized in Table 10.

#### Chlorine Trifluoride (CTF)(ClF<sub>3</sub>)\*\*\*

Like fluorine, chlorine trifluoride is among the most active chemicals known. It is a nearly colorless gas at atmospheric pressure and room temperature, but can be liquified by the application of slight pressures. CTF reacts violently with water and many organic compounds. It attacks the respiratory tract. The threshold-limit value of toxicity for CTF in air is 0.1 ppm.

\*FLOX: see References 224, 266, and 267.

\*\*Oxygen difluoride: see References 83, 124, 224, 238, 282, and 283.

\*\*\*Chlorine trifluoride: see References 17, 23, 40, 81, 82, 106, 110, 111, 112, 113, 114, 115, 116, 117, 118, 119, 151, 158, 200, 211, 243, 244, 269, and 286.

Chlorine trifluoride can be handled in many of the common metals shown in Table 11. As with fluorine, a coating is formed on metals, which provides protection from corrosive attack. Among the metals which are resistant to CTF are copper, brass, steel, Monel, and nickel. Of these, Monel and nickel are preferred because of their resistance to hydrogen fluoride and hydrogen chloride, which are formed by reaction of CTF with water. The organic materials that are compatible are limited to Teflon and Kel-F.

Although some surface staining has occurred in tests employing impact, shock, and perforation, there has been no ignition of low-carbon steel, stainless steel, aluminum, copper, magnesium, or titanium in liquid or gaseous CTF. (114, 115)

Cleaning and passivating treatments similar to those described for metals that are used to contain fluorine must be used for CTF systems to reduce the possibility of rapid reactions.

#### Bromine Trifluoride (BTF)(BrF<sub>3</sub>)\*

Bromine trifluoride is a colorless liquid that boils at 275 F. Its vapor pressure at 70 F is 0.15 psia. It is toxic and has a threshold limit value for toxicity in the atmosphere of 0.1 ppm.

Bromine trifluoride reacts violently with many organic compounds and vigorously with water. BTF, like CTF, appears to react with some metals to form a protective coating of the metal fluoride. As shown in Table 4, this coating permits the use of nickel up to about 1300 F, copper to 750 F, and steel to 480 F. The data are not sufficiently detailed to permit a more extensive classification than that shown in Table 12.

Metals which do not form protective coatings are vigorously attacked. Examples of this type of condition are molybdenum and tungsten, either as the pure metal or in an alloy. Titanium is also attacked by BTF. Boron, silicon, columbium, and sulfur all burn in BTF (liquid).

Materials of construction, equipment design, cleaning, passivation, and general handling practice for bromine trifluoride are the same as for chlorine trifluoride.

#### Bromine Pentafluoride (BPF)(BrF<sub>5</sub>)\*\*

Bromine pentafluoride is a colorless liquid that boils at 105 F. Its vapor pressure at 70 F is 7 psia. The toxicity threshold-limit value of BrF<sub>5</sub> in the atmosphere is 0.1 ppm.

Bromine pentafluoride reacts with most of the known elements except nitrogen, oxygen, and the rare gases. Under the proper conditions, it will react with most inorganic compounds except those containing fluorine in their highest valence state. Most organic compounds react violently with BPF at room temperature and atmospheric pressure. Detailed corrosion data for metals are not available.

Recommended materials of construction for BPF are the same as those for chlorine trifluoride. The same precautions for cleaning and passivation must be followed.

\*Bromine trifluoride: see References 82, 94, 106, 151, 211, 243, and 269.

\*\*Bromine pentafluoride: see References 243 and 269.

### Iodine Pentafluoride (IPF)( $IF_5$ )\*

Iodine pentafluoride is a colorless liquid. It boils at 207 F and has a vapor pressure of 0.4 psia at 70 F. Its toxicity threshold-limit value is 0.1 ppm.

Iodine pentafluoride is the least reactive of the halogen fluorides. Very few quantitative corrosion data for common materials of construction in IPF are available. However, it is reported that most metals are only slightly attacked by it at ordinary temperatures. The recommended materials of construction are the same as those for chlorine trifluoride.

IPF reacts violently with water. It also reacts with most organic compounds. Those rich in hydrogen will yield hydrogen fluoride and tend to ignite. Reaction with chlorine-containing compounds tends to release free iodine.

### Perchloryl Fluoride (PF)( $ClO_3F$ )\*\*

Perchloryl fluoride is a colorless gas with a sweet odor. It can be liquified at room temperature at pressures in excess of 150 psia. PF affects the respiratory tract and causes "burns" if the liquid is splashed onto the body. The toxicity threshold-limit for PF in air is 3 ppm.

Anhydrous PF is normally shipped in liquid form in steel containers. Table 13 shows other materials which resist PF quite well.

Reactions of PF with water are very slow up to temperatures of about 575 F. However, in the presence of water, PF becomes more corrosive as indicated in Table 14. Under moist conditions, Types 304, 310, and 314 stainless steels have shown relatively good resistance at room temperatures. Short-time tests also have shown the nickel alloys, Hastelloy C, titanium, and tantalum to have good resistance.

Grenade or cylinder-perforation tests resulted in detonation of titanium in liquid and gaseous perchloryl fluoride.(114,115) Titanium also ignited under impact in perchloryl fluoride but the burning was not sustained. Other metals which underwent these same tests but did not ignite were: steel, stainless steel, copper, magnesium, and aluminum.(114,115)

Teflon and Kel-F are very resistant to attack by PF. Other plastics which are suitable are unmodified phenolic resins and epoxy resins. Rubbers which are compounded with carbon black tend to be inflammable, while those with iron oxide fillers are not inflammable. The large surface areas of sponge rubbers make them inflammable. Ordinary oils, greases, and waxes should never be used with PF, however, fluorocarbon compounds are compatible. Many organic materials do not react with PF at room temperature, but if ignited will burn violently. Some inorganic materials react rapidly, e.g., mercury and "Indicating" Drierite.

### Mixtures With Perchloryl Fluoride

Compatibility data for materials in 25 percent perchloryl fluoride-75 percent chlorine trifluoride

\*Iodine pentafluoride: see References 243 and 269.

\*\*Perchloryl fluoride: see References 81, 82, 83, 110, 111, 112, 113, 114, 115, 116, 117, 118, 119, 151, 211, 219, and 224.

are summarized in Table 15. With the exception of columbium, molybdenum, and titanium, most metals perform well.

Compatibility data for materials in 50 percent perchloryl fluoride-50 percent tetrafluorohydrazine are presented in Table 16. All of the usual metals of rocket construction appear compatible in both the liquid and gas at -109 F. (Tetrafluorohydrazine boils at -100 F. It can be liquified at room temperature at pressures greater than 600 psi.)

### Halogenated Hydrocarbon Propellants\*

Halogenated hydrocarbon propellants are chiefly fluorinated and chlorinated methane and ethane. Many are well known as refrigerants under the trade names of Freon, Genetron, etc. They range from gases to liquids at room temperature with boiling points of about -200 F to about +200 F.

Compatibility data for some five propellants are summarized in Tables 17 to 21. Most of the common structural materials can be used satisfactorily with the halogenated hydrocarbons. At high temperatures, some metals promote catalytic breakdown of the propellants. The general order of this tendency is: Inconel<18-8 stainless steel<nickel<copper<340 steel<aluminum<bronze<brass<silver.(68) Aluminum alloys containing more than 2 percent magnesium and magnesium alloys are not recommended for use in halogenated hydrocarbons containing water.

Impact tests at 70 ft-lb in Propellants 113 and 114B2 have not caused ignition of aluminum, aluminum plus alumina sand, titanium, titanium plus titanium filings, or titanium plus alumina sand.

Tetrafluoroethylene and chlorotetrafluoroethylene plastics and orlon acrylic are generally suitable for use in halogenated hydrocarbon propellants. Polyvinyl alcohol resists these propellants but is sensitive to water. Phenolics, Delrin acetal resin, nylon, polyethylene and vinyls are suitable for use in many applications, but the behavior of different types may vary in different propellants, and thus should be thoroughly tested before use. Methacrylates and polystyrene are generally not suitable. No single elastomer has been found to be compatible in all halogenated hydrocarbon propellants, but a satisfactory combination can usually be found.(68)

### HYDRAZINES

#### Hydrazine ( $N_2H_4$ )\*\*

Hydrazine is a clear oily liquid with an odor similar to that of ammonia. Its vapor pressure at 80 F is 0.31 psia and it boils at 236 F. Hydrazine vapors affect the respiratory tract, nervous system, liver, and kidney, and cause "burns" when spilled on the skin. The toxicity threshold-limit value in the atmosphere is 1 ppm.

The information regarding the compatibility of various metals and nonmetals with hydrazine and

\*Halogenated hydrocarbon propellants: see References 51, 53, 68, 84, 136, 214, and 279.

\*\*Hydrazine: see References 3, 4, 7, 23, 70, 81, 82, 102, 110, 135, 145, 149, 151, 170, 171, 172, 173, 174, 183, 196, 211, and 216.

hydrazine-water mixtures is not completely consistent. These differences appear to be related to the specific application. For example, a metal may be satisfactory if air oxidation of the surface can be prevented. On the other hand, this same metal may be unacceptable for service in which prolonged exposure to air cannot be avoided.

In assessing the compatibility of a material with hydrazine, two major factors must be considered for any given exposure condition. They are:

- (1) The corrosion behavior of the material in contact with hydrazine
- (2) The effect of the material and/or corrosion products on the rate of decomposition of hydrazine.

This is particularly true for carbon and low-alloy steels, copper alloys, and molybdenum. From the corrosion standpoint, they are satisfactory. On the other hand, these metals and/or their corrosion products catalyze hydrazine decomposition at elevated temperatures. Explosions may occur. At one time, it was believed that Type 316 stainless steel (containing molybdenum) caused explosions when contacted by hydrazine at elevated temperature. However, it is now generally agreed that the hazard from misoperation in hydrazine and unsymmetrical hydrazine is no greater with 316 stainless steel than with any of the other 300 series stainless steels. (254)

Most metallic materials of construction are compatible with hydrazine. Data are summarized in Table 22.

Many plastics and rubbers are compatible with hydrazine at room temperature. Graphite and Graphitar are not suitable, since they tend to promote decomposition.

#### Monomethyl Hydrazine (CH<sub>3</sub>NNH<sub>2</sub>)\*

Monomethyl hydrazine is a clear liquid with the odor of ammonia. It has a vapor pressure of 1 psia at 80 F and boils at 189.5 F. Its toxic properties are somewhat similar to those of hydrazine. The recommended threshold limit is 0.5 ppm in the atmosphere.

The following materials can be used in the storage and handling of monomethyl hydrazine:

#### Metals

303 stainless steels  
304 stainless steels  
321 stainless steels  
347 stainless steels  
4130 steel  
Aluminum alloys to 160 F  
Aerobraze-I

#### Nonmetals

Tetrafluoroethylene resins  
High-density polyethylene  
Some silicone rubbers  
Some unplasticized trichlorofluorethylene

\*Monomethyl hydrazine: see Reference 154.

Copper, lead, zinc, and alloys containing more than 0.5 wt% molybdenum are not compatible with monomethyl hydrazine.

#### Unsymmetrical Dimethyl Hydrazine (UDMH) [(CH<sub>3</sub>)<sub>2</sub>NNH<sub>2</sub>]\*\*

UDMH is a clear liquid with the odor of ammonia. Its vapor pressure at 80 F is 8.4 psia and it boils at 146 F. Its toxicity is similar to that of hydrazine but not so severe. The toxicity threshold-limit value of UDMH is 0.5 ppm.

In general, unsymmetrical dimethyl hydrazine affects materials in much the same manner as hydrazine. Of the metals, low-alloy steels, aluminum, and stainless steels are commonly used to contain UDMH. Aluminum is attacked by UDMH if water is present with the attack being in direct proportion to the amount of water. Teflon, Kel-F (unplasticized), nylon, polyethylene, and Havgel 60 are among the plastic materials which are not attacked by UDMH.

Lubricants such as APS C-407, Parkerlube 5 PB, Molykote, and Peraline 12-4 may cause decomposition. On the other hand, litharge and glycerine paste, X-Pando, and Q-seal are compatible and can be used for thread compounds and other similar applications. Petroleum and silicone greases do not react, but are dissolved by the UDMH. Data for all materials are summarized in Table 23.

