

ASD-TDR-62-869

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**DEVELOPMENT OF A NICKEL BASE ALLOY SHEET
FOR HIGH TEMPERATURE APPLICATION**

TECHNICAL DOCUMENTARY REPORT NO. ASD-TDR-62-869

April 1963

Directorate of Materials and Processes
Aeronautical Systems Division
Air Force Systems Command
Wright-Patterson Air Force Base, Ohio

Project No. 7351, Task No. 735105



(Prepared under Contract No. AF 33(616)-7999 by Chance Vought Corp.,
Dallas, Texas; H. Greenewald, Jr., and T. J. Riley, Authors)

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FOREWORD

(21) Report on

This report was prepared by Chance Vought Corp., Aeronautics and Missiles Division, under USAF Contract No. AF 33(616)-7999. This contract was initiated under Project No. 7351, "Metallic Materials", Task No. 735105, "High Strength Metallic Materials". The work was administered under the direction of the Directorate of Materials and Processes, Deputy for Technology, Aeronautical Systems Division, with Captain L. F. Bubba, succeeded by Lt. F. L. Krempski, acting as project engineer.

This report covers work conducted from 15 March 1961 to 15 July 1962.

Mr. H. Greenwald, Jr., was the Chance Vought principal investigator. Mr. T. J. Riley cooperated in the research and in the preparation of this report. Mr. R. Calvert was in charge of all mechanical tests.

In addition to the authors, Mr. John Freche and Mr. William Waters of the Lewis Research Center, NASA, Cleveland, Ohio, have made substantial contributions to the work on the NASA Taz8 alloy.

FOREWORD

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In addition to the authors, Mr. John Freche and Mr. William Waters of the Lewis Research Center, NASA, Cleveland, Ohio, have made substantial contributions to the work on the NASA TaZ8 alloy.

ABSTRACT

The objective of this contract was to develop 15 to 30 mil nickel alloy sheet having 50,000 psi tensile strength at 1900°F, having good corrosion (oxidation) resistance, and good ductility. This objective was essentially attained by developing a new process of directly rolling thin cast slabs of nickel base alloy into sheet on a specially designed rigid rolling mill.

Two pre-existing nickel base casting alloys and a series of experimental compositions obtained by modifying the two starting alloys were initially investigated in the cast condition in this program. The two starting alloy compositions were Inco 713c developed by International Nickel Co. and the TaZ8 alloy developed by Lewis Research Center of NASA. Of the new experimental compositions, TaZ8 alloy and No. 429 alloy have 1900°F tensile strengths exceeding 50,000 psi at 1900°F in the as cast condition. Inco 713c has a tensile strength of about 40,000 psi in the as cast condition.

Procedures were developed on the rigid mill for hot rolling both Inco 713c and TaZ8 alloy into 15 to 30 mil sheet. Heat treatment procedures were developed for both alloys which enabled them to meet contract target strength properties. The maximum room temperature ductility obtained in rolled and heat treated Inco 713c sheet was 20% elongation compared to a maximum of 5% elongation for the rolled TaZ8 alloy. The good room temperature ductility of hot rolled Inco 713c was further indicated by the fact that this hot rolled sheet was cold rolled from 20 mil down to 3 1/2 mil foil. The oxidation resistance of TaZ8 alloy is adequate for limited periods of time at 1900°F; that of No. 429 alloy is substantially better; and that of Inco 713c is best of the three.

This technical documentary report has been reviewed and is approved.



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TABLE OF CONTENTS

		PAGE
I	Introduction.	1
II	Description of the Program.	2
III	Experimental Procedures and Tests	3
	A. Processing Techniques	3
	B. Evaluation Techniques	5
IV	Experimental Data and Discussion.	11
	A. Alloy Improvement by Varying Chemical Composition	11
	B. Alloy Sheet Improvement by Rolling.	35
	C. Alloy Sheet Improvement by Heat Treatment	57
	D. Rolled Alloy Sheet Evaluation	80
V	Conclusions and Recommendations	101
	A. Conclusions	101
	B. Recommendations	102
VI	Bibliography.	103
	APPENDIX A.	
	Melting and Casting Techniques and Equipment.	104
	APPENDIX B.	
	The Rigid Rolling Mill.	117
	APPENDIX C.	
	Elevated Temperature Tensile Test Procedure	123
	APPENDIX D.	
	Complete Casting Data Summary	125
	APPENDIX E.	
	Vendor Typical Chemical Analysis for Melt Charges	128
	APPENDIX F.	
	Nickel Alloy Rolling Data	129
	APPENDIX G.	
	Tensile Data of Nickel Alloys	141

LIST OF FIGURES

FIGURE	PAGE
1. Sheet Tensile Specimens Used for Room-Temperature Tests	6
2. Typical Elevated Temperature Tensile Specimens	6
3. Typical Tensile Test Setup for Resistance Heating	8
4. Relationship of Specimen to Quartz Lamps Used for Radiant Heat Tensile Test.	8
5. Electron Micrographs of Inco 713c in the As-Cast Condition, Melt 405. Etch No. 1	15
6. Electron Micrograph Showing TiC Precipitate in a Grain Boundary of As-Cast Inco 713c. 9000X. Heat No. 56	15
7. Microstructure of As-Cast Al Modified Inco 713c (19.01% Al, Melt 430). Etch No. 3	16
8. Microstructure of As-Cast Al Modified Inco 713c (27.64% Al, Melt 431). Etch No. 3	16
9. Microstructure of As-Cast Al-Ti Modified Inco 713c (9.62% Al, 2.68% Ti, Melt 438). Etch No. 4	16
10. Microstructure of As-Cast Ta-C Modified (16.26% Ta, 1.08% C) Inco 713c, Melt 388.	16
11. Microstructure of As-Cast Ta-C Modified (16.26% Ta, 1.08% C) Inco 713c, Melt 390. Etch No. 3.	17
12. Microstructure of As-Cast 2(Ta-C) Modified (26.52% Ta, 1.76% C) Inco 713c, Melt 392. Etch No. 3	17
13. Microstructure of As-Cast W-C Modified Inco 713c (13.05% W, 1.45% C, Melt 403). Etch No. 3.	17
14. Microstructure of As-Cast Ta-Mo-C Modified Inco 713c (9.79% Ta, 17.27% Mo, 1.66% C, Melt 428). Etch No. 3.	17
15. Microstructure of As-Cast Ta-Cr-C-W Modified Inco 713c (7.58% Ta, 17.17% Cr, 2.46% C, 17.75% W, Melt 429). Etch No. 3	18
16. Microstructure of As-Cast Zr-C Modified Inco 713c (7.21% Zr, 0.54% C, Melt 432). Etch No. 3.	18

LIST OF FIGURES (Continued)

FIGURE	PAGE
17. Microstructure of As-Cast Ti-C Modified Inco 713c (11.90% Ti, 1.11% C, Melt 433). Etch No. 3.	18
18. Microstructure of As-Cast Mo-C Modified Inco 713c (14.85% Mo, 2.45% C, Melt 434). Etch No. 3.	18
19. Microstructure of As-Cast Cr-W-C Modified Inco 713c (17.30% Cr, 15.83% W, 1.14% C, Melt 435). Etch No. 3.	19
20. Microstructure of As-Cast Cr-Ti-C Modified Inco 713c (14.07% Cr, 7.19% Ti, 1.15% C, Melt 436). Etch No. 3.	19
21. Microstructure of As-Cast Cb-Ta-W-C Modified Inco 713c (3.12% Cb, 0.83% Ta, 7.76% W, 2.66% C, Melt 437). Etch No. 3.	19
22. Comparison of Elevated Tensile Strengths of Several Nickel Base Alloys between 1900 ^o F and 2300 ^o F.	23
23. Comparison of Microstructure of As-Cast Inco 713c Sheet with that of As-Cast NASA. Etch No. 3.	24
24. Comparison of Microstructure of NASA Alloy, Melt 385, with that of Ta-C Modified Inco 713c (16.26% Ta, 1.08% C, Melt 388), Both Heat Treated at 2000 ^o F for 16 Hours and Aged at 1500 ^o F for 28 Hours. Etch No. 1.	25
25. Comparison of Microstructure of NASA Alloy, Melt 385, with that of Ta-C Modified Inco 713c (16.26% Ta, 1.08% C, Melt 388), Both Heat Treated at 2200 ^o F for 16 Hours and Aged at 1500 ^o F for 28 Hours. Etch No. 1.	25
26. Effects of Variations in Tantalum and Carbon Additions on the Microstructure of As-Cast Ta-C Modified Inco 713c. Etch No. 3.	26
27. Comparison of As-Cast Microstructure of Inco 713c (Melt 405), Ta-C Modified Inco 713c (40.56% Ta, 2.63% C, Melt 395), No. 429 Alloy (Melt 457) and NASA Alloy (Melt 387) as shown by Electron Micrographs at 25000X	27
28. Microstructure of As-Cast Cr Modified NASA Alloy Sheet (17.68% Cr, Melt 426). Etch No. 3.	31
29. Microstructure of As-Cast Cr-Al Modified NASA Alloy (17.17% Cr, 8.56% Al, Melt 427). Etch No. 3.	31

LIST OF FIGURES (Continued)

FIGURE		PAGE
30.	Modified NASA Alloys as Cast (Melt 529, 530, 531, 532, 533, 534). Etch No. 4. 500X	32
31.	Macro-Etched, Rolled Inco 713c Showing that Cracking During Rolling Tends to Occur at the Grain Boundaries. 1.5X.	33
32.	Typical Examples of NASA Alloy Rolled at M & C	40
33.	Work Hardening of Inco 713c, Heat No. 404 During Cold Rolling.	43
34.	Work Hardening of Ta-C MOD Inco 713c, Heat No. 388 During Cold Rolling	43
35.	Work Hardening of NASA Alloy, Heat No. 425 During Cold Rolling.	43
36.	Effects of Cold Rolling and Annealing on the Microstructure of Cast Inco 713c, Melt 404. Etch No. 3. 500X	44
37.	Effects of Cold Rolling (20%) and Annealing at 2000°F on the Microstructure of Ta-C Modified Inco 713c, Melt 388. Etch No. 3. 500X	44
38.	Effects of Cold Rolling and Annealing on the Microstructure of NASA Alloy, Melt 419. Etch No. 3. 500X.	45
39.	NASA Alloy in Stainless Steel.	46
40.	Rolled Inco 713c Sheet (Fine Grain).	49
41.	Rolled Inco 713c Sheet (Coarse Grain).	49
42.	Inco 713c, Rolled At Vought. .025"Thick	50
43.	Comparison of the Effects of 90% Reduction in Thickness by Hot Rolling on Inco 713c, NASA, and No. 429 Alloys. Etch No. 4. 500X	51
44.	Inco 713c, Melt 550, as Rolled at Room Temperature, Etch No. 4. 500X. (Note 200 gm Micro-Hardness Indentations)	52
45.	NASA Alloy, Hot Rolled to Various Reductions in Thickness at Several Temperatures. Etch No. 4. 500X	53

LIST OF FIGURES (Continued)

FIGURE	PAGE
46. Comparison of Microstructures of As-Cast and As-Rolled No. 429 Alloy. Etch No. 4. 500X.	54
47. Comparison of Microstructures of As-Cast and As-Rolled Ta-C Modified (16.26% Ta, 1.08% C) Inco 713c, Melt 388. Etch No. 1. 500X	54
48. Electron Micrographs, No. 429 Alloy, as Rolled, 60% at 2150°F, Melt 457. Etch No. 1.	55
49. NASA Alloy, Rolled 60% at 2000°F, Melt 462.	55
50. Inco 713c Foil Rolled at Vought. Melt No. 550.	59
51. NASA Alloy Rolled at 1850°F.	60
52. Diagram Showing Strain Distribution in Sheet Being Rolled On a Rigid Mill with Resultant Edge Cracks.	61
53. Diagram Showing Resultant Strain and Crack Distribution During Two Successive Passes on Conventional Mill.	61
54. Homogenization of As-Cast Dendritic Structure in Inco 713c by Heat Treatment at 2150°F. Etch No. 1. 100X	67
55. Effect of Solution Heat Treatment of Cast Inco 713c. Etch No. 2. 500X	67
56. TiC Precipitate in Grain Boundary of As-Cast Inco 713c. Note: Micro-Hardness Indentor Shattered the Precipitate, Indicating Its Brittle Nature. Etch No. 1. 1000X	67
57. Inco 713c Section Cut From One-Inch Thick Specimen Heat Treated at 2150°F for 3 Hours. Etch No. 1. 100X.	67
58. Effect of Solution Heat Treatment on Grain Boundary in Cast Inco 713c, Melt 67. Etch No. 2. 4000X.	68
59. Comparison of Macro-Etching Characteristics of As-Cast and Heat-Treated Inco 713c.	69
60. Effects of Heat Treatment on the Microstructure of Hot Rolled NASA Alloy. Etch No. 4. 500X.	70
61. Comparison of Microstructure of NASA Alloy in the As-Cast, As-Rolled, and As-Rolled + Heat-Treated Condition.	71

LIST OF FIGURES (Continued)

FIGURE	PAGE
62. Effect of Various Prolonged Solution Heat Treatments on Rolled NASA Alloy at 2200°F, Air Cooled. Etch No. 4. 500X.	76
63. Electron Micrograph Study of the Effects of Two Different Heat Treatments on Cast NASA Alloy, Melt 385. Etch No. 1.	77
64. Variation of Microstructure within a Single Casting of NASA Alloy, Melt 385 As-Cast. Etch No. 1. 100X	77
65. Optical Microscope Study of the Effects of Two Different Heat Treatments on Cast NASA Alloy	78
66. Electron Micrographs of NASA Alloy Hot Rolled 60%, then Aged at 1500°F for 63 Hours. Etch No. 1	79
67. Ultimate Tensile Strength and Percent Elongation Variation of Inco 713c Rolled Sheet at Various Temperatures (Condition of Metal, Rolled + H.T. at 2150°F for 40 Hours + 24 Hours at 1600°F.	85
68. Calculated Average Strain Rate versus Ultimate Tensile Strength of Inco 713c Sheet at 1900°F Condition, Rolled + H.T. 40 Hours at 2150°F + 24 Hours at 1600°F.	86
69. Total Time of Tensile Test versus Ultimate Tensile Strength of Inco 713c Sheet at 1900°F. (Conditions are the same as Figure 68)	86
70. Increase in Weight of Specimen (With Oxide Scale) After Exposure at 2100°F.	91
71. Net Change in Weight of Specimen After Exposure at 2100°F in Air (After Removing Loose Oxide).	91
72. Increase in Weight of Specimen (With Oxide Scale) After Exposure at 2100°F.	92
73. Net Change in Weight of Specimen After Exposure at 2100°F in Air (After Removing Loose Oxide).	92
74. Oxidation Tests Run on As-Cast Inco 713c, NASA Alloy and No. 429	93
75. Effects of Heat-Treatment Temperatures Near the Solidus Temperature on the Microstructure of NASA Alloy. Etch No. 4. 100X.	95

LIST OF FIGURES (Continued)

FIGURE	PAGE
76. Effects of Heat-Treatment Temperatures Near the Solidus Temperature on the Microstructure of Ta-C Modified (16.26% Ta, 1.06% C) Inco 713c, Melt 390, Water Quenched, Etch No. 4.	96
77. Specimens Heated to Various Temperatures for One-Half Hour and Quenched to Determine Solidus Point for Two Alloys.	97
78. Vought Sheet Specimen Grinding Apparatus.	100
79. Melting and Casting Furnace in Empty Position	111
80. Melting and Casting Furnace with Mold in Prepouring Position.	111
81. Melting and Casting Furnace Loaded and Ready to Pour.	111
82. Melting and Casting Furnace in the Poured Position.	111
83. Pressed and Cured Ceramic Mold Half	112
84. Typical Mold Preheat Cycle.	112
85. Cleaned Casting with Cast Sheets Still Attached to the Riser.	113
86. NASA Alloy .100" Thick Sheet (as Cast & Cleaned).	113
87. Solidification in a Typical Simple Eutectic System.	114
88. Typical Solidification Mode in a Casting.	114
89. Solidification in a Simple Solid Solution System.	114
90. Typical Redistribution of Macroseggregation in a Rolled Metal by Heat Treatment.	115
91. Typical Redistribution of Macroseggregation in a Cast Metal by Rolling.	115
92. Redistribution of Macroseggregation in a Cast Metal by Heat Treatment	115
93. Mode of Tensile Failure in a Cast Structure	116
94. Mode of Tensile Failure in a Rolled Structure	116
95. Tensile Failure in a Rolled and Heat-Treated Structure.	116
96. View of Vought Rolling, Mill and Preheat Furnace.	119
97. Vought Rolling Mill in 4-High Configuration	120
98. Diagram Showing Crowning as a Result of Two Successive Passes on Conventional Mill.	122
99. Diagram Showing Crowning Caused by Bending of the Rolls	122

LIST OF TABLES

TABLE	PAGE
1. Standard Compositions and Strength of INCO 713C and NASA Alloy...	14
2. Modified INCO 713C Alloy Compositions and Strength Properties....	14
3. 1900°F to 2300°F Comparison of Tensile Strengths of INCO 713C, MOD INCO 713C and NASA Alloys.....	22
4. Modified NASA Alloy Composition & Strength.....	30
5. Modifications of INCO 713C Made to Improve Wrought Properties....	30
6. Melt Losses in High Tantalum Content Nickel Base Alloys Melted in Magnesia Crucibles.....	36
7. Magnesium Content of Nickel Alloys in Magnesia Crucibles.....	36
8. Boron Content of Nickel Alloys Melted in Magnesia Crucibles.....	37
9. Refractory Materials Used in Making Crucibles For This Program...	37
10. Nitrogen Content of Nickel Alloy Melted in The Plasma Resistance Furnace.....	37
11. Work Hardening Test on Melt 388 Ta-C Modified INCO 713C.....	41
12. Work Hardening Test on Melt 404 INCO 713C.....	41
13. Work Hardening Test on Melt 425 NASA Alloy.....	42
14. Comparison of Room and Elevated Temperature Strength.....	65
15. Effect of Solution Heat Treatment on Rolled INCO 713C Sheet.....	66
16. Effect of Aging Heat Treatments on Rolled and Solution Heat Treated INCO 713C Sheet.....	66
17. Effect of Solution Heat Treatment on Cast NASA Alloy Sheet.....	75
18. Effect of Solution Heat Treatment on Rolled NASA Alloy Sheet.....	75

LIST OF TABLES (Continued)

TABLE	PAGE
19.	Effect of Solution Heat Treatment and Aging on Rolled NASA Alloy Sheet. 75
20.	Tensile Data on Nickel Alloys 82
21.	Tensile Properties of Rolled and Heat Treated INCO 713C . . . 83
22.	Variation in 1900 ^o F Tensile Strength of INCO 713C with Varying Strain Rate 84
23.	Variation of 1900 ^o F Tensile Strength of NASA Alloy and Various Strain Rates (Melt 418). 84
24.	Oxidation Data. 88
25.	No. 429, NASA Alloy and 713C X-Ray Diffraction Data Run on Scaled Oxide 90
26.	Specific Gravity of No. 429 and NASA Alloy. 98

I. INTRODUCTION

There is a critical need for oxidation resistant metal structures having usefully high tensile strength at temperatures approximating 2000°F. Existing nickel base superalloy sheet materials have adequate oxidation resistance at these temperatures but have relatively low tensile strength properties. Refractory metals have adequate strength at these temperatures but are subject to rapid destruction by oxidation attack. This program has been devoted to solving this problem by developing nickel base sheet materials having improved strength in the 1900-2000°F temperature range.

The primary objective of this program was to develop 15 to 30 mil thick nickel base alloy sheet with short term tensile strength of 50,000 psi at 1900°F, adequate corrosion (oxidation) resistance, and ductility. These target properties represent a substantial improvement over the best commercially available nickel alloy sheet, René 41. René 41 sheet has a tensile strength of 50,000 psi at 1700°F and only about 20,000 psi at 1900°F.

The secondary objective of this program was to demonstrate the feasibility of producing improved nickel alloy sheet in sizes up to 10" x 48". Another purpose of this program was to determine the feasibility of directly rolling sheet from relatively thin cast slabs of high strength superalloys.

The original objectives of this program have been essentially met by two different alloy sheet materials. These two materials are: (1) Inco 713c and (2) NASA TaZ8 alloy sheet. In rolled and heat treated sheet form the Inco 713c sheet has better ductility and better oxidation resistance at 1900°F than the NASA TaZ8 alloy sheet. Both materials have about the same 1900°F tensile strength. Therefore, on balance, the Inco 713c rolled and heat treated sheet must be considered the more desirable engineering material insofar as short time tensile properties are concerned.

Manuscript released by authors November 1962 for publication as an ASD Technical Documentary Report

II. DESCRIPTION OF THE PROGRAM

The program objective was pursued in three ways as follows:

- (a) Improvement by changes in chemical composition.
- (b) Improvement through mechanical working by rolling of the sheet.
- (c) Improvement by heat treatment of the rolled sheet.

The alloys used in this program were formulated by melting in a plasma-resistance furnace. The molten alloy was then cast into hot ceramic molds to directly form cast sheet blanks about 1/8 inch thick. All alloys cast were evaluated in the as cast condition. Alloys having useful properties in the as cast condition were rolled into sheet. This rolled sheet was then evaluated in both the as-rolled and heat-treated conditions.

III. EXPERIMENTAL PROCEDURES AND TESTS

A. Processing Techniques

1. Melting and Casting Techniques

All metal evaluated in this program was melted and cast in the plasma-resistance furnace developed previously at Vought. All castings used in this program were cast in hot ceramic molds. This molding and casting technique permits the casting of relatively large, thin sheets of nickel base superalloys, as well as a wide range of other alloys. Most of the cast sheets used for the program were about 0.1 inch thick and approximately 5" x 8" in size. Larger sized cast sheets approximately 8" x 16" were also made. This larger sheet represents the largest mold which can be made on the existing molding press at Vought, but is not believed to represent any maximum size limitation of the process as a whole. It was subsequently shown that sheets approximately 10" x 16" x .025" can be rolled from the 5" x 8" cast sheet. Efforts have been made to roll thicker cast slabs and slices from ingots in this program. These efforts have been unsuccessful, demonstrating the practical need for casting the starting sheet in thicknesses approximately 0.1 inch. A more detailed description of the melting and casting techniques used is given in Appendix A, Melting and Casting Procedures and Equipment.

2. Rolling Procedures

a. Sub-Contract Rolling - Metals and Controls

Virtually all the rolling work done at Metals and Controls Division, Texas Instruments, Inc., Attleboro, Massachusetts, was done on two rolling mills: (1) a two high mill with 7 inch diameter rolls; and (2) a two high mill with 20 inch diameter rolls. Most of the rolling done was performed at room temperature with periodic intermediate annealing heat treatments at temperatures of from 2000°F to 2150°F.

One test was made rolling the as cast NASA alloy at 1500°F and another test was made rolling the NASA alloy at 1500°F after the sheet had been previously cold rolled and annealed at 2150°F. All cold rolling was performed with reductions in thickness per pass of only one to two thousandths of an inch. All hot rolling was done with reductions per pass of under 0.005 inch, although this was much harder to control. Hardness readings were usually taken on the sheet after

each pass through the mill and the sheet was measured for thickness with a micrometer after each pass. The design and construction of the rolling mills is such that it was not possible to determine the actual applied rolling loads. Most of the specimens rolled were about 2 inches wide by 4 inches long. In addition to rolling at room temperature and at 1500°F, the NASA alloy was rolled at liquid nitrogen temperature (-320°F). Rolling tests were performed on the Ta-C modified Inco 713c (Melt No. 388), the Inco 713c and the NASA alloy compositions to determine the work hardening characteristics of the alloys during cold rolling. The rolled alloys were also subjected to a variety of heat treatments to determine their annealing characteristics and metallographic specimens were taken as appeared advisable.

b. In House Rolling - Chance Vought

Difficulties encountered in rolling these alloys at Metals and Controls resulted in Vought designing and constructing a unique rigid rolling mill in support of this program. Future reference to the "rigid" rolling mill will connote the Vought constructed mill. A detailed description of this rolling mill and the major factors affecting its design are given in Appendix B, The Rigid Rolling Mill. The following procedure was used in rolling sheet on the rigid mill in this program.

1. With the rolling mill in the two high configuration, set the gap between the two rolls so that the cast sheet to be rolled would just fit.
2. Preheat the cast sheet to from 1750°F to 2150°F and immediately put the hot sheet through the rolling mill.
3. Replace the rolled sheet in the preheat furnace and adjust the roll gap downward by a predetermined amount of from 3 to 6 mils. Remove the sheet from the furnace, pass it through the rolls, and replace the sheet in the furnace to reheat. Note the separating force experienced by the mill by reading the strain gauge recorder.
4. Repeat this cycle until the sheet is reduced to from 40 to 60 mils.

5. Convert the rolling mill to the 4 high configuration.
6. Resume hot rolling, still using 3 mils reduction per pass as per the procedure in 3 above. Continue this until a sheet thickness of about 15 mils is reached.

3. Heat Treatment Procedures

The first heat treatment attempted on cast NASA Ta28 alloy was done in air at 2200°F. This resulted in very serious oxidation and scaling of the specimens. All subsequent heat treatment was done under an inert atmosphere or by encapsulating the specimens in stainless steel or Inconel. The encapsulation procedure worked very well provided there were no leaks in the stainless steel or Inconel box and provided the stainless steel box did not scale through. Both leaks in the welds and scaling through of boxes and of welds have given trouble at times with the encapsulation process. Also, it is difficult to control cooling rates in the specimen while it is encapsulated. Therefore, the normal heat treatment procedure is to heat the specimen in an Inconel retort under a low flow rate of argon and with a slight positive pressure of argon maintained at all times in the retort. This procedure has worked very well for all alloys and for all heat treatments used in this program.

B. Evaluation Techniques

1. Room Temperature Tests

The test specimen used for room temperature tests is shown in Fig. 1. Specimens were loaded in a universal test machine with a loading accuracy of 1% of the applied load.

Strain was measured over a one inch gage length with a Baldwin microformer type extensometer complying with ASTM method E 83-57T strain accuracy classification B-2. The load and the strain were plotted autographically on a Baldwin Load-Strain recorder. The specimens were strained at a rate of $0.005 \pm .002$ inches per inch per minute through yield. After yield a strain rate of .05 inches per minute head travel was used. Total elongation by caliper was determined over a one inch gage length.

2. Elevated Temperature Tests

The test specimen configuration is shown in Fig. 2. Specimens were loaded in a universal test machine with a

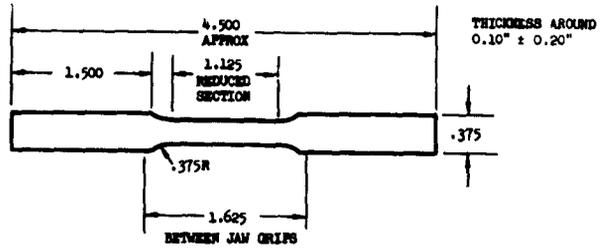
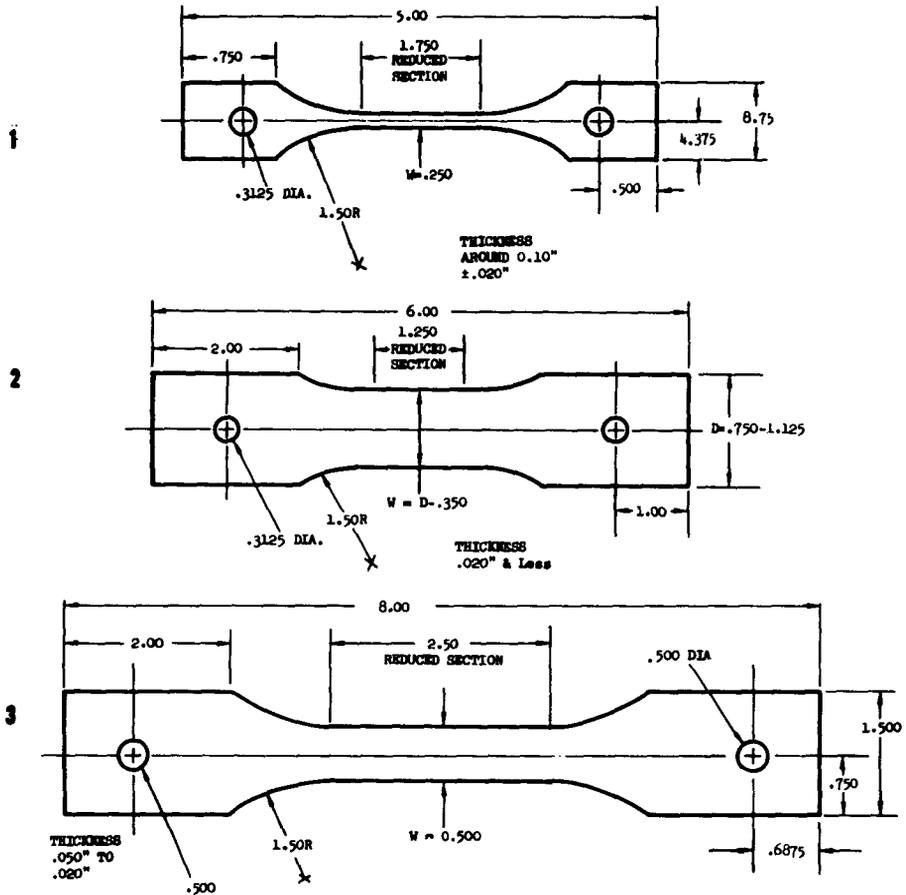


Figure 1. Sheet Tensile Specimens used for Room-Temperature Tests.



1. Tensile Specimen used for All Elevated Temperature Tests on As-Cast Sheet Prior to Melt 428.
2. Tensile Specimen used for Most Elevated Temperature Tests on Rolled Sheet, Performed using Resistance and Radiant Heat.
3. Tensile Specimen used for Elevated Temperature Tests, Performed in a Tube Furnace.

* All Dimensions are Approximate.

Figure 2. Typical Elevated Temperature Tensile Specimens.

loading accuracy of 1% of the applied load. Specimen heating was accomplished with either quartz lamps or resistance heating.

The power source for both methods of heating was an ignitron power controller. For resistance heating the output of the power controller was supplied to a Kirkhof welding transformer which in turn supplied power directly to the test specimen.

The method of attaching the power leads to the test specimen for resistance heating is shown in Fig. 3. For quartz lamp heating, controlled voltage was supplied to two lamp reflector assemblies, each containing five lamps, placed equal distances from each surface of the specimen, Fig. 4.

Specimen temperatures were sensed with 36 gauge chromel-alumel thermocouples spot welded to the test specimen and recorded on a Brown strip chart multipoint recorder.

Specimens were loaded at a rate of 16,000 psi per minute. Total elongation was determined by caliper over a one inch gage length.

A more detailed description and discussion of the elevated temperature testing procedure is given in Appendix C, Elevated Temperature Tensile Procedure.

3. Oxidation Testing

a. Determination of Weight Change after Exposure to Still Air at High Temperature

The specimen of sheet to be oxidation tested is belt sanded down to clean metal on all sides. The specimen is accurately measured to determine total surface area and weighed. The specimen is placed in a sillimanite crucible of known weight and the crucible with the specimen in it is placed in a globar furnace with a still air atmosphere for a predetermined length of time. After exposure to the air in the furnace, the crucible with the specimen in it is removed from the furnace and allowed to cool in air with a lid over the crucible. It is necessary to have a lid over the specimen while it is cooling since the oxide scale formed on some alloys spalls off vigorously upon cooling and is lost. After the specimen has cooled, the net weight of the specimen plus the oxide scale is determined and the weight gain in terms of milligrams per square centimeter of area is

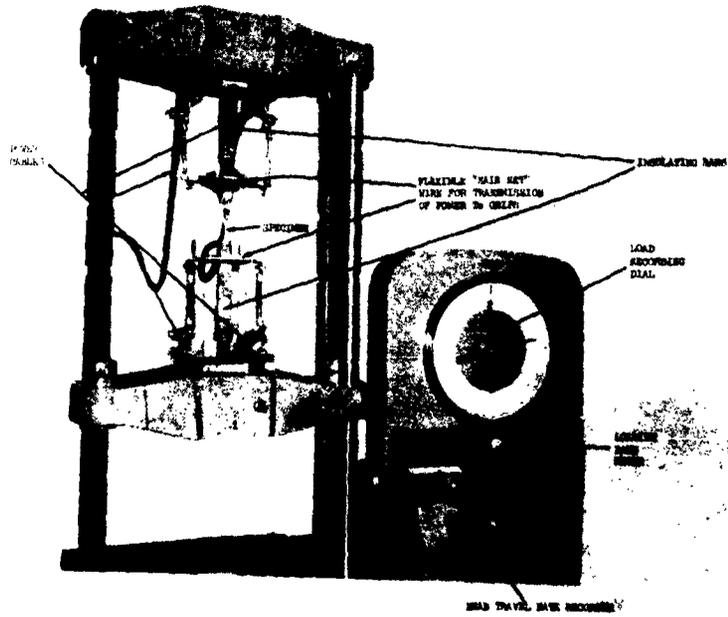


Figure 3. Typical Tensile Test Setup for Resistance Heating.



Figure 4. Relationship of Specimen to Quartz Lamps Used for Radiant Heat Tensile Tests.

calculated. The oxide scale on the specimen is removed by hard rubbing with a towel. The specimen is reweighed and the weight loss in terms of milligrams per square centimeter is determined.

These tests give an indication of the relative oxidation resistance of the metals and alloys tested. They do not take into account important factors such as:

1. Depletion of surface layers of the alloy with consequent weakening of the structure.
2. Intergranular attack.
3. Diffusion of oxygen and nitrogen into the alloy with consequent damage to properties and effect on weight gain measurement.
4. Volatility of some of the oxides which might be formed on these alloys, affecting the weight gain measurements.
5. Effects of air velocity during oxidation.
6. Effects of pressures other than atmospheric.
7. Effects of temperature cycling.

The results of the oxidation tests performed in this program should therefore be taken as an indication of relative values for certain particular conditions. More detailed testing should be related to a particular end use environment.

b. Identification of Oxidation Products by X-ray Diffraction

A General Electric XRD-5 X-ray diffractometer, equipped with a cobalt tube was used to determine the identity of the oxides formed on these alloys. The SPG-2 detector tube was used with the scaler-recorder, and pulse height discrimination was not used. A Fe_2O_3 filter, double layer, was used to eliminate the K-beta radiation.

The oxides were ground up to less than 325 mesh using an agate pestle and mortar. This powder was then pressed into the powder specimen holder and mounted for diffraction patterns. The diffraction pattern covered the region of 6° to 147° theta. The interpretations of the patterns were based on the ASTM File of X-ray Powder Data.