#### Hydrazine-Unsymmetrical Dimethyl Hydrazine Mixtures\*\*

Much of the information on properties and compatibility of 50:50 N<sub>2</sub>H<sub>4</sub>:UDMH (Aerozine 50) is summarized in the Titan II Storable Propellant Handbook. (167, 168, 169) Most of the common metallic materials of construction are compatible with N<sub>2</sub>H<sub>4</sub>:UDMH at room temperature, including aluminum alloys, steel, stainless steel, nickel alloys, and titanium alloys. As described under "Hydrazine", Type 316 and molybdenum-containing stainless steels are no longer believed to pose an explosion hazard with 50:50 N<sub>2</sub>H<sub>4</sub>:UDMH. (254) Of the plastic materials only some of the fluorocarbons, polyethylene, polypropylene and polyolefins are Class 1 materials. Compatibility data are summarized in Tables 24 and 25.

#### HYDROGEN\*\*\*

Liquid hydrogen boils at -423 F. Hydrogen is not toxic in the usual sense but will cause "burns" if the cold liquid contacts the skin. Hydrogen is readily ignited in air at concentrations of 4 to 74 vol%.

Liquid hydrogen and gaseous hydrogen at low temperatures are both considered to be noncorrosive. Embrittlement of metals by the low temperature of the liquid or gas is a more important factor. As can be seen in Table 26, a number of metals can be

\*Unsymmetrical dimethyl hydrazine: see References 3, 7, 23, 24, 54, 60, 61, 62, 63, 79, 81, 82, 92, 102, 110, 145, 151, 155, 156, 157, 164, 171, 172, 173, 174, 182, 196, 198, 211, 227, 228, 281, and 296.

\*\*Hydrazine-unsymmetrical dimethyl hydrazine mixtures: see References 4, 31, 47, 48, 49, 50, 165, 166, 167, 168, 169, 170, 198, 205, and 254.

\*\*\*Hydrogen: see References 34, 46, 74, 80, 81, 82, 93, 151, 181, 199, 211, and 221.

rated compatible (Class 2) with liquid hydrogen; among these are the 300 series stainless steels, Type 410 stainless steel, aluminum and most of its alloys, some nickel alloys, cobalt alloys, and molybdenum. The use of organic materials is limited because of the effect of the low temperature on their physical properties. To avoid this temperature effect, "warm joints" are used, in which the gasket material is kept at a higher temperature so that only hydrogen gas contacts a joint. Table 26 lists some of the organic materials that can be used with liquid hydrogen.

#### HYDROGEN PEROXIDE (H<sub>2</sub>O<sub>2</sub>)\*

Hydrogen peroxide is a colorless liquid that boils at 303 F. Its toxicity threshold-limit value in the atmosphere is 1 ppm.

When considering materials of construction for handling and containing concentrated hydrogen peroxide, both the effect of the H<sub>2</sub>O<sub>2</sub> on the construction material and the effect of the construction material on the H<sub>2</sub>O<sub>2</sub> must receive equal attention. If corrosion of the material takes place, usually the H<sub>2</sub>O<sub>2</sub> will also decompose, although the reverse is not true, for some materials catalytically decompose H<sub>2</sub>O<sub>2</sub> without much corrosive attack occurring.

One means of reducing the decomposition is to passivate the construction material before use.

#### Aluminum Passivation

An accepted passivation procedure for aluminum consists of several steps.

Step 1, thoroughly clean the metal. This step consists of degreasing with trichlorethylene, perchlorethylene, or a detergent wash, or both, depending upon the type of contaminating dirt, followed by thorough rinsing with clean water.

Step 2, treat with 5 percent nitric acid for 1 or 2 days. Rinse with tap water. Spots or areas which are not passivated can be readily identified. These spots will not have the uniform dull, velvety finish characteristic of passivated aluminum.

Step 3, reclean unpassivated areas and dig out areas containing iron or other inclusions in the aluminum.

Step 4, repeat the cleaning and nitric acid treatments until the aluminum is satisfactorily passivated.

Step 5, treat with stabilized 35 percent H<sub>2</sub>O<sub>2</sub> for 1 to 3 days. The passivity of the aluminum can be checked by the amount of decomposition of the 35 percent H<sub>2</sub>O<sub>2</sub>, by gas bubbles, and by the warming of the solution.

Step 6, rinse with distilled or deionized water.

Step 7, expose to 90 percent H<sub>2</sub>O<sub>2</sub>. During the first 16 to 24 hours of exposure to strong H<sub>2</sub>O<sub>2</sub>, equipment must be carefully watched to be sure that the H<sub>2</sub>O<sub>2</sub> is not decomposing.

\*Hydrogen peroxide: see References 23, 28, 35, 36, 37, 38, 41, 44, 75, 76, 77, 81, 82, 94, 101, 102, 105, 108, 110, 129, 144, 147, 148, 149, 151, 186, 187, 198, 206, 208, 210, 211, 213, 221, 229, 230, 234, 245, 246, 249, 250, 258, and 293.

Modifications of the procedure and other treatments which produce the same result may be used.

#### Stainless Steel Passivation

The passivating procedure for the 300 series stainless steels is similar to that for aluminum.

Step 1, clean with appropriate solvents and detergent solutions to remove dirt, grease, and other contamination and rinse with tap water.

Step 2, treat with 70 percent nitric acid for 4 or 5 hours and rinse with distilled water. An alternative treatment for 17-7PH steel uses a 2 percent Na<sub>2</sub>Cr<sub>2</sub>O<sub>7</sub>·2H<sub>2</sub>O-20 percent HNO<sub>3</sub> solution for 1/2 hour at 120-130 F.

Step 3, condition the stainless steel in H<sub>2</sub>O<sub>2</sub> of the concentration it will be required to handle. This operation should be observed closely to determine whether decomposition is taking place.

Step 4, if decomposition takes place, repeat the cleaning and passivating steps.

Some stainless steels may not respond to passivating treatments without prior cleaning by pickling in a 3% HF-10% HNO<sub>3</sub> solution.

#### Pretreatment of Plastics

Plastic materials must be cleaned with the appropriate solvents or detergent solutions and rinsed. Next, they should be pre-exposed for 16 to 24 hours in the H<sub>2</sub>O<sub>2</sub> solution in which they will be used.

#### Discussion

Another method which effectively reduces the decomposition is the addition of sodium stannate to H<sub>2</sub>O<sub>2</sub> in about 2.0 ppm concentration. This treatment is effective in reducing decomposition after refilling a container with fresh stannate-free H<sub>2</sub>O<sub>2</sub>.

In general, the acceptability of a material for H<sub>2</sub>O<sub>2</sub> service is based mostly upon the amount of active oxygen that is lost by decomposition rather than upon a corrosion rate of the construction material. Accordingly, the "class" ratings used for materials for H<sub>2</sub>O<sub>2</sub> have been based on decomposition limits. Table 27 describes these limits in detail for four classes of materials. The table lists the material in each class immediately under the description.

Aluminum, some of its alloys, tantalum, and zirconium are the only metals included in Class 1.

Many aluminum alloys, stainless steel alloys, silicon, and tin fall into Class 2.

Class 3 contains other aluminum alloys, a variety of stainless steel alloys, Inconel X, Alloy H-075, and Refractalloys 26 and 70.

In general, the presence of copper in an aluminum alloy greatly reduces its compatibility with H<sub>2</sub>O<sub>2</sub>. The 1060 alloy is most widely used in 90 percent H<sub>2</sub>O<sub>2</sub> service; however, several other alloys are considered to be Class 1 materials.

The attack of an aluminum alloy is usually of the smooth overall type, but pitting occurs

occasionally. Pitting is usually attributed to the presence of impurities which cause local breakdown of the  $H_2O_2$ . Chloride ions also accelerate the pitting attack. The addition of small amounts of nitrate ion, such as sodium nitrate, tends to reduce the action of the chloride ion. However, the presence of 10 mg  $Cl^-$  per liter causes accelerated attack even with nitrate present. Anodizing of the metal also reduces the attack by  $H_2O_2$  with chloride, but damage of the anodic coating localizes the pitting in the damaged area. The addition of compounds to  $H_2O_2$  which change the pH in either direction from the neutral point may accelerate attack.

Galvanic coupling of aluminum to stainless steel results in increased attack on the aluminum. Chloride ions in turn increase the galvanic effect.

Many of the higher strength aluminum alloys are not compatible with  $H_2O_2$ ; therefore, one alternative is to use a strong alloy clad with a compatible grade. Of course, special attention must be given to welds, to insure complete covering of the base alloy.

Many cases of decomposition of  $H_2O_2$  in aluminum or other compatible metals is traced to soluble or suspended contamination of the  $H_2O_2$  and not to an effect of the container material.

The table shows that the 300 series stainless steels cannot be rated as Class 1, but give very good Class 2 service. The 300 series steels are used for high-pressure flowing systems and may be welded. Chloride contamination, at the 10-mg-per-liter level, does not appear to cause pitting in stainless steels. The galvanic effect in aluminum-stainless steel couples tends to protect the steel. The 17-7PH grades of stainless steel are satisfactory with  $H_2O_2$ , but the 400 series is not. A 120-grit finish on the 17-7PH steels improved their service.

Lower concentrations of  $H_2O_2$  (52 to 90 percent) require the same materials of construction as 90 percent  $H_2O_2$ . Higher strength peroxide (98 percent) is, in general, more stable than 90 percent  $H_2O_2$  in contact with metals. Aluminum alloys 1060, 5052, and 7072 are rated Class 1. Stainless steels Types 304, 316, and 347, and aluminum alloys 6061 and 356 are rated Class 2.

In the transporting of high-strength  $H_2O_2$ , 1060 aluminum has been used in tank cars and 43 aluminum or 300 series stainless steels in self-priming centrifugal pumps, while valves, fittings, and instruments are usually made of 300 series stainless steel. (208)

Of the plastics and rubbers, molded Teflon, Kel-F, and Mylar B are rated Class 1 (unrestricted use). Koroseal 700 has been used extensively as gasketing material. Many lubricants exhibit impact sensitivity in  $H_2O_2$ . The fluorinated and chlorinated lubricants appear most promising with a Class 2 rating and no impact sensitivity. Compatibility data for nonmetals in  $H_2O_2$  are presented in Table 28.

#### METHYLENE CHLORIDE ( $CH_2Cl_2$ )\*

Methylene chloride is a colorless liquid. It boils at 104 F and exerts a vapor pressure of 380 mm at 72 F. The toxicity threshold limit for  $CH_2Cl_2$  in the atmosphere is 50C.

\*Methylene chloride: see References 127, 129, and 211.

Liquid methylene chloride is compatible with copper, steel, austenitic stainless steels, Hastelloy B, Hastelloy C, asbestos, and graphite. Gaseous methylene chloride is more corrosive, being compatible with Northite and Durimet 20. Compatibility data are summarized in Table 29.

#### NITRIC ACID ( $HNO_3$ )\*

The nitric acid used for propellants is usually in the concentrated form referred to as "fuming nitric acid". In general, the fuming acids contain less than 5 percent water. If the acid contains dissolved oxides of nitrogen, it is known as "red fuming nitric acid" or "RFNA". The  $NO_2$  content normally varies from 7 to 30 percent.

Nitrogen dioxide is not present in "white fuming nitric acid" or "WFNA", which contains a minimum of 97 percent  $HNO_3$ .

Hydrofluoric acid may be added to either RFNA or WFNA as a corrosion inhibitor. Listed below are the Military Specification compositions of inhibited and noninhibited acid (MIL-N-7254 C, July 19, 1956):

	<u>White Fuming Nitric Acid</u>	
	<u>Type I</u>	<u>Type I A (WFNA)</u>
Nitric acid ( $HNO_3$ )	97.5% min.	96.8% min.
Nitrogen dioxide ( $NO_2$ )	0.0 + 0.5%	0.0 + 0.5%
Water ( $H_2O$ )	2.0% max.	2.0% max.
Hydrofluoric acid (HF)	0.0	0.6 ± 0.1%

  

	<u>Red Fuming Nitric Acid</u>	
	<u>Type III</u>	<u>Type III A (RFNA)</u>
Nitric acid ( $HNO_3$ )	82.0 - 85%	81.3-84.5%
Nitrogen dioxide ( $NO_2$ )	14.0 ± 1.0%	14.0 ± 1.0%
Water ( $H_2O$ )	2.5 ± 0.5%	2.5 ± 0.5%
Hydrofluoric acid (HF)	0.0	0.6 ± 0.1%

#### Red Fuming Nitric Acid

Red fuming nitric acid is a highly corrosive material; therefore, the choice of construction materials is based upon the corrosion resistance of the material rather than on the catalytic decomposition of the acid. Aluminum and stainless steel alloys are usually used to handle RFNA. The compatibility data in Table 30 indicate that at room temperature, the corrosion rate for aluminum alloys is slightly higher than for the 300 series stainless steels. It should be noted that at 160 F, aluminum alloys are more resistant than stainless steels. Stainless steels and aluminum alloys are usually attacked in a uniform manner. However, selective attack in the heat-affected zone near welds is sometimes produced. Knife-line attack at welds may occur in aluminum alloys above 120 F. The 1060 alloy appears to be free from this attack to higher temperatures. The low-carbon grades of the 300 series stainless steels and those containing columbium or titanium are less susceptible to attack at welds than are the regular grades. Aluminum, when coupled to stainless steel, acts as a sacrificial anode to protect the steel.

\*Nitric acid: see References 18, 19, 29, 30, 32, 39, 64, 69, 81, 82, 87, 89, 93, 95, 96, 97, 102, 104, 110, 127, 128, 130, 133, 135, 150, 151, 179, 181, 185, 188, 189, 190, 192, 193, 198, 204, 211, 214, 221, 222, 223, 239, 240, 241, 242, 248, 260, 287, 298, and 300.

As little as 0.1 percent HF added to RFNA greatly reduces the corrosion rate of both aluminum and stainless steel. If this change from RFNA to IRFNA is made in stainless steel equipment and selective attack has already started, the inhibitors are not effective. The HF inhibitor reduces the selective attack at welds, permitting use at higher temperatures.

Titanium and tantalum are both resistant to RFNA; however, caution must be used with the titanium alloys. A pyrophoric reaction may occur with titanium alloys in RFNA which contains less than 1.5 to 2.0 percent H<sub>2</sub>O. Both titanium and tantalum are attacked much more rapidly by IRFNA than by the acid without the HF addition.

A number of other alloys are compatible with red fuming nitric acid. These include cobalt alloys, Types 430 and 446 stainless steels, chromium, and for some applications, nickel and some nickel alloys. Platinum, gold, tin, and zirconium may be used. Low-alloy steels, lead, copper, and magnesium are rapidly attacked by either RFNA or IRFNA.

#### White Fuming Nitric Acid

White fuming nitric acid is similar to red fuming nitric acid with respect to compatibility with construction materials. It is not so stable as RFNA, but compatibility is largely dependent upon corrosion properties rather than on decomposition.

The corrosion behavior of metals in white fuming nitric acid is much the same as that in RFNA. The same materials are resistant and the same materials are severely attacked. However, Table 31 shows that the temperature limits are somewhat lower in the white fuming nitric acid.

#### Concentrated Nitric Acid

Table 32 has been included to show the compatibility of materials in somewhat less concentrated acids than the fuming grades. It can be seen that acids from 80 percent up to the fuming range are much more corrosive than the more concentrated ones. Stainless steels are the best materials of construction for these acids.

#### NITROGEN TETROXIDE (N<sub>2</sub>O<sub>4</sub>)\*

Nitrogen tetroxide is an equilibrium mixture of dinitrogen tetroxide and nitrogen dioxide (N<sub>2</sub>O<sub>4</sub> ↔ 2NO<sub>2</sub>). It is a heavy brown liquid that boils at 70.1 F. The liquid causes severe burns on body tissue. The toxicity threshold-limit value in the atmosphere is 5 ppm as NO<sub>2</sub> or 2.5 ppm as N<sub>2</sub>O<sub>4</sub>.

Much of the information on properties and compatibility of N<sub>2</sub>O<sub>4</sub> is summarized in the Titan II Storable Propellant Handbook, (167,168,169)

Dry (less than 0.2 percent H<sub>2</sub>O) nitrogen tetroxide can readily be contained by several metals and their alloys. It is normally handled in aluminum, mild steel, cast iron, or stainless steel, although there have been reported instances of intergranular attack in welded 2014-T6 aluminum. Compatibility data are summarized in Table 33.

\*Nitrogen tetroxides: see References 3, 9, 10, 11, 12, 13, 23, 31, 47, 48, 49, 50, 59, 72, 81, 82, 88, 102, 110, 126, 131, 137, 145, 151, 165, 166, 167, 168, 169, 170, 171, 172, 173, 174, 181, 196, 198, 205, 211, 220, 221, 245, 256, 257, 287, 288, and 292.

Titanium is resistant to N<sub>2</sub>O<sub>4</sub> except under impact. It has been found that titanium impacts sporadically under reasonably well controlled test conditions; the ignition frequency is increased markedly by titanium filings or glass particles on the impact surface; increasing the impact-energy level increases the ignition frequency; and increasing the water content of the N<sub>2</sub>O<sub>4</sub> to 2 to 5 percent lowers the ignition frequency. (256,257,292) The ignitions do not spread beyond the impact area.

Moist N<sub>2</sub>O<sub>4</sub> is, in general, more corrosive because of the nitric acid formed. As shown in Table 34, at levels of up to 1 percent moisture in N<sub>2</sub>O<sub>4</sub>, most common metals are still Class 1 at room temperature. With 3.2 percent moisture in N<sub>2</sub>O<sub>4</sub> the corrosion resistance of steel and aluminum alloys drops off markedly even at slight increases in temperature. Data for the latter are summarized in Table 35.

Limited data on the compatibility of materials with flowing N<sub>2</sub>O<sub>4</sub> are presented in Table 36.

The most resistant plastic materials are polymerized fluorinated hydrocarbons such as Teflon and unplasticized Kel-F. Other plastics such as Koroseal, Saran, polyethylene, and Tygon are suitable for short-time exposures. Vinyl plastics, in general, do not hold up well in N<sub>2</sub>O<sub>4</sub>.

Asbestos and graphite are satisfactory for packing materials and graphite-waterglass for thread compound.

Impact tests at 60 ft-lb in liquid N<sub>2</sub>O<sub>4</sub> have resulted in detonations of polydimethylsiloxane. Similar tests at 70 ft-lb did not cause detonation in polychloroprene, branched or linear polyethylene, polypropylene, polyvinylidene fluoride, polyvinylidene fluoride-hexafluoropropylene copolymer. (126)

#### OXYGEN\*

Oxygen is a light blue transparent liquid at -297.4 F. In the liquid state it can cause "burns" if spilled on the skin. Oxygen supports combustion and accidental contact should be avoided with oxidizable materials.

Of the elemental materials, oxygen is next to fluorine in reactivity. It will form compounds with all elements except the rare gases. However, the reactivity of liquid oxygen is very low compared with that of gaseous oxygen.

Liquid oxygen is considered to be noncorrosive to most metals. In particular, nickel, Monel, Inconel, copper, aluminum, the 300 series of stainless steels, brass, and silver solder are used in liquid-oxygen-handling equipment.

Several instances have been reported of violent reactions of titanium and liquid oxygen which appeared to be related to impact. The impact sensitivity of titanium in LOX has been investigated rather extensively.\*\* It appears that the ignition of titanium under impact occurs in the following sequence: (138,140)

\* Oxygen: see References 1, 5, 6, 7, 8, 21, 22, 25, 26, 27, 33, 42, 43, 45, 65, 66, 67, 71, 78, 80, 81, 82, 98, 102, 107, 125, 126, 138, 139, 140, 141, 142, 151, 176, 177, 180, 181, 195, 198, 209, 211, 218, 231, 232, 235, 236, 255, 259, 262, 284, and 285.

\*\* Ignition in LOX: see References 7, 22, 27, 42, 43, 65, 66, 71, 78, 126, 138, 139, 140, 141, 142, 176, 177, 180, 198, 218, 235, and 236.

- (1) The impact exposes fresh metal and results in some gaseous oxygen being formed at the point of impact.
- (2) The gaseous oxygen reacts with the fresh metal in an exothermic reaction.
- (3) The heat generated raises the metal temperature sufficiently to result in localized dissolution of any  $TiO_2$  film that might form.
- (4) Thus a protective oxide film does not build up and the reaction proceeds rapidly between the base metal and oxygen.

Ignition of massive titanium is observed in gaseous oxygen at liquid-oxygen temperatures at pressures of about 100 psi and above. This critical pressure limit is lowered only slightly as the temperature of the oxygen is raised to ambient temperature.

Massive aluminum exhibits ignition under severe detonation in LOX. The frequency is not so great nor is the propagation so severe as it is with titanium under the same conditions. Magnesium also ignites under detonation at shock levels higher than those for titanium but lower than those for aluminum.

Organic materials should be avoided with both liquid and gaseous oxygen because of possibilities of explosions. Currently, there is no single test or group of tests which gives a reliable compatibility rating for organic materials in liquid oxygen service. It is recommended that organic materials be avoided wherever possible and used only with caution. No completely compatible lubricants have been found. Thread antiseize sealants of graphite in chlorinated organic carrier and halogen paraffin oils with pour points as low as -100 F have been used in LOX systems. Teflon, Mylar, and certain chloroprene and Buna-N compounds have been used as static seals while Kel-F-300, Kel-F-500, Kel-F-240, Fluorothene FYTD, Fluorothene FYTS, and certain chloroprene and Buna-N compounds have been used in dynamic seals.<sup>(284,285)</sup>

Many organic and plastic materials exhibit impact sensitivity in LOX including:<sup>(71,126)</sup>

- (1) Synthetic elastomers and Thiokols
- (2) Cellulose-based papers
- (3) Silicone- and silicate-based oils and greases
- (4) Thermoplastics such as nylon and phenolics
- (5) Thermo-setting resins (phenolics, silicones, epoxies, etc.)
- (6) Petroleum-based oils and greases.

The fluorocarbon plastics are probably the best choice with respect to impact sensitivity. These, however, should not be used with aluminum. A number of other organic materials might be used, but specific conditions should be carefully studied. The list of references on ignition in LOX contains the results of many impact tests on organic materials, which can be used as a guide to selection.

## OZONE

Ozone is colorless in gaseous and liquid state. It boils at -168 F. The toxicity threshold-limit value for ozone in the atmosphere is 0.1 ppm.

There are few data on the compatibility of materials with ozone. It has been shown that 100 percent gaseous ozone can be stored up to 50 days at 5 atm pressure and dry ice temperature (-109 F) in stainless steel, aluminum, and glass with no decomposition of the ozone.<sup>(280)</sup>

## SOLID PROPELLANTS

### ANP-2639AF

There are few or no data published in the open literature on the compatibility of materials with solid propellants. Bent beam specimens of the following materials bonded to propellant ANP-2639AF and stressed to 75 percent of the 0.2 percent off-set yield strength have survived over 100 days' exposure at room temperature and 180 F.<sup>(289)</sup>

Ladish D6AC  
300 M  
Vascojet 1000  
AM-355 (longitudinal)  
PH 15-7Mo  
BI20VCA titanium  
(longitudinal and transverse)

### Nitronium Perchlorate ( $NO_2ClO_4$ )

Nitronium perchlorate is a white crystalline powder at room temperature which has a faint odor of chlorine and nitrogen oxide. Its vapor pressure is less than 0.05 mm at 68 F. It melts and decomposes at 250 to 285 F.

The following materials are reported to be compatible with nitronium perchlorate:<sup>(55)</sup>

Metals:	stainless steel (mild steel if system is dry)
Nonmetals:	glass, Teflon, unplasticized polyvinyl chloride, polyethylene
Lubricants:	Hooker Fluorolube Grease GR-54 Hooker Fluorolube Oil X-30 3M Kel-F Polymer Oil, Grade No. 1 Halocarbon Oil Series B-21.