4. Metallographic Examination

a. Optical Metallography

Conventional mounting and polishing techniques were used to prepare metallographic specimens for examination in this program. The following etching solutions were used to etch the specimens for examination on the optical and electron microscopes:

- (1) Conc. HCl & FeCl₃ (etchant #1)
- (2) A mixture of Conc. HCl & H₂SO₄ (etchant #2)
- (3) Conc. HCl & drops of H₂O₂ (30%), (etchant #3)
- (4) NASA electrolytic etch (etchant #4)

4 parts H₂O

4 parts Glycerine

2 parts Conc. HNO₃

1 part Conc. HF

b. Electron Microscope Metallography

The electron microscope replicas used in this report were prepared in accordance with the following technique:

Cellulose acetate was softened with acetone and placed on the polished and etched surface to be replicated, allowed to dry to stiffness, and then carefully removed.

The negative plastic replica produced in this step was placed in a vacuum evaporator and shadowed with chromium at an angle of 30°, the carbon was deposited normal to the replica surface until the desired thickness was obtained.

The double replica was then removed from the evaporator, cut into small squares. The small sections of the double replica were then placed in a dish of acetone until the plastic dissolved, leaving a chromium shadowed carbon positive replica of the original surface. This positive replica was studied on a Norelco Model 100B electron microscope.

IV. EXPERIMENTAL DATA AND DISCUSSION

A. Alloy Improvement by Varying Chemical Composition

1. Choice of Starting Alloy Composition

The objective of this contract was to obtain nickel base alloy sheet of improved elevated temperature tensile properties. At the time of the initiation of this contract in 1961, the nickel base alloys having the best elevated temperature tensile strength were all casting alloys and were not available in sheet form. Two of the best of these casting alloys were selected as starting compositions for this investigation. One alloy selected was Inco 713c. This alloy is among the best of the commercially available superalloys, and has been proven as a casting alloy in many applications over a period of several years. The other alloy selected was the NASA TaZ8 alloy. This alloy had the highest tensile strength at 1900°F of any superalloy on which data was available. This alloy was a new and experimental alloy having great promise, but was not commercially available and had no previous history as an engineering material. The standard compositions and previously reported room temperature and 1900°F tensile strengths of these two starting alloys are given in Table 1.

These two starting alloys were modified in this program by making changes in chemical composition, by rolling, and by heat treatment. All three of these techniques resulted in substantial changes in the metallurgical structure of both alloys. For complete casting data refer to Appendix D.

2. Modification of Starting Alloy Compositions to Improve As-Cast Properties

a. Modification of Inco 713c

The primary strengthening mechanism in Inco 713c is the precipitation of γ prime. This is a complex intermetallic composed of nickel, aluminum, and titanium. The precipitation of titanium and other carbides is also believed to contribute to the good high temperature strength of Inco 713c. Figure 5 shows the structure of Inco 713c γ prime precipitates in cast thin sheet. Figure 6 shows both the γ prime precipitate and the titanium carbide

precipitate occurring in the grain boundary of Inco 713c cast into a relatively thick section. The titanium carbide precipitate has a marked tendency to occur along the grain boundaries in the as cast Inco 713c, and very little carbide precipitate is noted elsewhere in the structure.

In this program efforts were made to modify substantially the structure of the as cast alloy by increasing the amount of gamma prime producing elements and by increasing the amount of carbide producing elements in the alloy. In the first case, aluminum and titanium contents were increased either together or separately. In the second case, carbon content was raised and strong carbide forming elements such as tantalum and tungsten were added simultaneously. Table 2 summarized the various modified Inco 713c compositions investigated. All of these alloys had low ductility at room temperature and all were difficult to machine.

The two melts containing very high percentages of aluminum were extremely brittle, and both proved to be unmachinable because of this brittleness. Inco 713c modified with smaller aluminum additions plus a small titanium addition was far less brittle than the high aluminum alloys, but even this alloy had such poor ductility that several tensile specimens were lost in the machining process. It should be pointed out here that all of the modified Inco 713c alloys were substantially more difficult to machine than Inco 713c, either because of a tendency to break during machining because of low ductility or because of the hardness of the precipitated carbides in the alloys. Later in the program, machining techniques were developed for making tensile bars which could have saved a number of the tensile bars lost during machining at this stage in the program. The loss of a number of specimens during machining accounts largely for the limited amount of tensile data available for these alloys.

Three efforts were made to strengthen Inco 713c by increasing the nickel aluminide precipitate. These efforts are contained in Melt Nos. 430, 431, and 438. Melt Nos. 430 and 431 were so brittle that no usable tensile bars were obtained. Melt 431 was so brittle that the casting broke up into small pieces during vapor blasting. Melt No. 438 was much less brittle; however, only one usable tensile bar was obtained from Melt No. 438 and the rest broke during machining.

Unfortunately, the sole surviving tensile bar was a room temperature bar and was not an elevated temperature test bar. Limitation of available time prevented the making of another melt of this composition, so 1900°F tensile strength was not determined on any of the aluminum and titanium modified Inco 713c alloys. Figures 7, 8 and 9 show the microstructures of these three alloys in the as cast condition. All three structures show a relatively coarse structure which might be improved by heat treatment. However, the observed properties did not appear to warrant this additional investigation.

A total of eleven different usable melts were made of carbide modified Inco 713c. One melt (Melt No. 395) was attempted having a very high tantalum-carbon content, and this melt reacted with the crucible to such an extent that no usable casting resulted. Another melt, No. 403 resulted in cast-sheet having excessive gas porosity which made it unusable for tensile bars. Of the eleven usable melts, Melt No. 434 was so brittle that all tensile test bars broke during machining. All of the remaining melts, with the exception of No. 429, had about the same room temperature ductility as the as cast NASA TaZ8 alloy. The 1900°F tensile strengths of all compositions tested were equal to or superior to that of René 41, but only No. 429 had 1900°F strength meeting program target requirements. The microstructures of twelve alloys cast are given in Figures 10 through 21. All of these microstructures show substantial amounts of precipitated carbides and most show microstructures substantially different from either Inco 713c or NASA TaZ8 alloy.

There is much more variation in 1900°F tensile strength among these alloys than can be explained by the variation in the amounts of carbide precipitate observed in the microstructures. For example, Melt No. 429 has about 50% greater 1900°F tensile strength than Melt No. 435 although the total amount of carbide precipitate in the two alloys appears to be about the same. Furthermore, the only compositional difference between the two alloys is that 429 has more tantalum and more carbon. It is, therefore, evident that it is not only the quantity of carbide precipitated, but also the composition of the carbide that is important. Furthermore, it is interesting to note that alloys containing up to 1.66% carbon still have room temperature and 1900°F ductility about the same as Inco 713c, which

TABLE 1

Standard Compositions and Strength of INCO 713C & MASA ALLOY
(AS CAST)

Alloy	Composition														Tensile Properties					
	Ni	Ta	Ta	Cr	Al	W	Mo	V	Zr	C	Ti	Fe	Si	Mn	S	ROOM TEMPERATURE UTS (psi)	EL (psi)	1800°F UTS (psi)	EL (psi)	
MASA(3)	8.0	---	6.0	6.0	4.0	4.0	4.0	2.5	1.0	.125	---	---	---	---	---	---	134,400	5.2	54,300	9.5
INCO 713c(1,2)	8.0	---	6.0	6.0	4.0	4.0	4.0	2.5	1.0	.125	---	---	---	---	---	---	134,400	5.2	54,300	9.5
INCO 713c(1,2)	8.0	---	6.0	6.0	4.0	4.0	4.0	2.5	1.0	.125	---	---	---	---	---	---	134,400	5.2	54,300	9.5
INCO 713c(1,2)	8.0	---	6.0	6.0	4.0	4.0	4.0	2.5	1.0	.125	---	---	---	---	---	---	134,400	5.2	54,300	9.5
INCO 713c(1,2)	8.0	---	6.0	6.0	4.0	4.0	4.0	2.5	1.0	.125	---	---	---	---	---	---	134,400	5.2	54,300	9.5
INCO 713c(1,2)	8.0	---	6.0	6.0	4.0	4.0	4.0	2.5	1.0	.125	---	---	---	---	---	---	134,400	5.2	54,300	9.5
INCO 713c(1,2)	8.0	---	6.0	6.0	4.0	4.0	4.0	2.5	1.0	.125	---	---	---	---	---	---	134,400	5.2	54,300	9.5
INCO 713c(1,2)	8.0	---	6.0	6.0	4.0	4.0	4.0	2.5	1.0	.125	---	---	---	---	---	---	134,400	5.2	54,300	9.5
INCO 713c(1,2)	8.0	---	6.0	6.0	4.0	4.0	4.0	2.5	1.0	.125	---	---	---	---	---	---	134,400	5.2	54,300	9.5

T A B L E 2

Modified INCO 713C Alloy Compositions and Strength Properties

Melt Composition
(Determined by Chemical Analysis)

MELT No.	ALLOY	Ni	Ta	Cr	Al	W	Mo	Si	Zr	C	Ti	Fe	Cb	MECHANICAL PROPERTIES 1500°F				
														ROOM TEMPERATURE UTS	EL	ROOM TEMPERATURE UTS	EL	
388	Ta-C Mod	63.94	10.87	10.98	5.36	---	3.60	0.19	0.07	0.84	0.89	1.30	1.68	119,300	110,700	2.0	45,400	8.0
390	INCO 713-C	64.08	11.06	10.83	5.59	---	3.64	0.20	0.16	0.29	0.83	1.34	1.72	63,600	---	2.0	40,700	3.0
392	INCO 713-C	67.54	5.77	12.01	5.84	---	3.47	0.19	0.72	1.16	0.35	1.55	0.78	117,900	112,700	2.0	37,000	10.0
428	INCO 713-C	52.76	9.79	9.57	4.99	---	17.27	0.15	0.17	1.66	0.64	1.15	1.38	108,950	---	2.0	28,400	4.5
429	Ta-C-Cr-W	43.87	7.58	17.17	4.16	17.75	3.50	0.16	0.09	2.46	0.52	0.93	1.07	71,600	---	---	53,200	1.5
430	Mod INCO-713C	57.59	0.75	11.37	19.01	---	3.51	0.15	0.10	0.31	2.05	1.35	1.54	---	---	---	---	---
431	INCO 713-C	52.30	0.18	9.81	27.64	---	3.11	0.16	0.12	0.13	0.81	1.27	1.57	---	---	---	---	---
432	INCO 713-C	64.48	0.45	11.09	6.64	---	3.75	0.22	7.21	0.54	1.26	1.41	1.83	91,600	---	2.0	27,100	10.0
433	Zr-C Mod	60.84	0.17	11.12	5.59	---	3.36	0.15	1.29	1.11	11.90	1.37	1.50	---	---	---	24,300	2.0
434	INCO 713-C	60.02	0.13	11.25	5.07	---	14.85	0.17	0.40	2.45	2.46	1.31	1.54	---	---	---	---	---
435	INCO 713-C	50.50	0.16	17.30	4.55	15.83	4.65	0.16	0.32	1.14	1.80	1.07	1.40	82,900	---	1.0	37,000	2.0
436	Cr-W-C Mod	63.29	0.22	14.07	5.83	---	3.89	0.16	0.13	1.15	7.19	1.34	2.45	90,500	---	1.0	39,400	4.5
437	INCO 713-C	60.46	0.83	12.33	5.32	7.76	4.21	0.15	0.21	2.66	1.35	1.22	3.12	116,400	---	1.0	34,400	2.5
438	Mod INCO-713C	66.52	0.15	12.20	9.62	---	3.95	0.19	0.16	0.30	2.68	1.45	1.76	107,400	---	1.0	---	---
438	INCO 713-C	66.52	0.15	12.20	9.62	---	3.95	0.19	0.16	0.30	2.68	1.45	1.76	107,400	---	1.0	---	---



2,250X.



8,000X.

Figure 5. Electron Micrographs of Inco 713c in the As-Cast Condition, Melt 405. Etch No. 1.

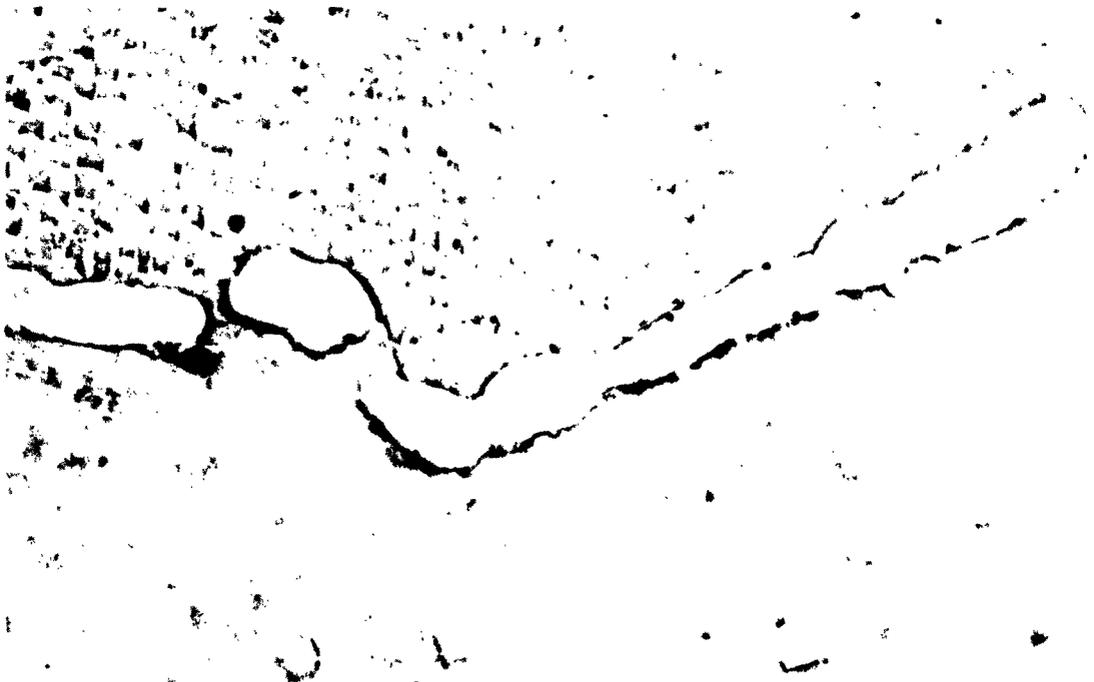
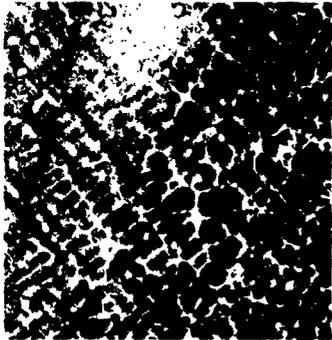
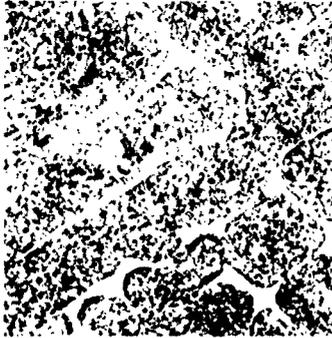


Figure 6. Electron Micrograph Showing TiC Precipitate in a Grain Boundary of As-Cast Inco 713c. 9,000X. Heat No. 56.



100X.

Figure 7. Microstructure of As-Cast Al-Modified Inco 713c (19.01% Al, Melt 430). Etch No. 3.



500X.



100X.

Figure 8. Microstructure of As-Cast Al-Modified Inco 713c (27.64% Al, Melt 431). Etch No. 3.



500X.



100X.

Figure 9. Microstructure of As-Cast Al-Ti-Modified Inco 713c (9.62% Al, 2.68% Ti, Melt 438). Etch No. 4.



500X.



Etch No. 1. 100X.

Figure 10. Microstructure of As-Cast Ta-C Modified (16.26% Ta, 1.08% C) Inco 713c, Melt 388.

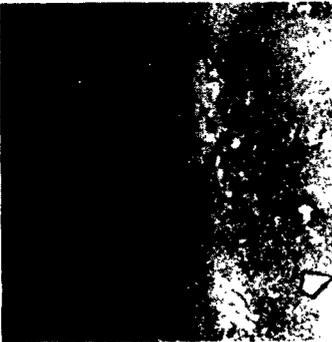


Etch No. 3. 500X.



100X.

Figure 11. Microstructure of As-Cast Ta-C Modified (16.26% Ta, 1.08% C) Inco 713c, Melt 390. Etch No. 3.



500X.

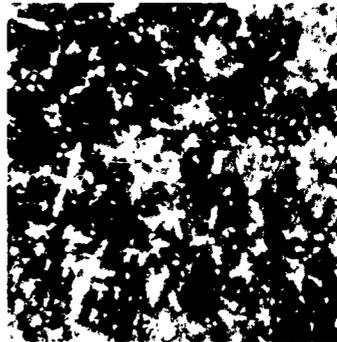


100X.

Figure 12. Microstructure of As-Cast 2(Ta-C) Modified (26.52% Ta, 1.76% C) Inco 713c, Melt 392. Etch No. 3.

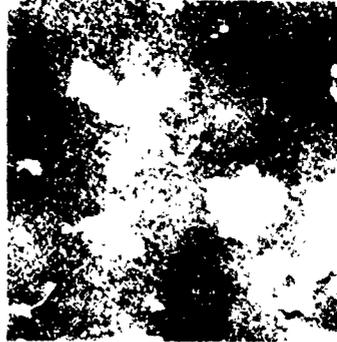


500X.



100X.

Figure 13. Microstructure of As-Cast W-C Modified Inco 713c (13.05% W, 1.45% C, Melt 403). Etch No. 3.



500X.

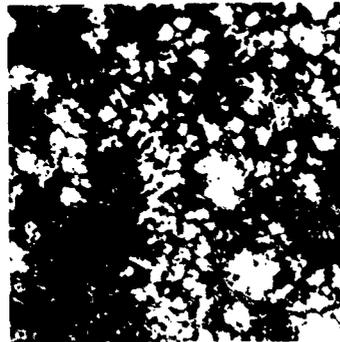


100X.

Figure 14. Microstructure of As-Cast Ta-Mo-C Modified Inco 713c (9.79% Ta, 17.27% Mo, 1.66% C, Melt 428). Etch No. 3.

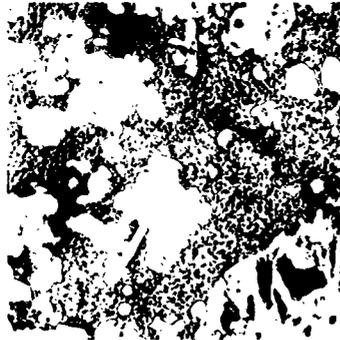


500X.



100X.

Figure 15. Microstructure of As-Cast Ta-Cr-C-W Modified Inco 713c (7.58% Ta, 17.17% Cr, 2.46% C, 17.75% W, Melt 429). Etch No. 3.



500X.

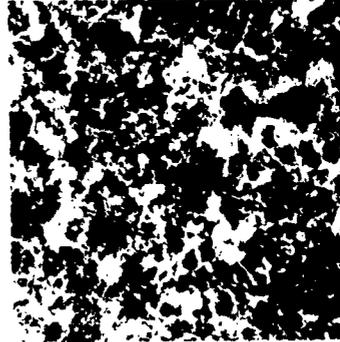


100X.

Figure 16. Microstructure of As-Cast Zr-C Modified Inco 713c (7.21% Zr, 0.54% C, Melt 432). Etch No. 3.



500X.



100X.

Figure 17. Microstructure of As-Cast Ti-C Modified Inco 713c (11.90% Ti, 1.11% C, Melt 433). Etch No. 3.

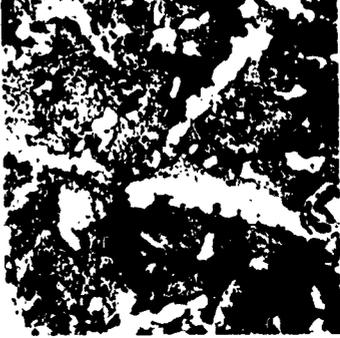


500X.

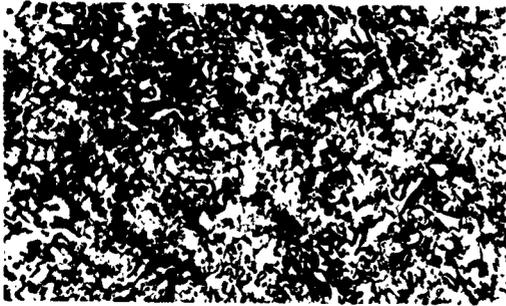


100X.

Figure 18. Microstructure of As-Cast Mo-C Modified Inco 713c (14.85% Mo, 2.45% C, Melt 434). Etch No. 3.



500X.

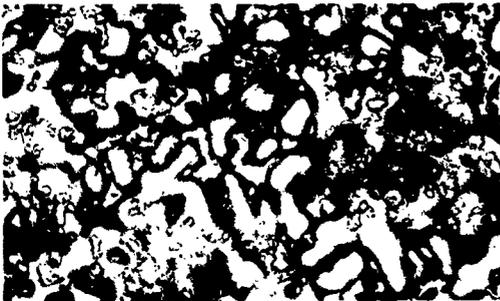


100X.

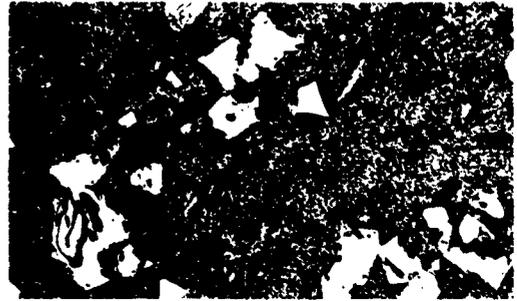


500X.

Figure 19. Microstructure of As-Cast Cr-W-C Modified Inco 713c (17.30% Cr, 15.83% W, 1.14% C, Melt 435). Etch No. 3.



100X.

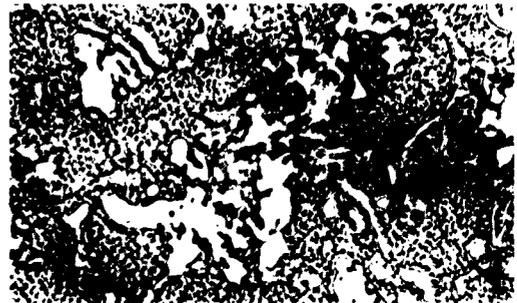


500X.

Figure 20. Microstructure of As-Cast Cr-Ti-C Modified Inco 713c (14.07% Cr, 7.19% Ti, 1.15% C, Melt 436). Etch No. 3.



100X.



500X.

Figure 21. Microstructure of As-Cast Cb-Ta-W-C Modified Inco 713c (3.12% Cb, 0.83% Ta, 7.76% W, 2.66% C, Melt 437). Etch No. 3.

has only about 0.16% carbon. Considering the low atomic weight of carbon and the high molecular weight of the carbides formed, 1.66% carbon corresponds to a high atomic percentage of carbide present. It would therefore appear that an alloy strengthened at high temperature by carbide precipitation is not of necessity brittle.

Accurate determination of the solidus temperature of these complex alloys is very difficult. However, specimens cut from each of these alloys were exposed to a variety of temperatures from 2200°F to 2500°F. The reproducibility of results was poor and so no accurate tabulation is possible. However, it appeared that the alloys showing the lowest strength at 1900°F had the lowest solidus temperature and that Melt No. 429 had a solidus temperature of about 2500°F. This would indicate that further improvement in 1900°F strength will require the development of alloys having higher solidus temperatures. Since equilibrium diagrams are not available for these very complex alloys, further development in this direction must be along rather empirical lines.

Table 3 and Figure 22 show a comparison of the tensile strengths of four alloys at temperatures of 1900°F and above. The data given for the rolled Inco 713c is based on duplicate tensile tests, while the data for the other alloys is all based on single tensile bar results. In spite of the extremely limited number of tests, the available data does appear to fit a logical pattern. Firm conclusions must await the acquisition of further test data.

Table 1 shows that the Inco 713c and NASA TaZ8 alloy compositions are substantially different. Figure 23 shows that as cast Inco 713c and NASA TaZ8 alloys vary substantially in microstructure. Figures 10, 11, and 12 show that Inco 713c with tantalum and carbon additions have as cast microstructures similar to that of the NASA TaZ8 alloy and dissimilar to that of Inco 713c. Figures 24 and 25 show comparisons between the microstructures of cast and heat treated NASA TaZ8 alloy (Melt 385) and the microstructure of cast and heat treated Ta-C modified Inco 713c alloy (Melt 388). Figure 26 shows that relatively large variations in the tantalum and carbon content of Ta-C modified Inco 713c do not greatly alter the microstructure of the alloy. Table 2 shows that the highest 1900°F

strength in Ta-C modified Inco 713c (Melt 388) is achieved when both tantalum and carbon additions are high. Figure 27 shows a comparison of the microstructure of Inco 713c (Melt 405), Ta-C modified Inco 713c (Melt 395), No. 429 alloy (Melt 457) and NASA alloy (Melt 387) in the as cast condition at 25,000 diameters magnification. Each of the four microstructures is appreciably different from each of the others. The differences between the NASA, Inco 713c, and No. 429 alloys were apparent at lower magnifications. The Ta-C modified Inco 713c (Melt 395) is seen to be different from the NASA TaZ8 alloy at this magnification, whereas optical microscope studies had failed to reveal this difference. The difference between the two alloys appears to be more one of precipitate size than a difference in basic structure. In each instance, the NASA TaZ8 alloy is markedly similar in microstructure to the various Ta-C modified Inco 713c alloys. It appears that the characteristic structure of the NASA TaZ8 alloy is caused by the presence of significant amounts of tantalum and carbon. In both the as cast and heat treated conditions the NASA TaZ8 alloy has substantially better 1900°F tensile strength than the Ta-C modified Inco 713c compositions. It is believed that the significant compositional difference here is the presence of 4% tungsten in the NASA TaZ8 alloy. Here, as in the No. 429 alloy, the concurrent presence of tantalum, tungsten, and carbon appear to be necessary to the attainment of maximum 1900°F tensile strength. Additional tests beyond the scope of this program will be required to characterize this relationship, or even to prove beyond doubt that it exists.

The No. 429 alloy has a high carbon content. Metal-crucible reactions are a serious problem when this alloy is melted in either a magnesia or a zirconia crucible. Two efforts were therefore made to melt this alloy in a graphite crucible. Both efforts were unsuccessful. In spite of the high carbon content of the No. 429 composition, the alloy dissolved still more from a graphite crucible. This erodes the crucible, changes the composition of the alloy, and raises the melting point of the alloy to a point where it becomes impossible to pour a casting. It would therefore appear advisable to attempt melting this alloy in a water cooled copper crucible, but this was beyond the scope of this program.

TABLE 3
 1900°F TO 2300°F COMPARISON OF TENSILE STRENGTHS
 OF INCO 713C, MOD INCO 713C & NASA ALLOY

ALLOY	MELT	CONDITION	TEST TEMPERATURE °F	ULTIMATE TENSILE STRENGTH (psi)	% ELONGATION
INCO 713c	405	As cast	1900	46,300	4.0
	404	As cast	1900	50,000	10.0
	550	Rolled	2000	26,900	5.0
	550	Rolled	2100	15,400	16.0
	550	Rolled	2200	5,300	11.0
	392	As cast	1900	37,000	10.0
2 (Ta-C) modified INCO 713c		As cast	2100	8,600	--
Ta-C-Cr-W modified INCO 713c	429	As cast	1900	53,200	1.5
	457	As cast	2100	19,000	7.0
	457	As cast	2200	6,700	3.0
	457	As cast	2300	418	2.5
NASA	393	As cast	1900	46,200	--
	394	As cast	1900	53,200	2.0
	393	As cast	2000	28,900	--
	394	As cast	2100	7,900	3.0

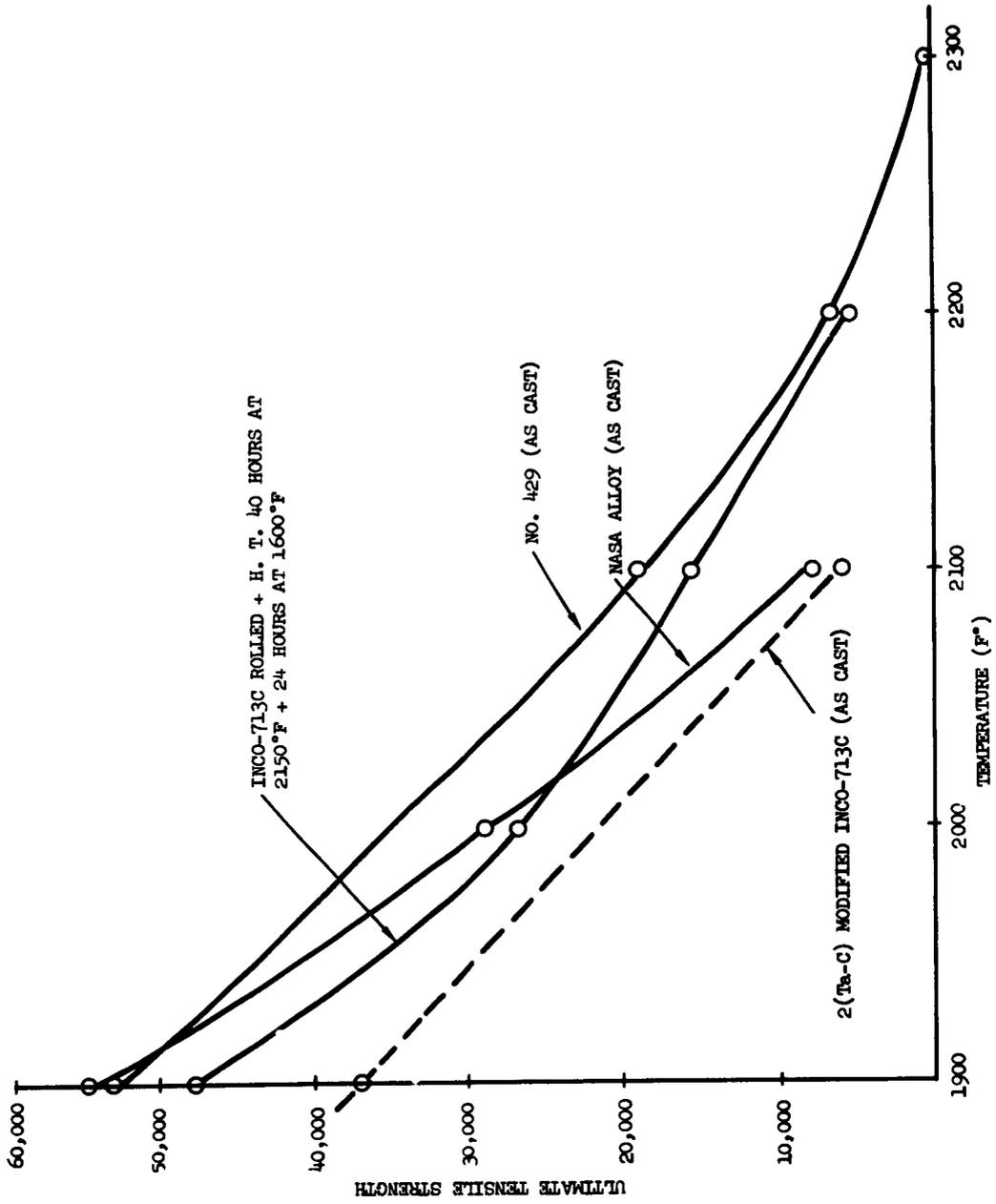


Figure 22. Comparison of Elevated Tensile Strengths of Several Nickel Base Alloys between 1900° and 2300°F.

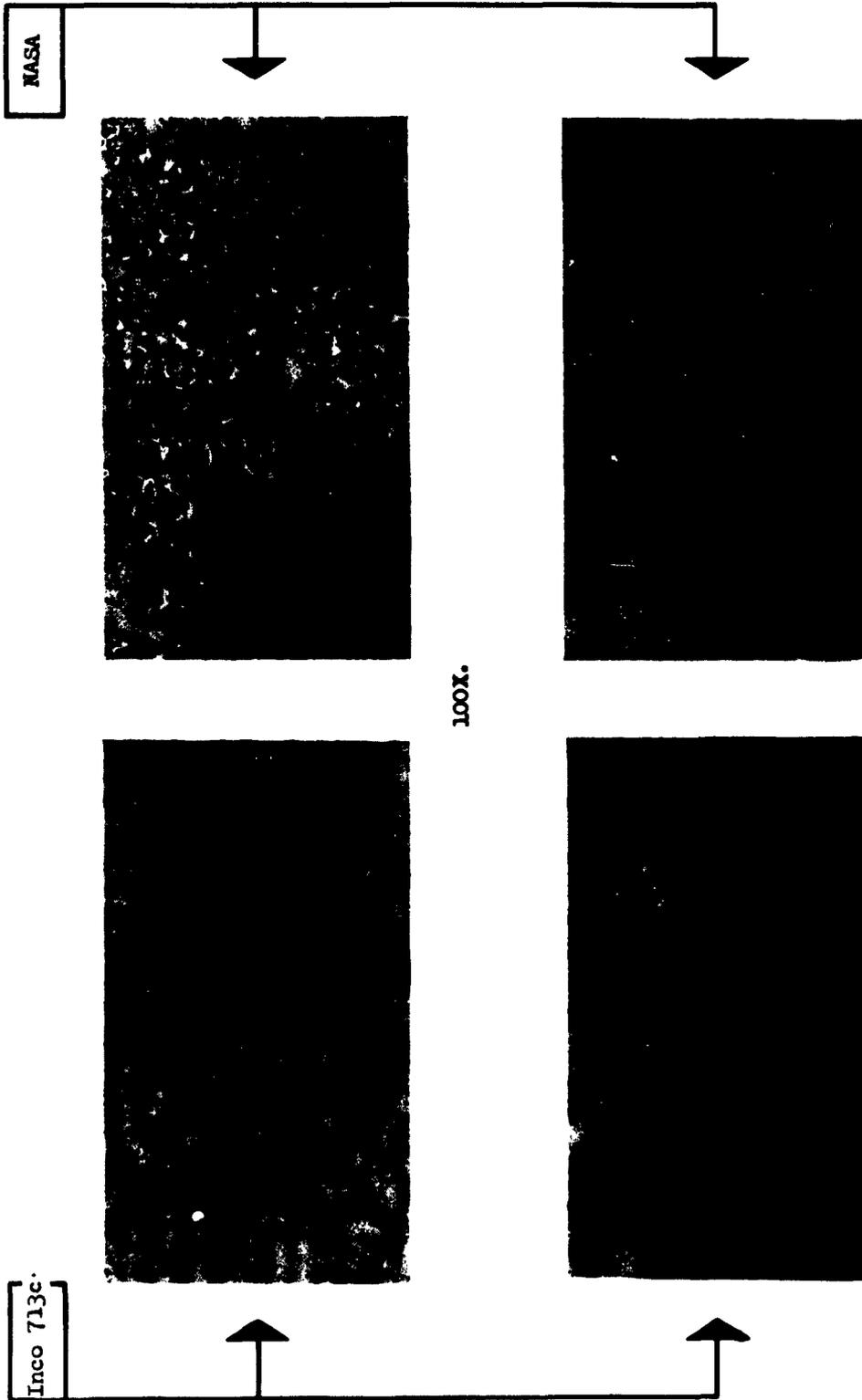


Figure 23. Comparison of Microstructure of As-Cast Inco 713c Sheet with That of As-Cast NASA. Etch No. 3.



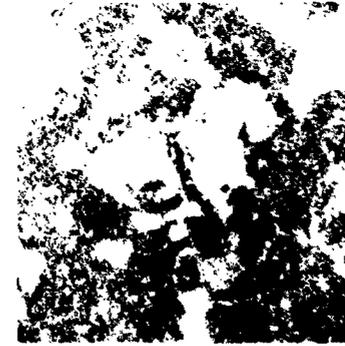
NASA Alloy. 100X.

Ta-C-Modified Inco 713c. 100X.

NASA Alloy. 500X.

Ta-C-Modified Inco 713c. 500X.

Figure 24. Comparison Of Microstructure Of NASA Alloy, Melt 385, With That Of Ta-C-Modified Inco 713c (16.26% Ta, 1.08% C, Melt 388), Both Heat Treated At 2000°F For 16 Hours And Aged At 1500°F For 28 Hours. Etch No. 1.



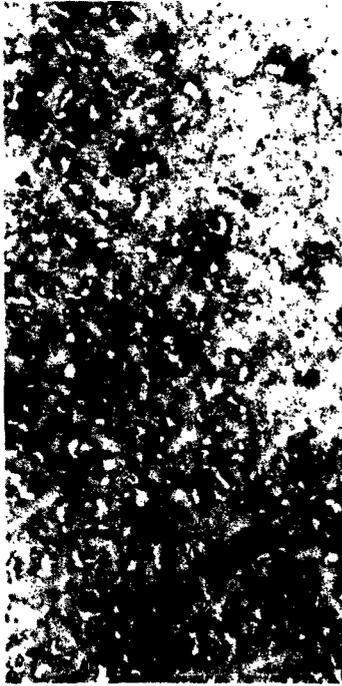
NASA Alloy. 100X.

Ta-C-Modified Inco 713c. 100X.

NASA Alloy. 500X.

Ta-C-Modified Inco 713c 500X.

Figure 25. Comparison Of Microstructure Of NASA Alloy, Melt 385, With That Of Ta-C-Modified Inco 713c (16.26% Ta, 1.08% C, Melt 388), Both Heat Treated At 2200°F For 16 Hours And Aged At 1500°F For 28 Hours. Etch No. 1



Melt 390. (11.06% Ta, 0.29% C)

100X.



Melt 392. (5.77% Ta, 1.16% C)



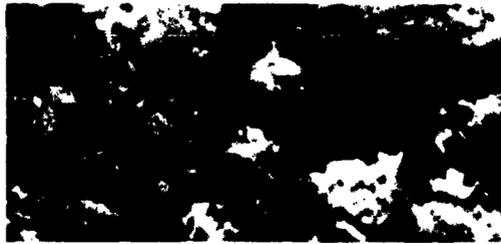
Melt 390.

500X.

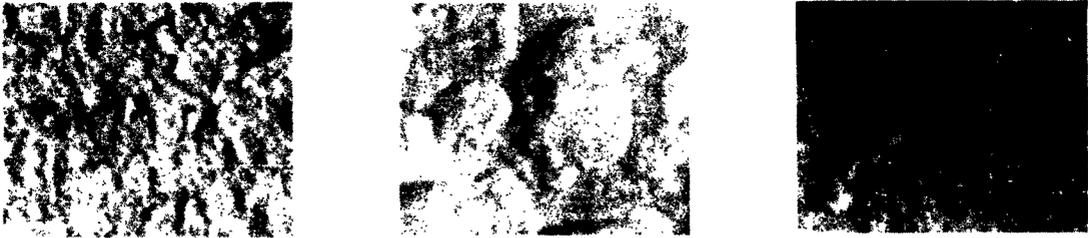


Melt 392.

Figure 26. Effects Of Variations In Tantalum And Carbon Additions
On The Microstructure Of As-Cast Ta-C-Modified Inco
713c. Etch No. 3.



As-Cast Inco 713C.



Tantalum-Modified Inco 713C, As Cast.



No. 429



NASA Alloy

Figure 27. Comparison Of As-Cast Microstructure Of Inco 713c (Melt 405), Ta-C-Modified Inco 713c (40.56% Ta, 2.63%C, Melt 395), No. 429 Alloy (Melt 457) And NASA Alloy (Melt 387) As Shown By Electron Micrographs At 25,000X.

b. Modification of NASA Alloy

The as cast 1900°F tensile strength of the NASA Ta₂8 alloy met the contract target of 50,000 psi tensile strength at 1900°F from the beginning. However, this alloy has relatively low room temperature ductility and relatively poor oxidation resistance. While the modifications in Inco 713c alloy composition were made in an effort to improve 1900°F tensile strength, the modifications in the NASA alloy composition were made in an effort to improve ductility and oxidation resistance.

Table 4 summarizes the results obtained in all melts made of the modified NASA alloy compositions. Melt Nos. 426 and 427 were made in an effort to improve oxidation resistance of the alloy without diminishing high temperature strength. Both modifications resulted in unacceptably low 1900°F tensile strength so further investigation of these modifications was abandoned. Figures 28 and 29 show the as cast microstructure of these two modified NASA alloys. In both alloys the characteristic NASA alloy structure has been substantially changed.

Melt Nos. 529 through 534 represent NASA alloy modifications designed to improve room temperature ductility. The most significant changes in composition in this series were the addition of small amounts of iron. These modifications seemed to result in a small increase in room temperature ductility, but at the cost of a substantial reduction in 1900°F tensile strength. The effects of these minor changes in composition on microstructure appear to be quite great. Figure 30 shows the microstructures of this series of modified NASA alloys. Some of these structures appear to more closely approach the structure of Inco 713c than that of normal NASA alloy. Since these melts were not chemically analyzed, it is possible that higher than usual melt losses of key elements, such as tantalum, have had a controlling effect on the microstructure and properties rather than the relatively minor additions made deliberately. In any case, none of these melt compositions appear promising as improvements over the Ta₂8 NASA alloy.

3. Modification of Inco 713c Alloy to Improve Rolling Properties

The NASA alloy, and the first Inco 713c alloy cast, rolled satisfactorily on the rigid rolling mill. After the first

Inco 713c sheet to be rolled had been evaluated and found to be promising, additional melts of Inco 713c were made. These additional Inco 713c castings were found to be substantially more difficult to roll than the original melts of Inco 713c.

A series of melts was made varying slightly the chemical composition of Inco 713c in an effort to obtain cast sheets having the rollability of the first Inco 713c castings made. (Refer to Table 5). The chemical composition was varied by adding trace elements, by increasing the content of the strengthening elements within specification limits, and by adding small percentages of tungsten. Melt Nos. 547, 548 and 549 represented modifications of Inco 713c by the addition of trace amounts of boron, zirconium, and cerium. None of these modified alloys showed any improvement in rolling characteristics. Melt Nos. 550 and 551 were made increasing the tantalum, aluminum and molybdenum contents to near the maximum allowable under the Inco 713c specification. Melt Nos. 552 and 553 were made with small tungsten additions plus increasing the tantalum and molybdenum contents to near maximum Inco 713c specification limits. Melt No. 550 rolled very well and the tensile properties of the rolled and heat treated sheet were approximately the same as those of the first Inco 713c sheet rolled. The castings from Melt Nos. 551 and 553 rolled as well as Melt 550. The castings from Melt No. 552 rolled well but showed a somewhat greater tendency toward edge cracking than Melt 550.

The most serious problem in rolling Inco 713c has been the tendency on the part of some Inco 713c castings to crack at the grain boundaries during the later stages of hot rolling (see Figure 31). Increasing the content of the strengthening elements (Ta, Mo, and Al) apparently strengthens the grain boundaries to the point where they cease to fail during rolling. This also has the effect of increasing 1900°F tensile strength. The reason for adding small percentages of tungsten to the last two melts was that the first good melts of Inco 713c were made in the same crucible immediately after a high tungsten melt and are thought to have contained small percentages of tungsten as a result. Although these melts were not specifically analyzed for tungsten the analysis did report "significant quantities" of tungsten present.

TABLE 4
Modified NASA Alloy Composition & Strength
(AS CAST)

Melt	Alloy	Composition													Mechanical Properties					
		Ni	Ta	Cr	Al	W	Mo	V	Zr	C	Fe	Si	OTHER	UTS (psi)	Room Temperature		UTS (psi)	SEI	UTS (psi)	SEI
															UTS (psi)	SEI				
426(1)	Cr	59.83	7.15	17.68	4.56	3.50	3.40	1.83	0.73	0.187	0.13				111,600	1.0	20,600	3.0		
427(1)	Cr-Al modified	57.15	7.17	17.15	8.56	3.10	3.18	1.57	0.73	0.25	0.07				116,200	1.0	19,400	11.5		
529(2)		67.2	7.9	7.0	6.0	5.0	4.0	2.5	0.09	0.09	0.02	0.01 B	83,700	105,000	2.0	28,500	1.0			
530(2)		67.0	8.0	8.0	6.0	4.0	4.0	2.5	0.09	0.09	0.02	0.3 Misch	115,500	130,000	1.7	37,300	1.0			
531(2)		65.6	8.1	7.0	5.0	4.0	4.0	2.5	1.0	0.09	0.88	"	116,100	127,000	4.0	37,000	2.0			
532(2)		66.0	8.1	7.0	6.1	4.0	4.0	2.5	1.0	0.09	0.88	"	95,400	95,600	2.0	36,600	1.0			
533(2)		66.5	7.9	6.9	6.0	5.0	4.0	2.5	0.18	0.18	0.88	0.01 B	51,300	54,400	1.0	25,900	1.0			
534(2)		65.4	8.2	7.1	6.1	4.1	4.1	2.6	1.0	0.19	0.88	0.01 B	59,700	59,700	1.7	33,100	1.5			

Note: 1 Composition determined by chemical analysis
2 Composition determined by charge analysis, no chemical analysis available

T A B L E 5

Modifications of INCO 713-C Made to Improve Wrought Properties

MELT INCO 713-C Modifications	Chemical Composition (Calculated from Charge Analysis)											Ti	Fe	Si	B	Ce	Remarks
	Ni	Ta	Cr	Al	W	Mo	Zr	C	Fe	Si	B						
547 High Zr	72.18	2.15	12.70	5.70	---	4.17	0.18	0.15	0.83	1.90	0.14	0.01	0.01	---	---	---	---
548 Ce Mod	72.18	2.15	12.70	5.70	---	4.17	0.08	0.15	0.83	1.90	0.14	0.01	0.10	---	---	---	---
549 High Zr B Mod	72.18	2.15	12.70	5.70	---	4.17	0.18	0.15	0.83	1.97	0.14	0.04	---	---	a.	---	---
550 High Zr-Al Ta-B Mod	70.97	2.60	12.49	6.10	---	5.08	0.18	0.15	0.82	1.94	0.14	0.04	---	---	a.	---	---
551 High Al-Ta B-Mod	70.97	2.60	12.49	6.10	---	5.08	0.06	0.15	0.83	1.94	0.14	0.04	---	---	a.	---	---
552 High Ta-Mo B-W Mod	70.42	2.58	12.39	5.56	0.97	5.04	0.08	0.15	0.81	1.92	0.14	0.04	---	---	a.	---	---
553 High Ta-Mo B-W Mod	69.88	2.57	12.30	5.52	1.93	5.01	0.08	0.15	0.80	1.91	0.14	0.04	---	---	a.	---	---

a. Part of the iron is added with the boron as ferro boron.

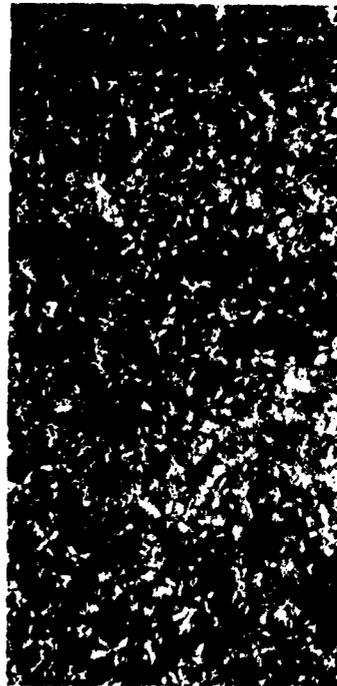


100X.



500X.

Figure 28. Microstructure of As-Cast Cr Modified NASA Alloy Sheet (17.68% Cr, Melt 426). Etch.No. 3.



100X.

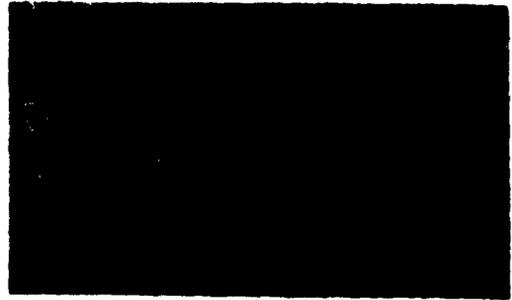


500X.

Figure 29. Microstructure of As-Cast Cr-Al Modified NASA Alloy (17.17% Cr, 8.56% Al, Melt 427). Etch No. 3.



Melt No. 529. NASA alloy, Less 1% Zr,
Plus 3% Misch Metal +.03% FeB + 1% W.



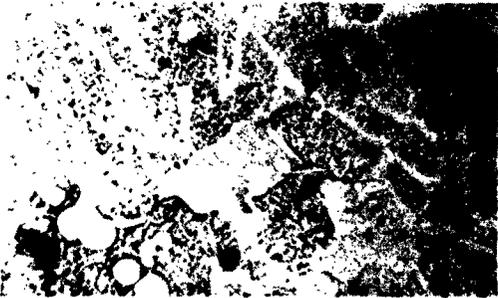
Melt No. 530. NASA alloy, Less 1% Zr,
Plus .3% Misch + .03% FeB.



Melt No. 531. NASA alloy, +.88% Fe +
+ .030% FeB.



Melt No. 532. NASA alloy, +.030% FeB
+ 3% Ce + .88% Fe.



Melt No. 533 NASA alloy, Less 1% Zr
+.09%C+.03 FeB+.88% Fe+.3% Misch+1% W.



Melt No. 534. NASA alloy, +.09% C +
.03 FeB + .88% Fe +.3% Misch.

Figure 30. Modified NASA Alloy As Cast (Melt 529, 530, 531, 532, 533, 534).
Etch No. 4. 500X.



Figure 31. Macro-Etched, Rolled Inco 713c Showing That Cracking During Rolling Tends To Occur At The Grain Boundaries. 1.5X.

4. Effects of Melting Conditions on Alloy Composition

a. Tantalum Melt Losses

Table 6 lists 16 melts containing high percentages of tantalum and giving both charge analyses and chemical analyses showing the tantalum loss incurred for each alloy. All of these melts were made in magnesia crucibles. Although up to 26.52 percent of tantalum was added to these melts, the maximum tantalum retained in the castings was 11.06 percent. As a result of these tests, it would appear that no more than about 11% tantalum can be held in a nickel alloy melt made in a magnesia crucible. Any tantalum added beyond this equilibrium percentage apparently reacts with the magnesia in the crucible and is either vaporized as the oxide or forms a slag.

b. Magnesium Pickup

Table 7 lists four melts which had the magnesium content of the resultant castings determined by chemical analysis. All four melts were made in magnesia crucibles. Since magnesium metal has a boiling point below the melting temperatures of the nickel alloys, any magnesium metal found in the casting represents in effect a dissolved gas in the metal at the time of pouring. This residual magnesium content varied from 50 to 100 parts per million in the melts analyzed. It is interesting to note that the melts that are high in magnesium are also high in nitrogen content. Both magnesium and nitrogen contents would therefore appear to be dependent on the same factor; namely, the efficiency with which the gaseous constituents are removed from the metal by bubbling argon gas through the melt. The contents of both gases appear to be independent of alloy composition. The magnesium content observed does not appear to have any large adverse effect on the properties of the metal. Later melts made in zirconia crucibles would be expected to pick up some zirconium from the crucible. Since these alloys normally are specified to contain some zirconium, this would not be expected to have a detrimental effect. It would be preferable to have a crucible which did not react with the melt, since crucible-metal reactions would not appear to be the ideal way to make alloy additions.

c. Boron Pickup

Table 8 shows that the boron content in four melts of nickel base alloy made in magnesia crucible varies

from 14 ppm to 82 ppm. This boron might conceivably have come from one of the items of melting stock as an impurity. Some of this boron in the modified Inco 713c alloy almost certainly came as one of the alloying constituents of the Inco 713c base material melted. On the other hand, the magnesia crucibles contain from 1500 to 2000 ppm of boron (see Table 9). Some of the boron content probably came from a metal-crucible reaction. Again it would appear desirable to make whatever boron additions are desirable by a means other than crucible-metal reactions.

d. Nitrogen Content

Table 10 shows that the nitrogen content of four nickel alloy melts made in a magnesia crucible and bubbled with argon gas ranges from 37 ppm to 82 ppm. The nitrogen content appears to be independent of alloy composition and in particular independent of chromium content. It would appear to be a function of the effectiveness of the argon bubbling performed in the given melt. The effect of small amounts of nitrogen in the alloy would be expected to be analogous to that of carbon; notably, to decrease ductility and raise strength. In view of the small amount present in these melts, the nitrogen content does not appear likely to be harmful. If larger amounts of nitrogen were present, it might cause gas porosity in the castings. This was not a problem in castings made in this program. Specifically, the argon bubbling procedure used throughout this program was originally developed for the express purpose of eliminating gas porosity in castings and it has been successful in doing this.

B. Alloy Sheet Improvement by Rolling

1. Initial Rolling Trials at Metals and Controls

The original plan of operation for this contract called for all melting and casting operations to be performed at Chance Vought and for all rolling operations to be performed at Metals and Controls. The initial rolling trials were performed at Metals and Controls.

Most of the work at Metals and Controls was devoted to efforts to cold roll Inco 713c, NASA alloy, and Ta-C modified Inco 713c. Initial cold rolling trials showed that all three nickel alloys cracked severely after from 5-20% reduction. This cracking occurred at the edge and also as a multitude of small cracks distributed over the

T A B L E 6

Melt Losses in High Tantalum Content Nickel Base Alloys Melted in Magnesia Crucibles

MELT	ALLOY	Composition											S1	Cb	Mn	Ta	Change in Ta Content in Melting				
		Ni	Cr	Al	W	Mo	V	Zr	C	Ti	Fe	S						B	Mg	N2	
385	charge analysis	NASA	67.58	6.02	7.01	4.01	4.01	2.23	1.00	0.13	---	---	---	---	---	---	---	---	8.02	-0.27	
388	chemical analysis	Ta-C Mod	69.00	5.81	6.03	3.88	4.06	1.80	1.07	0.12	---	---	---	---	0.12	<10ppm	53ppm	0.005	37ppm	7.75	
388	charge analysis	INCO 713-C	60.75	10.71	4.92	---	3.66	---	0.08	1.08	0.82	1.28	---	---	---	---	---	---	16.26	-5.39	
389	chemical analysis	NASA	63.94	10.98	5.99	---	3.60	---	0.07	1.84	0.89	1.30	<10ppm	82ppm	0.010	82ppm	0.19	1.68	0.02	10.87	
390	charge analysis	Ta-C Mod	67.36	6.00	6.36	4.00	4.00	2.51	1.00	0.13	---	---	---	---	---	---	---	---	8.00	+0.34	
390	charge analysis	INCO 713-C	69.14	5.67	6.03	3.69	4.05	2.10	0.49	0.12	---	---	---	---	0.07	<10ppm	14ppm	0.010	60ppm	8.34	
392	chemical analysis	Ta-C Mod	60.75	10.71	4.29	---	3.66	---	0.08	1.08	0.82	1.28	---	---	---	---	---	---	16.26	-5.20	
392	charge analysis	INCO 713-C	64.08	10.83	5.59	---	3.64	---	0.16	0.29	0.83	1.34	<10ppm	73ppm	0.005	41ppm	0.20	1.72	0.02	11.06	
412	chemical analysis	NASA	67.54	12.01	5.84	---	3.47	---	0.72	1.16	0.35	1.55	---	---	---	---	---	0.19	0.78	---	20.75
414	chemical analysis	NASA	67.37	6.00	7.00	4.00	4.00	2.50	1.00	0.125	---	---	---	---	---	---	---	---	5.77	+0.10	
414	charge analysis	NASA	66.68	6.78	7.08	4.00	3.84	1.80	0.90	0.104	---	---	---	---	---	---	---	0.05	---	---	8.00
415	chemical analysis	NASA	67.34	4.97	7.35	4.07	4.12	2.04	1.00	0.116	---	---	---	---	---	---	---	0.06	---	---	8.00
415	charge analysis	NASA	67.32	6.97	6.03	4.00	4.00	2.50	1.00	0.19	---	---	---	---	---	---	---	0.06	---	---	8.18
422	chemical analysis	NASA	67.37	6.25	6.42	4.01	4.01	1.76	1.00	0.14	---	---	---	---	---	---	---	0.06	---	---	8.00
422	charge analysis	NASA	67.32	6.97	6.03	4.00	4.00	2.50	1.00	0.19	---	---	---	---	---	---	---	0.06	---	---	7.91
423	chemical analysis	NASA	67.10	7.26	6.11	3.96	3.96	2.05	0.88	0.17	---	---	---	---	---	---	---	0.03	---	---	8.00
423	charge analysis	NASA	67.32	6.97	6.03	4.00	4.00	2.50	1.00	0.19	---	---	---	---	---	---	---	0.03	---	---	7.96
424	chemical analysis	NASA	67.09	7.07	6.04	3.88	3.80	1.97	0.72	0.16	---	---	---	---	---	---	---	0.13	---	---	8.00
424	charge analysis	NASA	67.32	6.97	6.03	4.00	4.00	2.50	1.00	0.19	---	---	---	---	---	---	---	0.13	---	---	7.95
425	chemical analysis	NASA	67.17	6.98	6.08	3.95	3.94	2.01	0.98	0.17	---	---	---	---	---	---	---	0.04	---	---	8.00
425	charge analysis	NASA	67.32	6.97	6.03	4.00	4.00	2.50	1.00	0.19	---	---	---	---	---	---	---	0.04	---	---	8.00
426	chemical analysis	Cr Mod NASA	66.53	7.87	6.02	3.83	3.93	1.63	0.87	1.73	---	---	---	---	---	---	---	0.04	---	---	8.00
426	charge analysis	Cr Mod NASA	59.34	17.89	5.43	3.53	3.53	2.20	0.88	0.10	---	---	---	---	---	---	---	0.04	---	---	7.77
427	chemical analysis	Cr-Al Mod	59.87	17.68	4.56	3.50	3.40	1.83	0.73	0.187	---	---	---	---	---	---	---	0.13	---	---	7.05
427	charge analysis	Cr-Al Mod	57.11	17.55	8.73	3.39	3.39	2.13	0.85	0.15	---	---	---	---	---	---	---	0.13	---	---	7.15
428	chemical analysis	Ta-C-Mo Mod	57.15	17.15	8.56	3.10	3.18	1.57	0.73	0.25	---	---	---	---	---	---	---	0.07	---	---	6.79
428	charge analysis	INCO 713-C	49.58	8.75	4.05	---	17.70	---	0.07	4.57	0.70	1.05	---	---	---	---	---	0.07	---	---	7.17
429	chemical analysis	Ta-C-Cr-W Mod	52.76	9.57	4.99	---	17.27	---	0.17	1.66	0.64	1.15	---	---	---	---	---	0.15	1.38	---	13.26
429	charge analysis	INCO 713-C	39.90	17.52	3.23	21.00	2.40	---	0.05	2.96	0.50	0.84	---	---	---	---	---	0.15	1.38	---	9.79
429	chemical analysis	INCO 713-C	43.87	17.17	4.16	17.75	3.50	---	0.09	2.46	0.52	0.93	---	---	---	---	---	0.16	1.07	---	10.70
429	charge analysis	INCO 713-C	43.87	17.17	4.16	17.75	3.50	---	0.09	2.46	0.52	0.93	---	---	---	---	---	0.16	1.07	---	7.58

T A B L E 7

Magnesium Content of Nickel Alloys Melted in Magnesia Crucibles

MELT ALLOY	Magnesium Content of Nickel Alloys Melted in Magnesia Crucibles											S1	Cb	Mn	N2	B	Mg	Mn	S1	Cb	Mg	Mn	Content	
	Ni	Ta	Cr	Al	W	Mo	V	Zr	C	Ti	Fe													S
385 NASA	69.00	7.75	5.81	6.03	3.88	4.06	1.80	1.07	0.12	---	---	---	---	---	---	---	---	---	---	---	---	---	---	0.005
388 Ta-C Mod INCO 713C	63.94	10.87	10.98	5.99	3.60	---	3.60	---	0.07	1.84	0.89	1.30	<10ppm	82ppm	0.010	82ppm	0.02	0.12	---	---	---	---	0.005	
389 NASA	69.14	5.67	6.03	3.69	4.05	---	3.66	---	0.08	1.08	0.82	1.28	<10ppm	14ppm	0.010	60ppm	0.02	0.12	---	---	---	---	0.010	
390 Ta-C Mod INCO 713C	64.08	11.05	10.83	5.59	---	3.64	---	0.16	0.29	0.83	1.34	<10ppm	73ppm	0.005	41ppm	0.02	0.20	1.72	---	---	---	---	0.005	

T A B L E 8
Boron Content of Nickel Alloys Melted in Magnesia Crucibles

MELT ALLOY	Composition														Boron Content			
	Ni	Ta	Cr	Al	W	Mo	V	Zr	C	Ti	Fe	S	N ₂	Mg		Mn	Si	Cb
385 NASA	69.00	7.75	5.81	6.03	3.88	4.06	1.80	1.07	0.12	---	0.12	<10ppm	37ppm	0.005	---	0.12	---	53ppm
388 Ta-C Mod	63.94	10.87	10.98	5.36	---	3.60	---	0.07	0.84	0.89	1.30	<10ppm	82ppm	0.010	0.02	0.19	1.68	82ppm
389 NASA	69.14	8.34	5.67	6.03	3.69	4.05	2.10	0.49	0.12	---	0.07	<10ppm	60ppm	0.010	---	0.12	---	14ppm
390 Ta-C Mod	64.08	11.06	10.83	5.59	---	3.64	---	0.16	0.29	0.83	1.34	<10ppm	41ppm	0.005	0.02	0.20	1.72	73ppm

T A B L E 9

Refractory Materials Used in Making Crucibles for This Program

Type of Refractory	Vendor	Vendor Designation	Grain Size	Composition (Typical)	Vendor Recommended Max. USE Temp.
Zirconia	Zirconium Corp. of America	RMF4C	-8 mesh	92.34% ZrO ₂ +HfO ₂ 0.08% SiO ₂ 5.56% CaO 0.08% MgO 0.07% Fe ₂ O ₃ 0.30% Al ₂ O ₃ 1.05% TiO ₂	4400°F
Magnesia	Norton Co.	Magnorite x	-30 mesh and -100 mesh mixed in equal quantities	96.39% MgO 1.91% CaO 1.21% SiO ₂ 0.23% Fe ₂ O ₃ 0.26% Al ₂ O ₃ 1500-2000 ppm Boron	4000°F
Magnesia	Norton Co.	1152	-14 mesh	93% MgO 0.9% Al ₂ O ₃ 3.4% SiO ₂ 2.0% CaO 0.2% Fe ₂ O ₃ 0.1% Alkali 1500-2000 ppm Boron	3400°F

T A B L E 10
Nitrogen Content of Nickel Alloy Melted in the Plasma Resistance Furnace

MELT ALLOY	Ni	Ta	Cr	Al	W	Mo	V	Zr	C	Ti	Fe	S	B	Mg	Mn	Si	Cb	Nitrogen Content
385 NASA	69.00	7.75	5.81	6.03	3.88	4.06	1.80	1.07	0.12	---	0.12	<10ppm	53ppm	0.005	---	0.12	---	37ppm
388 Ta-C Mod	63.94	10.87	10.98	5.36	---	3.60	---	0.07	0.84	0.89	1.30	<10ppm	82ppm	0.010	0.02	0.19	1.68	82ppm
389 NASA	69.14	8.34	5.67	6.03	3.69	4.05	2.10	0.49	0.12	---	0.07	<10ppm	14ppm	0.010	---	0.12	---	60ppm
390 Ta-C Mod	64.08	11.06	10.83	5.59	---	3.64	---	0.16	0.29	0.83	1.34	<10ppm	73ppm	0.005	0.02	0.20	1.72	41ppm

entire surface of the sheet (see Figure 32). Tables 11, 12 and 13 show the work hardening of these three alloys by cold rolling. Figures 33, 34, and 35 show the same work hardening data in chart form. Inco 713c work hardens from Rc 42 to Rc 52 with about 20% reduction; Ta-C modified Inco 713c work hardens from Rc 39 to Rc 51 in about 30% reduction, while the NASA alloy work hardens from Rc 43 to Rc 57 with about 35% reduction. After 20% reduction, the respective increases in hardness are 10 Rc points for Inco 713c, 11 points for Ta-C modified Inco 713c, and 12 points for NASA alloy.

Figures 36, 37, and 38 show comparisons of the microstructures of as cast, as rolled, and rolled and heat treated Inco 713c, Ta-C modified Inco 713c (Melt No. 388), and NASA alloy. It will be noted that the Inco 713c is the only one of the three alloys which changes microstructure to an appreciable extent as a result of cold working and annealing. Inco 713c recrystallizes in five minutes at 2150°F after cold working. Neither of the other alloys shows any recrystallization after even 70 to 100 minutes of heat treatment at the same temperature.

Subsequent efforts to roll NASA alloy sheet by repeated cold rolling with heat treatment of the metal between cold rolling trials were all unsuccessful in reducing the NASA alloy sheet below about 70 mils. In all cases, crowning of the sheet during rolling became a serious problem. Furthermore, the as cast sheet has some variations in thickness. After successive rolling passes these variations in thickness remained, even though the over-all thickness of the sheet has been reduced as much as 20%. Since crowning added greater thickness variations, the as rolled sheet had substantially greater variations in thickness than did the as cast sheet. In some cases, these thickness variations in as rolled sheet were as high as 40 mils on a 70 mil sheet. Metal rolled on the mill with 20 inch diameter rolls showed a greater tendency to crown than the metal rolled on the 7 inch diameter rolls. Crowning in both cases was excessive and resulted in the rapid destruction of the entire sheet by the propagation of edge cracks across the whole width of the sheet (see Figure 32).

NASA had reported somewhat higher ductility in the NASA alloy at liquid nitrogen temperatures (-320°F). Therefore an effort was made at Metals and Controls to roll NASA alloy chilled to liquid nitrogen temperature. This test resulted in the severe cracking at about the same reduction as sheet rolled at room temperature.

NASA alloy cast sheet was encapsulated in 300 series stainless steel and cold rolled at Metals and Controls. This resulted in the complete breakup of the NASA alloy sheet into small pieces (see Figure 39).

Another piece of NASA alloy sheet was encapsulated in 300 series stainless steel and rolled at 1500°F. This also resulted in the complete breakup of the NASA alloy sheet.

The relative location of metal preheating furnaces and rolling mills in the Metals and Controls plant was such that the maximum practical rolling temperature possible was 1500°F. Hot rolling of the NASA alloy at 1500°F was attempted at Metals and Controls, but this was unsuccessful in reducing the sheet beyond the point obtainable by cold rolling.

Because of the continued difficulties in rolling thin sheet on the conventional rolling equipment available at Metals and Controls, together with the limitation on the maximum practical rolling temperature, further rolling trials at Metals and Controls were suspended.

The detailed results of all of the rolling trials performed at Metals and Controls are included in Appendix F which is a summary of all the rolling trials made in this program.

2. Hot Rolling of Sheet on the Rigid Mill

The results of all the rolling trials done on the rigid mill are reported in Appendix F. The results of these tests can be summarized as follows:

- a. The optimum procedure for all nickel base alloys rolled in this program requires a reduction per pass in mill setting of 3 mils per pass. This reduction per pass applies to both the two high and the four high configurations of the rolling mill and to all gauges of materials being rolled down to a minimum of 20 mils. Below 20 mil sheet thickness, smaller reductions are taken.
- b. The roll speed used for most tests was 97.5 surface feet per minute.

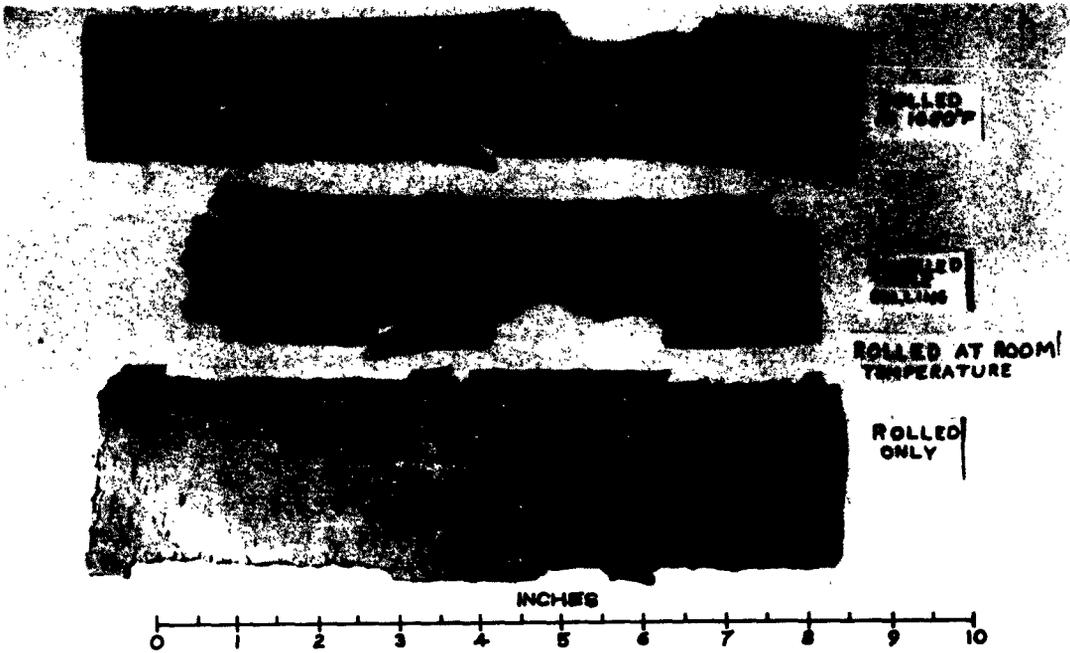


Figure 32. Typical Examples of NASA Alloy Rolled at M and C.