TABLE 3. COMPATIBILITY OF MATERIALS WITH AMMONIA

Material	Temperature, F								References	
	Gas				Liquid					
	Class 1	Class 2	Class 3	Class 4	Class 1	Class 2	Class 3	Class 4		
<b>Metals</b>										
Aluminum	212	100	175	>175				175	175	211,224,287
302 Stainless Steel	75			<900						82
304 Stainless Steel	600			<900						58,73,109, 217,221, 58,73,217, 221
316 Stainless Steel	600			<900						221
347 Stainless Steel			75							109
410 Stainless Steel	600			800						58,73,221
430 Stainless Steel	600			<900						58,73,217, 221
Marbrite	75			<900						82
Duralmet 50	75			<900						221
Carpenter 20	600			<900						58
Mild Steel	600			<900		75				211,221,287
Cast Iron	75			<900		75				211,221,287
Si-Iron	212	75						75		211,287
Mn-Cast Iron, Low Cu	75							160		211,287
Mn-Cast Iron, High Cu	75									All: 211,287
Nickel	500		500	<1100						211,287
Inconel	700		1100	>1100		75				211,221,287
Monel	500		500	<1100					75	211,287
Hastelloy A	600	600		>1000						58,127,128, 211
Hastelloy C	600	600		>1000						58,127,128, 211
Hastelloy D	500	600		>1000						58,127,128, 211
Hastelloy F		600		>1000						127,129
Chlorimet 20	75									221
Nickel-Copper	75									82
Copper		75		High					Low	127,217,221, 287
Yellow Brass		75		High					Low	127,221,287
Red Brass		75		High					Low	127,221,287
Tin Bronze		75		High					Low	127,221,287
Al Bronze		75		High					Low	127,221,287
Si Bronze		75		High					Low	127,221,287
Cu-Nickel		75		High					Low	82,127,221, 287
Gold	212					High				211,221
Lead		75		>160			75			211
Dow Metal C		Low								82
Dow Metal F-1		Low								82
Dow Metal H		Low								82
Dow Metal J-1		Low								82
Dow Metal M		Low								82
Platinum	212				High					211,221
Ir-Platinum	High				High					221
Rh-Platinum	High				High					221
Silver	75								Hot	211
Ag-Cu					All				All	82
Titanium	175									211
Tantalum	212				High	212			High	221
Zinc	75									211
Zirconium	175									211
<b>Organic Materials</b>										
Rubber, Hard Linings				75					75	82
Rubber, Soft Linings				75					75	211
Rubber, Natural		75		Hot						211
GRS		75		Hot						82
Neoprene		75		Hot						82
Butyl Rubber		75		Hot						82
Thiokol			Cold							82
Glass Fabric and Silicone Elastomers*	Hot									82
Silicone Greases	Hot									82
Neveg 41 Epon	212									211
Silicone Elastomer		75								82
Silicone Resins										82
Teflon	Hot									82
Cork										82
Vinyl Copolymers		Hot				Hot				82
Phenolics		Hot				Hot				82
Furans		Hot				Hot				82
Polyethylene		Hot				Hot				82
Kel-F		Hot				Hot				82
Vinylidene Chloride			Cold					Cold		82
Sulfur Cement			Cold					Cold		82
Bituminous Composition			Cold					Cold		82
Polystyrene				75						219
Polyesters				75						299
Phenol Formaldehyde				75						259
<b>Nonmetals</b>										
Glass	212									211
Stoneware	212									211
Kaolinite	>2000									82
Carbon	>2000									82
Graphite	>2000									211

\* Glass Fabric and Silicone Rubber

TABLE 4. COMPATIBILITY OF MATERIALS WITH HI-Cal-3 (82)

Material	Remarks
<b>Class 1</b>	
Mild steel	Below 120 F, probably higher
40/50 carbon steel	Below 120 F, probably higher
Hot-rolled primer steel	Below 120 F, probably higher
Type 304 stainless steel	Below 120 F, probably higher
Beryllium copper	Below 120 F, probably higher
Cupro-nickel	Below 120 F, probably higher
Naval brass	Below 120 F, probably higher
Phosphor-bronze	Below 120 F, probably higher
Phosphorized copper	Below 120 F, probably higher
Nickel	Below 120 F, probably higher
Inconel	Below 120 F, probably higher
Monel	Below 120 F, probably higher
Niobel	Below 120 F, probably higher
Incoloy	Below 120 F, probably higher
Aluminum	Below 120 F, probably higher
Lead	Class 2 at 120 F
Titanium	Below 120 F, probably higher
Tantalum	Below 120 F, probably higher
<b>Class 2</b>	
<b>Organics</b>	
Bakelite	No change at 120 F
Easton PVC plastic pipe	No change at 120 F
Epoxy resin	No change at 120 F
Graphite bearing	No change at 120 F
Kel-F 300	No change at 120 F
Kel-F 500	No change at 120 F
Teflon	No change at 120 F
Hycar 1001-520-39-5-4	No change at 120 F
Viton 4411A-58	No change at 120 F
Hycar 1000x88-520-39-20-3	No change at 120 F
Hycar 1001-520-37-83-1	No change at 120 F
IF4 Fluororubber	No change at 120 F
LS-53 Fluorosilicone rubber	Slightly less resilient at 120 F
Hycar 1001-520-37-83-5	No change at 120 F
X-Pando pipe dope	No change at 120 F
Nylon Zytel 101-KC-10	No change at 120 F
<b>Class 3</b>	
<b>Organics</b>	
Hycar 1001-520-39-5-2	120 F, Slightly stiffened
Polyethylene tubing	120 F, Turns yellow
Garlock 8748 (Buna-W binder)	120 F, Stiffened
Garlock 7021 (GRS - high sulfur binder)	120 F, Stiffened and roughened
Garlock 900 (GRS binder)	120 F, Stiffened
African Blue Asbestos packing	120 F, Darkened
Teflon asbestos packing	120 F, Weight gain
<b>Class 4</b>	
<b>Organics</b>	
Hycar 1001-520-39-5-5	120 F, Blistered
Compressed asbestos gasket	120 F, Fibers loosened
Garlock 7228 (neoprene binder)	120 F, Blistered
Hycar 1042-520-24-144-1	120 F, Brittle, crazed
Tygon tubing	120 F, Hardened
Garlock 7705 (GRS - blue asbestos)	120 F, Stiffened
Plexiglas	120 F, Became soft and sticky
Fairprene 5051 (neoprene on duck)	77 F, Stiffened
Fairprene 5039 (neoprene on nylon)	77 F, Became brittle
Natural rubber	120 F, Softened, easily torn
National 846, O-ring	120 F, Stiffened
Neoprene	120 F, Softened
Hycar 1001-520-39-5-1	120 F, Stiffened
Hycar 1001-520-39-5-3	120 F, Stiffened
Silicone rubber	120 F, Deteriorated to a powder

TABLE 5. COMPATIBILITY OF MATERIALS WITH PENTABORANE (204)

Material	Temperature, F								References	
	Gas				Liquid					
	Class 1	Class 2	Class 3	Class 4	Class 1	Class 2	Class 3	Class 4		
<b>Metals</b>										
Aluminum, 2024 T-3					75					
Aluminum, 3003 H-14					75					
Aluminum, 3052-H					75					
Aluminum, 6061 T6					75					
Aluminum, 7075 T-6					75					
Aluminum, 286 T-6					75					
Aluminum, Chromated					75					
Cadmium-Plated Steel					75					
Cadmium-Coated Aluminum					75					
Copper					75					
Brass					75					
Iron					75					
Steel					75					
302 Stainless Steel					75					
304 Stainless Steel					75					
321 Stainless Steel					75					
347 Stainless Steel					75					
Magnesium Alloy, Fed QQ					75					
Nickel-201					75					
Nickel-201					75					
Titanium, C-110H					75					
Titanium, C-130H					75					
<b>Nonmetals</b>										
Asbestos, Graphite-impregnated					75					
Asbestos, Hercules No. 571 K					75					
Pure Carbons					75					
Graphite No. 39					75					
Nylon					75					
Viton					75					
Viton-A					75					
Kel-F No. 500C					75					
Spinel and Glass Cloth					75					
Microplex 1					75					
Fluorocarbon Rubber					75					
Foamless									75	
Dow Corning A-7602 Foam									75	
Dow Corning A-7603 Foam									75	
Mopco F 10 Foam									75	
Mopco B 40									75	
Natural Rubber									75	
Nitrile Rubber or Nylon									75	
Dow Corning 9483 Rubber									75	
Low-Density Plastic 15G 150-101									75	
Baron									75	
Nylon									75	
Nylar									75	
Dyne									75	
Kubetek G-2077K									75	
Kubetek A-103J									75	
Fiberglass, No. 35F									75	
Fiberglass, No. 51F									75	
Dow Corning Silastic No. 80-24-48C									75	
Geacica Silastic W-20									75	
<b>Composites</b>										
Rockwell Nordstrom Lube No. 921					75					
Molybdenum Disulfide					75					
Rockwell Nordstrom Lube No. 833									75	
Rockwell Nordstrom Lube No. 8-21									75	
Rockwell Nordstrom Lube No. 850									75	
Rockwell Nordstrom Lube No. 356									75	
Rockwell Nordstrom Lube No. 852mS									75	
Rockwell Nordstrom Lube No. 8-20									75	
Rockwell Nordstrom Lube No. 942-B									75	
Water-based Lubricants									75	

TABLE 6. COMPATIBILITY OF MATERIALS WITH FLUORINE

Material	Temperature, F								References	
	Gas				Liquid					
	Class 1	Class 2	Class 3	Class 4	Class 1	Class 2	Class 3	Class 4		
<b>Metals</b>										
Aluminum 1100	500	600	800	<673	>320				-310	7,73,132, 143,153,159, 207,211,274
Aluminum 2017										
Aluminum 2024	700	>1000	1000	>1000			-320		-310	143
Aluminum 3003	700	700	1000	>1000			-320		-310	120,143
Aluminum 5052										
Aluminum 5154	700	>1000	1000	>1000			-320		-310	120,274
Aluminum 6061										
Aluminum 7075							-320		-310	143,274
Beryllium				75						27
304 Stainless Steel	400	400	500	>900	-320	-320	-310			58,211,274
304 L Stainless Steel	270			>400						143
309 Stainless Steel	570			750						7
309 Gb Stainless Steel	500			>900						207
310 Stainless Steel	570	660	750							7,207
316 Stainless Steel	400				-320					58,211,274
321 Stainless Steel										120
347 Stainless Steel	390	570	750		-320			-310		120,217,274
410 Stainless Steel					-320					58,153,159
430 Stainless Steel					-320					274
430 Stainless Steel	400	400	500	600						7,207,211
Carpenter 20					-320					38
PH15-7Mo					-320					153,274
AM-350-C					-320					274
AM-350-CK					-320					274
AM-350-D					-320					274
AM-350-DE					-320					274
Cast Iron				75						211
Armco Iron	750		500	>500						7,159,160,207
Stainless Iron				75						211
Iron (S.704S1)		200	400	>400						159,160
Iron (S.7451)		100	300	>300						159,160
Steel: Steel	750	840		930						7
SAE 1010	100	200	400	>400				-310		207
SAE 1015	300	400	570	>500						7,159,160, 207,211
SAE 1020	390	840		930						7,207
SAE 1130	750		840	930						7,207
Carbon Steel	700			930						82
Music Wire		570								7
Nickel	1200	900	1300	>1300	-320			-310		7,73,120,143 153,159,160, 211,274
D-Nickel		1200	1300	>1300						82
L-Nickel		800	1000	>1300				-310		82
Nickel (Low Carbon)	1100	1200	1300	>1300						82
Nickel (Electrolytic)	1150	1300	1300	>1300						82
Duronicel	1100	1200	1300	>1300						82
Steel	1300	1300	1300	>1300	-320	-320	-310			7,73,143, 153,159,160, 207,211,274
Cast Iron	400	800	1000	>1000				-310		82
Cast Iron	600	800	1200	>1200						82
Inconel	930	1170	500	>500				-310		7,207,211
Hastelloy B		75	212							58,127,128, 211
Hastelloy C	95		212							58,127,128, 211
Hastelloy D			212							58,211
Copper	200	400	800	>800	-320			-310		7,153,159, 207,211,274
Deoxidized Copper		130	1290							7
Copper ETp	200	400	800	>800						82
Brass (70-30)	200	400								159,160
Brass (80)	200	400	600	>600						73,211
Brass (243)	200	400	500	>500				-310		82
Brass (Low-leaded)										120
Brass (Yellow)					-320					274
Brass (Cartridge)					-320					274
Brass (Castings)										274
Brass	200	400	700					-320		73,159,160, 211
Copper-Nickel	200	400		>800						155
Copper-12% Nickel					-320					274
Copper-30% Nickel					-320					274
Everdur 1010					-320					274
Chromium Plate	400									159,160
Lead			100	>100						211
Magnesium		500	700							84
Magnesium M1A	653	<1000								143
Magnesium AZ1A-T6	691	<1000								143
Magnesium ZB1G-T6		400						-310		82
Magnesium AZ1G-T6	<1000									143
Magnesium HE-31			1100					-310		274
Magnesium HE-31A-M24										82
Magnesium HE-31 (Coated with Dow-17)								-320		274
Magnesium HE-31										153
Magnesium AZ-31								-320		153,274
Magnesium AZ-31 (Coated with Dow-17)										274
Magnesium MA (1.205n)		200								159,160
Magnesium PS-1A		200								159,160
Magnesium J-1H	140									159,160
Magnesium H-1A		600	1000							82
Magnesium Dow Metal G	570	600	600							