TABLE 11

WORK HARDENING TEST ON MELT 388 Ta-C MOD. INCO 713C

Pass No.	Thickness (.001")	Hardness (R _c)	Pass No.	Thickness (.001")	Hardness (R _c)
0	109 ⁺¹ ₋₃	39 ⁺⁰ ₋₁	7	79 ⁺² ₊₂	51 ⁺¹ ₋₁
1	104 ⁺¹ ₋₂	43 ⁺⁰ ₋₁	8	75 ⁺² ₋₁	51 ⁺⁰ ₋₁
2	99 ⁺² ₋₁	44 ⁺² ₋₁	9	72 ⁺² ₋₂	51 ⁺¹ ₋₂
3	96 ⁺¹ ₊₁	47 ⁺¹ ₋₀	10	69 ⁺² ₋₁	51 ⁺¹ ₋₁
4	92 ⁺² ₋₂	49 ⁺⁰ ₊₀	11	65 ⁺¹ ₋₁	51 ⁺¹ ₋₀
5	87 ⁺² ₋₁	50 ⁺⁰ ₋₀	12	61 ⁺¹ ₋₁	52 ⁺⁰ ₋₀
6	83 ⁺² ₋₁	50 ⁺¹ ₋₁	13	56 ⁺¹ ₋₁	52 ⁺¹ ₋₁

TABLE 12

WORK HARDENING TEST ON MELT 404 INCO 713C

Pass No.	Thickness (.001")	Hardness R _c
0	101 ⁺³ ₋₃	42 ⁺⁰ ₋₀
1	99 ⁺² ₋₂	43 ⁺¹ ₋₀
2	96 ⁺¹ ₋₁	47 ⁺⁰ ₋₁
3	94 ⁺⁰ ₋₂	50 ⁺¹ ₋₁
4	89 ⁺² ₋₁	50 ⁺⁰ ₋₁
5	85 ⁺² ₋₁	51 ⁺¹ ₋₁
6	81 ⁺² ₋₀	51 ⁺¹ ₋₁

TABLE 13

WORK HARDENING TEST ON Melt No. 425 NASA ALLOY

Pass No.	Thickness (.001")	Hardness R_c
0		
1	100 ⁺² ₋₂	43 ⁺⁰ ₋₀
2	97 ⁺¹ ₋₁	47 ⁺¹ _{-1.5}
3	96 ⁺⁰ ₋₁	48 ⁺² ₋₁
4	94 ⁺⁰ ₋₁	50 ⁺¹ _{-.5}
5	92 ⁺¹ ₋₀	50 ⁺¹ ₋₁
6	91 ⁺¹ ₋₁	51 ⁺² ₋₀
7	89 ⁺¹ ₋₀	52 ⁺² ₋₁
8	87 ⁺⁰ ₋₁	54 ⁺² ₋₁
9	85 ⁺⁰ ₋₁	54 ⁺³ _{-1.5}
10	82 ⁺¹ ₋₁	54 ⁺¹ _{-.5}
11	80 ⁺¹ ₋₁	55 ⁺² ₋₁
12	76 ⁺¹ ₋₁	55 ⁺¹ ₋₂
13	74 ⁺¹ ₋₂	56 ⁺¹ ₋₁
14	70 ⁺¹ ₋₂	55 ⁺¹ ₋₂
15	65 ⁺¹ ₋₁	57 ⁺² ₋₀
16	62 ⁺¹ ₋₁	57 ^{+1.5} ₋₁
17	58 ⁺¹ ₋₁	58 ^{+2.5} ₋₁
18	55 ⁺⁰ ₋₁	57 ^{+2.5} ₋₁

Figure 33. Work Hardening Of Inco 713c,
Heat No. 404 During Cold Rolling.

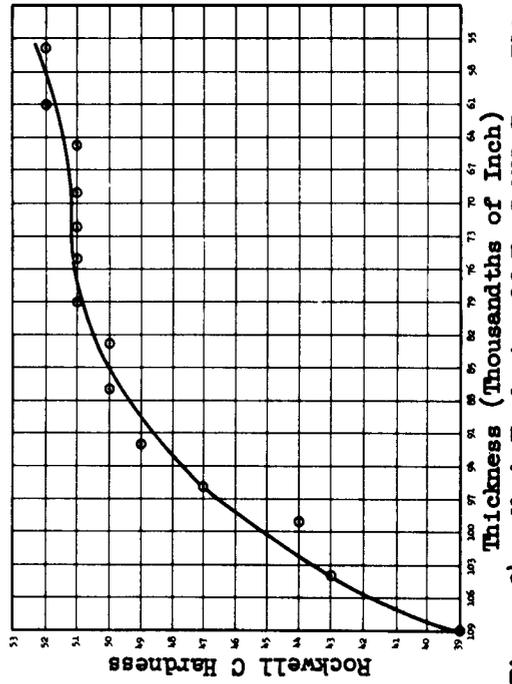
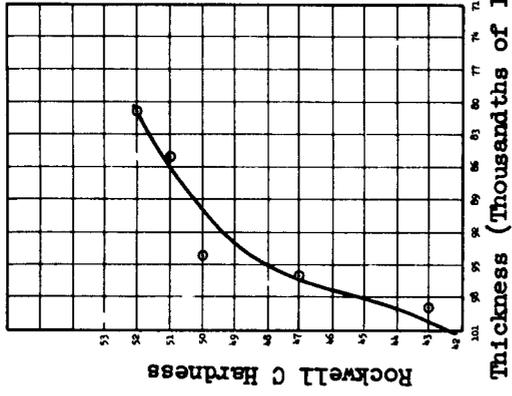


Figure 34. Work Hardening Of Ta-C MOD Inco 713c,
Heat No. 388 During Cold Rolling.

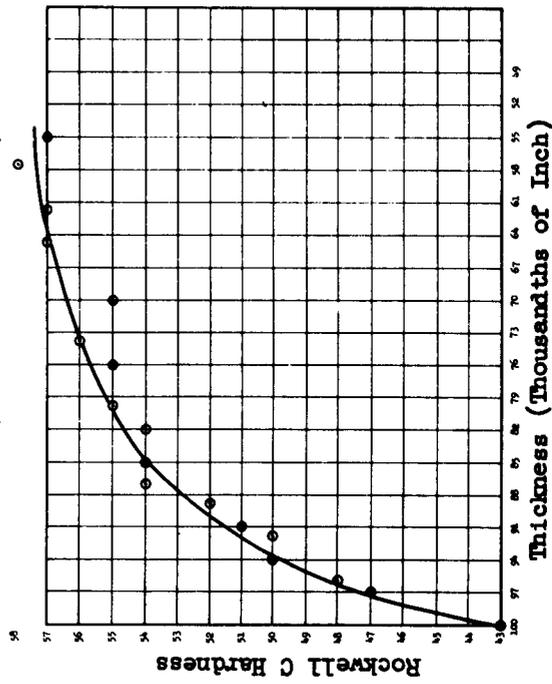
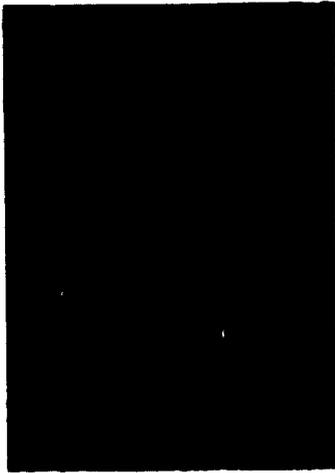


Figure 35. Work Hardening Of NASA Alloy, Heat
No. 425 During Cold Rolling.



As Cast



Cold Rolled 20%, Annealed for 5 Minutes
in Air at 2150°F, Water Quenched.



Same as Photo at Left, Except
Annealed for 70 Minutes.

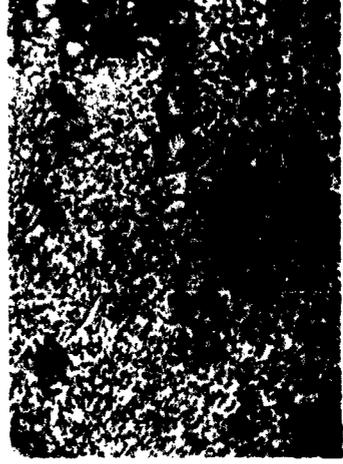
Figure 36. Effects of Cold Rolling and Annealing on the Microstructure
of Cast Inco 713c, Melt 404. Etch No. 3. 500X.



As Cast.

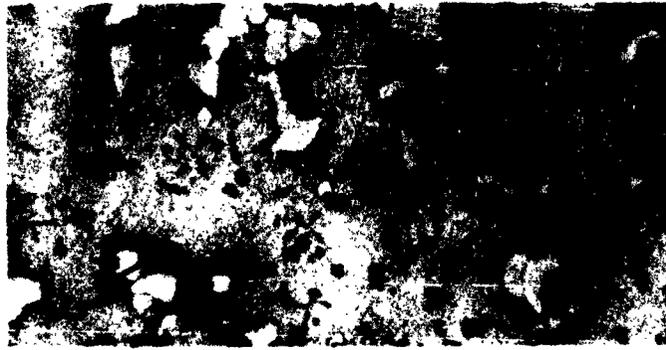


As Rolled + 5 Minutes Anneal.



As Rolled + 70 Minutes Anneal.

Figure 37. Effects of Cold Rolling (20%) and Annealing at 2000°F on
the Microstructure of Fe-C Modified Inco 713c, Melt 388.
Etch No. 3. 500X.



As Cast.



Cold Rolled 14%, Annealed 10 Minutes in Air at 2150° F, Water Quenched.



Cold Rolled 14%, Annealed 100 Minutes in air at 2150° F,
Water Quenched.

Figure 38. Effects of Cold Rolling and Annealing on the Microstructure
of NASA Alloy, Melt 419. Etch No. 3. 500X.

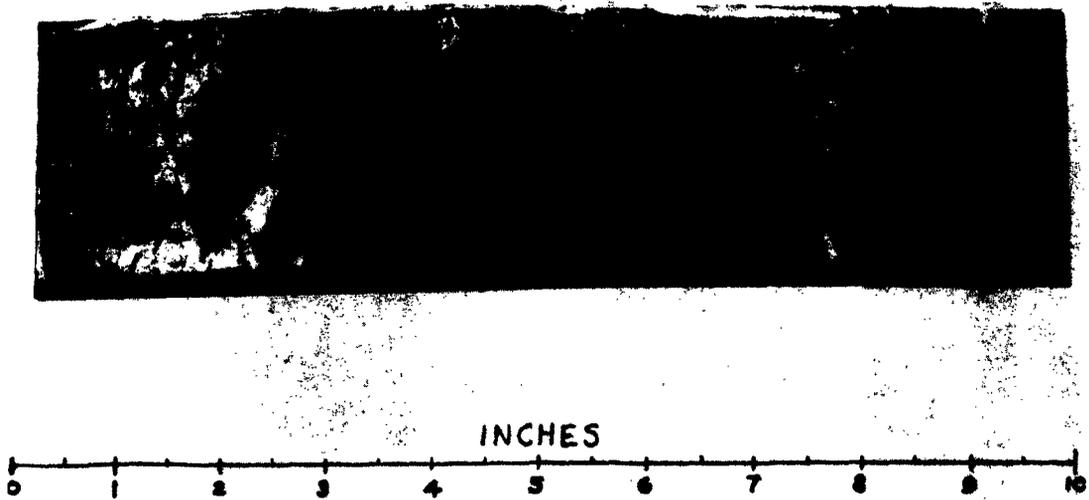


Figure 39. NASA Alloy Rolled in Stainless Steel

- c. Optimum rolling temperatures for each alloy rolled are as follows:

Inco 713c	1950°F
NASA	1900°F
Ta-C modified Inco 713c (Melt No. 388)	2000°F
No. 429 Alloy	2100°F

- d. The two high mill configurations (6" diameter working rolls) should be used down to 40 mil sheet thickness.
- e. The four high configurations (1 1/2" diameter working rolls) should be used from 40 mil sheet thickness on down to foil gauges.
- f. The as cast sheet should not vary in thickness more than about 20 mils. The less the variation, the less likely it is that cracks will appear in the sheet during rolling.
- g. Fine as cast grain size metal rolls better than coarse grain metal.
- h. Inco 713c rolls better and with much less tendency to crack if the percentages of strengthening elements are kept on the high side of the specification, or if small percentages of tungsten are added.
- i. Shrinkage porosity which does not open to the outside will be sealed up during rolling, but shrinkage porosity of any kind will contribute to cracking of the sheet during rolling if other conditions are adverse.
- j. Surface pits and defects will not initiate cracking or failure during rolling but they may weaken the sheet during testing.
- k. Avoidance of crowning or curling of the sheet during rolling is vital if serious cracking is to be avoided.
- l. Prolonged solution heat treatment of the sheet either before rolling or after a few rolling passes causes rapid failure of the sheet by cracking in almost all instances. The reason for this is not directly apparent.

- m. Proper setting of the rolls prior to making the initial pass is important. Parallelism of the roll axes of rotation must be maintained within .001" to .003" at all times.

Refer to Figures 40 through 42 for typical examples of sheet rolled at Vought. Figure 43 shows the microstructure of hot rolled Inco 713c, NASA alloy and No. 429 alloy, all reduced 90%. Inco 713c exhibits a fine grained, typically hot rolled structure. NASA alloy has a heavily fibered structure typical of a cold rolled metal, even though it was rolled at 1825°F. The No. 429 alloy shows no evidence of either recrystallization or of cold rolling in its structure, even though it had been reduced in thickness 90%. Figure 44 shows the microstructure of Inco 713c foil cold rolled from hot rolled sheet. Figure 45 shows the change in microstructure in NASA alloy with various reductions in thickness and with rolling at several temperatures. NASA alloy shows no appreciable change in microstructure in comparing as cast with 20% reduction. Further reductions to 50% and 68% at 1750°F and 2000°F show typically cold rolled structures with fibering becoming increasingly evident. Hot rolling at 2250°F to 68% reduction still shows considerable evidence of fibering, but it also shows some tendency toward relief and recrystallization. If higher rolling temperatures could be achieved, true hot rolling might occur, but this is prevented by the fact that the solidus temperature is only a little above 2250°F. Figure 46 shows a comparison of the structures of as cast No. 429 alloy with the same alloy in the hot rolled condition. Some of the carbide precipitates appear to be broken in the rolled structure but there is no evidence of fibering or of true recrystallization. Figure 47 shows a comparison of as cast and hot rolled Ta-C modified Inco 713c (Melt No. 388) microstructures. The carbide precipitate is apparently broken up into elongated groups of precipitates by rolling. There is no evidence of recrystallization in the rolled structure. Figures 48 and 49 show electron micrograph studies of the structures of hot rolled No. 429 and NASA alloys. The No. 429 alloy has a great deal more areas of massive precipitate than the NASA alloy although both have areas of massive precipitate. The structure in the No. 429 alloy is very difficult to resolve. It is clear from the microstructures that both alloys employ precipitation hardening as one of their strengthening mechanisms.

3. Analysis of Reasons for Failure of Rolling Trials on Conventional Mills

Efforts were made to roll NASA alloy, Inco 713c, and Ta-C modified Inco 713c on conventional rolling mills both at Metals and Controls and elsewhere. None of these efforts

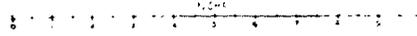
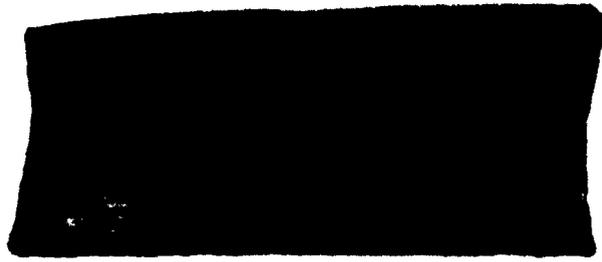


Figure 40. Rolled Inco 713c Sheet (Fine Grain).



Figure 41. Rolled Inco 713c Sheet (Coarse Grain).

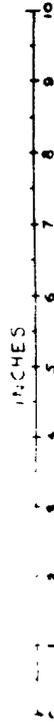


Figure 42. Inco 713c, Rolled at Vought .025" Thick.



713C, Melt 404, Reduced From As-Cast To .015", 90% Reduction At ,2000°F, As Rolled.



NASA Alloy, Melt 462, After Tensile Test At .050", Rolled To .015" (Total Reduction 90%) At 1825°F, As Rolled



No. 429 Alloy, Melt 457, After Tensile Test At .050", Reduced To .015" (Total Reduction 90%) At 2100°F, As Rolled.

Figure 43. Comparison of the Effects of 90% Reduction In Thickness By Hot Rolling On Inco 713c; NASA, And No. 429 Alloys. Etch No. 4. 500X.



NASA Alloy, Melt 462,
Rolled 20% At 1500°F
As Rolled.



NASA Alloy, Melt 462,
Rolled 50% At 1750°F
As Rolled.



NASA Alloy, Melt 462,
Rolled 68% At 2000°F
As Rolled.

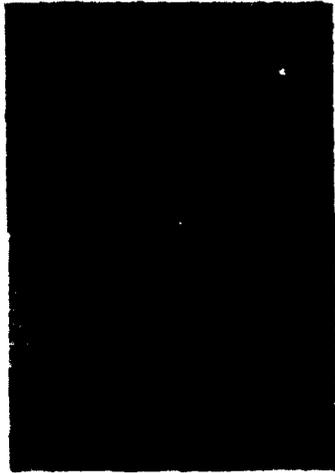


NASA Alloy, Melt 462,
Hot Rolled At 2250°F
68% As Rolled.

Figure 45. NASA Alloy, Hot Rolled To Various Reductions In Thickness At Several Temperatures. Etch No. 4. 500X.



As Cast.



As Rolled 50% At 2200°F.



As Rolled 60% At 2100°F.

Figure 46. Comparison Of Microstructures Of As-Cast And As-Rolled No. 429 Alloy. Etch No. 4. 500X.



As Cast



Rolled at 1800°F, 63%

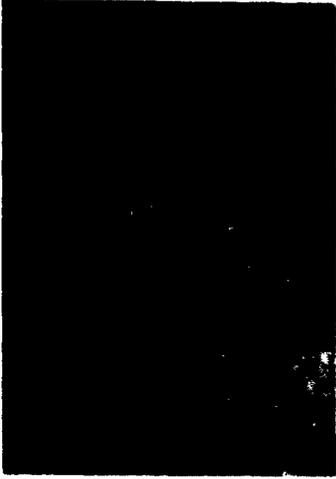
Figure 47. Comparison of Microstructures of As-Cast and As-Rolled Ta-C Modified (16.26% Ta, 1.08% C) Inco 713c, Melt 388. Etch No. 1. 500X.



3,000X.



50,000X.

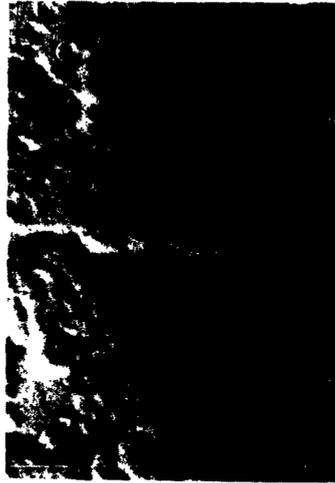


50,000 X. (Different Area)

Figure 48. Electron Micrographs, No. 429 Alloy, as Rolled, 60% at 2150°F, Melt 457. Etch No. 1.



3,000X.



25,000X.



50,000X.

Figure 49. NASA Alloy, Rolled 60% at 2000°F, Melt 462.

succeeded in rolling these alloys into sheet having any substantial, uniform reduction in thickness from the as cast sheet. On the other hand, procedures were developed on the rigid rolling mill which permitted the hot rolling of 15 to 30 mil sheet from NASA alloy, Inco 713c, Ta-C modified Inco 713c, and No. 429 alloy. Experimental trials on the rigid mill indicate that hot rolled Inco 713c sheet could be cold rolled down to 3.5 mil foil in widths up to three inches. This foil was cold rolled from 20 mil sheet previously hot rolled on the rigid mill. Figure 50 shows four pieces of Inco 713c foil rolled on this mill. Once hot rolling procedures had been developed at Vought, an effort was made to use these procedures to hot roll NASA alloy sheet on a conventional rolling mill having the same size working rolls. Under these conditions, the conventional rolling mill produced a severely cracked sheet with 10 mil crown (see Figure 51).

It is believed that the primary reason for the failure of rolling trials on conventional mills is the presence of excessive crowning of the rolled material.

The ability of the rigid mill to roll alloy sheet which was not rollable on the conventional rolling mills used in this program is believed due to the unique design features of the Vought mill which minimize sheet crowning. This mill is designed to maintain a greater uniformity of gap between the working rolls with variations in rolling load than is possible with conventional design. A more detailed discussion of the design of this mill is given in Appendix B.

Crowning of the sheet during rolling is a direct result of variations in the gap between the working rolls caused by the separating force generated by the working of the metal sheet passing between the rolls. Crowning is the direct cause of sheet failure during rolling by the deep penetration of edge cracks. In rolling relatively brittle materials, a slight degree of edge cracking always occurs. This is caused by the fact that in the center of the sheet the total effect of diminishing the thickness of the sheet goes into elongating the sheet. At the very edge of the sheet, the effect of diminishing the thickness of the sheet by rolling goes partly into the elongation of the sheet and partly into displacement of the metal along the axis of the roll, thus widening the sheet somewhat. Since this is true, the sheet at its very edge does not elongate as much as the same sheet a little way in from the edges. Unless the metal has substantial tensile ductility, this soon results in small edge cracks appearing to relieve the tensile stress caused by this greater elongation of the main portion of the sheet. (See Figure 52). In the absence of crowning, these

edge cracks will normally penetrate about an eighth of an inch and then stop. If crowning is taking place, first one side of the sheet will be placed under abnormally high tensile stress by the elongation of the other side of the sheet during rolling; and then the process will be reversed as the cocking of the rolls reverses itself. As a result of these abnormally high tensile stresses, the edge cracks will rapidly penetrate to the center of the sheet, thus destroying the sheet (see Figure 53).

C. Alloy Sheet Improvement by Heat Treatment

1. Inco 713c

a. Recrystallization

Rolled Inco 713c sheet having as little as 20% reduction recrystallizes completely at 2150°F in five minutes. Figure 36, page 44, shows such a recrystallized structure. Figure 43, page 51, shows the microstructure of Inco 713c reduced 90% by rolling at 2000°F. This structure is also completely recrystallized. It is, therefore, evident that rolled Inco 713c sheet will rapidly recrystallize at 2000°F although a specific minimum time cannot be stated on the basis of available data.

b. Solution Heat Treatment

The object of solution heat treatment is not to recrystallize the alloy, although this may be an incidental result of the heat treatment. The object of solution heat treatment is to dissolve the carbide and other precipitates in the matrix alloy and redistribute these precipitate constituents by diffusion while they are in solution. After this redistribution has taken place, the alloy is cooled rapidly to retain in solution the precipitate forming constituents as an undercooled non-equilibrium structure. In this case, subsequent heating to intermediate temperatures will produce "aging" reactions where the precipitate will be formed under controlled conditions. If the solution heat treated alloy is permitted to cool more slowly, the precipitate forming constituents will precipitate out during cooling in less controlled manner. In general, it would be considered more desirable metallurgically to rapidly quench the solution heat treated metal and then form the precipitated phases through controlled "aging" heat treatment. In practice, very rapid quenching of metal structures will result in greater warpage of the structure than

slower cooling and may result in cracking of the metal due to thermal shock. As a result, the most useful cooling rate from solution heat treatment temperature is usually a compromise between the most desirable metallurgically and that most desirable for minimizing warpage and cracking. In this program, water quenching of NASA alloy from solution heat treatment temperature was found to frequently cause cracking of the alloy. Therefore, most of the quenching of NASA alloy in this program was done by air cooling. Air cooling was also used for the other alloys investigated throughout most of the program in order to preclude cracking problems, and in order to maintain a uniform procedure for all alloys.

Both macro and micro segregation of substantial degree occur of necessity in cast alloys of complex composition. These reasons are explained in detail in Appendix A. The only method possible for reducing segregation in a cast alloy is the use of prolonged solution heat treatments. The length of time at solution temperature required for an effective heat treatment depends on the diffusion rates of the elements involved and on the distance which the atoms in question must move by diffusion to achieve the desired end. In the instance of interstitial alloying elements, such as carbon, nitrogen and hydrogen, diffusion rates are quite high and relatively short solution heat treatment times are required to homogenize even badly segregated alloys. Where the segregated alloy constituents are substitutional elements of large atomic size, such as tungsten, tantalum, molybdenum, etc., diffusion rates are very low at the highest temperatures possible without melting the alloy. Hence, these alloys require long solution heat treatment times in order to homogenize the alloy. In normal alloys the grain boundaries form a continuous phase and the grains themselves a discontinuous phase. Therefore, the strength of the alloy structure as a whole is largely affected by the strength of the grain boundary metal composition. In a normal alloy casting, the grain boundary metal invariably has a lower melting point than the center of the grain. In castings of complex alloys, the grain boundary metal may have a melting point appreciably lower than that indicated by an equilibrium diagram for the alloy as a whole. It is therefore particularly important to minimize such grain boundary segregation in complex alloys designed for use near their melting points. This applies especially to the nickel base superalloys investigated in this program.

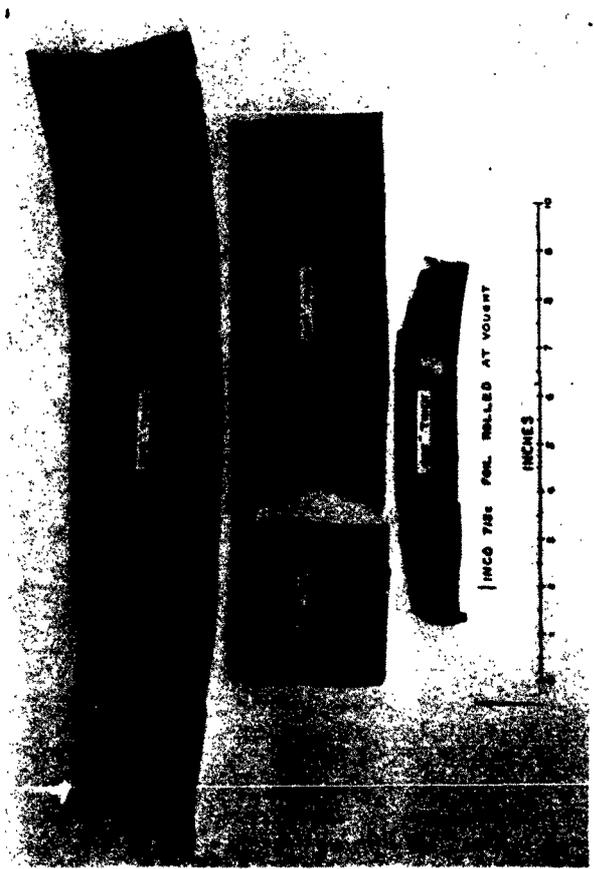
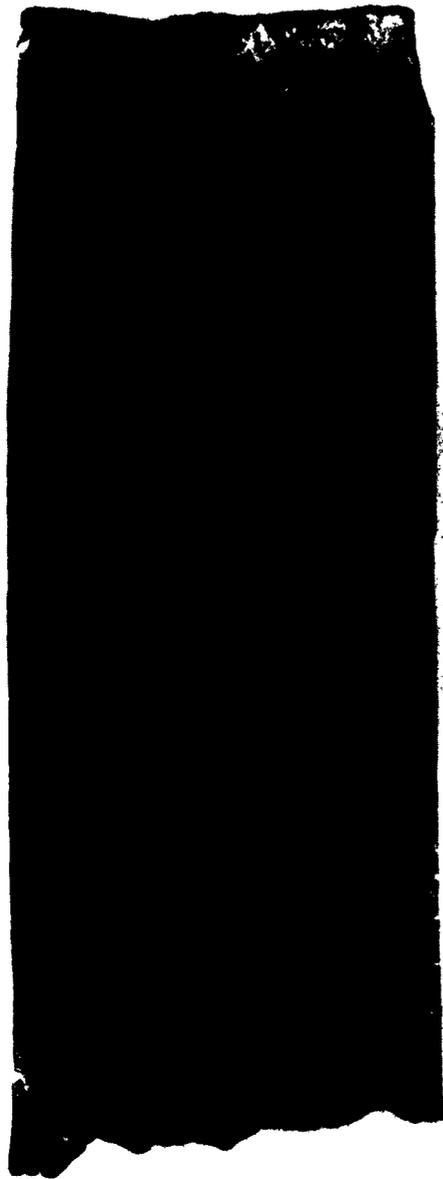


Figure 50. Inco 713c Foil Rolled at Vought. Melt No. 550.



VOUGHT
MILL



CONVENTIONAL
MILL

Figure 51. NASA Alloy Rolled At 1850°F.

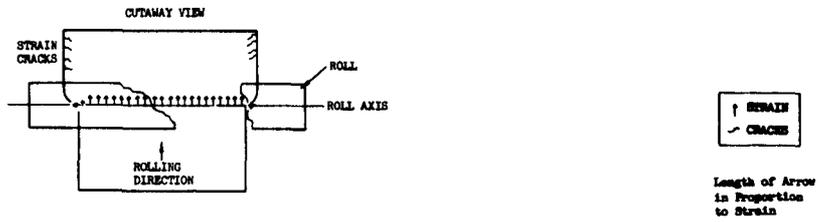


Figure 52. Diagram Showing Strain Distribution in Sheet Being Rolled on a Rigid Mill with Resultant Edge Cracks.

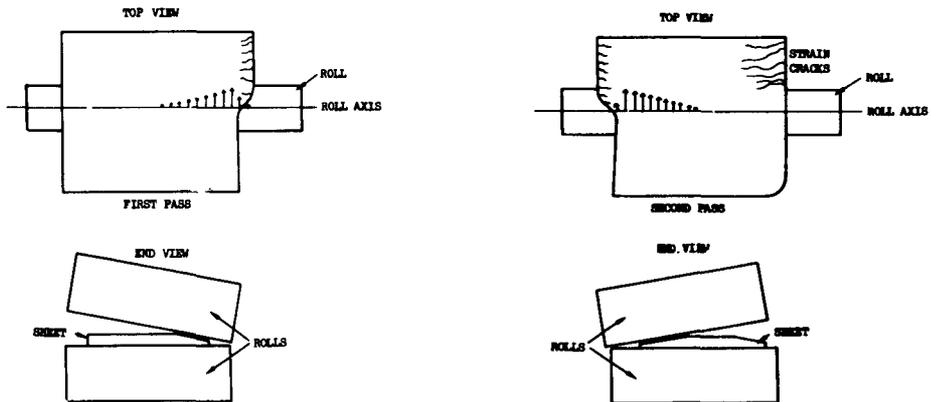


Figure 53. Diagram Showing Resultant Strain and Crack Distribution During Two Successive Passes on Conventional Mill.

Solution heat treatments of up to 24 hours at 2150°F indicate substantial improvement of both the room temperature and elevated temperature strength of cast Inco 713c. Table 14 shows a comparison of both room temperature and elevated temperature strength for as-cast and heat treated Inco 713c. Figure 54 shows the homogenization of a dendritic, as cast, Inco 713c microstructure obtained by a 3 hour solution heat treatment at 2150°F. Figure 55 shows the progressive change in grain boundary conditions resulting from increasing times of solution heat treatment for cast Inco 713c. The precipitate in the grain boundaries was determined to be TiC by preferentially dissolving the matrix metal chemically and determining the crystal parameters of the precipitate by X-ray diffraction. It will be noted in Figure 55 that a 3 hour heat treatment has reduced the amount of precipitate at the grain boundary, but a 21 hour treatment has not only cleared the grain boundary of precipitate but has also changed the character of the grain boundary. The longer heat treatment corresponds to improved strength at both room and elevated temperature. Figure 56 shows TiC precipitate at the grain boundary of Inco 713c which has been fractured by a microhardness indenter. This indicates the brittleness of this precipitate at room temperature.

Figure 57 shows the microstructure of Inco 713c cast into a one-inch thick section and heat treated for 3 hours at 2150°F. The specimens of Inco 713c previously shown had been cast into sections one-half inch thick or less. It will be noted in Figure 57 that the degree of grain boundary segregation, even after 3 hours solution heat treatment, is still substantial when compared with the as cast structure of a thinner casting. It is apparent from this, that the heavier the section cast the more difficult it is to remedy grain boundary segregation by solution heat treatment.

Figure 58 shows electron micrograph studies of the effects of solution heat treatment at 2150°F on the TiC precipitate in the grain boundaries of cast Inco 713c. Figure 59 shows the effects of varying times of solution heat treatment of Inco 713c on its macroetching characteristics. The macroetched as cast Inco 713c shows a ridge of TiC precipitate remaining unattacked along the grain boundary with the depleted zone on each side of this ridge preferentially attacked by the etchant. This shows a marked tendency for preferential attack along the grain boundary by chemical etchants in the as cast condition. It is

reasonable to deduce from this that there would also be preferential attack in an analogous manner by oxidation at elevated temperature. The Inco 713c solution heat treated for 3 hours at 2150°F shows preferential attack along the grain boundaries, but it no longer shows a retained ridge of TiC at the grain boundaries. This evidence, coupled with the microstructures previously discussed, clearly indicates that the TiC precipitate has been largely redistributed out of the grain boundaries by the 3 hour solution heat treatment.

Carbon is an interstitial element having a high diffusion rate and titanium is a substitutional alloying element having a relatively small atomic diameter and hence would be expected to have a relatively high diffusion rate. It is therefore reasonable to believe that the TiC has diffused out of the grain boundaries, but that macrosegregation of other substitutional elements at the grain boundaries still exists in significant amount after 3 hours heat treatment. This would explain the continued preferential grain boundary attack after three hours heat treatment and the absence of the TiC ridge. After 21 hours solution heat treatment the preferential attack of the macroetching solution on the grain boundaries has completely disappeared. This would indicate that for this particular sample of cast Inco 713c, a solution heat treatment time of 21 hours at 2150°F is adequate to essentially homogenize the structure. It should be pointed out that this solution heat treatment might not be adequate for cast Inco 713c having coarser grain size or a greater amount of macrosegregation. These tests do indicate that a prolonged solution heat treatment time is required to homogenize complex nickel base alloys. These tests also provide some indication that solution heat treatment might minimize the tendency for preferential grain boundary attack by oxidation at elevated temperatures.

Table 15 shows the effects of long time solution heat treatments on the room temperature and elevated temperature tensile properties of rolled Inco 713c sheet. The most marked improvement in properties obtained by extending solution heat treatment time beyond 24 hours is in room temperature ductility. Solution heat treatment for at least 24 hours appears to be necessary to attain the best combination of elevated temperature tensile strength and room temperature ductility.

c. Aging Heat Treatments

The solution heat treated Inco 713c rolled sheet essentially met the target properties of the program so relatively little work was done on investigating aging heat treatments for this alloy. Only two aging temperatures were used; 1600°F and 1900°F. The effects of these aging heat treatments on solution heat treated rolled Inco 713c sheet are given in Table 16. One-half hour aging at 1900°F has had no appreciable effect on tensile properties at 1900°F. Aging heat treatments at 1600°F tend to improve elevated temperature ductility, but they tend to lower room temperature ductility. Aging at 1600 F does not appear to appreciably affect tensile strength. The aging heat treatments used in this program on Inco 713c do not appear to have had any appreciable effect on tensile properties other than to provide some improvement in elevated temperature ductility.