TABLE 12. COMPATIBILITY OF MATERIALS WITH PERCHLORYL FLUORIDE (ClO<sub>2</sub>F)

Material	Temperature, F								References	
	Gas				Liquid					
	Class 1	Class 2	Class 3	Class 4	Class 1	Class 2	Class 3	Class 4		
<b>METALS</b>										
Aluminum, 1060					85				111,115	
Aluminum, 1100	160				85				111,115,219	
Aluminum, 1100 (Welded)	85				85				117	
Aluminum, 2014 (Welded)	85				85				111,115,219	
Aluminum, 2024	160				85				111,115	
Aluminum, 3003	160				85				111,115,219	
Aluminum, 5052	160				85				219	
Aluminum, 6061	160				85				117	
Aluminum, 6061 (Welded)	85				85				111,115	
Aluminum, 7075					85				111,115	
Copper, ETP	85				85				111,115	
Copper, DP					85				111,115	
Beryllium Copper, BK					85				111,115	
3% Copper Bronze, BK	85				85				111,115	
Aluminum Bronze, BK	85				85				111,115	
Yellow Brass	85				85				111,115	
Red Brass	85				85				111,115	
101C Steel	85				85				111,115	
101C Steel (coated with Passbond 40)					85				111,115	
101C Steel (coated with Passbond 27)					85				219	
Carbon Steel	160				85				111,115	
403 Stainless Steel	85				85				219	
43C Stainless Steel	160				85				219	
531 Stainless Steel	160				85				219	
2C Stainless Steel	160				85				111,115,219	
2C2 Stainless Steel	160				85				219	
302 Stainless Steel	160				85				219	
304 Stainless Steel	160				85				219	
310 Stainless Steel	160				85				111,115,219	
314 Stainless Steel	160				85				219	
316 Stainless Steel	160				85				111,115,219	
321 Stainless Steel	75	160			85				219	
329 Stainless Steel	160				85				111,115	
347 Stainless Steel	160				85				117	
347 Stainless Steel (Welded)	85				85				117	
41C Stainless Steel (Welded)	85				85				117	
41C2C (Welded)	85				85				111,115	
PH15-70 (RH99.9)	85				85				111,115	
PH15-70 (RH100)	85				85				111,115	
PH15-70 (Welded)	85				85				111,115	
Carpenter 20-Cb	85				85				111,115	
Lead	75	80							219	
Magnesium AZ31B		85			85				111,115	
Magnesium HK21A		85			85				111,115	
Magnesium HK31A		85			85				111,115	
<b>Alloys</b>										
Inconel	160	85			85				111,115,219	
Incoloy	85	85			85				111,115	
Incoloy B	160				85				219	
Incoloy C	85				85				219	
Nickel 2C (Welded)	85				85				117	
Monel 40C (Welded)	85				85				111,115	
Cupro-Nickel, 30K	85				85				111,115	
Silver Sinter	75	80							219	
Nickel Silver, 18K	85				85				111,115	
Tin	75	80							219	
Titanium 100A	95	80			85(1)				111,115,219	
Titanium G-120AV	95				85(1)				111,115,219	
Zinc	75								111,115,219	
<b>Organic Materials</b>										
Teflon	75	300							82,111,115	
Kel-F	75	300							82,111,115	
Phenolic Resins		300							82	
Epoxy Resins		300							82	
Silicone	75	80							82	
Fluorosilicone Rubber (Iron Oxide Filler)		300							82,219	
GRS Rubber	300								82	
Kel-F Elastomer	300								82	
Silicone Rubber (On Glass Cloth)		300							82	
Polychlorotrifluoroethylene	300								82	
Oil or Grease		80							82	
Cellulose		80							82	
Cellulose Acetate		80							82	
Ethyl Cellulose		80							82	
Casein		80							82	
Hexamine Formaldehyde		80							82	
Methyl Styrene		80							82	
Nylon		80							82	
Polystyrene		80							82	
Polyethylene Chloride		80							82	
Polystyrene		80							82	
Polyethyl Methacrylate		80							82	
Polycrylonitrile		80							82	
Polyvinyl Pyrrolidone		80							82	
Polyisobutylene		80							82	
Polyethylene	75								82,219	
Nylon		80							82	
Xylylene Glycol Polyester		80							82	
Cellulose		82,219							82	
Nylon	75								82,219	
Plexiglas	75								82,219	
Dacron		300							82	
Oxlon		300							82	
Neel		300							82	
Silk		300							82	
Alkyl Resins		300							82	
Alkyl Enamel		300							82	
Modified Phenolic Resin		300							82	
Silicone Glass Cloth		300							82	
Waxed Glass Cloth		300							82	
Varnished Paper		300							82	

TABLE 13. (Continued)

Material	Temperature, F								References	
	Gas				Liquid					
	Class 1	Class 2	Class 3	Class 4	Class 1	Class 2	Class 3	Class 4		
Butyl Rubber with Carbon									300	82
Natural Rubber									300	82
Form Rubber									300	82
Nylon Carbon Filled									300	82
Neprene Carbon Filled									300	82
Oronite Tape									300	82
Perfluorobutyl Acrylate (Carbon Filled)									300	82
Natural Rubber									300	82
(Carbon Filled)									300	82
Silicone Rubber									300	82
Boresite									300	82
Carbylon Wax									300	82
Lubri-L Grease									300	82
Silicone Stopcock Grease									300	82
Transformer Oil	75								300	82,219
Chlorinated Naphthalene Fluorolube	75								300	82
<b>Other Materials</b>										
Kaoline Clay									300	82
Mercury									300	82
Drierite (Indicating)									300	82
Mica Board									300	82
Kraft Board									300	82
Linoleum	75								300	82,219
White Lead Pipe Dope									300	82
Permatex Pipe Dope									300	82
Chemical Pipe Dope									300	82
Loak Back Pipe Dope	75								300	82,219
Paper									300	82
Plastic									300	82
Alumina		300							300	82
Activated Silica		300							300	82
Sodium Metasilicate		300							300	82,219
Graphite	75								300	82
Glass									300	82
Asbestos Paper									300	82
Transite Board									300	82
Cork	75								300	82,219

(a) Ignites under impact, but burning is not sustained.

TABLE 14. COMPATIBILITY OF MATERIALS WITH PERCHLORYL FLUORIDE CONTAINED IN 1% MOISTURE

Material	Temperature, F								References	
	Gas				Liquid					
	Class 1	Class 2	Class 3	Class 4	Class 1	Class 2	Class 3	Class 4		
<b>METALS</b>										
Aluminum, 1100		85								115,219
Aluminum, 2024										219
Aluminum, 3003										219
Aluminum, 5052										219
Aluminum, 6061										219
Copper		75						85		115,219
Brass		75						85		219
Aluminum Bronze, BK		85						85		115
Yellow Brass		85						85		115
101C Steel					85(1)					115
Carbon Steel					75					219
403										







TABLE 24. COMPATIBILITY OF MATERIALS WITH 50:50 HYDRAZINE-  
UNSATURATED DICHLOROPOLYMER (50:50)

Material	Temperature, F								References
	Gas				Liquid				
	Class 1	Class 2	Class 3	Class 4	Class 1	Class 2	Class 3	Class 4	
<b>Metals</b>									
Aluminum, 1100					60				21,165,169
Aluminum, 1100-O					60				60
Aluminum, 2014-T4					60				20,169
Aluminum, 2014-T6	160				160				20,165
Aluminum, 2014-T6 (Welded)	160	60			160	60			169
Aluminum, 2014-T6 (Spot Welded)							60		169
Aluminum, 2014-T6 (Extrusion)	160				160				20,169
Aluminum, 2014-T6 (Extrusion Stressed to 30,000 psi)					60				169
Aluminum, 2014-T6 (Welded and Stressed to 30,000 psi)					60				169
Aluminum, 2014-T6 (Hard Anodize)					60				20,169
Aluminum, 2014-T6 (H2SO4 Anodize)	160				160				169
Aluminum, 2014-T6 (Iridite)					60				20,169
Aluminum, 2014-T6 (Alodine)			160			160			169
Aluminum, 2014-T6 (Fluoride)					60				169
Aluminum, 2024-T6	160				160				20,165,169
Aluminum, 2219-T81					60				169
Aluminum, 2219-T81 (Welded)	160				160				169
Aluminum, 3003-H14	160				160				169
Aluminum, 3003-H16	160				160				169
Aluminum, 3003-H18					160				169
Aluminum, 3003-H19					160				169
Aluminum, 3003-H22					160				169
Aluminum, 3003-H24					160				169
Aluminum, 3003-H26					160				169
Aluminum, 3003-H28					160				169
Aluminum, 3003-H32					160				169
Aluminum, 3003-H34					160				169
Aluminum, 3003-H36					160				169
Aluminum, 3003-H38					160				169
Aluminum, 3003-H40					160				169
Aluminum, 3003-H42					160				169
Aluminum, 3003-H44					160				169
Aluminum, 3003-H46					160				169
Aluminum, 3003-H48					160				169
Aluminum, 3003-H50					160				169
Aluminum, 3003-H52					160				169
Aluminum, 3003-H54					160				169
Aluminum, 3003-H56					160				169
Aluminum, 3003-H58					160				169
Aluminum, 3003-H60					160				169
Aluminum, 3003-H62					160				169
Aluminum, 3003-H64					160				169
Aluminum, 3003-H66					160				169
Aluminum, 3003-H68					160				169
Aluminum, 3003-H70					160				169
Aluminum, 3003-H72					160				169
Aluminum, 3003-H74					160				169
Aluminum, 3003-H76					160				169
Aluminum, 3003-H78					160				169
Aluminum, 3003-H80					160				169
Aluminum, 3003-H82					160				169
Aluminum, 3003-H84					160				169
Aluminum, 3003-H86					160				169
Aluminum, 3003-H88					160				169
Aluminum, 3003-H90					160				169
Aluminum, 3003-H92					160				169
Aluminum, 3003-H94					160				169
Aluminum, 3003-H96					160				169
Aluminum, 3003-H98					160				169
Aluminum, 3003-H100					160				169
Aluminum, 3003-H102					160				169
Aluminum, 3003-H104					160				169
Aluminum, 3003-H106					160				169
Aluminum, 3003-H108					160				169
Aluminum, 3003-H110					160				169
Aluminum, 3003-H112					160				169
Aluminum, 3003-H114					160				169
Aluminum, 3003-H116					160				169
Aluminum, 3003-H118					160				169
Aluminum, 3003-H120					160				169
Aluminum, 3003-H122					160				169
Aluminum, 3003-H124					160				169
Aluminum, 3003-H126					160				169
Aluminum, 3003-H128					160				169
Aluminum, 3003-H130					160				169
Aluminum, 3003-H132					160				169
Aluminum, 3003-H134					160				169
Aluminum, 3003-H136					160				169
Aluminum, 3003-H138					160				169
Aluminum, 3003-H140					160				169
Aluminum, 3003-H142					160				169
Aluminum, 3003-H144					160				169
Aluminum, 3003-H146					160				169
Aluminum, 3003-H148					160				169
Aluminum, 3003-H150					160				169
Aluminum, 3003-H152					160				169
Aluminum, 3003-H154					160				169
Aluminum, 3003-H156					160				169
Aluminum, 3003-H158					160				169
Aluminum, 3003-H160					160				169
Aluminum, 3003-H162					160				169
Aluminum, 3003-H164					160				169
Aluminum, 3003-H166					160				169
Aluminum, 3003-H168					160				169
Aluminum, 3003-H170					160				169
Aluminum, 3003-H172					160				169
Aluminum, 3003-H174					160				169
Aluminum, 3003-H176					160				169
Aluminum, 3003-H178					160				169
Aluminum, 3003-H180					160				169
Aluminum, 3003-H182					160				169
Aluminum, 3003-H184					160				169
Aluminum, 3003-H186					160				169
Aluminum, 3003-H188					160				169
Aluminum, 3003-H190					160				169
Aluminum, 3003-H192					160				169
Aluminum, 3003-H194					160				169
Aluminum, 3003-H196					160				169
Aluminum, 3003-H198					160				169
Aluminum, 3003-H200					160				169
Aluminum, 3003-H202					160				169
Aluminum, 3003-H204					160				169
Aluminum, 3003-H206					160				169
Aluminum, 3003-H208					160				169
Aluminum, 3003-H210					160				169
Aluminum, 3003-H212					160				169
Aluminum, 3003-H214					160				169
Aluminum, 3003-H216					160				169
Aluminum, 3003-H218					160				169
Aluminum, 3003-H220					160				169
Aluminum, 3003-H222					160				169
Aluminum, 3003-H224					160				169
Aluminum, 3003-H226					160				169
Aluminum, 3003-H228					160				169
Aluminum, 3003-H230					160				169
Aluminum, 3003-H232					160				169
Aluminum, 3003-H234					160				169
Aluminum, 3003-H236					160				169
Aluminum, 3003-H238					160				169
Aluminum, 3003-H240					160				169
Aluminum, 3003-H242					160				169
Aluminum, 3003-H244					160				169
Aluminum, 3003-H246					160				169
Aluminum, 3003-H248					160				169
Aluminum, 3003-H250					160				169
Aluminum, 3003-H252					160				169
Aluminum, 3003-H254					160				169
Aluminum, 3003-H256					160				169
Aluminum, 3003-H258					160				169
Aluminum, 3003-H260					160				169
Aluminum, 3003-H262					160				169
Aluminum, 3003-H264					160				169
Aluminum, 3003-H266					160				169
Aluminum, 3003-H268					160				169
Aluminum, 3003-H270					160				169
Aluminum, 3003-H272					160				169
Aluminum, 3003-H274					160				169
Aluminum, 3003-H276					160				169
Aluminum, 3003-H278					160				169
Aluminum, 3003-H280					160				169
Aluminum, 3003-H282					160				169
Aluminum, 3003-H284					160				169
Aluminum, 3003-H286					160				169
Aluminum, 3003-H288									