2. NASA TaZ8 Alloy

a. Recrystallization

The NASA alloy is much more difficult to recrystallize than Inco 713c. Rapid recrystallization of the NASA alloy does not take place until a temperature of 2200°F is reached. Furthermore, even at this temperature recrystallization has been observed only on sheet which has received reductions in thickness by rolling of approximately 50%. Figure 38 on page 45 shows that heat treatment at 2150°F does not recrystallize NASA alloy cold rolled 14%. Figure 45 on page 53 shows that NASA alloy hot rolled at 2250°F has not recrystallized. Figure 60 shows that NASA alloy rolled to 45% reduction and then solution heat treated for 1/2 hour at 2200°F has undergone some recrystallization, although recrystallization does not appear complete. Figure 61 shows a comparison of the microstructure of as cast, as rolled, and recrystallized NASA alloy as shown by electron micrograph studies. The rolled and recrystallized NASA alloy had been hot rolled at 1650°F to 60% reduction and heat treated at 2200°F for 1/2 hour. Recrystallization appears complete, but the basic structure of the rolled and recrystallized alloy is substantially different from that of the as cast alloy. Since the 1900°F strength of the as cast alloy is appreciably greater than that of the best rolled and heat treated NASA alloy, it would appear that the as cast structure is to be preferred for high temperature strength.

TABLE 14

COMPARISON OF ROOM AND ELEVATED TEMPERATURE STRENGTH

OF

AS CAST AND HEAT TREATED INCO 713C

Melt	Condition	Test Temperature	Yield Strength (psi)		Ultimate Tensile Strength (psi)		Elongation Percent Room Temperature	1700°F STRESS RUPTURE	
			Room Temperature	Room Temperature	Room Temperature	Room Temperature		Stress.	Life Elongation (hours) (%)
56	As Cast	Room	118,100		126,300		2.5		
56	As Cast + (A)	Room	122,100		130,600		3.0		
56	As Cast	1700						30,000	10.8
56	As Cast + (A)	1700						30,000	26.1
56	As Cast + (B)	1700						30,000	15.7
67(C)	As Cast	Room	94,500		110,600		8.0		
67(C)	As Cast + (A)	Room	119,000		142,900		9.0		
67(C)	As Cast	1700						30,000	20.2
67(C)	As Cast + (A)	1700						30,000	111.2
102(C)	As Cast	Room	109,800		121,900		5.0		
	As Cast + (A)	Room	118,400		131,900		4.0		
	As Cast + (D)	Room	115,500		122,500		2.0		
	As Cast + (E)	Room	127,000		131,800		2.0		
	As Cast	1700						30,000	20.8
	As Cast + (A)	1700						30,000	3.5
	As Cast + (D)	1700						30,000	16.3
	As Cast + (E)	1700						30,000	112.3

(A) H.T. 3 hours at 2150°F air cool
 (B) H.T. 3 hours at 2150°F + 24 hours at 1550°F
 (C) .01% niobium added
 (D) H.T. 3 hours at 1950°F Air Cool
 (E) H.T. 24 hours at 2150°F Air Cool

TABLE 15
Effect of Solution Heat Treatment on Rolled INCO 713C Sheet

Melt	Roll Number	Heat Treatment	Test Temperature	Yield Strength		Ultimate Tensile Strength		Elongation (%)	
				ROOM TEMPERATURE	1900°F	ROOM TEMPERATURE	1900°F	ROOM TEMPERATURE	1900°F
548	210	as rolled	1900 °F						
550	216	41 hrs at 2150°F A.C.	1900 °F						
		" " " "	1900 °F						
404	122	1/2 hr at 2200°F A.C.	1900 °F						
404	122	3 hrs at 2200°F A.C.	ROOM & 1900 °F	143,900	178,200	12.0	42,800	10.0	4.0
		3 hrs at 2200°F, 170°F ROOM & 1900 °F oil quench	ROOM & 1900 °F	124,600	164,200	14.0	38,200	6.0	6.0
404	129	24 hrs at 2200°F A.C.	ROOM & 1900 °F	135,700	187,300	18.0	51,400	10.0	10.0
544	191	63 hrs at 2200°F A.C.	ROOM	123,700	177,100	18.0			

TABLE 16
Effect of Aging Heat Treatments on Rolled and Solution Heat Treated INCO 713C Sheet

Melt	Roll Number	Heat Treatment	Test Temperature	Ultimate Tensile Strength		Elongation (%)
				ROOM TEMPERATURE	1900°F	
548	210	as rolled	1900	44,000	3.0	
		41 hrs at 2150°F + 24 hrs at 1600°F	1900	50,800	10.0	
550	216	41 hrs at 2150°F + 61 hrs at 1600°F	1900	50,000	6.0	
547	211	40 hrs at 2150°F + 24 hrs at 1600°F + 1/2 hr at 1900°F	1900	47,800	9.0	
		1/2 hr at 2200°F + 24 hrs at 1600°F	1900	49,400	12.0	

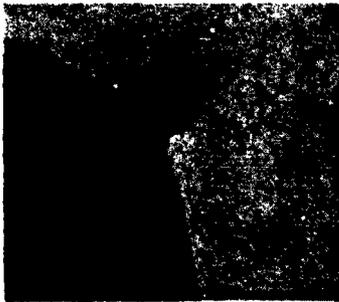


As Cast



Solution Heat Treated at
2150°F for 3 Hours and Air Cooled.

Figure 54. Homogenization of As-Cast Dendritic Structure in Inco 713c by Heat Treatment at 2150°F. Etch No. 1. 100X.



As Cast, Melt 56.



Heat No. 56, Solution
Heat Treated At 2150°F
For 3 Hours, Air Cooled.



Heat No. 104, Solution
Heat Treated At 2150°F
For 21 Hours, Air Cooled.

Figure 55. Effect Of Solution Heat Treatment Of Cast Inco 713c. Etch No. 2. 500X



Figure 56. TiC Precipitate in Grain
Boundary of As-Cast Inco 713c.
Note: Micro-Hardness Indentor
Shattered the Precipitate,
Indicating its Brittle Nature.
Etch No. 1. 1,000X.

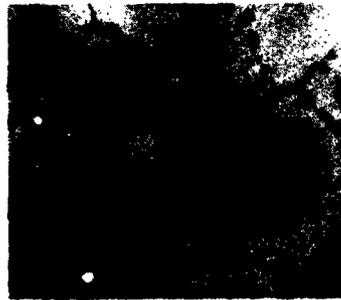
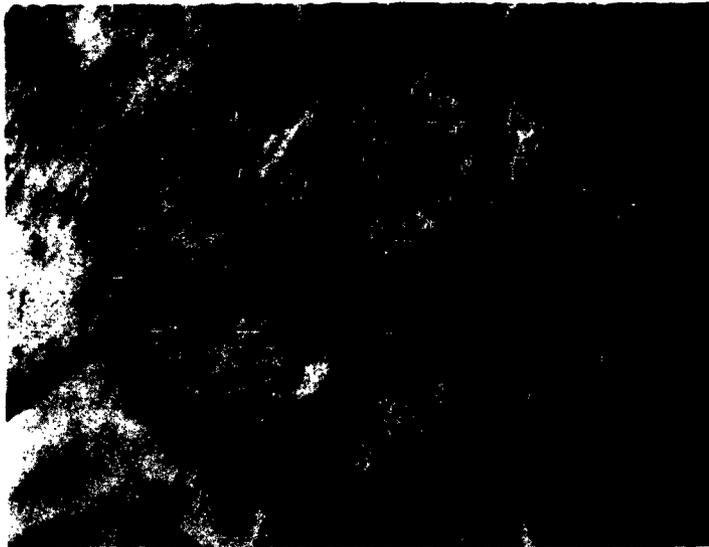


Figure 57. Inco 713c Section Cut
from One-Inch Thick
Specimen Heat Treated
at 2150°F for 3
Hours. Etch No. 1.
100X.

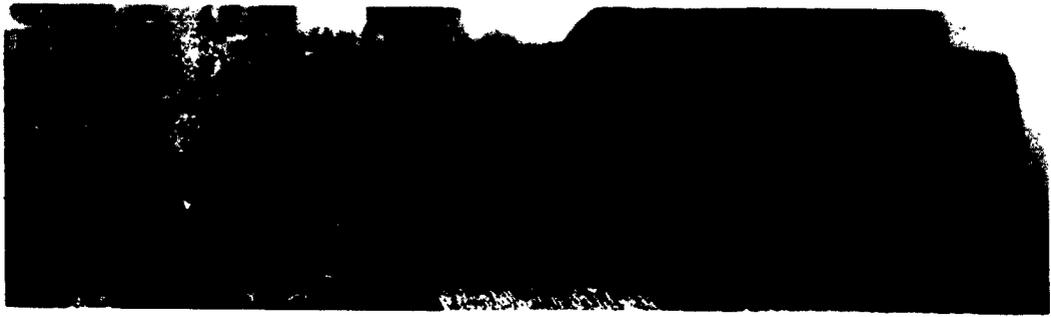


As Cast



Solution Heat Treatment at 2150°F
for 3 Hours, Air Cooled.

Figure 58. Effect of Solution Heat Treatment on Grain Boundary in
Cast Inco 713c, Melt 67. Etch No. 2. 4,000X.



As Cast. 6X.



3 Hours At 2150°F, Air Cool. 6X.

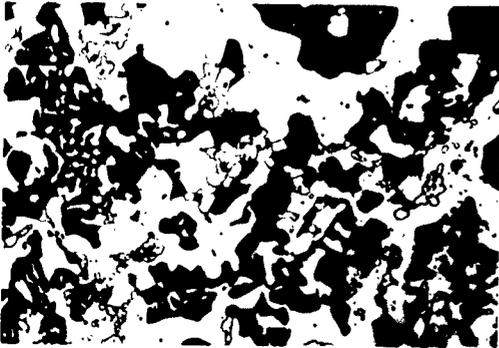


21 Hours At 2150°F,
Air Cool. 4X.

3 Hours At 2150°F,
Air Cool. 4X.

As Cast. 4X.

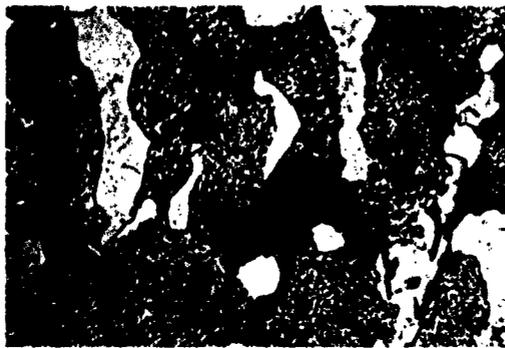
Figure 59. Comparison Of Macro-Etching Characteristics Of As-Cast And Heat-Treated Inco 713c.



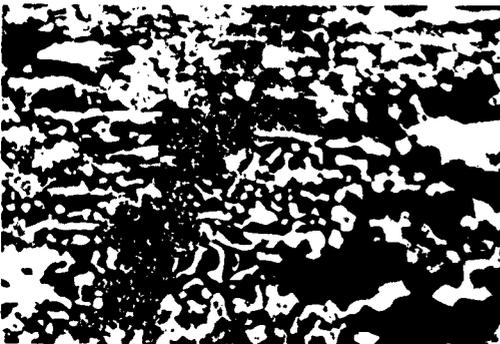
NASA Alloy, Melt 462, Reduced 45% at 1650°F as Rolled.



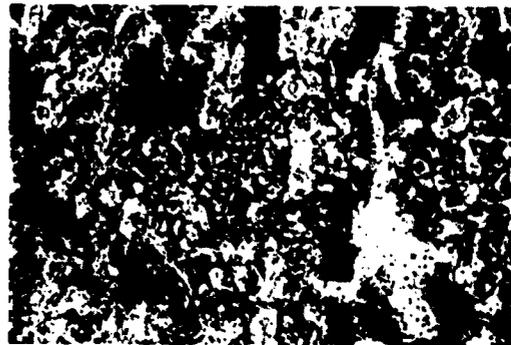
NASA Alloy, Melt 462, Reduced 45% at 1650°F, then Heat Treated 1/2 Hour at 2200°F, Air Cooled.



NASA Alloy, Melt 462, Reduced 60% at 1750°F, then Heat Treated 63 Hours at 1500°F, Air Cooled.



NASA Alloy, Melt 418, Reduced 50% at 1825°F, then Heat Treated 1/2 Hour at 2200°F, Air Cooled.



Same as Photo at Left, with Additional Heat Treatment of 24 Hours at 1600°F.

Figure 60. Effects of Heat Treatment on the Microstructure of Hot Rolled NASA Alloy. Etch No. 4. 500X.

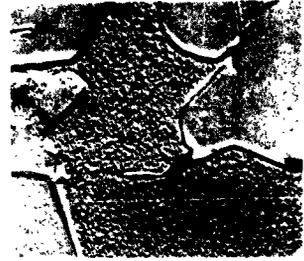
As Cast



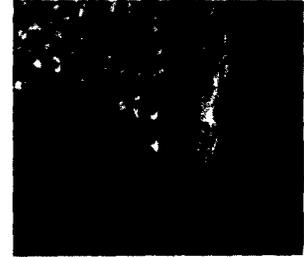
As Rolled
At 1650°F.



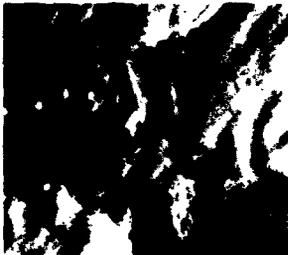
As Rolled At 1650°F
+ 1/2 HR At 2200°F
Air Cooled.



3,000X.



25,000X.



50,000X.

Figure 61. Comparison Of Microstructure Of NASA Alloy In The As-Cast, As-Rolled, And As-Rolled + Heat-Treated Condition.

In summation, rolled NASA alloy will recrystallize completely in 1/2 hour at 2200°F provided it has had 60% reduction in thickness by rolling before heat treatment. With 45% reduction in thickness, recrystallization is partial after 1/2 hour at 2200°F. Lesser reductions or lesser heat treatment temperatures do not result in appreciable recrystallization.

b. Solution Heat Treatment

Prolonged solution heat treatments do not materially affect the microstructure of NASA alloy as seen in the optical microscope. (See Figure 62). Figure 63 shows the effects on the as cast NASA alloy structure of 2000°F and 2200°F solution heat treatments as shown in electron micrograph studies. These solution heat treatments apparently cause the massive precipitate to break-up into finer form, and the higher the heat treatment the more complete the breakup. Table 17 gives a comparison of both room temperature and 1900°F tensile strengths of NASA alloy in the as-cast condition and after two different solution heat treatments. It will be noted that solution heat treatment increases room temperature tensile strength somewhat, but otherwise has no significant effect on the mechanical properties tested. A comparison of Figure 64 with Figure 65 shows that the variation in microstructure of NASA alloy caused by solution heat treatment is relatively small compared with the variation in microstructure of the same alloy from one part of a given casting to another part of the same casting. These differences in as cast structure are apparently due to differences in cooling rate of one part of the casting relative to other parts of the same castings.

Table 18 shows the effects of various solution heat treatment times on the mechanical properties of NASA alloy. The most noticeable improvement in properties resulting from prolonged solution heat treatments is a small increase in room temperature ductility. Tensile strength at 1900°F appears to decrease somewhat with longer solution heat treatment times, although this may be due to oxidation attack on the surface of the metal caused by impurities in the atmosphere used in heat treatment. At best, there is no significant improvement in 1900°F tensile strength as a result of long time solution heat treatment. Since the solidus point of the NASA alloy has been determined to be about 2300°F, and since temperatures appreciably below 2200°F

will not recrystallize the rolled structure, it would appear unlikely that any practical solution heat treatment can be devised to substantially improve the properties of the NASA alloy.

c. Aging Heat Treatment

Table 19 shows the effects of various aging heat treatments on solution heat treated, rolled NASA alloy. Aging heat treatments after solution heat treatment do substantially improve the mechanical properties of the NASA alloy. The aging treatment giving the best room temperature properties is 24 hours at 1750°F. This heat treatment gives somewhat lower room temperature tensile strength than some others, but most important it is the only heat treatment found to give usable room temperature ductility to rolled NASA alloy sheet. This heat treatment gives 4% to 5% elongation at room temperature, which is probably a practical minimum allowable figure if the alloy is to find practical use in rolled sheet form. The best 1900°F tensile properties were obtained with an aging heat treatment of 24 hours at 1600°F. It is likely that an intermediate aging temperature might result in an optimum balance of properties.

Figure 66 shows the results of an electron microscope study of rolled NASA alloy which was aged at 1500°F for 63 hours. This study shows that aging at this temperature without prior solution heat treatment does not materially change the structure from that existing in the as rolled condition.

3. No. 429 Alloy

Only a very limited amount of work has been done on the rolling and heat treatment of No. 429 alloy. The microstructure of No. 429 alloy is complex and difficult to adequately resolve due to the greatly varying rate of attack of standard etching solutions on the various microconstituents. No. 429 alloy has been hot rolled at 2100°F and at 2200°F. None of the alloy rolled has received subsequent heat treatment. There is more variation to be observed in the microstructure of a single casting of the alloy than there is variation between the microstructures of the alloy in the as cast and the hot rolled conditions. Only one tensile test was run on sheet rolled from this alloy, therefore no conclusions can be drawn as to the strength of No. 429 in the as rolled condition. (Text continued on Page 80.)

TABLE 17
Effect of Solution Heat Treatment on Cast NASA Alloy Sheet

Melt	Heat Treatment	Test Temperature	Yield Strength		Ultimate Tensile Strength		Elongation (%)	
			ROOM TEMPERATURE	ROOM TEMPERATURE	ROOM TEMPERATURE	ROOM TEMPERATURE	ROOM TEMPERATURE	ROOM TEMPERATURE
386 and 394	as cast	ROOM & 1900 °F	120,900		121,100		1.0	2.0
393	16 hrs at 2200 °F W.Q.	ROOM & 1900 °F			92,600		1.0	2.0
394	16 hrs at 2200 °F W.Q.	ROOM & 1900 °F	129,700		137,600		1.0	1.0

TABLE 18
Effect of Solution Heat Treatment on Rolled NASA Alloy Sheet

Melt	Roll Number	Heat Treatment	Test Temperature	Yield Strength		Ultimate Tensile Strength		Elongation (%)	
				ROOM TEMP	ROOM TEMP	ROOM TEMP	ROOM TEMP	ROOM TEMP	ROOM TEMP
418	109	as rolled	1900 °F				37,700		5.0
425, 385	123, 125	1/2 hr at 2200 °F A.C.	ROOM & 1900 °F	153,600		163,400		1.0	11.5
425, 385	123, 125	3 hrs at 2200 °F A.C.	ROOM & 1900 °F			128,300		2.0	8.0
		3 hrs at 2200 °F, 170 °F oil quench	ROOM & 1900 °F	149,200		162,100		2.0	6.0
425	123	24 hrs at 2200 °F A.C.	ROOM & 1900 °F	138,600		164,500		3.0	3.0
385	125	64 hrs at 2200 °F A.C.	ROOM			144,800		6.0	6.0
462, 418	120, 148	64 hrs at 2200 °F, re-tempered and cooled	ROOM & 1900 °F	134,400		161,700		1.0	4.0
425, 385	123, 125	3 hrs at 2250 °F A.C.	ROOM & 1900 °F			136,500		0.0	1.0

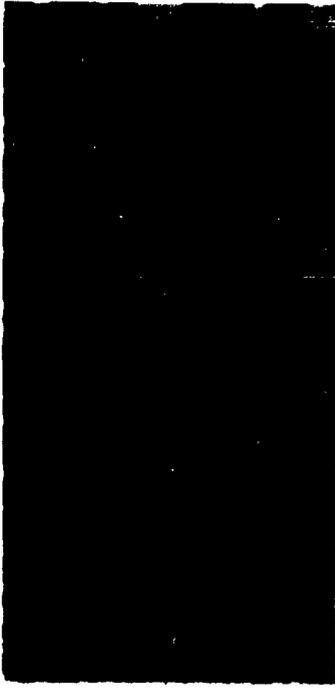
TABLE 19
Effect of Solution Heat Treatment and Aging on Rolled NASA Alloy Sheet

Melt	Roll Number	Heat Treatment	Test Temperature	Yield Strength		Ultimate Tensile Strength		Elongation (%)	
				ROOM TEMP	ROOM TEMP	ROOM TEMP	ROOM TEMP	ROOM TEMP	ROOM TEMP
424, 425	150, 199	as rolled	1900 °F						5.0
		64 hrs @ 2150 °F + 64 hrs @ 1450 °F	Room & 1900 °F	207,000 (A)		217,000 (A)		0.0	2.0
424, 418	143, 148	61 hrs @ 2150 °F + 24 hrs @ 1600 °F	Room & 1900 °F	189,400 (A)		226,600 (A)		2.0	4.5
387, 385	154, 155	61 hrs @ 2150 °F + 24 hrs @ 1750 °F	Room & 1900 °F	135,200 (A)		197,400 (A)		5.0	2.0
385, 385	152, 152	61 hrs @ 2150 °F + 64 hrs @ 1950 °F	Room & 1900 °F	128,300 (A)		175,200 (A)		3.0	7.0
		1/2 hr @ 2200 °F + 24 hrs @ 1600 °F	1900 °F						14.0
424, 415	126, 114	64 hrs @ 2200 °F + 24 hrs @ 1450 °F	Room & 1900 °F			121,000			8.0
387, 419	119, 171	64 hrs @ 2200 °F + 64 hrs @ 1600 °F	Room & 1900 °F	123,500		131,500		1.0	12.0
462, 419	120, 171	as before + 2 1/2 hrs @ 2000 °F	Room & 1900 °F	120,200		134,400		1.0	16.0
418	147	64 hrs @ 2200 °F + 24 hrs @ 1675 °F	1900 °F						3.0
425	122	64 hrs @ 2200 °F + 24 hrs @ 1900 °F	Room & 1900 °F	123,300		144,900			10.0

(A) Because of some oxidation depletion and scaling during heat treatment, thickness of specimen was measured with pointed micrometer.



1/2 HOUR



3 HOUR



24 HOUR



64 HOUR

Figure 62. Effect of Various Prolonged Solution Heat Treatments on Rolled NASA Alloy at 2200°F, Air Cooled. Etch No. 4. 500X.



Heated to 2000°F for 16 Hours and
Water Quenched. 25,000X.



Heated to 2200°F for 16 Hours and
Water Quenched. 25,000X.

Figure 63. Electron Micrograph Study of the Effects of Two Different
Heat Treatments on Cast NASA Alloy, Melt 385. Etch No. 1.



Sheet No. 2.



Sheet No. 4.



Heavy Bar Section.

Figure 64. Variation of Microstructure within a Single Casting of
NASA Alloy, Melt 385 as Cast. Etch No. 1. 100X.



NASA Alloy Heated to 2000°F for 16 Hrs and Water Quenched Heat No. 385 100X FeCl₃ + HCl Etch.



NASA Alloy Heated to 2200°F for 16 Hrs and Water Quenched Heat No. 385 100X FeCl₃ + HCl Etch.



NASA Alloy Heated to 2000°F for 16 Hrs and Water Quenched Heat No. 386 100X FeCl₃ + HCl Etch.



NASA Alloy Heated to 2200°F for 16 Hrs and Water Quenched Heat No. 386 100X FeCl₃ + HCl Etch.

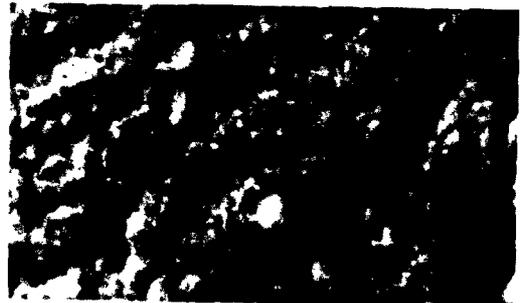
Figure 65. Optical Microscope Study of the Effects of Two Different Heat Treatments on Cast NASA Alloy.



3,000X.



25,000X.



50,000X.

Figure 66. Electron Micrographs Of NASA Alloy Hot Rolled 60%, Then Aged At 1500°F, For 63 Hours. Etch No. 1

4. Ta-C Modified Inco 713c (Melt No. 388)

Figure 37 shows that heat treating rolled Ta-C modified Inco 713c at 2000°F does not result in recrystallization of the alloy. This alloy therefore has a higher recrystallization temperature than Inco 713c. Work on this alloy was stopped relatively early in the program to provide more time for work on Inco 713c and NASA alloy, since the latter two alloys had better 1900°F tensile strength in the as cast condition. Hence, the recrystallization temperature of this alloy is not known, and the effects of any such recrystallization heat treatment are not known.

D. Rolled Alloy Sheet Evaluation

1. Tensile Properties

a. Inco 713c Rolled Sheet

Rolled and heat treated Inco 713c sheet has a tensile strength of approximately 50,000 psi at 1900°F and approximately 170,000 psi at room temperature. The ductility of this alloy ranges from 12% to 20% elongation at room temperature depending on heat treatment and from 6% to 10% at 1900°F depending on heat treatment. Figure 67 shows a plot of tensile strength and elongation versus test temperature for the range from room temperature to 2300°F. Table 21 gives a summary of the tensile strengths of rolled and heat treated Inco 713c in various conditions.

The rate of straining of the test specimen has a substantial effect on the tensile strength reported for nickel base alloys in the temperature range approximating 1900°F. Standard elevated temperature tensile test procedures generally call for a strain rate of 0.005 inches per inch per minute during the elastic portion of the test and from 0.05 to 0.1 inches per minute of head travel thereafter. The tests run at elevated temperature throughout most of this program were run at these or somewhat lower strain rates. Since only ultimate tensile strength and elongation were required, an extensometer was not used in making these tests. Therefore, direct control of strain rate was not possible. As is explained in Appendix C, strain rate was indirectly controlled by controlling load rate. The load rate used in this program results in a rate of straining less than standard rather than greater than standard. In order to determine the actual effect of straining rate during testing on the

reported tensile strength of Inco 713c, tests were run on comparable Inco 713c test bars at three substantially different rates of loading. Table 22 summarized the results of these tests and Figure 68 shows a plot of reported tensile strength versus calculated average strain rate. It has been determined by experiment that the total head travel of the machine during a test is approximately twice the actual strain in the gauge length of a standard 1" gauge length test bar used in this program. Figure 69 gives a plot of reported tensile strength versus time spent under load in the given test. It will be seen from these tables and charts that the true standard tensile strength of Inco 713c at 1900°F would be slightly higher than the results of the tensile test run in this program would indicate. This discrepancy is small and is not thought to be significant. It will be noted that large variations in strain rate will produce a variation of almost 300% in the reported value for the tensile strength of this alloy at 1900°F.

Table 23 shows the effect of strain rate on NASA alloy tensile strength at 1900°F. The data in Tables 22 and 23 therefore show that strain rate during testing substantially affects the reported tensile strength for both Inco 713c and NASA alloy.

b. NASA TaZ8 Alloy Rolled Sheet

NASA TaZ8 alloy sheet which has been rolled, solution heat treated at 2200°F and then aged at 1600°F has a 1900°F tensile strength of about 50,000 psi with an elongation of about 4%. The same alloy sheet will have a room temperature strength of about 170,000 psi and 4% to 5% elongation if it is aged at 1750°F instead of 1600°F. The tensile data on which these conclusions are based is given in Table 19 on page 75. No tensile data was obtained on rolled sheet for any temperature other than room temperature and 1900°F.

c. Other Alloys

A few isolated tensile tests were run on sheet rolled from Ta-C modified Inco 713c (Melt No. 388) but these tests are inadequate in number and in quality to use as the basis for any conclusions. No tensile tests were run on any other of the alloys rolled.

TABLE 20
TENSILE DATA ON NICKEL ALLOYS
(Various Temperatures, Other Than room and 1900°F)

ALLOY AND ROLL NUMBER	MELT NUMBER	CONDITION	TEST TEMPERATURE (°F)	THICKNESS (0.001")	AREA square inches	ULTIMATE TENSILE LOAD (#)	ULTIMATE TENSILE STRENGTH (PSI)	ELONGATION (%)
NASA	393	as cast	2000	107.2	.02627	760	27,900	---
	394	as cast	2100	96.4	.02384	188	7,900	3.0
2(Ta-C) Modified INCO 713C	392	as cast	2100	94.1	.02302	198	8,600	---
Ta-C-Cr-W Modified INCO 713C	457	as cast	2100	116.3	.02805	534	19,000	7.0
	457	as cast	2200	108.8	.02646	176	6,700	3.0
	457	as cast	2300	108.2	.02628	11	418	2.5
INCO 713C, 201	545	as rolled	Room	49.1	.01395	1788	156,100	3.0
INCO 713C, 201	545	rolled + a.	1200	50.2	.01783	2560	128,200	3.0
INCO 713C, 201	545	rolled + a.	1200	50.2	.01783	2560	143,600	---
INCO 713C, 201	545	rolled + a.	1500	8.6	.01234	1340	108,600	2.0
INCO 713C, 221	550	rolled + a.	1500	19.6	.00649	812	125,100	7.0
INCO 713C, 230	550	rolled + a.	1700	17.9	.01449	1060	73,200	4.0
INCO 713C, 231	550	rolled + a.	1700	19.5	.00995	860	86,400	7.0
INCO 713C, 232	550	rolled + a.	2000	18.2	.01713	460	26,900	15.0
INCO 713C, 231	550	rolled + a.	2000	18.7	.01013	260	22,700	5.0
INCO 713C, 230	550	rolled + a.	2100	19.1	.01484	228	15,400	16.0
INCO 713C, 221	550	rolled + a.	2100	19.8	.01043	116	11,100	8.0
INCO 713C, 230	550	rolled + a.	2200	17.1	.01387	60	4,300	10.0
INCO 713C, 230	550	rolled + a.	2200	18.8	.01511	80	5,300	11.0

(a) H.T. 40 hrs at 2150°F, +16 HRS at 1600°F.

TABLE 21
Tensile Properties of Rolled and Heat Treated INCO 713c

Remarks	Yield Strength (FSI)		Ultimate Tensile Strength (FSI)		Elongation (%)	
	Room Temperature	Room Temperature	Room Temperature	1900°F	Room Temperature	1900°F
as rolled				41,800		3.0
as rolled				42,900		3.0
rolled + HT 24 hrs at 2200°F, A. C.	minimum	121,900	170,400	51,400	12.0	10.0
	average	131,200	182,600	(single test)	15.0	
rolled + HT 3 hrs @ 2200°F, A. C.	minimum		178,200	28,800	12.0	6.0
	average		(single test)	35,800		8.0
rolled + HT 3 hrs @ 2200°F, 170°F oil quench (single test)		124,600	164,200	38,200	14.0	6.0
rolled + HT 1/2 hr @ 2200°F A. C. (single test)				45,300		4.0
same as above + 24 hrs @ 1600°F A. C. (single test)				49,400		12.0
rolled + HT 64 hrs @ 2200°F A. C.	minimum	122,600	170,200		18.0	
	average	123,300	172,800		19.0	
rolled + HT 41 hrs @ 2150°F A. C.	minimum			41,900		2.0
	average			46,400		6.5
as above + 24 hrs @ 1600°F	minimum			49,500		10.0
	average			50,200		11.0
as above + additional 37 hrs @ 1600°F	minimum			45,300		6.0
	average			47,700		7.1
rolled + HT 40 to 60 hrs @ 2150°F + 24 hrs at 1600°F	minimum	141,200	156,100	45,700	3.0	8.0
	average	(single test)		46,600		9.0

TABLE 22

Variation in 1900° F Tensile Strength of INCO 713c with Varying Strain Rate
 (All specimens rolled and HT 40 hrs @ 2150° F + 24 hrs at 1600° F)

Time of Test (Seconds)	Average Strain Rate Inches/Inch/Minute	Ultimate Tensile Strength	Elongation %
864	0.0083	30,700	12.0
434	0.0083	31,500	6.0
720	0.012	24,700	13.0
234	0.021 (A)	46,400	8.0
204	0.026 (A)	47,800	9.0
1.5	2.100	60,000 (B)	7.0
1.8	2.700	65,300	8.0
3.6	2.500	69,700	15.0

(A) The normal rate used in tests in this report

(B) Minimum tensile strength, load indicating dial moved before recording of data.

TABLE 23
 VARIATION OF 1900° F TENSILE STRENGTH OF NASA ALLOY AND VARIOUS STRAIN RATES (MEET 4.18)

Average Strain Rate Inches/Inch/Minute	Ultimate Tensile Strength	% Elongation
.01 (A)	41,900	3.0
.006	24,400	2.0
.004	29,400	

Rolled & HT 64 Hrs. at 2200° F
 & 24 Hrs. at 1675° F

(A) This rate corresponds to normal rate used in all other tensile tests in this program.

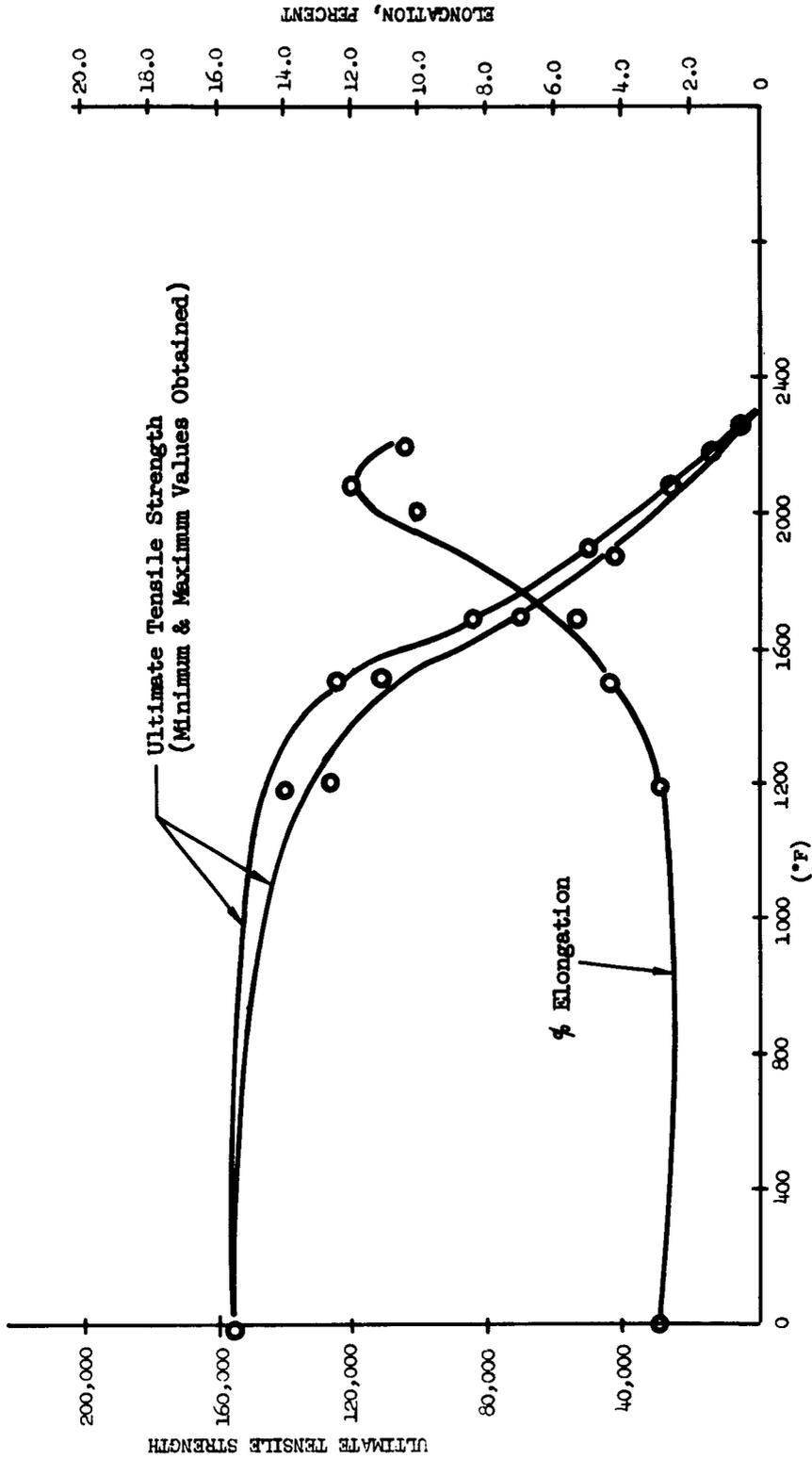


Figure 67. Ultimate Tensile Strength and Percent Elongation Variation of Inco 713c Rolled Sheet at Various Temperatures (Condition of Metal, Rolled + H.T. At 2150°F for 40 Hours + 24 Hours at 1600°F).