TABLE 24. (Continued)

Material	Temperature, F								References
	Gas				Liquid				
	Class 1	Class 2	Class 3	Class 4	Class 1	Class 2	Class 3	Class 4	
<b>Rubbers</b>									
Chicago Rawhide-Sirvone 9623								145	92
Chicago Rawhide-Sirvone 9694								145	92
Chicago Rawhide-Sirvone 9617								85	92
Chicago Rawhide-Sirvone 20316								85	92
Commercial Hard Rubber Company 3601								85	92
Enjay 035					80			75	92
Enjay 218								140	92
Enjay 268								60	92, 169
Enjay 551								60	92
Enjay CR617								85	92
Pirestone Rubber D-404								85	92
Pirestone Rubber D-430								85	92
Pirestone Rubber D-431								85	92
Pirestone Rubber D-432								145	92
Pirestone Rubber D-406								145	92
Pirestone Rubber D-400								145	92
Pirestone Rubber D-408								145	92
Pirestone Rubber D-409								145	92
Pirestone Rubber D-410								85	92
Parker Appliances 37-24								85	92
Plastics and Rubber Products 805-70								85	92
Plastics and Rubber Products 805-90								85	92
Precision Rubber Products 907-90								85	92
Precision Rubber Products 925-70								85	92
St. Iman RR 613-75					80			100	92, 169
Stuber Rubber 85-00								85	92
Synthetic Rubber Products 5028606								85	92
Synthetic Rubber Products 50223								85	92
Thiokol C 4280-1								145	92
Thiokol C 5923								100	92
De Pont Neoprene (1156)								100	92
D.F. Goodrich Neoprene 091								60	92, 169
Neoprene Mycar 2202								85	92
Myca 520-41-125-1								85	92
Myca 1043 Std. B. 1								85	92
Myca 4001								100	92
Myca G4								100	92
Chemigum 86 12								75	92
Chemigum 86								75	92
Kal-F 3700								60	92, 169
Kal-F 5000								60	92, 169
Fluorubber 1F4								60	92, 169
Via rei								85	92
Viton A								85	92
Viton A-247B								85	92
Viton A-44-11 A-35								85	92
Viton B								80	165, 169
Millman Fluororubber 82 821-470								60	92, 169
Precision Rubber 18007, 18057								180	169
Hydropol V								140	92
Hydropol T								145	92
Toson X3603								75	92
Saran Rubber								75	92
Saran Rubber 300 Unhard								100	92
Saran Rubber 300 Cured								100	92
Dow Silicone LB-63								100	92
Dow Silicone DC-132								75	92
G.E. Nitrite Silicone MS805402								85	92
G.E. Nitrite Silicone 8270								130	92
Nicols Teflon A Silicone LB-63								85	92
Parco 823-70								80	165, 169
Parco 805-70								80	165, 169
Precision Rubber 9357								80	165, 169
Precision 214-607-9								80	169
Precision Rubber 9407								80	165, 169
940 X 399								160	169
Parker 8480-7								80	169
Parker 8496-7								60	169
Parker 318-7								160	169
Cusman 1357								80	169
Linear 7806-70								80	169
Medbar 82800-71								150	169
Formula 120 (Resin Cured)								160	169
Formula 121 (Resin Cured)								160	169
Acushnet 888-442								160	165, 169
Acushnet 888-849								160	169
Acushnet 888-850								160	169
Acushnet 888-851								160	169
Millman EX 904-90 LB3								160	165, 169
Medbar 58789-2307								60	165, 169
T-1021 300X St.								130	92
Derlock 900								60	169
Derlock 22								60	92, 169
Derlock 500								60	92, 169
Perco 8318-7								60	169
Adiprene B1136								100	92
Adiprene B1157								100	92

TABLE 24. (Continued)

Material	Temperature, F								References
	Gas				Liquid				
	Class 1	Class 2	Class 3	Class 4	Class 1	Class 2	Class 3	Class 4	
<b>LUBRICANTS AND SEALANTS</b>									
LESM Lube								80	169
S No. 58-4								80	169
LOX Safe								80	169
Amok C								80	169
ED-11								80	169
Microseal 100-1 (Dry Lubr)								80	169
Rockwell Nordstrom 147								75	92
Rockwell Nordstrom 421								75	92
Rockwell Nordstrom 561								75	92
Rockwell Nordstrom 921								75	92
Rockwell Nordstrom 950								80	169
Wolve Seal A								80	92
Flare Graphite DC-35								80	169
DC-41 Vacuum								80	169
Molykote Z								80	169
Kal-F 40								75	92
Kal-F 80								75	92, 165, 169
Kal-F 200								75	92
Drilube 703								60	165, 169
Bayco-32								60	169
Electrofilm 66-C								60	169
Polylycol Oils PL-65								60	169
Relison L								80	92, 169
RC 150								80	169
Fluorolube 80-600								80	169
Fluorolube Hg-1200								80	169
Fluorothene G								80	169
Androl L237								75	92
Canus 200								75	92
Fluoropak								75	92
Lubriscal								75	92
Worm Stopcock Grease								75	92
Microseal 15								75	92
Silicone DC 11								75	92
Kaddy Lube 100								80	169
Kaddy Lube 200								80	169
Water Glass Graphite								80	169
Drylube Sealant								80	169
Nydax A								80	169
Teflon Tape								80	169
<b>Paints and Compounds</b>									
PR 1422									60
RTV 20									60
Paraplex P-43									60
Proseal 793									60
Polyprene 8159									60
Crystal 8807									60
<b>Adhesives</b>									
Armstrong A-6									60
BT 847									60
BT 434									60
Epon 422								80	92, 169
Epon 4-3								80	169
<b>Ceramics</b>									
Temporel 1300								60	92, 169
Sauerstein P-1								60	92, 169
Sauerstein 31								60	169
Sauerstein 47								75	169
Resulfur								75	169
<b>Coatings</b>									
Epoxy No. 1									60
Modified Epoxy No. 2									60
Epoxy No. 7									60
Epoxy No. 9									60
Epoxy No. 6809									60
Alkyd No. 4									60
Polyurethane									60
Acrylic Nitrocellulose									60
Vinyl									60
Primer MIL-P-6889									60
Enamel									60
Tyger K									160
Dopolymer P-2000									160
EA 9747 Primer									160
Corrosite Clear 581									160
Proseal 323				</					

TABLE 25. COMPATIBILITY OF MATERIALS WITH 50:50 HYDRAZINE (UNSYMMETRICAL DIMETHYL HYDRAZINE) PLUS UP TO 20% WATER (50)

Material	Temperature, F								References
	Gas				Liquid				
	Class 1	Class 2	Class 3	Class 4	Class 1	Class 2	Class 3	Class 4	
<b>Metals</b>									
Aluminum, 1100-O					>60				
Aluminum, 5656-0/121					>60				
Aluminum, 7075-T6					>60				
Aluminum, 356-T6	>60								
Ni-Span-C					>60				
Titanium, SA-44					>60				

TABLE 26. COMPATIBILITY OF MATERIALS WITH LIQUID HYDROGEN (82,211,221)

Class 1 or 2 Materials		Class 4 Materials
Aluminum, 1100	Inconel	Austenitic, 40 E
Aluminum, 3052	Low-Carbon Steel	Magnesium
Aluminum, 4043	High-Nickel Steels	Lead
Aluminum, 2024-T3	Titanium	Zinc
Aluminum, 1100-T	Nitrile Rubber	Iron
300 Series Stainless Steel	Silicone Rubber	High-Carbon Steel
410 Stainless Steel	Teflon	Natural Rubber
Haynes 21	Gerrick Packing	Neoprene Rubber
Molybdenum	Bavelite	Cork
Nickel	Micarta	Wood
Monel	Lucite	Polymethyl Chloride
	Graphite	Saran
		Polystyrol Acetrol

TABLE 27. COMPATIBILITY OF METALS WITH 90 PERCENT HYDROGEN PEROXIDE

Aluminum Alloys	Stainless Steel Alloys	Nickel Alloys	Cobalt Alloys	Other Alloys	Other Pure Metals	Reference
<b>Class 1 Metals</b>						
Maximum per cent of active oxygen loss (AOL) by H <sub>2</sub> O <sub>2</sub> in 1 week: at 30 C (76 F) - 0.2%, at 66 C (150 F) - 5.0%						
Minimum stability of H <sub>2</sub> O <sub>2</sub> after test - 95% stable in 24-hour test at 212 F in glass						
No other effect on H <sub>2</sub> O <sub>2</sub> or metal						
1060	None	None	None	None	Tantalum	35, 36, 37, 38, 149, 210
1100					Zirconium	35, 36, 132, 149, 210
1160						35, 210
1260						35, 210
7072						35, 36, 37, 38, 210
B-356						35, 36, 38, 210
<b>Class 2 Metals</b>						
Maximum per cent of active oxygen loss (AOL) by H <sub>2</sub> O <sub>2</sub> in 1 week: at 30 C (76 F) - 6.0%, at 66 C (150 F) - 80%						
Minimum stability of H <sub>2</sub> O <sub>2</sub> after test - 90% stable in 24-hour test at 212 F in glass						
No other effect on H <sub>2</sub> O <sub>2</sub> ; slight bronzing of the metal allowable but no corrosion						
3003	Type 202	None	None	None	Silicon	35, 36, 210
4043	Type 302				Tin (cp)	35, 36, 149, 210
5052	Type 304					35, 36, 37, 38, 149, 210
5054	Type 304 ELC					35, 210
5056	Type 309					35, 210
5652-0	Type 310					35, 36, 210
5254-0	Type 316					35, 36, 37, 38, 210
6061	Type 316 ELC					35, 36, 37, 38, 210
3063	Type 317					35, 210
6363	Type 318					35, 36, 210
150	Type 321					35, 210
214E	Type 322					35, 36, 37, 38, 210
214F	Type 347					35, 36, 210
356F	17-7PH, 37-45 R <sub>c</sub>					35, 36, 37, 210
5052 (anodized)	17-7PH, 45 R <sub>c</sub> (buffed)					35, 210
6061 (anodized)	17-7PH (unhardened)					35, 210
6061 (HNO <sub>3</sub> Pass.)	Hasco-O-Seven					35, 210
6061 (detergent wash)	Malin-Wilabrite					35, 36, 210
6061 (WFNA Pass.)						35

Table 27. (Continued)