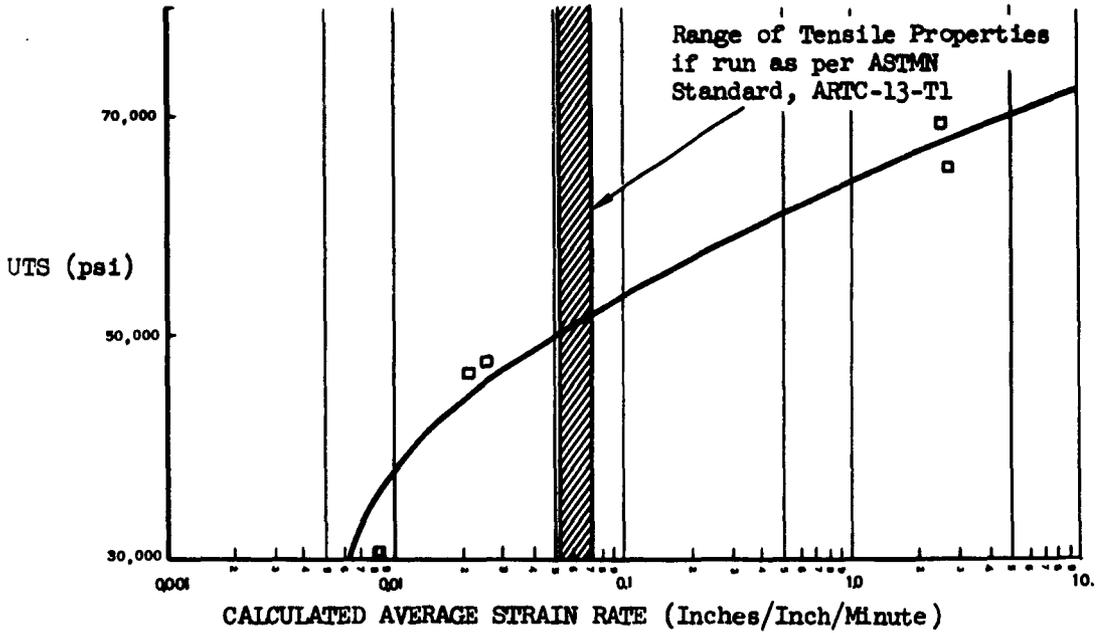


Figure 68. Calculated Average Strain Rate versus Ultimate Tensile Strength of Inco 713c Sheet at 1900°F. Condition, Rolled +HT 40 Hours at 2150°F + 24 Hours at 1600°F.

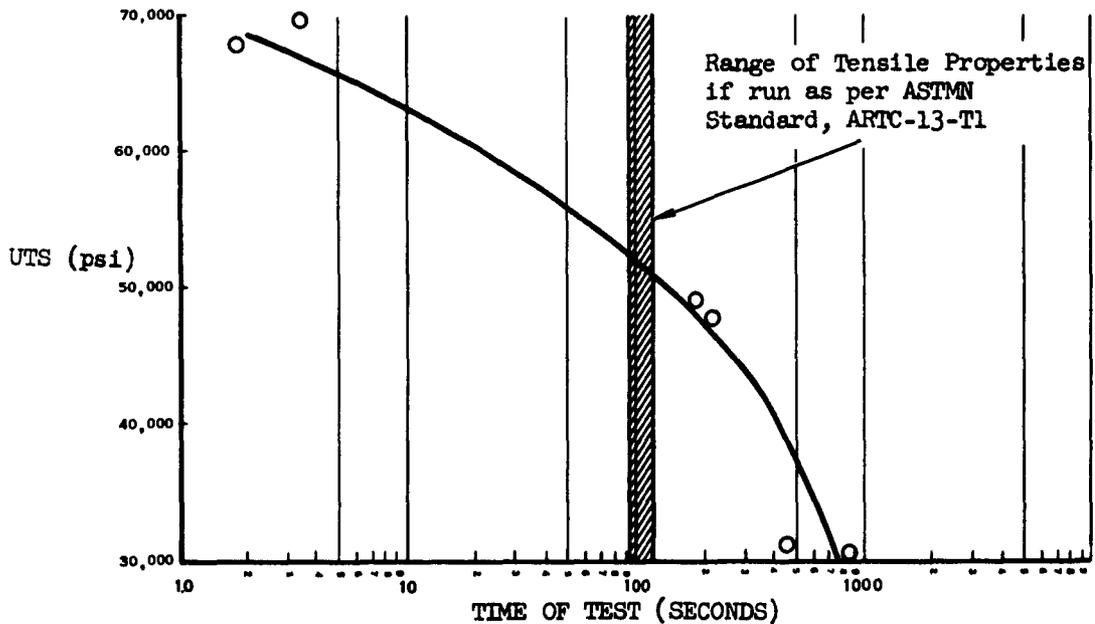


Figure 69. Total Time of Tensile Test versus Ultimate Tensile Strength of Inco 713c Sheet at 1900°F. (Conditions are the Same as Figure 68.)

2. Oxidation Resistance

a. Rate of Oxidation Attack

Oxidation tests in still air were run on NASA alloy, Inco 713c and No. 429 alloy in both as cast and rolled form. Oxidation tests were simultaneously run on René 41 sheet for comparison purposes. Table 24 and Figures 70 through 74 show the results of these oxidation tests. Briefly summarized, the tests show that Inco 713c alloy has about the same oxidation resistance at temperatures of up to 2100°F as René 41. NASA alloy has much poorer oxidation resistance than Inco 713c or René 41 and the No. 429 alloy is intermediate in oxidation resistance between NASA and Inco 713c alloys. Furthermore, the oxide scale formed on the NASA alloy has a marked tendency to explosively spall off of the metal surface upon cooling to room temperature, while the oxide scales formed on the other alloys are relatively adherent.

b. Identification of Oxidation Products

Samples were taken from the oxide scales formed on all of the alloys tested and these samples were analyzed by X-ray diffraction techniques. Table 25 gives the results of these analyses. It will be noted that the major constituent of the scale formed on the NASA alloy is NiO and the major constituent of the scale formed on the other two alloys is Cr₂O₃. This would indicate that chromium oxide is far more protective than nickel oxide. Since the Inco 713c has much better oxidation resistance than the No. 429 alloy, and since both oxide scales are primarily chromium oxide, it is obvious that other factors are also of importance in determining oxidation resistance.

3. Determination of Solidus Temperature

Nickel base superalloys have solidus temperatures much lower than their liquidus temperatures, and this wide mushy zone makes it difficult to determine solidus and liquidus temperatures by thermal analysis. In this program, an effort was made to determine the solidus temperature and to obtain an indication of the liquidus temperature by heating specimens of the alloy to temperatures within the temperature range expected to include solidus and liquidus temperatures and then water quenching these specimens. This was done with two alloys, NASA (Melt 387) and Ta-C (Melt 390)

TABLE 24
OXIDATION DATA

ALLOY	CONDITION	TEST TEMPERATURE (°F)	TIME (min.)	Weight Change (mg/cm ²) Specimen with Loose Oxide 1900°F	Weight Change (mg/cm ²) Specimen Alone 1900°F	Weight Change (mg/cm ²) Specimen Alone 2100°F
INCO 713C	as cast	1900	3840	+0.84-(a)	+0.17-a	+0.39-a
INCO 713C	as cast	1900	3840	+0.98-(b)	+0.20-b	+0.20-b
INCO 713C	as cast	2100	180			+0.37-b
INCO 713C	as cast	2100	180	+0.51-a		-1.26-b
INCO 713C	as cast	2100	10	+0.55-b		-1.19-b
INCO 713C	as cast	2100	100	+1.81-b		-0.13-b
INCO 713C	as cast	2100	1000	+0.12-b		
INCO 713C	as rolled	2100	10			
INCO 713C	33% Reduction	2100	100	+2.30-b		-3.68-b
INCO 713C	as rolled	2100	1000	+3.97-b		-0.88-b
INCO 713C	33% Reduction	2100	1000			
NASA	as cast	1900	3840	-6.72-a	-24.8-a	-23.00-a
NASA	as cast	1900	3840	+6.26-b	-24.5-b	-21.40-b
NASA	as cast	2100	180			-8.50-b
NASA	as cast	2100	180	+6.76-a		-7.14-b
NASA	as cast	2100	10	+9.02-b		-25.00-b
NASA	as cast	2100	10	+4.85-b		
NASA	as cast	2100	100	+11.73-b		-28.53-b
NASA	as cast	2100	1000	+31.95-b		-84.80-b
NASA	as rolled	2100	10	+4.35-b		
NASA	50% Reduction	2100	100	+10.73-b		
NASA	as rolled	2100	1000	+30.40-b		-79.30-b
NASA	50% Reduction	2100	1000			
NASA	as rolled	2100	1000			
NASA	50% Reduction	2100	1000			
No. 429 Alloy	as cast	1900	3840	+6.81-a	+6.68-a	+4.76-a
	as cast	1900	3840	+6.37-b	+6.32-b	+5.34-b
	as cast	2100	180			
	as cast	2100	180	+4.54-a		
	as cast	2100	180	+5.33-b		

TABLE 24 (Continued)

ALLOY	CONDITION	TEST TEMPERATURE (°F)	TIME (min.)	OXIDATION DATA	
				Weight Change (mg/cm ²) Specimen With Loose Oxide 1900°F	Weight Change (mg/cm ²) Specimen Alone 1900°F
No. 429 ALLOY	as cast	2100	10		
	as cast	2100	100	+1.72-b	+1.19-b
	as cast	2100	1000	+5.65-b	+4.16-b
	as rolled	2100	10	+17.49-b	-31.58-b
	63.5% Reduction	2100	10	+0.00-b	+1.50-b
	as rolled	2100	100	+7.48-b	+2.78-b
	63.5% Reduction	2100	1000	+37.90-b	-10.10-b
	as rolled	2100	10		
	63.5% Reduction	2100	1000		
RENE 41	as received	2100	10	+1.82-b	+0.68-b
	as received	2100	100	+1.87-b	+1.21-b
	as received	2100	1000	+3.84-b	-3.63-b

Footnotes:

- (a) Specimen in open crucible.
 (b) Specimen in closed crucible.

TABLE 25

NO. 429, NASA ALLOY AND 713c X-RAY DIFFRACTION DATA RUN ON SCALED OXIDE
(Run on Both Powder Camera and Diffractometer)

Alloy and Condition		Indicated Presence and Method		
		NiO	Cr ₂ O ₃	α-Al ₂ O ₃
No. 429	3 hrs. at 2100°F	Neither	Diffractometer indicates, major constituent	Diffractometer indicates presence
NASA	3 hrs. at 2100°F	Major constituent diffractometer & powder camera	Powder camera indicates possibilities of presence.	Neither method indicates presence
NASA	64 hrs. at 1900°F	Major constituent diffractometer & powder camera	Powder camera indicates possibilities of presence	Neither method indicates presence
713c	3 hrs. at 2100°F	Powder camera indicates possibility of presence	Major constituent powder camera only	
713c	64 hrs. at 1900°F	Powder camera indicates presence	Major constituent powder camera only	Powder camera indicates possibility of presence

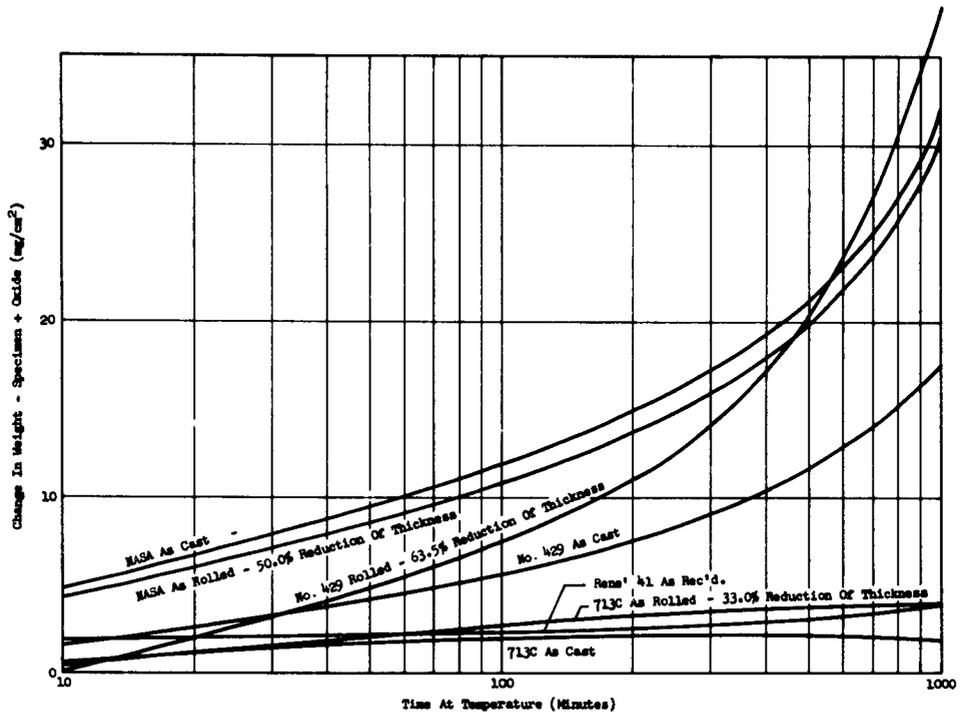


Figure 70. Increase in Weight of Specimen (With Oxide Scale) After Exposure at 2100°F.

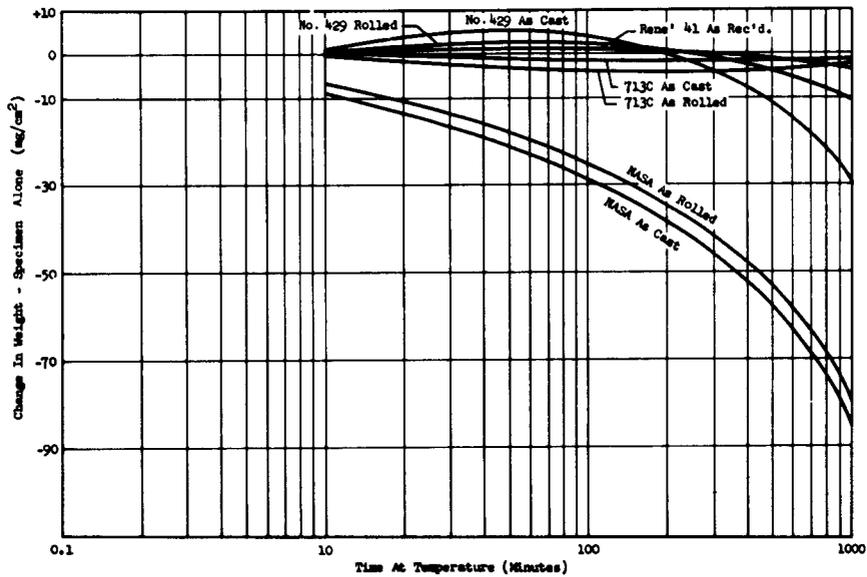


Figure 71. Net Change in Weight of Specimen After Exposure at 2100°F in Air. (After Removing Loose Oxide.)

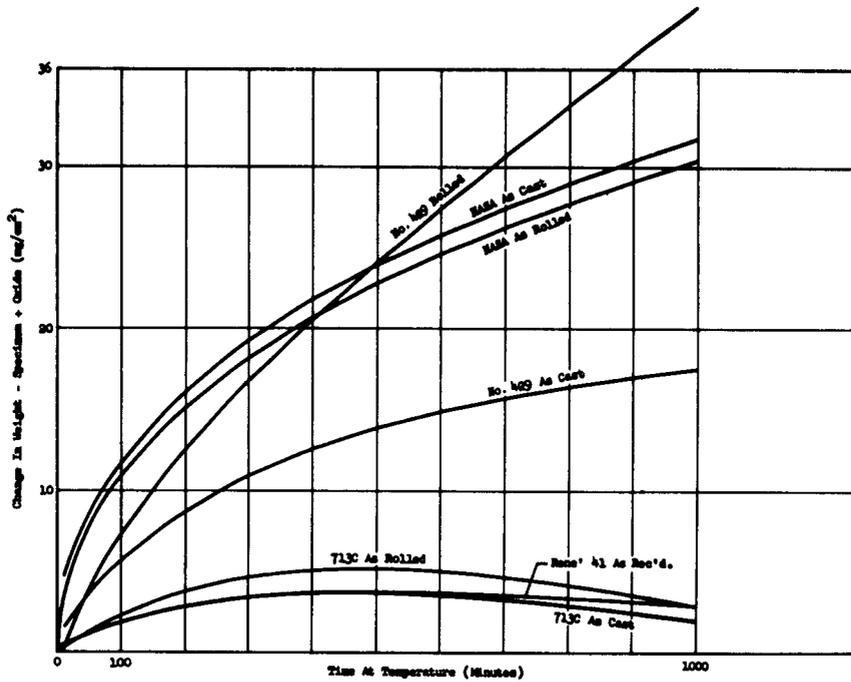


Figure 72. Increase in Weight of Specimen (With Oxide Scale) After Exposure at 2100°F.

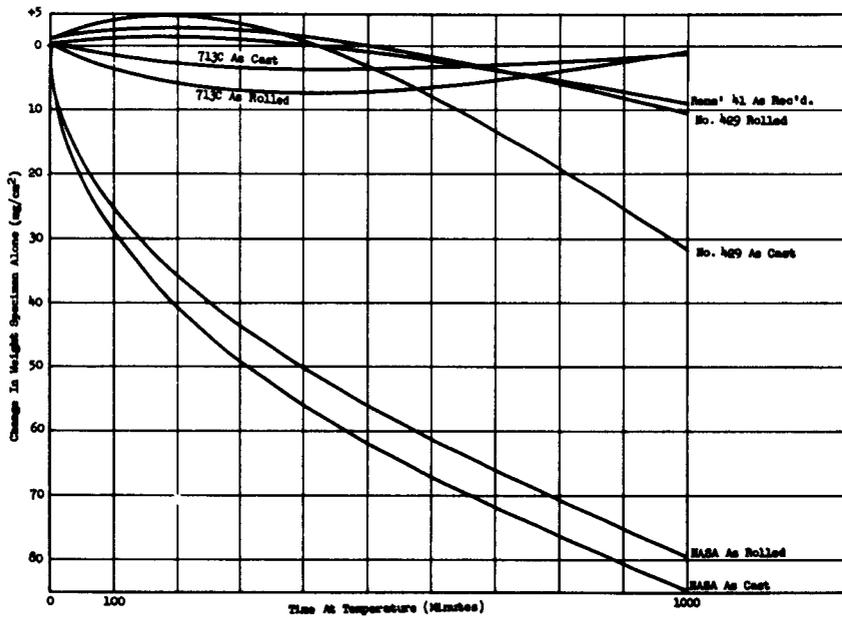


Figure 73. Net Change in Weight of Specimen After Exposure at 2100°F in Air. (After Removing Loose Oxide.)

 OPEN TO AIR
 COVERED WITH LID

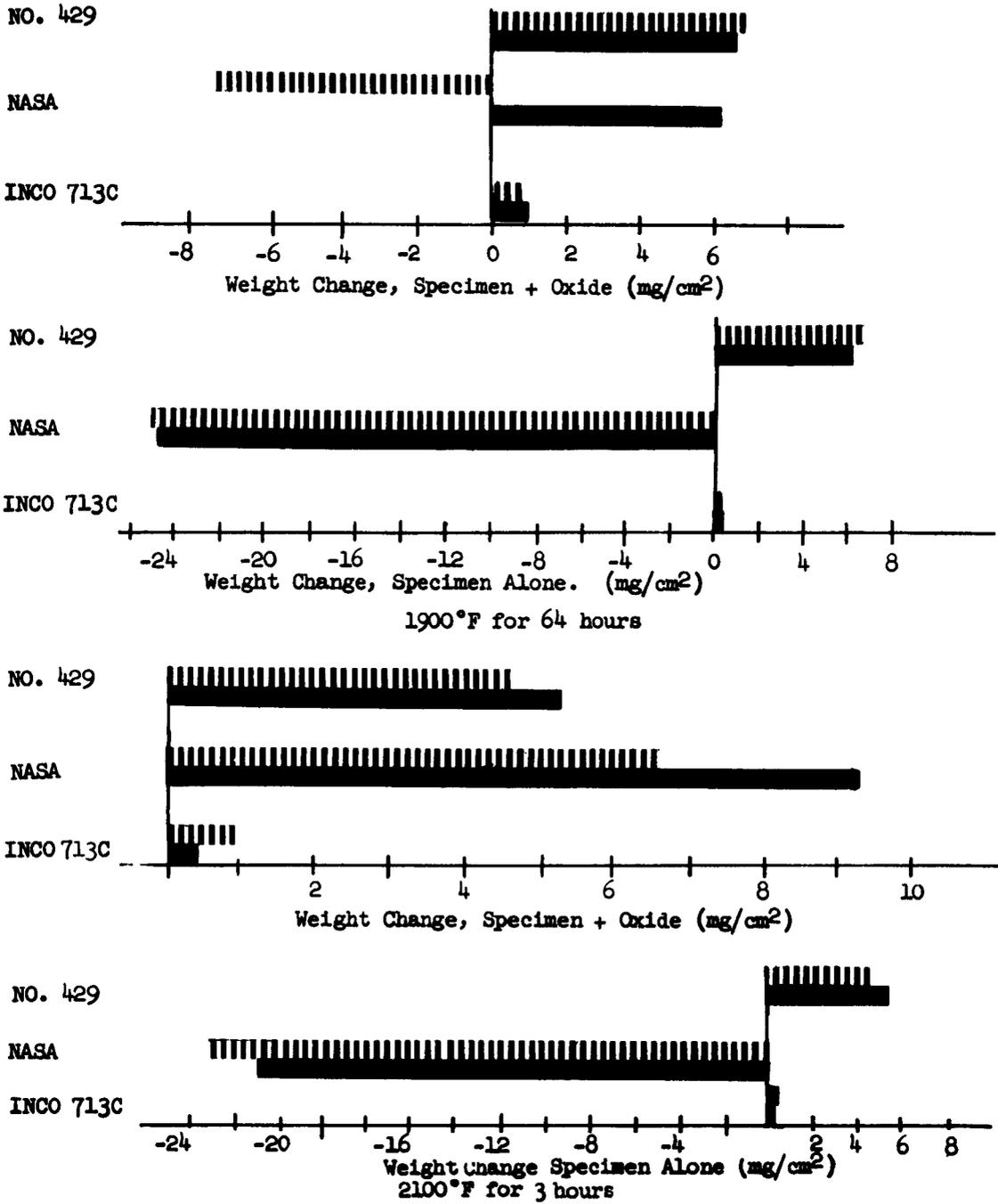


Figure 74. Oxidation Tests Run On As-Cast Inco 713c, NASA Alloy and No. 429.

modified Inco 713c. Figures 75 and 76 show the effects of heating these alloys to temperatures in the 2250°F and 2500°F range on the micro-structure of the alloys. It would appear from this data that the NASA alloy has a solidus temperature of between 2250°F and 2300°F and that the Ta-C modified Inco 713c has a solidus temperature of about 2250°F. Figure 77 shows photographs of the specimens heat treated in air in the 2250°F to 2500°F temperature range for these same two alloys. (It will be noted that the NASA alloy has a far greater tendency to scale at these temperatures than the Ta-C modified Inco 713c). It would also appear that the liquidus temperature of the Ta-C modified Inco 713c (Melt 390) approximates 2450°F and that the liquidus temperature of the NASA alloy is in excess of 2500°F.

4. Phase Identification

Both No. 429 and NASA alloy have large amounts of precipitated phase in their microstructure. Samples of both alloys were digested in chemical solutions designed to preferentially dissolve the matrix metal leaving the precipitate phase as an undissolved residue. Efforts to determine the chemical composition of the precipitate by X-ray diffraction means failed because the observed data did not match any of the compounds listed in the standard catalog. However, the specific gravity of the base alloys and of their precipitated phases were determined together with the percentages of precipitated phase. This data is summarized in Table 26.

5. Preliminary Fabricability Evaluation

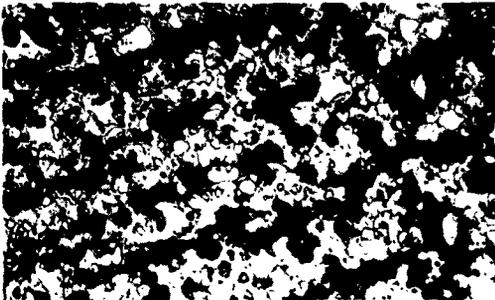
a. Investigation of Methods of Machining Test Bars

Initial efforts to machine tensile test bars from these nickel base alloys using high speed tool steel cutters were totally unsuccessful due to extremely rapid tool breakdown. Carbide cutters were successful in machining some of the nickel alloys, but resulted in a high proportion of ruined specimens in some other experimental alloys notably alloy No. 429. In all cases, tool wear was quite rapid and both tool and specimen vibration were serious problems. As a result of these difficulties, several other methods of tensile test specimen preparation were investigated.

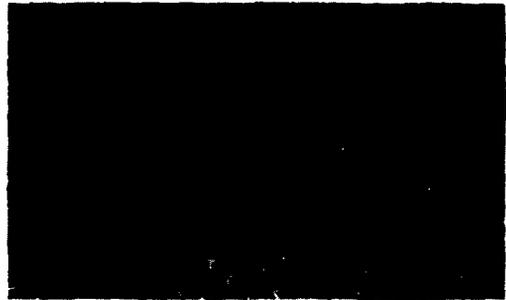
A number of trials were made of electric spark discharge machining of alloy No. 429 using two different machines and a variety of settings on both machines.



1/2 Hour at 2250°F, Melt 387,
Water Quench.



1/2 Hour at 2300°F, Melt 387,
Water Quench.



1/2 Hour at 2400°F, Melt 387,
Water Quench.

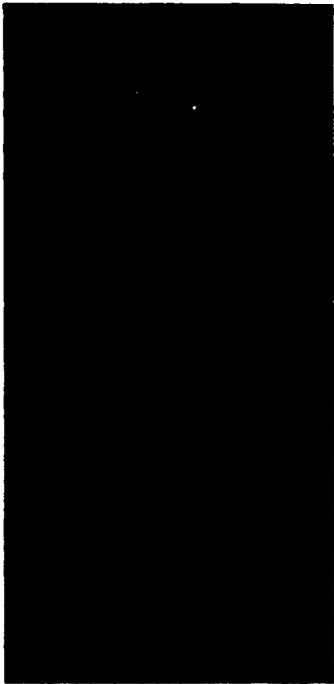


1/2 Hour at 2450°F, Melt 387,
Water Quench.

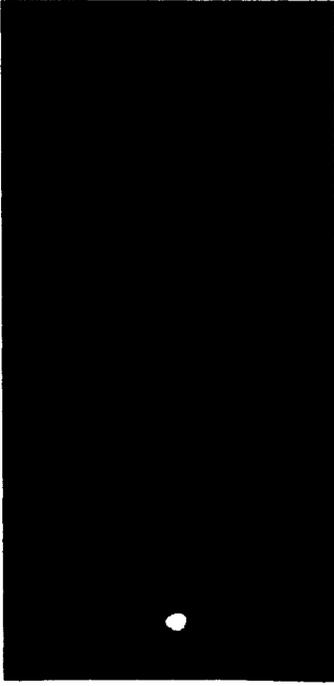


1/2 Hour at 2500°F, Melt 387,
Water Quench.

Figure 75. Effects of Heat-Treatment Temperatures Near the Solidus Temperature on the Microstructure of NASA Alloy. Etch No. 4. 100X.



2250°F.



2300°F.



2350°F.



2400°F.

Figure 76. Effects of Heat-Treatment Temperature Near the Solidus Temperature on the Microstructure of Ta-C Modified (16.26% Ta, 1.06% C) Inco 713c, Melt 390, Water Quenched. Etch No. 4.

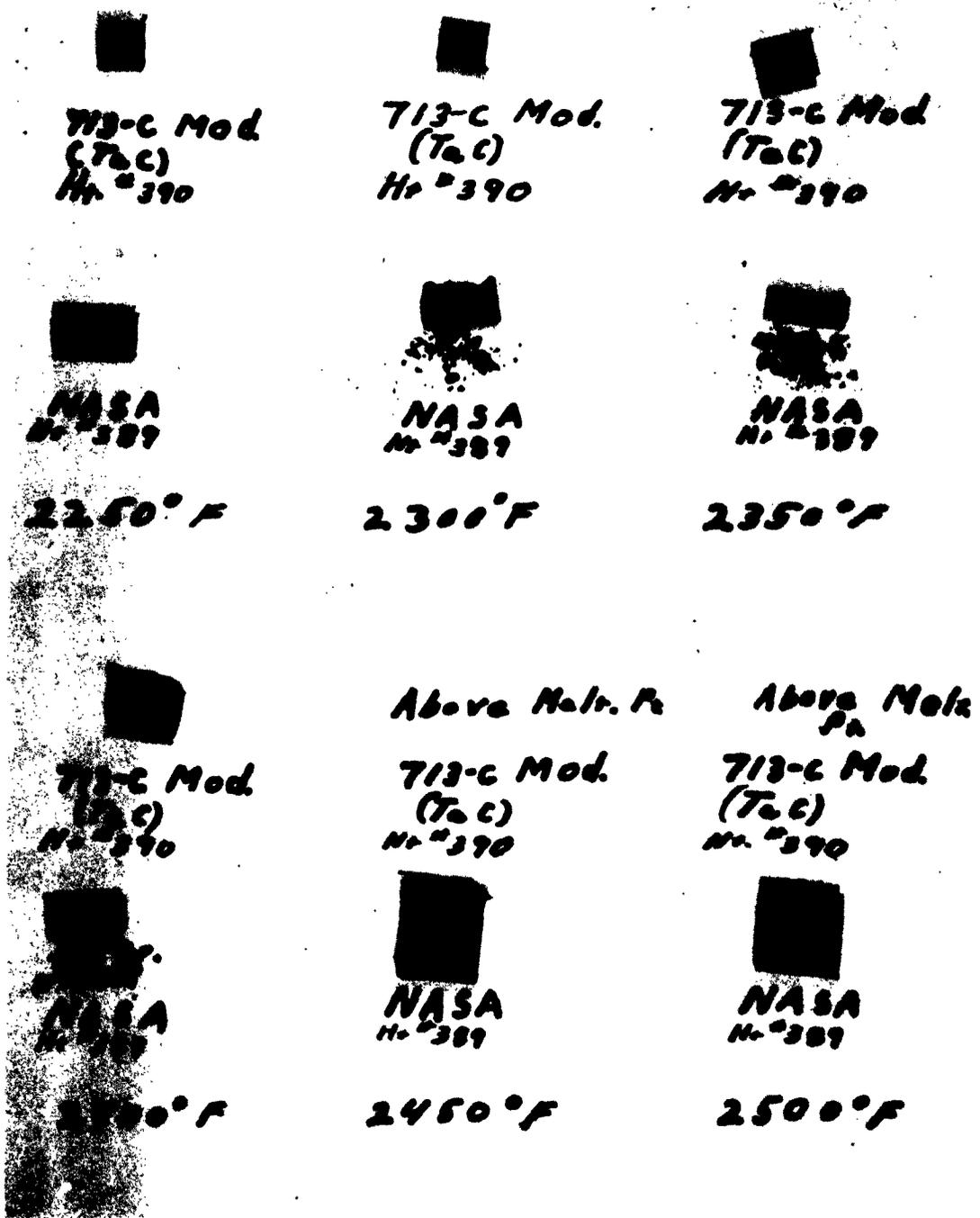


Figure 77. Specimens Heated to Various Temperatures for One-Half Hour and Quenched to Determine Solidus Point for Two Alloys.

TABLE 26
SPECIFIC GRAVITY OF NO. 429 AND NASA ALLOY

Alloy and Condition	Specific Gravity	Remarks
No. 429 as cast	8.847	
No. 429 precipitated phase ^a	7.284	35% by wt. 42.4% by volume
No. 429 dissolved portion ^b	9.97	65% by wt. 57.6% by volume
NASA alloy as cast	8.397	
NASA alloy rolled	8.472	
NASA alloy precipitated phase ^a	4.041	13.9% by wt. 28.8% by volume
NASA alloy dissolved portion ^b	10.12	86.1% by wt. 71.2% by volume

a The residue collected on 2 micron filter after dissolving as cast specimens in concentrated HCl + ferric chloride solution

b By calculation

A variety of cutting electrode compositions were tried including the following:

1. Brass
2. 4340 Steel
3. Graphite
4. Silver-tungsten carbide alloy

All of these electrodes eroded rapidly during the test without substantially cutting the No. 429 alloy. This investigation failed to develop any practically useful methods of electric spark discharge machining of this alloy.

A number of test specimens were successfully produced by the following grinding procedures:

1. Saw blanks out with a reinforced alundum cutoff wheel (Norton No. TBNX24) under a flood of coolant.
2. Rough grind to test bar contour by hand on a silicon carbide grit bench grinder wheel.
3. Belt grind to approximate final contour using a silicon carbide grit belt.
4. Finish grind to size on a surface grinder using a silicon carbide grinding wheel.
5. Drill required holes with a carbide tipped drill.

The most rapid procedure developed to date for cutting tensile test bars from these nickel base alloys is as follows:

1. Saw blanks with a silicon carbide cutoff wheel under a flood of coolant.
2. Using a belt grinder of special design constructed for the purpose at Vought, Figure 78, cut the contour of the test bar in a single operation using wet or dry, 40 grit, silicon carbide, abrasive belts under a flood of coolant.
3. Drill required holes with a carbide tipped drill using a rigid drill fixture to position the drill.



Figure 78. Vought Sheet Specimen Grinding Apparatus.

V. CONCLUSIONS AND RECOMMENDATIONS

A. Conclusions

The work done under this contract has resulted in the following developments:

1. It has been demonstrated that thin cast slabs of nickel base superalloys can be directly hot rolled into sheet without the need for intermediate forging or extrusion operations. This greatly simplifies the procedure for obtaining sheet of a new cast alloy.
2. Inco 713c rolled and heat treated sheet has a 1900°F ultimate tensile strength of about 50,000 psi at strain rates less than ASTM standards, with good room temperature strength and ductility, and good oxidation resistance. This represents about a 200°F increase in useful temperature over the best previously available superalloy sheet, Rene' 41. The rate of strain during tensile tests has a substantial effect on the tensile strength reported, especially 1900°F. The strain rate in the elastic range used in most of the tensile tests in this program was calculated to be less than .005 in/in/minute as established by ASTM standard procedure. (Note: See Appendix C, pg.123). Higher strain rates were observed to give higher strengths.
3. NASA Ta28 rolled and heat treated sheet has a 1900°F tensile strength of about 50,000 psi with poorer ductility and oxidation resistance than the Inco 713c sheet. On the other hand, NASA Ta28 alloy has a higher recrystallization temperature and hence might have higher stress rupture strength at temperatures of 1900°F and above.
4. Nickel base alloys can be strengthened by precipitation hardening using complex refractory metal carbides as the precipitating constituent rather than nickel aluminate.
5. The use of a rigid rolling mill, which does not depend on hold down screws for positioning of the movable roll, makes it possible to roll metals and alloys having lower ductility at rolling temperature than is possible with rolling mills of conventional design. Indications are that this mill is capable of rolling high temperature alloys down to foil thickness.
6. Long time solution heat treatments gave the best improvement in the ductility and tensile strength of Inco 713c of all the heat treatments tried. The strength and ductility of

of NASA alloy is affected more by aging heat treatments than solution heat treatment.

7. A limiting factor in this program, with respect to further improvements in alloy compositions, was the crucible materials employed. Specifically, when high percentages of tantalum were added to melts, reaction between the molten metal and the crucible occurred.
8. Thin cast slabs of Inco 713c can be directly rolled into 25 mil sheets up to 10 x 16 inches in size.

B. Recommendations

1. It is recommended that the basic techniques of casting thin slabs into sheet be applied to the new NASA nickel base alloys and to the refractory metals. It is believed that this technique will permit a great reduction in the time required to get a new alloy from discovery to commercial use.
2. The tensile properties of Inco 713c sheet obtained in this program indicate that its strength at 1900°F offers the potential of a substantial advancement over presently available nickel base alloy sheet and foil. It is recommended that investigations be initiated to obtain additional mechanical, and physical property data needed for further evaluation purposes. These investigations should include tensile and stress rupture strength, notch sensitivity, tensile strength as a function of strain rate, compressive strength, and limited fabricability evaluations.
3. The complex refractory metal carbide precipitation hardening system used in alloy No. 429 should be further investigated as a possible means of producing superalloy sheet which would have usable strength at temperatures up to 2400°F combined with good oxidation resistance.
4. Improved melting furnace equipment is urgently needed to eliminate crucible-metal reactions. One possible approach is the use of water cooled metal crucibles for melting the alloy prior to casting.

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APPENDIX A

MELTING AND CASTING TECHNIQUES AND EQUIPMENT

A. Melting Techniques

All melting of nickel alloy done in this program was done in the Vought designed Plasma-Resistance furnace. The particular furnace used in this program is shown in Fig. 79 and has a melting capacity of about 25 pounds of nickel base alloy. This furnace operates at all times under an argon atmosphere which can be at atmospheric pressure or at reduced pressures down to 30 microns. All melting operations done under this contract were performed under an atmospheric pressure of argon, although the argon pressure in the furnace was reduced to the 30 to 1000 micron level for the casting operation. In addition to melting under an argon atmosphere, argon gas is bubbled through the melt from the time melting commences until pouring operations start. The purpose of this is twofold; one, to mix the alloying constituents in the melt; and two, to remove gaseous impurities, such as oxygen, hydrogen, and nitrogen. (See Table 10, page 37).

The heat source for preheating the furnace prior to charging of the metal is a low voltage, high amperage carbon arc. The arc length under these conditions is approximately 1/2 inch and in general the arc behaves normally in spite of the argon atmosphere. After the furnace interior reaches a temperature of 2800°F, the electrical characteristics of the argon gas change. As a result of this change, the gap between the electrodes is increased to about 8 inches without increasing voltage or decreasing amperage. The consumption of the graphite electrodes also essentially ceases. Under these conditions the flow of current through the furnace can be interrupted and restarted without change in the electrode position, and without the use of high frequency or high voltage starting currents. Experience with this furnace over a period of time has shown that alloys melted under these conditions will not pick up any carbon from the furnace atmosphere. In fact, carbon losses are frequently encountered due to the removal of oxygen from the melt, apparently as carbon monoxide.

During this program both magnesia and zirconia refractory crucibles have been used in this furnace for melting nickel base alloys. In both cases, there is evidence that there is some reaction between the molten metal and the crucible. In the case of the zirconia crucible, this reaction results in an increase in the zirconium content of the metal. Since zirconium is generally considered to be a desirable alloying element in nickel

base alloys in small percentages, this reaction is not detrimental unless excessive amounts of zirconium are added. In the case of the magnesia crucible, magnesium metal and boron are picked up by the metal as a result of the crucible reaction. Magnesium metal has a high vapor pressure at the temperatures used in melting nickel alloys, and for the most part is vaporized. The small amount of residual magnesium left in the alloy (See Table 7) does not appear to be seriously detrimental. The boron pickup in the alloy is in the range generally considered to be desirable for nickel base superalloys. (See Table 8). Therefore, for the particular nickel base alloys primarily concerned in this program, both magnesia and zirconia crucibles appear satisfactory. (Tables 7 and 8 on pages 36 and 37).

The crucibles used in this program were all fabricated at Chance Vought from purchased refractory grain. The refractory materials used in this program are identified in Table 9, page 37. The crucibles were fabricated by the following process:

- a. Mix dry grain with water to form a damp sand.
- b. Hard ram or press the damp refractory into a steel flask around a wood form to form the crucible shape desired.
- c. Place the rammed crucible on a flat silicon carbide plate and remove both the wood form and the steel flask.
- d. Dry the crucible on the silicon carbide plate for four hours at 500°F.
- e. Fire the dried crucible in an air atmosphere in a furnace whose temperature is gradually raised from 500°F to 2500°F over a period of 8 hours.
- f. Furnace cool the crucible and install in the melting furnace.

Of these crucibles, those made from Magnorite X were the most generally satisfactory and had the longest service life. The zirconia crucibles were relatively fragile and subject to damage from thermal shock. As a result, their life in the furnace was quite limited. The crucibles made from RM 1152 tended to soften at service temperatures and underwent post-firing shrinkage. This caused early failure of the crucible roof with resultant short crucible life.

In most of the alloys melted, crucible reactions were not serious problems. However, in the few high tantalum nickel base alloys melted crucible reactions were very serious and resulted in the removal of much of the tantalum from the alloy and serious damage to the crucible. (See Table 6, page 36). Melt Nos. 395 and 396 are not shown in Table 6 because they resulted in the loss of both crucible and

melt and hence no analysis of the melts was possible. Melt Nos. 395 and 396 had 40.56% tantalum and a total of 43.08% refractory metal content. The trend in improving superalloys appears to be in the direction of increasing refractory metal content. It is apparent that any substantial increase in refractory metal content is going to require the development of improved crucible materials or different melting techniques. Water cooled metal crucibles may be an answer to this problem, although conventional consumable arc melting of ingots in water cooled copper crucibles does not appear to be a satisfactory answer.

The general procedure used in making all melts in this program is as follows:

- a. Preheat the melting furnace to about 3000°F.
- b. Charge and melt the major constituent. In alloys being made from individual constituents, this would be nickel. In alloys made by modifying an existing alloy, this would be the starting alloy; for example, purchased Inco 713c vacuum melted bar.
- c. Charge and melt alloying constituents, adding the refractory metals first and the volatile elements last.
- d. After all alloying additions are made, maintain the melt at temperature and bubble with argon gas for 15 to 20 minutes to thoroughly mix the alloy and to remove gaseous impurities.
- e. The metal is then raised to pouring temperature and poured.

B. Casting Technique

All castings made in this program were made in the plasma-resistance melting furnace using preheated ceramic molds press formed on a molding press. This molding and casting technique permits the casting of relatively large, thin sheets of nickel base superalloys, as well as a wide range of other alloys.

The ceramic molds used in this program were press formed on a molding press from both silica and zircon base refractory powders. No evidence of metal-mold reaction has been found with either molding composition, if pouring temperature is controlled in the 2700-2900°F range. In practice these powders are mixed with ceramic and chemical bonding agents in minor percentages, dampened to form a workable material, and pressed into mold halves under a mold face pressure of about 500 pounds per square inch. The pressed mold half is quite weak at this point and must be cured prior to further handling. This curing is normally done by baking the mold at from 300 to 500°F for

3 to 8 hours depending on size and the particular bonding materials being used. This cured mold is quite hard and strong and can be sawed, drilled, or otherwise machined as may prove advisable. Through this program the gates and risers were machined into the cured mold. Initially the cavity for the formation of the sheet casting itself was formed into the mold at the time of pressing by use of a pattern. Fig. 83 shows such a pressed and cured mold half. Later in the program castings were made in molds where the cavity for the sheet itself was also machined. The cured ceramic molds are easily and rapidly machined using carbide tools or abrasive wheels and belts. It was found that under some conditions some slight warpage and distortion of the molds occurred during curing. It was found that the highest degree of uniformity of thickness in the cast sheet could be obtained by machining the sheet cavity after curing of the mold. No further warpage of the mold during the preheating and pouring operations has been observed.

The cured and machined mold halves are assembled into stacks suitable for pouring, and the outside surfaces of these mold stacks are coated with a slurry of refractory grain, southern bentonite, and water. The purpose of this coating is to prevent the mold halves from shifting relative to each other during handling, preheat and pouring. These mold stacks are placed in the preheat furnace and preheated to from 1200°F to 1800°F on a pre-determined heating cycle. This heating cycle is calculated to permit the uniform heating of the mold stack without mold cracking due to thermal shock. (See Fig. 84). The preheated mold stack is placed in the melting furnace, which contains a charge of molten metal ready for pouring and clamped in place. Fig. 80 shows the mold stack in this position. The vacuum tight access door is closed on the furnace and the pressure inside the furnace chamber reduced to from 30 to 1000 microns. Fig. 81 shows the furnace in this condition. The entire furnace chamber is then rapidly rotated 180 degrees, almost instantly pouring the molten metal charge through a 3 inch diameter pouring neck into the evacuated mold. Atmospheric pressure of argon gas is then restored in the furnace chamber to accelerate mold cooling. This combination of almost instantaneous pouring with the use of a hot evacuated mold permits the filling of relatively large areas of sheet down to 30 mil thicknesses. Fig. 82 shows the furnace in the poured position. After pouring the furnace remains in the poured position for about 15 minutes to be certain that the casting has completely solidified. At the end of this time the furnace is returned to its original position, the vacuum tight access door is opened, and the poured mold stack is removed.

The poured mold stack is allowed to cool in air to room temperature. It has been found that this minimizes warpage and distortion of the cast sheet. After cooling to room temperature the ceramic mold is broken away from the casting and the cast sheets are sawed off from the gates and risers. These cast sheets are sand blasted, X-rayed, Zyglo inspected, and are then ready for rolling. The "tree" of

cast sheets, after the ceramic mold is broken away and before the sheets are sawed off, is shown in Fig. 85. Fig. 86 shows another larger cast sheet with the riser still attached. This sheet is about 0.1 inch thick and approximately 8" x 16" in size. This larger sheet represents the largest mold which can be made on the existing molding press at Vought but is not believed to represent any maximum size limitation of the process as a whole. The risering shown in Fig. 86 gives a higher degree of X-ray soundness in the sheet as a whole than does the risering shown in Fig. 85. Fig. 86 represents a later risering procedure.

Efforts have been made to roll thicker cast slabs and slices from ingots in this program. These efforts have been unsuccessful, demonstrating the practical need for casting the starting sheet in thicknesses approximating 0.1 inch.

This necessity of starting with thin cast sheets has a good technical explanation. Figure 87 shows a typical simple eutectic type of phase diagram. The vertical dashed line represents a particular alloy composition being cooled from the liquid state to room temperature. As this alloy cools in the form of a sheet casting (Shown in Fig. 88) the first crystallites of solid metal to form are composition A. The remaining liquid at this point has a composition of A'. If cooling continued slowly enough for equilibrium conditions to occur, the composition of the solidified metal would be consecutively represented by B, and C plus C'. Under conditions of rapid solidification, as it occurs in a casting, the first metal to freeze has composition A. As cooling continues under these conditions, the composition of subsequent layers of metal deposited on the initial crystal of metal have compositions represented by B and C and the points of the solidus curve intermediate between these points. As this metal freezes out of the liquid under these non-equilibrium conditions the composition of remaining liquid shifts to the right of the equilibrium compositions represented by B' and C'. The last metal to freeze will be in the grain boundaries, and will have a composition represented by a point to the right of C'. Figure 89 shows a simple solid solution phase diagram with a dashed line representing a particular alloy composition cooling from above the liquidus temperature to room temperature. The actual effect on the metallurgical structures of the casting remains the same as is shown in Figure 88. The significance of all this is that any casting of an alloy composition will have substantial variations in chemical composition within each grain, and that the grain boundaries always represent the minimum melting point composition. The diagrams shown are for relatively simple phase diagrams. In the case of more complex phase diagrams, particularly those involving peritectic reactions, the depression of the melting point at the grain boundaries can be even greater. Diffusion is extremely rapid in liquid metals and relatively speaking, very slow in solid metals at any temperature. Therefore, the larger the grain size of the casting the greater the degree of macrosegregation

resulting from the mechanism of solidification. The larger the section cast, the slower will solidification take place, and the larger the grain size. With this larger grain size a larger zone of low melting point constituent will be formed at the grain boundaries. In nickel base alloys at least, this lower melting point grain boundary constituent appears to be substantially weaker at elevated temperatures. Hence, when efforts are made to hot roll these alloys cast in relatively thick sections, failure occurs along the grain boundaries.

In any alloy casting there will be some degree of segregation. If the casting is relatively fine grained and is a nickel alloy, this cast alloy sheet can be hot rolled. Figure 90 shows the relative structure of the cast alloy as shown in Figure 88 after it has received a substantial amount of reduction in thickness by rolling. The S spacing between the centers of segregation has been greatly reduced from Figure 88 to Figure 90 by virtue of the rolling operation even without any effect of heat treatment. Figure 91 and 92 show these same alloys after heat treatment. It will be noted that the physical form of the segregation has tended to change and to be diminished. However, even after heat treatment the rolled structure has a closer spacing, S_r , between points of segregation. Also the rolled structure has the higher melting point constituents lined up in the direction of rolling, while in the cast structure the higher melting point constituents are lined up normal to the sheet surface. As can be seen in Figure 93 alignment of segregation in the cast structure favors early failure under tensile stress and the path of crack formation is short. This tends to result in low tensile strength and ductility, particularly at room temperature. Figures 94 and 95 show a rolled structure under tensile load. The strong segregated structure in this case is in the direction of stress and the path of crack formation is relatively long. This favors higher tensile strength and greater ductility. Furthermore, in the heat treated rolled structure, since the S spacing is already much smaller as a result of rolling, the time to effect equilibrium conditions by diffusion is far shorter than it would be on an as cast structure. Since the rate of diffusion for substitutional elements is usually an exponential related to distance, this smaller S value for the rolled alloy can result in an adequate heat treatment time for a rolled alloy being several orders of magnitude less than it is for a cast alloy. In both cases, there is no effective substitute for keeping original macrosegregation to a minimum by rapidly freezing the metal in a fine grain size. In the case of complex nickel base alloys, this would appear to be best done by casting thin sheets.

The proper selection of mold preheat and metal pouring temperatures are vital to the achievement of useful castings. In the case of the NASA TaZ8 alloy, it was found that a mold preheat temperature of 1800°F resulted in serious hot tearing of the sheet casting while mold preheat temperatures of 1600°F and below resulted in cast sheets

free of hot tearing. Excessively high or uncontrolled pouring temperatures resulted in mold-metal reactions that damaged both the surface finish and the properties of the cast sheet. In general, the optimum conditions for casting sheet from nickel base alloys has been found to be a mold preheat temperature of from 1200 to 1600°F and a metal pouring temperature of about 2800 to 2850°F.

Most castings made in this program were made in silica base molds because of previous good experience in casting nickel base alloys in molds of this composition. Silica base molds have a greater tendency to crack due to thermal shock during mold preheat than zircon or alumina base molds. Therefore both alumina and zircon base molds were used for a limited number of melts in this program. As a part of this investigation a series of Inco 713c melts were cast into zircon molds. All of these castings showed substantial gas porosity. This is thought to be due to occluded gas in the mold caused in part by the very fine particle size of the available zircon. The original castings had been made in silica molds and when the use of silica molds was resumed with melt No. 544 the gas porosity problem disappeared. Earlier zircon molds made with coarser zircon flour had been successfully used to make NASA alloy castings. This tends to corroborate the gas occlusion explanation for the gas porosity in the Inco 713c castings made in fine zircon flour molds.

Figure 79. Melting and Casting Furnace in Empty Position.



1. Furnace shell containing the crucible.
2. Vacuum tight door.
3. Chamber for containing the mold stack during pouring.
4. Air clamps for clamping the mold in place during pouring.
5. Vacuum hose connection to vacuum pump for evacuating the furnace chamber.
6. Fixed support stand in which the furnace shell rotates during pouring.
7. Machined face of the furnace against which the vacuum tight door (2) fits during the pouring operation.
8. Electrode holder electrically insulated from the main shell.
9. Trunnions of the furnace which rotate during the pouring operation.

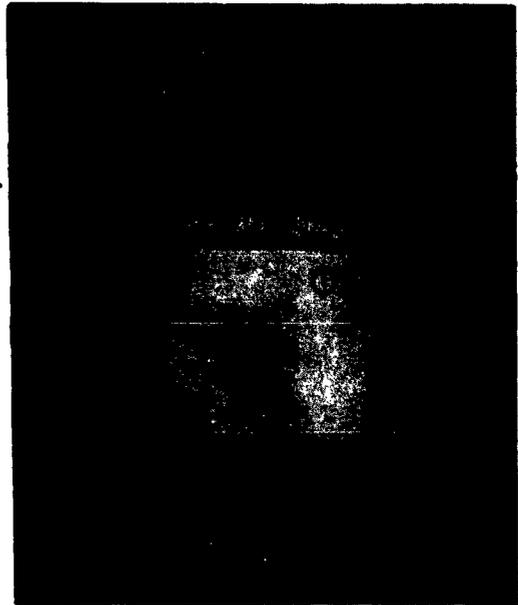
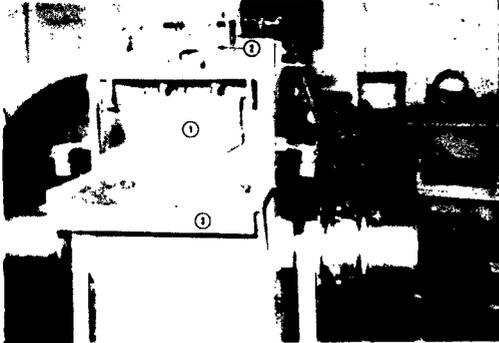


Figure 80. Melting and Casting Furnace with Mold in Prepouring Position.

1. Hot mold stack clamped in pouring position.
2. Air clamps clamping the mold in position.
3. Vacuum tight door.

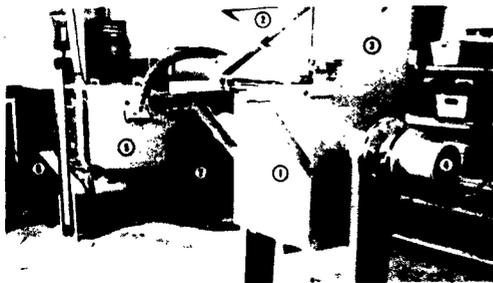


Figure 81. Melting and Casting Furnace Loaded and Ready to Pour.

1. Furnace shell containing the crucible.
2. Vacuum tight door locked in place and furnace chamber evacuated.
3. Portion of the furnace shell containing the hot mold stack.
4. Vacuum bell seals over the electrode holders.
5. Cold trap in the vacuum line.
6. Mechanical vacuum pump used to rapidly evacuate the furnace chamber.
7. Welding transformer used to supply power for the melting furnace.

Figure 82. Melting and Casting Furnace in the Poured Position.

1. Portion of the furnace shell containing the hot poured mold stack.
2. Vacuum bell seals over the electrode holders.
3. Argon inlet line into the furnace chamber.
4. Vacuum hose connection to the furnace chamber.
5. Furnace shell containing the crucible.



Figure 83. Pressed and Cured Ceramic Mold Half.

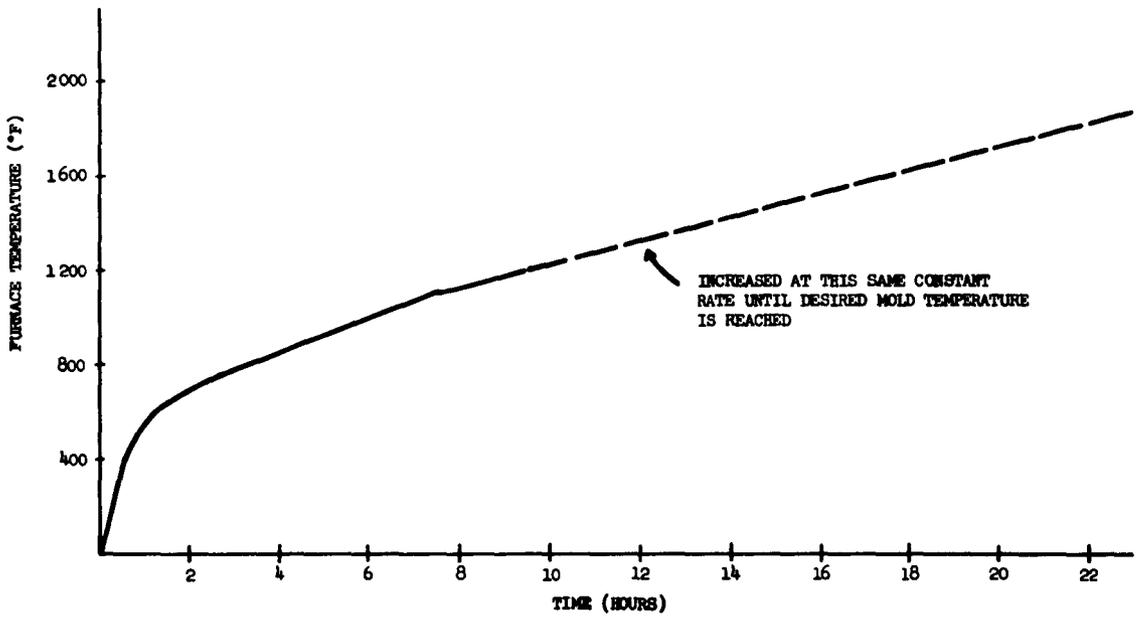


Figure 84. Typical Mold Preheat Cycle.



Figure 85. Cleaned Casting with Cast Sheets Still Attached to the Riser.

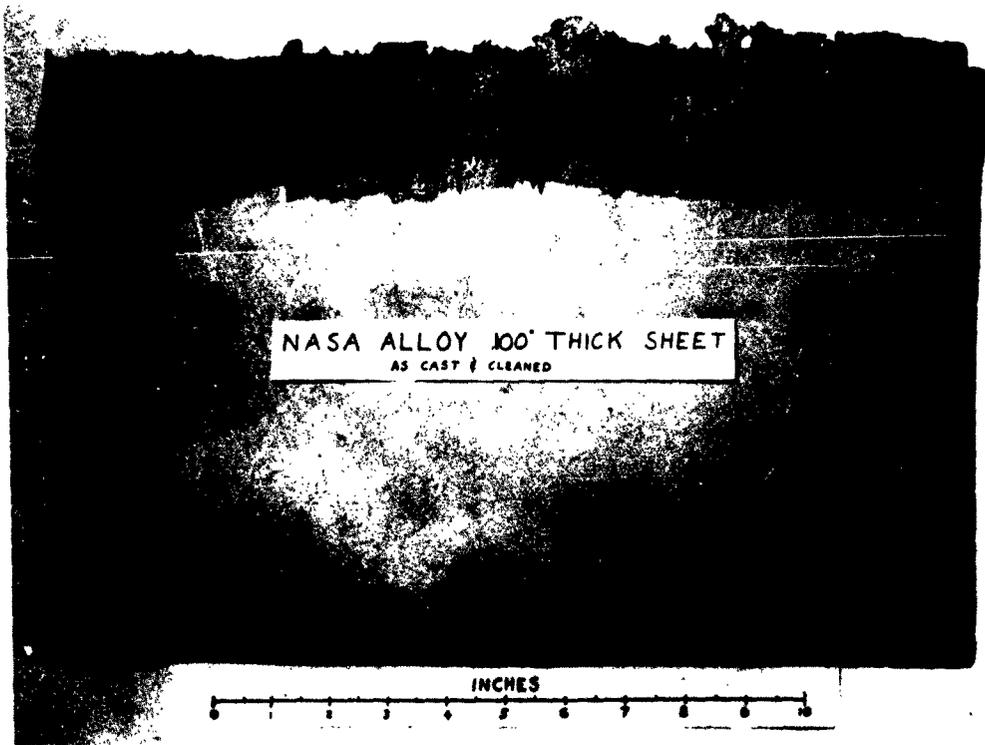


Figure 86. NASA Alloy .100" Thick Sheet (as Cast and Cleaned).

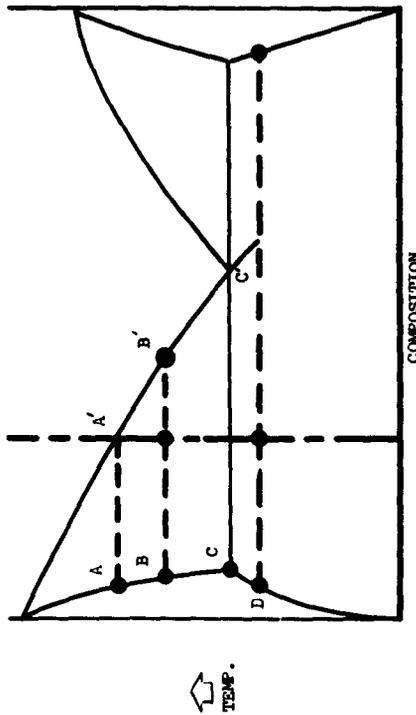


Figure 87. Solidification in a Typical Simple Eutectic System.

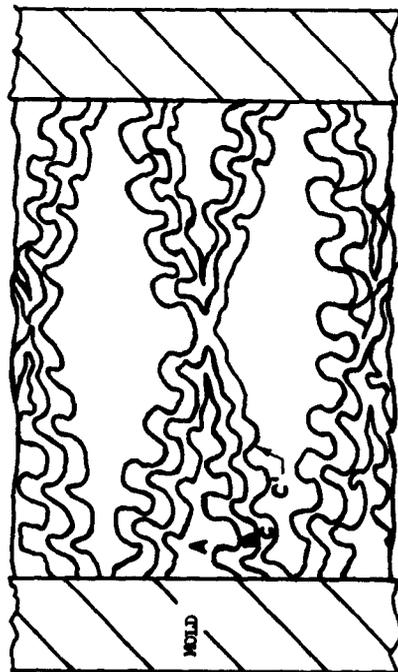


Figure 88. Typical Solidification Mode in a Casting.

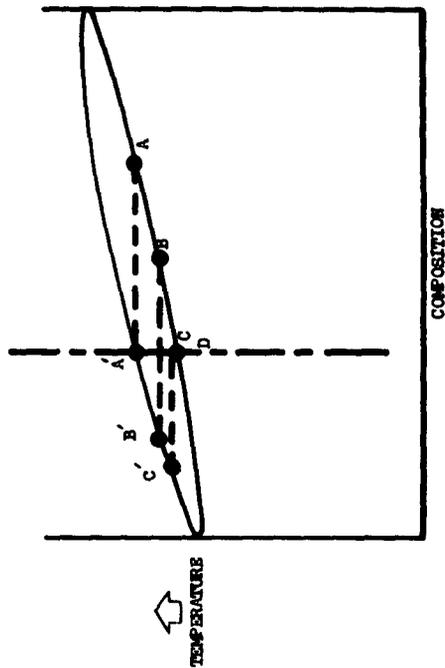


Figure 89. Solidification in a Simple Solid Solution System.

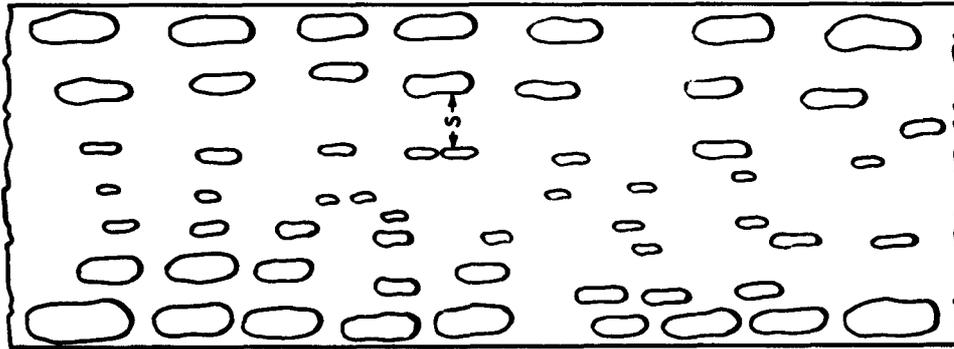


Figure 90. Typical Redistribution of Macrosegregation in a Rolled Metal by Heat Treatment.

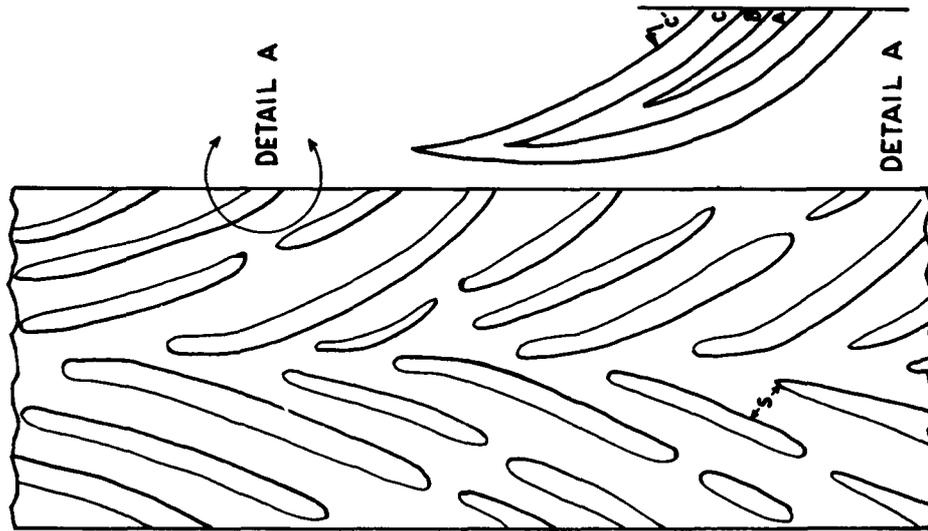


Figure 91. Typical Redistribution of Macrosegregation in a Cast Metal by Rolling.

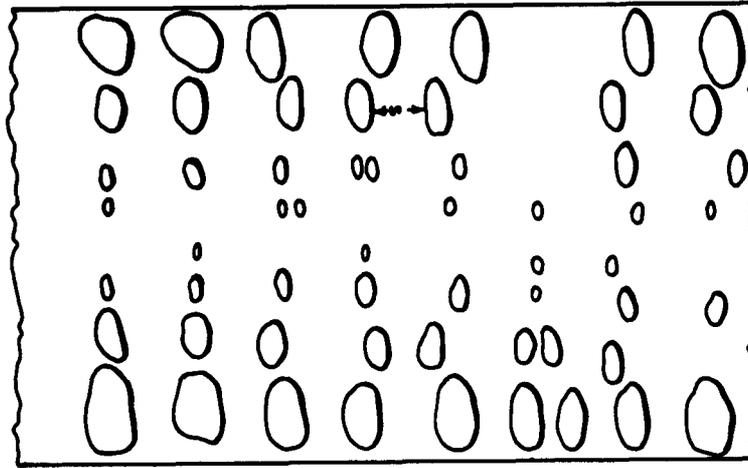


Figure 92. Redistribution of Macrosegregation in a Cast Metal by Heat Treatment.

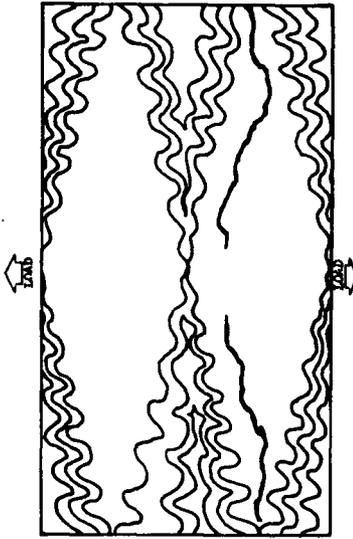


Figure 93. Mode of Tensile Failure in a Cast Structure.

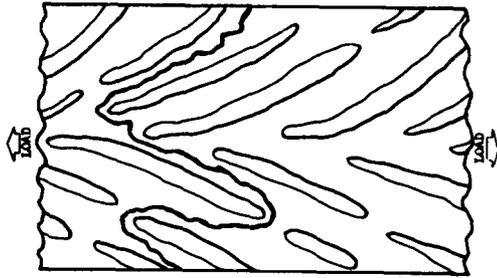


Figure 94. Mode of Tensile Failure in a Rolled Structure.

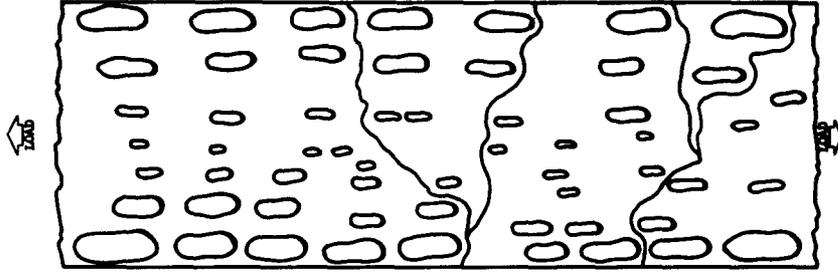


Figure 95. Tensile Failure in a Rolled and Heat-Treated Structure.