Aluminum Alloys	Stainless Steel Alloys	Nickel Alloys	Cobalt Alloys	Other Alloys	Other Pure Metals	Reference
<u>Class 3 Metals</u>						
Maximum per cent of active oxygen loss (AOL) by H <sub>2</sub> O <sub>2</sub> in 1 week: at 30 C (76 F) - 11.0%, at 66 C (150 F) - 100% in 24 hours						
Minimum stability of H <sub>2</sub> O <sub>2</sub> after test - 15% after 30 C (76 F) test						
Bronzing and staining, but no rusting or other corrosion products; slight attack may be allowed.						
2024	329	Inconel X	None	H-975	None	35, 37, 38, 210
B214F	AM 350			Refractalloy 26		35, 210
A360	17-7PH, 45 Rc			Refractalloy 70		35, 210
6061	17-7PH, 45 Rc					35
(anodized)	(electropolished)					
2017	17-4PH					35
(anodized)						
2024	Carpenter 20					35, 36, 210
(anodized)	Durimet 20					35, 210
6061	AM 355					35, 210
(hard coat)	19-9DL					35, 210
	16-25-6					35
	Preloy Type					35, 210
	302 porous wire					
	Worthite					35, 210
<u>Class 4 Metals</u>						
Maximum per cent of active oxygen loss (AOL) by H <sub>2</sub> O <sub>2</sub> in 1 week - not specified						
Minimum stability of H <sub>2</sub> O <sub>2</sub> after test - not specified.						
Pitting or corroding during or after test						
2014	AM 350	Nickel	Cobalt	Mild steel	Copper	35, 36, 149, 210
2017	410	Inconel	Haynes 25	Be-bronze	Zinc	35, 36, 149, 210
7075	416	Monel	Star "J"	Chromaloy H-3	Tungsten	35, 149, 206, 210
40E	420	Hastelloy	Haynes 3	Duriron	Titanium	35, 36, 210
218	430	"B"	Haynes 6	(cast)	Sodium	35, 36, 210
355F	431	Hastelloy	Haynes 12	Fanweld "O"	Magnesium	35, 38, 149, 210
		"C"				
A750	440		S-588	Ni-Resist	Beryllium	35, 210
B750	443	Hastelloy	S-590	Tantung	Cadmium	35, 149, 210
2024	446	"D"		Dow metal JIA	Chromium	35, 149, 210
(anodized)	Rigimesh J	Be-nickel		Dow metal MA	Gold	35, 210
	SS porous wire	Illium "G"		Kennametal	Iron	35, 36, 210
	300 series SS	Chlorimet		K-138	Lead	35, 36, 210
	powder compact			Kennametal	Manganese	35, 210
	Type 302			K-3H	Mercury	35, 210
	powder compact			Kennametal	Molybdenum	35, 94, 149, 210
				K-501		
	Type 316				Platinum	35, 210
	powder compact			Kennametal	Silver	35, 36, 210
	Type 302B			K-M		
	powder compact			Multimet N-155		35, 210
	Type 316 + Cb					35
	powder compact					
	Utiloy 3					35, 210
	Utiloy 20					35, 210
	Utiloy H					35, 210
	Utiloy NH					35, 210
	Elgiloy					35, 210

TABLE 28: COMPATIBILITY OF NONMETALS WITH 90 PERCENT HYDROGEN PEROXIDE AT 150 F (35)

Material	Classification
<u>Plastics - Polyethylene and Halogenated Polyethylene Type</u>	
Fluoroflex I-TP1001	1
Fluoroflex I-TP1000 (black)	2
Fluran B-4100	3
Halgene	2
Hypalon S-2	4
Hypalon Gasket	4
Hypalon V-54-B (gray)	4
Hypalon V-56-A (gray)	4
Hypalon V-163-4 (black)	4
Hypalon "O" Ring (GRC 90-5)	3
Irrathene 101 (irradiated polyethylene)	2
Kel-F (unplasticized)	1
Kel-F 800 (Lot 5649)	1
Kel-F 820 (G4028)	2
Kel-F 3700 gum	3
50% Kel-F 3700-50% Kel F 800	2
Kel-F 5500 (unpigmented)	2
Kel-F 5500 gum	3
Kel-F 5500-121	2
Kel-F 5500-61	2
50% Kel-F 5500-50% Kel-F 800	1
75% Kel-F 5500-25% Kel-F 800	1
Kel-F O-Ring (CPD. 7761-70)	2
Polyethylene	2
Rulon (Teflon Base)	2
Teflon (white)	1
Viton A (411A4) (black)	3
<u>Plastics - Polyvinylchloride and Copolymers</u>	
Alanol tubing	3
Boltron 6200 (gray)	2
Geon 118	4
Geon 404 (yellow)	3
Koroseal 116	3
Koroseal 117 (molded)	3
Koroseal 700	2
Lucoflex (translucent)	3
Lucoflex (white)	3
Marvinol 218-200	4
Marvinol 218-201	4
Marvinol NG-3005	4
Marvinol NR-6010	4
Saran	2
Saran Rubber Q-187	4
Transflex Tubing	4
Tygon B-20	3
Tygon B-32	3
Tygon B-63	3
Tygon B-71	3
Tygon B-72	3
Tygon B-136	3
Tygon S-22-1	4
Tygon TL-103	4
Tygon 2807	4
Tygon 3400	4
Tygon 3603	4
Tygon 3604A	2(a)
Tygon 3604B	2(a)
Vinyl 79139	2
Vinylite VG 1914	2
Vinylite VU 1940	2
Vinylite VS 1310	3
Vinylite VU 1900	3
Vinylite UE 1907	3
Vinylite VU 1920	3

TABLE 28. (Continued)

Material	Classification
<u>Plastics - Polyvinylchloride and Copolymers</u>	
Vinylite VU 1930	3
Vinylite VU 1940	3
<u>Plastics - Silicone Rubber Compounds</u>	
Fluorosilicone LS-53	2
GE 407B-217-1	4
GE 1240	2
GE 81223	2
GE 12601	4
GE 12602	3
GE 12650 (unpigmented)	2
GE 12650 (pigmented red)	3
GE 12670	4
GE 12670 (pigmented brown)	4
GE 15060 (pigmented)	3
GE 15080	3
GE X-7181	3
Parkone White 467-1 O-ring	4
SE 450 (unpigmented)	2
Silastic 152	3
Silastic 160	3
Silastic 160 O-ring	4
Silastic 161	3
Silastic 181	3
Silastic 240	2
Silastic 250	4
Silastic 261	3
Silastic 675	3
Silastic 6-128	2
Silastic 7-180	3
Silastic 9711	2
Silastic HR-9711	2
Silastic 9711 welded with S-2200	2
Silastic S-2000-4-480	2
Silicone 407-B-217-1	3
Silicone 407-B-437-1	2
Silicone HT 656	3
Silicone SR 5550	2
Silicone SR 5570	2
SR 5550	2
SR 5570	2
X-7181	3
Silicone Y-1749	2
<u>Rubbers and Plastics - General</u>	
Acrylon Rubber BA-12	4
Acrylon EA-5	4
Adiprene C	4
Bisilon No. 50	2
Buna N	4
Butyl Rubber A 3405	4
Butyl Rubber SR-384	4
Cyclocac (natural color)	4
Garlock No. 5681 (Teflon-impregnated asbestos)	4
Hycar PA 47B-1-1 (black)	4
Haveg 41 (Asbestos filled phenolic)	4
Haveg 60 (phenolic)	4
Hysol 4-77C (clear)	4
Hysol 4-77D (amber)	4
Hysol 4-77E	4
Hysol 4-77F	4
Hysol 4-78A (white)	4
Hysol 4-78B (brown)	4
Hysol 4-78C (amber)	4
Hysol 4-78D (amber)	4
Hysol 6000 B (amber)	4

TABLE 28. (Continued)

Material	Classification
<b>Rubbers and Plastics - General (continued)</b>	
Kralite	3
Melmac No. 1077	4
Mylar A	1
Mylar B	1
Neoprene Pure Gum	4
Neoprene SR 365-B	4
Nylon	4
Phenol-Formaldehyde	4
Plexiglas	4
Polystyrene (Polyflex)	2
Thiokol EC-801-LP2	4
Thiokol 3000 FA	4
Thiokol 3000 ST	4
Thiokol 1620 AH	4
<b>Plastics - Laminates, Diaphragm</b>	
<b>Materials and Adhesives</b>	
Chemelic MI-411 (Teflon Fiberglas)	2(b)
Duroid 5600 (fiber-reinforced Teflon)	3
Fairprene PS57-167 (Viton A, 116 Glass)	3
Fairprene PS57-168 (Viton A, Dacron)	2
Fairprene (Viton A)	
5806	2
5807	2
5809	2
Kel-F-Dacron Diaphragm-VL-1101m4	4
Kel-F 5160 Diaphragm	2
Kel-F 5500 (gray) Diaphragm	2
Kel-F 5500 (gray) on Dacron diaphragm	2
Korda Flex (Teflon-coated glass fabric)	2
Silastic DC-9711 on Dacron diaphragm	3
Vinyl coated Fiberglas (gray-green)	3
9711 Silicone seal washer DC A 4094 adhesive (Dow Corning Silicate base) on aluminum	3
9711 Silicone seal washer DC Chemloc 607 adhesive on aluminum	3
<b>Porous Materials</b>	
Al-Si-Mag, Porous Ceramic No. 393	3
Aluminum Oxide, Porous-RA-98	2
Armalon-Teflon Felt (impregnated)	3
Armalon-	3
Dacron Cloth	2
Dac-2100	
Dac-2101	2
Dac-2102	2
Filtros C Stone (55 Micron)	2
Glass Cloth G-206-C	2
Poroloy--302SS Wire	3
Porous Kel-F (15-Micron Pore)	2
Porous Porcelain (1.4 Micron)	2
Porous Teflon (9-Micron Pore)	2
Rigimesh J SS, Wire	4
Sintered 300 Series SS Powder Compact	4
Sintered 302 SS Powder Compact	4
Sintered 316 SS Powder Compact	4
Sintered 302B SS Powder Compact	4
Sintered 316 and Cb SS Powder Compact	4
Teflon Cloth (25 Grade)	3
Teflon Cloth-Repeat (25 Grade)	2
Teflon Cloth (40 Grade)	3
Teflon Felt (impregnated)	
Teflon Cloth T-2300	3
Teflon Cloth T-2305	2

TABLE 29. (Continued)

Material	Classification	Impact Sensitive
<b>Lubricants</b>		
Alkaterge C	4	Yes
Amino Silane Oil and Grease	4	Yes
Aplezon Hardwax W	4	
Arochlor 1221	4	Yes
Arochlor 1232	4	Yes
Arochlor 1242	4	Yes
Arochlor 1248	4	Yes
Arochlor 1254	4	Yes
Bardahl	4	Yes
Carum 200	4	Yes
Ceresin Max	4	Yes
CFB-1	4	Yes
Dichloro-bis-tri-fluoromethyl benzene	3	No(c)
Dichlorohexafluorobutene	3	No(c)
Fluorolube FS	2	No(c)
Fluorolube FS plus 5% fluorolube light grease	2	No(c)
Fluorolube heavy grease 10214	2	No(c)
Fluorolube oil 10213	2	No(c)
Fluorolube S	2	No(c)
Fluorolube T	2	No(c)
Fluorolube Oil, S-30	2	No(d)
Fluorolube Grease, Hg-1200	2	No(d)
Fluorolube Grease, GR-660	2	No(d)
Formulation		Yes
F-9	4	
OS-16	4	Yes
OS-22	4	Yes
OS-23	4	Yes
OS-27	4	Yes
OS-28	4	Yes
OS-30	4	Yes
OS-32	4	Yes
OS-33	4	Yes
OS-34	4	Yes
OS-35	4	Yes
OS-37	4	Yes
CP-3898-2	4	Yes
Skydrol (uncolored)	4	Yes
Halocarbon Oil 8-25 AV	2	No(c)
Halocarbon Grease, Series 25-10	2	No(d)
Halocarbon Hi-Temp Stopcock Grease	2	No(d)
Halocarbon Oil 10-21	2	No(c)
Halocarbon Oil 11-14	2	No(c)
Halocarbon Stopcock Grease	2	No(d)
Hexachlorobutadiene	3	No(c)
Hexachloropropylene	4	
Hydraulic fluid RPM	4	Yes
Hydraulic Oil Houghton Safe 620	3	
Kel-F Alkane	2	No(d)
Kel-F Oil Cut No. 1	2	No(c)
Kel-F Oil No. 10	2	No(c)
Kel-F No. 90 Grease	2	No(c)
Lindol HF (tricresyl phosphate)	4	Yes
Lindol HFX	4	Yes
Liqui-Moly Concentrate	4	Yes
Lubri-Seal	4	Yes
Mineral Oil	4	Yes
Paraffin	4	Yes
Perfluorolube Grease FCD-759	2	No(c)
Perfluorolube Oil FC-331	2	No(c)
FC-332	2	No(c)
FC-333	2	No(c)
FC-334	2	No(c)
FC-335	2	No(c)
Petrolatum	4	Yes
Polychloropentane (stabilized)	4	Yes
Renex	4	Yes
Silicone XF 224	4	Yes
Silicone Oil DC-7	4	Yes
DC-44	4	Yes
DC-200	4	Yes
DC-550	4	Yes
DC-701	4	Yes
DC-702	4	Yes
DC-710	4	Yes
Silicone Oil GE 2V3733	4	Yes
GE 5134c	4	Yes
Tectyl	4	Yes
1,1,2,2, tetrafluoroethyl dodecyl ether	4	Yes
Tributyl Phosphate	4	Yes
Ucon Hydrolube U-4	4	Yes

TABLE 28. (Continued)

Material	Classification
<b>Ceramics, Refractories, and Miscellaneous</b>	
Agate (natural)	3
Agate (polished)	3
Al-Si-Mag (porcelain)	2
Alundum LA 116	2
Boron Nitride	4
Carboloy 44-A	4
Carboloy 55-A	4
Carboloy 78	4
Carboloy 999	4
Ceramic AB-2	2
Ceramic Al-200	2
Charcoal	4
Crystalor. (Silicon Carbide)	4
Graphitar No. 30	4
Graphite P5A6 Silver impregnated	4
Graphite P-55 Copper impregnated	4
Graphite P-59L Copper impregnated	4
Graphite P-692	4
Karbate	4
Norbide	2
Synthetic Sapphire (polished)	1

**Protective Coatings**

(A. Recommended for Long-Time Contact and Splash Resistance)

Teflon	1 (e)
Kel-F	1 (e)
Kel-F on 1060 Aluminum	1
Kel-F on 5234 Aluminum	1
Kel-F on 5652 Aluminum	1
Glass-lining (Clear) Light-Gray	1
Glass-lining (Cobalt) Cobalt-Colored Glass	1

(B. Recommended for Splash Resistance Service Only)

Tygon Paint 7286 TP-81-Clear	3 (e)
Tygon Paint 71253 TP-107B	3 (e)
Corrosite No. 521	3 (e)
Corrosite No. 551	3 (e)
Corrosite No. 581	3 (e)
Plastic Metal No. 22	3 (e)
Saran Rubber Q-1875	3 (e)
Mv-Type No. 150	3 (e)
Amercoat No. 1262	3 (e)
Hellex	3 (e)
P-5, Copolymer	3 (e)
Neolac Gray No. 8588	3 (e)
Steelcote Stainless Steel	3 (e)

(C. Not Recommended for 90% H<sub>2</sub>O<sub>2</sub> Service)

Geon Latex 31X	4 (e)
Flexcoat No. 1 Black	4 (e)
Lithgow LC-600 (Gray)	4 (e)
Amercoat Red	4 (e)
Prufcoat Medium Gray	4 (e)
Lithgow LC (600) (Brown)	4 (e)
Veloform F-10 CPP304	4 (e)
Cordo -255A	4 (e)
Cordo Plastic Coating (E-1 Resin + H-26 Activator)	4 (e)
Chromalloy	4 (e)
Unichrome Drum Lining B-124-17	4 (e)
Uclion, System E Coating	4 (e)
EX63B Paint	4 (e)
Sealer ECBC1 With Accelerator	4 (e)

TABLE 28. (Continued)

Material	Formulation	Classification	Impact Sensitive (f)
<b>Joint Sealing Compounds</b>			
"Alithon" E	Polyethylene resin	1(e)	Non-impact sensitive
Dispersate 1820-	Butyl rubber in alkaline dispersion	2(e)	Non-impact sensitive
D-Seal	Teflon	4(e)	Non-sensitive Impact
Calbar CB Pipe Seal		4(e)	sensitive
Grane Thread Lub.	Oxidizable Oil	4(e)	1 doubtful Impact from 10 trials
Fel-Pro, G-B	Colloidal Copper	4(e)	Impact sensitive 1 of 4 trials
Graphite Paste	Graphite dispersion	4(e)	Non-impact sensitive
Goop (Blue)		4(e)	Positive
Goop (Silver)		4(e)	Positive
Cyl-Seal		4(e)	Non-impact sensitive
Molybdenite Pipe Dope		4(e)	Not tested
Permatex Aviation Form A Gasket No. 3		4	Not tested
CS-18 Lubricant	Skyrol + 3% benton 34	4(e)	Impact sensitive
Pecora Cop.		4(e)	Impact sensitive
Plastic Metal	No. 22-aluminum filler, latex base. No. 333-55 filler, latex base	4(e)	Impact sensitive when wet
Rutland Pipe Dope		4(e)	Sensitive 2 of 3 trials
Skyrol	Mixture Skyrol (P-501) and talc	4(e)	Impact sensitive
E-Film	Teflon-water dispersion	4(e)	
Weco No-Gel.	No. 50 metallic zinc base	4(e)	Impact sensitive
X-Pards		4(e)	Non-impact sensitive
Kel-F Grease No. 90	Polychlorotrifluoroethylene		Non-sensitive
Tin Plating Or Aluminum 6061		4(e)	
Alcoa Thread Lubricant		4(e)	Not tested
Ractoseal No. 15		4	Not tested

- (a) Based on service experience.
- (b) After 24-hour screening at 66 F (150 F).
- (c) Non-impact sensitive to 1 kg at room temperature.
- (d) Non-impact sensitive to 3 kg at room temperature.
- (e) Tested at room temperature.
- (f) One-kg impact at room temperature.

TABLE 29. COMPATIBILITY OF MATERIALS WITH METHYLENE CHLORIDE (CH<sub>2</sub>Cl<sub>2</sub>)

Material	Temperature, F								References
	Gas				Liquid				
	Class 1	Class 2	Class 3	Class 4	Class 1	Class 2	Class 3	Class 4	
<b>Metals</b>									
1100 Aluminum			125				75	100	211
Copper			125		100		75		211
Sr Bronze			125				75		211
Al Bronze			125				75		211
St Bronze			125		175		75		211
Red Brass			125				75		211
Yellow Brass			125				75		211
Mild Steel			125				75		211
Cast Iron							75		211
Mi-Resist							75		211
Si-Iron							75		211
410 Stainless Steel							75		211
430 Stainless Steel					100		75		211
304 Stainless Steel					75		75		211
316 Stainless Steel					75		75		211
Norphite	175		212						211
Durimet 2C	175		212						211
Carpenter 2C					100				211
A-Nickel			75			100	75		211
Monel			125			100	75		211
Inconel			75			100	75		211
Hastelloy B					100				127, 129, 211
Hastelloy C					100		100		127, 129, 211
Ni-O-ne					75				211
Lead							75		211
Gold							75		211
Platinum							75		211
Tantalum							75		211
Silver							75		211
<b>Nonmetals</b>									
Glass							75		211
Stoneware							75		211
Rubber								75	211
Asbestos					75				211
Graphite					75				211







TABLE 33. (Continued)

TABLE 33. (Continued)

Material	Temperature, F								References	
	Gas				Liquid					
	Class 1	Class 2	Class 3	Class 4	Class 1	Class 2	Class 3	Class 4		
Kal-F									75	90
Kal-F 300 (unplasticized)									60	169
Kal-F 300 (Annealed)									60	169
Genetron GC									160	23,169
Genetron GCX-3B					65				80	169
Genetron III-2B					65				75	169
Talthane A									67	23,169
Alicar 191										75
Polyethylene									60	169
Polyethylene (low density)									60	169
Polyethylene (high density)									60	169
Polyethylene (irradiated)									60	50,169
Raythane X (irradiated Polypropylene)					65				60	169
Nylon									60	50,169
Zytel 101									60	169
Capton 391									60	23,169
Mylar									60	23,50,169
Kynar									60	169
Delrin									80	30,169
Lexan									60	30,169
Medlar									60	30,169
Merlex 50									60	30,169
MI-Far									60	169
Spandite									60	169
Kodapak II									60	169
Hypalon 20									60	169
H-Flex									60	169
Epon 1031 (With PMDA)									60	169
Flexiglas II									75	90
Flexiglas CR-39									75	90
Flexiglas 95									75	90
Silicone Laminate									60	30,169
Phenolic Laminate									60	30,169
Epoxy Laminate									60	30,169
Polyester Laminate									60	30,169
Acrylic Laminate									60	30,169
Polyvinylchloride									60	169
Ultron									60	169
Opalon 1219									60	169
Opalon 1020									60	169
Opalon 1414									60	169
Opalon 81222									60	169
W-1014									60	169
<b>Elastomers</b>										
Perlastone 14 (ethylene Propylene Rubber)									60	169
Formula 132 (Ethylene Propylene Rubber, XG)									60	169
E-612-2									60	169
E-622-1									60	169
X-7000-1 to 7 and 9 to 11									60	169
Parker 800-7									60	169
Parker 800-8									60	169
Parker 8406-7									60	169
Parker XB-1235-10									60	169
Enja, 268									60	169
Enja, 551									60	169
Hycon 2202									60	169
Stilman 58013-75									60	169
Stilman 11502-34									60	169
Stilman 104-6-12a									60	169
Precision 1330120									60	169
Precision Formula 120									60	169
Precision Formula 121									60	169
Parker XV-1235-2									60	169
Parker XV-1235-5									60	169
Parker XB-1235-10									60	169
Trifluoroethylene-methane tetrafluoroethylene (TFM-TFE)									60	169
Viton A									60	169
Viton B									60	169
Stilman EXT74M-1									60	169
Omni A FBF-4									60	169
Parker V454-7									60	169
Parker 71-045									60	169
EX92-470									60	169
Kel-F Elastomer									60	169
Kel-F-3700									60	169
Kel-F-9900									60	169
Stilman TH107									60	169
Fluorel									60	169
LS 53									60	169
LS 63									60	169
Natural Rubber									60	169
Buna A									60	169
B-318-7									60	169
Conlastic 500									60	169
Silicone									60	169
Gerick 22									60	169
Gerick 900									60	169
Neoprene									60	169
Hypalon 20									60	169
Polybutadiene									60	169
Tyloc 1050									60	169
Tyloc 1640C									60	169
Tyloc 1647C									60	169
Penton									60	169
Chloroprene									60	169

Material	Temperature, F								References	
	Gas				Liquid					
	Class 1	Class 2	Class 3	Class 4	Class 1	Class 2	Class 3	Class 4		
<b>Lubricants</b>										
DC 11									67	169
DC 55									67	169
DC 11 VAC									67	169
KG 150									67	169
Rayco-30 Grease									65	169
Kel-F 90									60	169
Polysyl Oil									67	169
FX 45									67	169
Molykote Z									60	169
DriLube 703									60	169
Electrofilm 66-C									60	169
Rayco-32 Grease									60	169
Halocarbon Grease									67	169
Hydrosol 1475 and 421									60	169
Microwax 100-1									67	169
FD-78									67	169
LOX Safe									60	169
Flake Graphite									60	169
Johns Manville No. 50									60	169
Graphite 2,14,29,30, and 86									67	169
CCP-72									67	169
Parker P3M and P3N									67	169
Valve Seal A									67	169
Aplon L									60	169
Aplon L M6000									60	169
Fluoroethane G									60	169
<b>Sealants and Polishing Compounds</b>										
Redd-Lube 100									67	169
Redd-Lube 200									67	169
Waterglass-Graphite									67	169
Vulca A									67	169
DryLube Sealant									60	169
Teflon Tape (Unstretchered)									60	169
28 1422									60	169
RTV 20									60	169
Epon 928									60	169
Formapex P-43									60	169
Formapex 703									60	169
Palprene 515F									60	169
Crytal M & D									60	169
DC 11001									60	169
Epon VI									60	169
Epon VIII									60	169
Epon 422									60	169
Epon 811									60	169
Epon 101									60	169
Epon 815									60	169
EC-150									60	169
EC-104									60	169
Resinex SC1008									60	169
Resinex SC1013									60	169
Resinex E120-15									60	169
US Polymeric Division									60	169
<b>Adhesives</b>										
Armstrong A-6									60	169
EG 847									60	169
HT 424									60	169
3M-AP-6									60	169
4-3									60	169
Epon 422									60	169
<b>Chemicals</b>										
Temperal 1500									60	169
Sauerstein P-1									60	169
Sauerstein 11									60	169
Sauerstein 47									60	169
<b>Coatings</b>										
Roctiflex									60	169
Resinex 1, 2, and 3									60	169
Alutrols									60	169
Epoxy No. 1									60	169
Epoxy No. 5									60	169
Epoxy No. 7									60	169
Epoxy No. 9									60	169
Epoxy 8409									60	169
Modified Epoxy No. 5									60	169
Axyd No. 4									60	169
Polystyrene									60	169
Detail Primer and Filler on Steel									67	169
Catalac Improved									67	169
Acrylic Microcellulose Vinyl									60	169
Primer MII-P-6889									60	



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