APPENDIX B

THE RIGID ROLLING MILL

A. Statement of the Problem

Difficulties encountered in rolling nickel alloy sheet on conventional rolling mills caused Vought to design and construct a rolling mill of new and unique rigid design in support of this program. Many of the problems encountered in rolling of nickel base alloys on conventional mills appeared to be related to "crowning" of the sheet during rolling and to an inability to adequately predetermine and control the position of the movable roll during the rolling operation. Crowning of the sheet occurred in the early stages of rolling nickel alloy sheet on conventional mills. In subsequent passes the thicker section in the center of the sheet tended to elongate more rapidly than the outside thinner sections. This put the outside sections under tensile stress and this in turn caused serious edge cracking which soon penetrated to the center of the sheet. This problem could be minimized by a process of repeatedly removing the thinner outside edge sections of the rolled sheet by sawing them off. However, this rapidly becomes self-defeating as the width of sheet available for rolling diminishes to the vanishing point. The inability to control the position of the movable roll contributes largely to the crowning problem and also makes it impossible to provide a uniform reduction per pass over the entire sheet of material being rolled. As a result there are substantial variations in the thickness of the rolled sheet even aside from the crowning problem. These variations contribute to later cracking problems. An analysis of the problem of rolling mill design shows that "crowning" of rolled sheet can be caused by two different conditions as follows:

1. Bending of the rolls caused by the separating force of the metal sheet passing between the rolls. For 6" diameter steel rolls, rolling nickel base superalloy sheet 12 inches wide, this crowning can be calculated to be approximately 3 mils maximum. The experimental results obtained on the rigid mill at Vought corroborate this quite closely.
2. Random, uncontrolled motion or cocking of the movable roll due to the inability of the hold down screw to seat precisely the same way each time or due to unequal deflection of the hold down screws. This feature of conventional mills has been observed to cause crowning of up to 20 mils in a 4 inch wide sheet.

The Vought mill was therefore designed to eliminate the hold down screw method of construction with all of its resulting problems.

B. Description of the Rigid Mill

The rigid mill can be used either as a two high mill with 6 inch diameter working rolls and no back up rolls, or as a four high mill with 1 1/2 inch diameter working rolls and 6 inch diameter back up rolls. This rolling mill is driven by a 30 horsepower electric motor operating through a chain drive. All four rolls in the four high configuration are driven. Figs. 96 and 97 show the rigid rolling mill and associated metal preheat furnace. The significant characteristics of the construction of this rolling mill are summarized below:

- a. The mill can be used as either a 2 high or a 4 high mill.
- b. In the 2 high configuration, the working rolls are 6" in diameter with a 12" working length. The rolls are hardened steel and the bearings are bronze sleeves having 4" ID.
- c. In the 4 high configuration, the working rolls are 1 1/2 inch diameter hardened steel with brass bushing blocks for bearings. The back up rolls are the same 6 inch diameter rolls used as working rolls in the 2 high configuration.
- d. The mill is capable of withstanding a separating force of approximately 900,000 pounds. The separating force experienced is recorded by means of suitable instrumentation connected to strain gauges attached to the columns of the mill. The maximum separating force experienced to date has been 167,000 pounds.
- e. The column area is 29.6 inches. The rolling mill does not have any hold down screws so this column area directly reflects the true rigidity of the mill.
- f. The maximum mill deflection at 900,000 pounds separating force is approximately 40.5 mils if no preloading is applied. The maximum mill deflection at the maximum separating force incurred to date (167,000 pounds) is approximately 7.5 mils if no preloading is applied. The maximum preloading force used to date has been 51,900 pounds. With this preload, a separating force of 62,280 pounds experienced in rolling 3 mil foil 3 inches wide corresponds to a mill deflection of .05 mils. The working rolls in the present rigid mill are 4340 steel hardened to Rc 38. When rolling superalloy foil, these rolls plastically deform, and this plastic deformation is the present limiting factor in reducing foil gauge. Tool steel or carbide work rolls would be expected to make possible the rolling

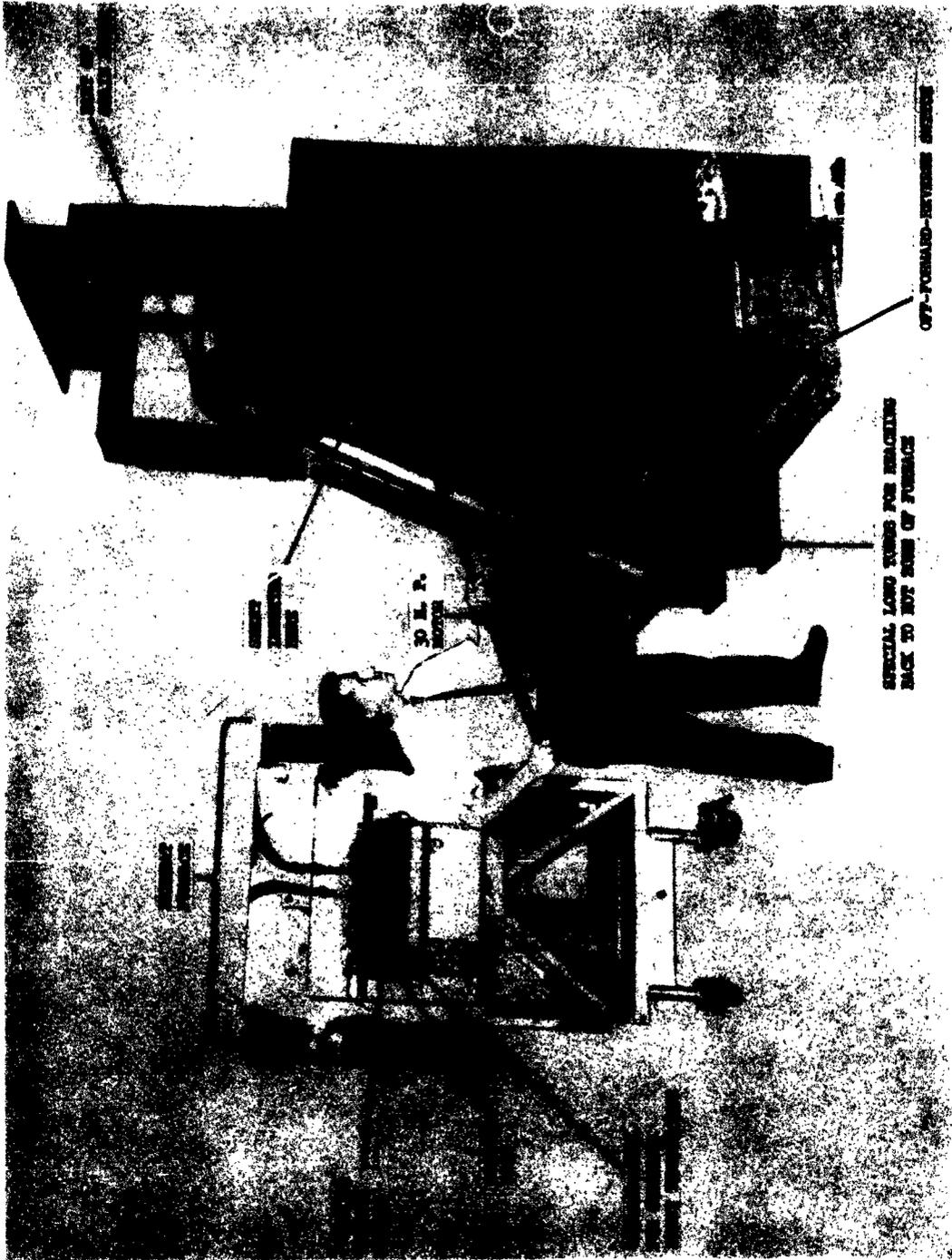


Figure 96. View of Vought Rolling, Mill and Preheat Furnace.



Figure 97. Vought Rolling Mill in 4-High Configuration.

of thinner gauge foil.

- g. Roll speeds of 97.5, 139.3 and 166.0 surface feet per minute are currently usable on the rigid mill. Other roll speeds can be obtained by changing one of the main drive sprockets, if desired.
- i. The position of the movable roll is determined by placing solid shims of predetermined thickness between the main bearing blocks of the movable roll and the mill frame.
- j. Preloading of the mill is accomplished by the use of tapered wedge blocks driving the movable roll toward the fixed roll.

C. Discussion of the Advantages of the Rigid Mill

This mill differs substantially from conventional rolling mills in several important ways. Rolling mills consist of two rolls or sets of rolls; one of which can be translated relative to the axis of rotation of the other roll or set of rolls. Both sets of rolls are rotated toward each other at the same speed. In a conventional rolling mill the movable roll or set of rolls is moved and then held in place by means of hold down screws connecting the bearing blocks of the movable roll with the rolling mill frame. In order for a screw to move it must have some clearance between the male and female portions of the screw mechanism. Due to this clearance, a screw will seat differently under load each time it is loaded, particularly if the load is not truly axial at all times. In a conventional rolling mill no load is normally applied to the hold down screw until the metal enters the rolls. Depending on how the metal enters the rolls, how uniform the entering metal is in thickness, the degree of metallurgical uniformity throughout the metal, and a number of other possible variables, the rolling load of the mill will be transmitted to the hold down screws in varying degrees of inequality and non-axiality. As a result, the true position of the movable roll with respect to the fixed roll can vary appreciably depending on how each of the two hold down screws may happen to seat under the load as applied. Hence, the movable roll may cock in one direction during one pass and in the opposite direction during the next pass. (See Fig. 98)

The hold down screws normally have a relatively small cross sectional area relative to the loads being applied and relative to the size of the mill frame. Hence, these hold down screws will compress elastically under load. Since this compression is elastic in nature, it will be proportionate to the particular load being momentarily applied to the particular hold down screw in question. Since the hold down screws are of relatively light construction, the elastic strain incurred will be a substantial portion of the total strain incurred in the mill as a whole. Both the random seating of the

hold down screw and its elastic deformation in compression serve to apply maximum rolling load to first one side of the sheet and then in subsequent passes to the other side of the sheet. This is the primary cause of "crowning."

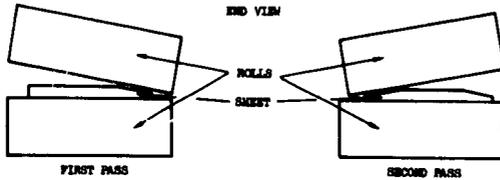


Figure 98. Diagram Showing Crowning as a Result of Two Successive Passes on Conventional Mill.

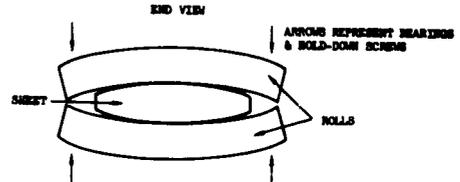


Figure 99. Diagram Showing Crowning Caused by Bending of the Rolls.

A secondary, but very minor, cause of crowning is the actual deflection of the roll itself. (See Fig. 99) When the actual rolling loads are determined experimentally by strain gauge measurements and the true deflection of the roll by bending during rolling is calculated, it is found that, for a 6 inch diameter working roll with no back up and with a 12 inch working width, bending deflection accounts for only a maximum crown of about 3 mils. This corresponds to the actual crown measured on an 11 inch wide sheet of Inco 713c rolled on the rigid mill. On a conventional mill 10 to 20 mils of crown was observed on a 4 inch wide sheet.

APPENDIX C

ELEVATED TEMPERATURE TENSILE TEST PROCEDURE

The test procedure used in tensile testing nickel base alloy sheet at elevated temperature is described and discussed below:

- A. A sheet tensile test specimen is gripped in the tensile test machine and three 36 gauge chromel-alumel thermocouples are spot welded to the surface of the specimen, one thermocouple being spot welded at each end of the gauge section and one at the center of the specimen. Each thermocouple consists of one chromel and one alumel wire, each separately spot welded to the specimen in close parallel position having the same longitudinal position on the specimen. The purpose of this is to assure the measurement of the specimen temperature and not just the temperature of a projecting bead. The thermocouples are connected to a Brown multipoint high speed strip chart recorder. Heating conditions are adjusted until all three thermocouples read the same temperature. Once test conditions have been standardized for a given set of specimens, further tests are conducted using only the center thermocouple in accordance with Standard Test Procedure ARTC-13-T-1, Rev. 6/1/59.(6)
- B. In gripping thin sheet metal specimens great care must be taken to assure that the method of gripping does not adversely affect the axiality of loading. When pin type grips are used, the specimen should be finally fastened in place while under a small tensile load with the specimen at room temperature. Otherwise, it is possible to cock the specimen in the grips during tightening of the pin bolt, resulting in nonaxial loading of the specimen.
- C. Where only tensile strength and elongation are required, elevated temperature tensile tests are run without the use of an extensometer. Hence, it is impossible to directly determine strain rates. Test machine head travel rates are not an adequate indication of strain, particularly during the early stages of the test. The reason for this is that the gripping and linkage devices used to attach the test specimen to the test machine strain when the testing load is applied to the specimen, and these may have a total strain substantially greater than that of the gauge length of the specimen.

In these tests, a constant loading rate of 16,000 pounds per square inch of test specimen per minute was used. Data published by Haynes Stellite for Rene' 41 (8), a nickel base superalloy, gives values of elastic modulus for this alloy of 16,200,000 at 1800°F and 8,200,000 at 2000°F. It would appear reasonable to assume an elastic modulus of about 12,000,000 at 1900°F. A load rate of 16,000 psi on a nickel alloy having an elastic modulus of 12,000,000 corresponds to a strain rate in the elastic range of 0.0013 inches per inch per minute. The standard strain rate specified for use with an extensometer is .005 inches per inch per minute. Therefore, the strain rate in the Vought tests would result in slightly lower reported strength values than the standard test using an extensometer. The Vought test results in a testing time of from 3 to 4 minutes. The procedure described in ARTC-13-T1 results in a testing time of from 1 1/2 to 2 minutes. ASTM testing procedure results in a testing time of from 1.2 to 2.1 minutes. The Vought testing procedure would, therefore, be expected to give slightly lower test results than the standard procedures cited.

- D. After the specimen is broken, total elongation is determined over a one inch gauge length by a caliper measurement.

APPENDIX D COMPLETE CASTING DATA SUMMARY

MELT COMPOSITIONS CALCULATED FROM CHANGE ANALYSES
(WEIGHT PERCENT BASIS)

MELT NO.	ALLOY	INCO 713-C	Bal	Mi	Ta	Cr	Al	W	Mo	V	Zr	C	Ti	Fe	Si	Mn	Cu	Cb	Other	Pouring Temp. °F	Mold Temp. °F	Remarks
56					13.32	6.28	6.28		4.82			0.13	1.13	1.18	0.79	0.14		2.38	Mischmetal-0.010 S 0.014	3000	1400	
57			70.63		13.95	5.58			4.36		0.210	0.140	1.006	0.122	0.015	0.996	0.248	1.77	S 0.006 B 0.776 Mischmetal-0.010	2800	1700	
102					13.32	6.28			4.82		0.010	0.130		1.180	0.790	0.140				2600	1600	
104			71.20		14.00	5.60			4.50		0.120	0.140	1.010	0.120	0.150	0.0100	0.025	1.80	S 0.0021 B 0.017	2450	1600	
385	NASA		67.58	8.02	6.02	7.01	4.01	4.01	4.01	2.23	1.00	0.13								2950	1800	
386	NASA		67.58	8.02	6.02	7.01	4.01	4.01	4.01	2.23	1.00	0.13								2950	1800	
388	INCO 713-C Ta-C Mod.	84.7	60.75	16.26	10.71	4.92			3.68		0.08	0.13	0.82	1.28						2950	1800	
389	NASA		67.36	8.00	6.00	6.99	4.00	4.00	4.00	2.51	1.00	0.13								2900	1800	
390	INCO 713-C Ta-C Mod.	84.7	60.75	16.26	10.71	4.29			3.66		0.08	0.13	0.82	1.28						2950	1800	
391	INCO 713-C Ta-C Mod.	73.5	Non Nickel Base Alloy						3.18		0.07	1.76	0.71	1.11						2950	1500	
392	NASA		53.09	26.52	9.29	4.27																
393	NASA		68.39	8.12	6.10	5.58	4.06	4.06	4.06	2.55	1.02	0.13								3000	1500	
394	NASA		68.39	8.12	6.10	5.58	4.06	4.06	4.06	2.55	1.02	0.13								3000	1500	
395	INCO 713-C Ta-C Mod.	58.2	41.74	40.56	7.36	3.37			2.52		0.05	2.73	0.57	0.88						2950	1200	
396	INCO 713-C Ta-C Mod.		Power was left on until furnace was turned, sheets filled with ceramic material.																			
397-400	INCO 713-C Ta-C Mod.	100	Non Nickel Base Alloy						4.32		0.09	0.16	0.97	1.51	0.17	0.10	0.26			2950	1500	
401	INCO 713-C		71.68	2.30	12.64	5.80			4.32		0.08	0.16	0.97	1.51	0.17	0.10	0.26			3050	1500	
402	INCO 713-C		71.68	2.30	12.64	5.80			4.32		0.08	0.16	0.97	1.51	0.17	0.10	0.26			3050	1500	
403	INCO 713-C		60.17	1.95	10.61	4.90	15.11	3.63			0.08	1.56	0.81	1.28								
404	INCO 713-C		71.68	2.30	12.64	5.80			4.32		0.09	0.16	0.97	1.51	0.17	0.10	0.26			3000	1500	
405	INCO 713-C		67.58	8.00	6.00	6.02	7.01	4.01	4.01	2.23	1.00	0.13								3000	1500	
406	NASA		67.58	8.02	6.02	7.01	4.01	4.01	4.01	2.23	1.00	0.13								2970	1700	
407	NASA		67.58	8.02	6.02	7.01	4.01	4.01	4.01	2.23	1.00	0.13								2970	1700	
408	NASA		67.36	8.00	6.00	6.99	4.00	4.00	4.00	2.51	1.00	0.13								2950	1500	
409	NASA		67.36	8.00	6.00	6.99	4.00	4.00	4.00	2.51	1.00	0.13								2950	1500	
410	NASA		67.36	8.12	6.10	5.58	4.06	4.06	4.06	2.55	1.02	0.127								2950	1500	
411	NASA		68.39	8.12	6.10	5.58	4.06	4.06	4.06	2.55	1.02	0.127								2950	1500	
412	NASA		68.39	8.12	6.10	5.58	4.06	4.06	4.06	2.55	1.02	0.127								2950	1500	
413	NASA		67.37	8.00	6.00	7.00	4.00	4.00	4.00	2.50	1.00	0.125								2950	1500	
414	NASA		67.37	8.00	6.00	7.00	4.00	4.00	4.00	2.50	1.00	0.125								2950	1500	
415	NASA		67.32	8.00	6.37	6.03	4.00	4.00	4.00	2.50	1.00	0.19								3050	1500	
416	NASA		67.32	8.00	6.37	6.03	4.00	4.00	4.00	2.50	1.00	0.19								2850	1500	
417	NASA		67.32	8.00	6.37	6.03	4.00	4.00	4.00	2.50	1.00	0.19								2850	1500	
418	NASA		67.32	8.00	6.37	6.03	4.00	4.00	4.00	2.50	1.00	0.19								2850	1500	
419	NASA		67.32	8.00	6.37	6.03	4.00	4.00	4.00	2.50	1.00	0.19								2850	1500	
420	NASA		67.32	8.00	6.37	6.03	4.00	4.00	4.00	2.50	1.00	0.19								2850	1500	
421	NASA		67.32	8.00	6.37	6.03	4.00	4.00	4.00	2.50	1.00	0.19								2850	1500	
422	NASA		67.32	8.00	6.37	6.03	4.00	4.00	4.00	2.50	1.00	0.19								2850	1500	
423	NASA		67.32	8.00	6.37	6.03	4.00	4.00	4.00	2.50	1.00	0.19								2750	1500	
424	NASA		67.32	8.00	6.37	6.03	4.00	4.00	4.00	2.50	1.00	0.19								2750	1500	
425	NASA Cr Mod		59.34	7.05	17.89	5.43	3.53	3.53	3.53	2.20	0.88	0.10								2850	1500	
426	NASA Cr Mod		59.34	7.05	17.89	5.43	3.53	3.53	3.53	2.20	0.88	0.10								2850	1500	
427	INCO 713C		57.11	6.79	17.55	8.73	3.39	3.39	3.39	2.13	0.085	0.15								2850	1500	
428	INCO 713C		49.58	13.26	8.75	4.05			17.70		0.07	4.57	0.7	1.05						2850	1500	
429	Ta-C Mod		39.90	10.70	17.52	3.23	21.00	2.40			0.05	2.96	0.5	0.84						2850	1500	
430	Ta-C Mod		49.73	1.93	10.53	21.53			3.60			0.13	0.80	1.27						2850	1500	
431	INCO 713C		52.38	1.69	9.23	31.15			3.15			0.12	0.71	1.08						2850	1500	

APPENDIX D (CONTINUED)

MELT COMPOSITIONS CALCULATED FROM CHARGE ANALYSIS

MELT NO.	ALLOY	INCO 713-C	Ni	Ta	Cr	Al	W	Mo	V	Zr	Fe	Si	Mn	Cu	Co	Other	Pouring Temp, F	Hold. Temp, F	Remarks	
432	INCO 713C	---	64.49	2.07	11.39	5.20	---	3.86	---	8.06	2.20	0.86	1.33	---	---	---	2900	1500	a.	
433	Zr-C Mod	---	57.73	1.87	10.18	4.70	---	3.48	---	0.08	5.09	15.27	1.22	---	---	---	3000	1500	a.	
434	Ti-C Mod	---	58.52	1.89	10.32	4.77	---	18.22	---	0.065	3.78	0.78	1.24	---	---	---	2850	1500	a.	
435	Mo-C Mod	---	47.05	1.51	18.77	3.81	21.00	2.82	---	0.059	2.99	0.64	0.98	---	---	---	2850	1500	a.	
436	Cr-M-C Mod	---	63.64	2.04	14.52	5.13	---	3.81	---	0.066	1.51	7.43	1.31	---	---	---	2850	1500	a.	
437	INCO 713C	88.01	63.10	3.06	11.15	5.08	6.52	3.78	---	0.065	3.00	0.85	1.30	---	1.69	---	2850	1500	a.	
438	INCO 713C	93.75	67.22	2.15	11.88	10.28	---	4.03	---	0.0694	0.14	2.29	1.39	---	---	---	2850	1500	a.	
439-495	INCO 713C	---	Non Nickel Base Alloy	---	---	---	---	---	---	---	---	---	---	---	---	---	2950	1500	---	
450	Ti-C Mod	---	41.84	8.03	17.67	3.38	20.92	2.52	---	0.052	3.83	0.56	0.88	---	---	---	3030	1500	---	
457	INCO 713C	---	41.82	8.03	17.65	3.38	20.95	2.52	---	0.054	3.83	0.57	0.88	---	---	---	---	---	---	
458	Ta-C-Cr-M Mod	---	---	---	---	---	---	---	---	---	---	---	---	---	---	---	---	---	---	---
459	INCO 713C	---	Non Nickel Base Alloy	---	---	---	---	---	---	---	---	---	---	---	---	---	---	---	---	---
460-461	NASA	---	67.34	7.99	6.97	6.02	4.00	4.00	2.50	1.00	0.19	---	---	---	---	---	2850	1500	---	
462	NASA	---	67.34	7.99	6.97	6.02	4.00	4.00	2.50	1.00	0.19	---	---	---	---	---	2850	1500	---	
463-526	INCO 713C	---	Non Nickel Base Alloy	---	---	---	---	---	---	---	---	---	---	---	---	---	---	---	---	---
527	LTV 429	86.6	39.50	10.70	17.52	3.23	21.0	2.40	---	0.050	2.96	0.50	0.84	---	---	---	---	---	f.	
528	LTV	58.5	43.20	8.17	18.20	3.42	21.6	2.50	---	0.054	0.090	0.490	1.404	0.060	0.030	---	2800	R.T.	f.	
529	NASA Mod	---	67.2	7.9	7.0	6.0	5.0	4.0	2.5	---	0.09	---	---	---	---	---	2800	R.T.	f.	
530	NASA Mod	---	67.0	8.0	8.0	6.0	4.0	4.0	2.5	---	0.09	---	---	---	---	---	2850	1600	e.	
531	NASA Mod	---	65.6	8.1	7.0	6.0	4.0	4.0	2.5	---	0.09	---	---	---	---	---	2850	1600	e.	
532	NASA Mod	---	66.0	8.1	7.0	6.1	4.0	4.0	2.5	1.0	0.09	---	---	---	---	---	2850	1600	d.e.	
533	NASA Mod	---	66.5	7.9	6.9	6.0	5.0	4.0	2.5	---	0.18	---	---	---	---	---	2800	1500	d.e.	
534	NASA Mod	---	65.4	8.2	7.1	6.1	4.1	4.1	2.6	1.0	0.19	---	---	---	---	---	2800	1500	b,d,e.	
535	INCO 713C	99.96	Bal.	2.15	12.70	5.70	---	4.17	---	0.08	0.15	0.83	1.90	0.10	0.05	---	2700	1500	c.e.	
536	Mischmetal INCO 713C	100	Bal.	2.15	12.70	5.70	---	4.17	---	0.08	0.15	0.83	1.90	0.10	0.05	---	2650	1500	b.	
537	INCO 713C	100	Bal.	2.15	12.70	5.70	---	4.17	---	0.08	0.15	0.83	1.90	0.10	0.05	---	2600	1500	b.	
538	INCO 713C	100	Bal.	2.15	12.70	5.70	---	4.17	---	0.08	0.15	0.83	1.90	0.10	0.05	---	2600	1500	b.	
539	INCO 713C	99.67	Bal.	2.15	12.70	5.70	---	4.17	---	0.08	0.15	0.83	1.92	0.10	0.05	---	2600	1500	b.	
540	Mischmetal INCO 713C	100	Bal.	2.15	12.70	5.70	---	4.17	---	0.08	0.15	0.83	1.90	0.10	0.05	---	2600	1500	b.e.	
541	INCO 713C	100	Bal.	2.15	12.70	5.70	---	4.17	---	0.08	0.15	0.83	1.90	0.10	0.05	---	2600	1800	b.	
542	INCO 713C	100	Bal.	2.15	12.70	5.70	---	4.17	---	0.08	0.15	0.83	1.90	0.10	0.05	---	2900	1800	b.	
543	INCO 713C	100	Bal.	2.15	12.70	5.70	---	4.17	---	0.08	0.15	0.83	1.90	0.10	0.05	---	2900	1800	b.	
544	INCO 713C	100	Bal.	2.15	12.70	5.70	---	4.17	---	0.08	0.15	0.83	1.90	0.10	0.05	---	2900	1800	a.	
545	INCO 713C	100	Bal.	2.15	12.70	5.70	---	4.17	---	0.08	0.15	0.83	1.90	0.10	0.05	---	2950	1500	a.	
546	INCO 713C	100	Bal.	2.15	12.70	5.70	---	4.17	---	0.08	0.15	0.83	1.90	0.10	0.05	---	3000	1500	a.	
547	INCO 713C	99.9	72.18	2.15	12.70	5.70	---	4.17	---	0.08	0.15	0.83	1.90	0.10	0.05	---	3000	1500	a.	
548	High Temp INCO 713C Cap Mod	99.9	72.18	2.15	12.70	5.70	---	4.17	---	0.18	0.15	0.83	1.90	0.10	0.05	---	3000	1800	a.	
549	INCO 713C High Temp	99.8	72.18	2.15	12.70	5.70	---	4.17	---	0.18	0.15	0.83	1.97	0.10	0.05	---	3000	1800	a.e.	

APPENDIX D (CONTINUED)

MELT COMPOSITIONS CALCULATED FROM CHARGE ANALYSES
(WEIGHT PERCENT BASIS)

MELT NO.	ALLOY	INCO 713C	Ni	Ta	Cr	Al	W	Mo	V	Zr	C	Ti	Fe	Si	Mn	Cu	Cb	Other	Fouring Temp., °F	Mold Temp., °F	Remarks
550	INCO 713C High Zirconia	97.6	70.97	2.60	12.49	6.10	---	5.08	---	0.18	0.15	0.82	1.94	0.14	0.10	0.05	---	S 0.005 B 0.040	2750	1200	a.e.
551	INCO 713C High Alumina	97.8	70.97	2.60	12.49	6.10	---	5.08	---	0.08	0.15	0.82	1.94	0.14	0.10	0.05	---	S 0.005 B 0.040	2800	1200	a.e.
552	INCO 713C W Mod	98.19	70.42	2.58	12.39	5.56	0.97	5.04	---	0.08	0.15	0.81	1.92	0.14	0.10	0.05	---	S 0.005 B 0.040	2600	1200	a.e.
553	INCO 713C W Mod	97.53	69.88	2.57	12.30	2.52	1.93	5.01	---	0.08	0.15	0.80	1.91	0.14	0.10	0.05	---	S 0.005 B 0.040	2600	1200	a.e.

- a. Silica Mold.
- b. Zirconia Mold.
- c. Alumina Mold.
- d. Iron added with the vanadium as ferro vanadium.
- e. Iron added with the boron as ferro boron.
- f. Melted in graphite crucible.
- g. Melts 485 thru 488 melted in Zirconia crucible.
- h. Melts 487 thru 482 melted in Zirconia crucible.
- i. Melts 529 thru 553 melted in Magnorite X crucible.

APPENDIX E

VENDOR TYPICAL CHEMICAL ANALYSIS FOR MELT CHARGES
WEIGHT PERCENT

PRINCIPAL METAL	Ni	Ta	Cr	Al	W	Mo	V	Zr	C	Ti	Fe	Si	OTHER
INCO 713C	71.68	2.30	12.64	5.80	-	4.32	-	0.09	0.16	0.97	1.51	0.17	0.10 Mn, 0.25 Cu
INCO 713C	Bal.	2.15	12.70	5.70	-	4.17	-	0.08	0.15	0.83	1.90	0.14	0.10 Mn, 0.05 Cu
Ta, Cr, Al, Ti unavailable													
W	.0005		.001		99.9	.0011					.0003	.0004	
Mo					99.0			.25			.25	.25	0.10 P, 0.03 S
V							99.8	.400 ppm					0.002 H ₂ , 0.06 O ₂ , 0.04 N ₂
Zr	0.02		0.03		50 ppm	50 ppm	50 ppm	Bel.		100 ppm	.15	100 ppm	2.5 Hf, 1 ppm B, 20 ppm Co, 50 ppm Pb, 80 ppm N ₂
C								98.5					1.0 volatile matter, 0.5 moisture
Ferrovanadium	0.01		0.10	0.80		0.02	70.5	.15			23.8	1.40	0.03 Cu, 0.025 S, 0.03 P, 0.15 Mn, 0.04 As 0.15 Co, 0.025 Ca

APPENDIX F

NICKEL ALLOY ROLLING DATA
ROLLED AT METALS & CONTROLS

ROLL #	ALLOY	MELT #	ROLLING TEMP. (°F)	MILL	INITIAL THICKNESS (.001")	FINAL THICKNESS (.001")	WIDTH (in.)	AVERAGE REDUCTION PER PASS (.001")	MAXIMUM LOAD (pounds)	MAX. LOAD PER IN. OF WIDTH (pounds)	FINAL SHEET CONDITION	REMARKS
73	NASA	416	Room	7" Dia. 2 high	100 to 114	60 to 66	--	3.0	--	--	Badly Cracked	Sheet annealed (between every 3 passes) at 2150°F for 1/2 hr. air cooled
74	NASA	416	Room on 4 passes 1670 on last 2	7" Dia. 2 high	96 to 111	74 to 82	--	3.5	--	--	Badly Cracked	Annealed at 2000°F for 1-1/2 hrs. before hot rolling.
75	NASA	419	Same as #74	7" Dia. 2 high	97 to 116	78 to 84	--	4.0	--	--	Badly Cracked	Annealed at 2150°F for 1-1/2 hrs. before hot rolling
76	NASA	411	Room	20" Dia. 2 high	97 to 106	72 to 75	--	4.0	--	--	Badly Cracked	Surface belt sanded before start annealed 9 hrs. at 2150°F after 4th pass air cooled.
77	Ta-C modifed INCO 713C	388	Room	7" Dia. 2 high	110	86 to 87	--	6.0	--	--	1/4" deep edge cracks	Surface belt sanded before start no anneal. Badly cracked throughout sheet.
78	INCO 713C	404	Room	7" Dia. 2 high	95 to 100	82 to 84	--	3.0	--	--	1/4" deep edge cracks	Same as #77
79	NASA	419	Room	7" Dia. 2 high for first 1/2, 20" Dia. 2 high last half	98 to 112	61 to 63	--	3.0	--	--	Badly Cracked	Annealed 1-1/2 hr at 2150, water quenched after 4th pass Annealed 1/2 hr at 2150, water quenched after 6th to 8th pass Annealed 1/2 hr at 2150 air cooled after 9th pass
80	NASA	422	Room	7" Dia. 2 high & 20" Dia. 2 high	95 to 102	72 to 75 on 1/2 83-86 on other	--	3.5	--	--	Badly Cracked	Annealed 1/2 hr. at 2150, air cooled after 3rd and 4th pass. Cut in two after 4th pass, 1/2 continued on 7" mill to .072" other half rolled on 20" mill to .063"
81	NASA	425	Room	7" Dia. 2 high & 20" Dia. 2 high	101 to 106	75-78 on 1/2 73-75 on other	--	2.5	--	--	Badly Cracked	All passes except 3rd done on 7" 2 high mill, annealed 1/2 hr. at 2150, water quenched after 2nd pass. Annealed 1/2 hr. at 2150, air cooled after 3rd pass cut in two, both halves rolled on 7" 2 high mill

APPENDIX F (Continued)
NICKEL ALLOY ROLLING DATA
ROLLED AT VOUJRET

ROLL #	ALLOY	MELT #	ROLLING TEMP. (°F)	MILL	INITIAL THICKNESS (.001")	FINAL THICKNESS (.001")	WIDTH (in.)	AVERAGE REDUCTION PER PASS (.001")	MAXIMUM LOAD (pounds)	MAX. LOAD PER IN. OF WIDTH (pounds)	FINAL SHEET CONDITION	REMARKS
82	Ta-C Modi- fied INCO 713c	388	1700, 1800 see re- mark	6" Dia. 2 high 5 H.P.	100	37 to 40	1		48,600	48,600	1/8 to 3/16" deep edge cracks	Temp. increased to 1800 on 8th pass (2" I.D. glo-bar furnace used for roll numbers 82 to 110 inclusive) Temp. increased to 1880 on 11th pass. Temp. decreased to 1800 on 15th pass.
83-1	NASA	459	1950 2000 2050 2100 2150 2250	Same	113 117	3- to 37	1	6 for 1/2, 3 for other half of run	73,900	73,900	A few 3/8" deep edge cracks	Temp. at 1950 for pass #9-11,16,17; Temp. at 2000 for pass #6-8,13; Temp. at 2050 for pass #12-14; Temp. at 2100 for pass #2,4,5; Temp. at 2150 for pass #1,3; Temp. at 2250 for pass #18-21; Furnace soaking time varied from 5 to 60 minutes
83-2	NASA	462	Same as 83-1	Same	98 - 113	33 - 36	1	6 for 1/2 3 for other half of run	73,900	73,900	Same as 83-1	Same as 83-1
84	NASA	462	2000 2050 2100 2150	Same	94-112 on one 40 on other	90-112 on one 40 on other	1	6	40,500	40,500	About the same as 83-1	(Two pieces) one specimen stopped on 5th pass due to lack of room in furnace. Temp. at 2000 for passes #2-5 Temp. at 2050 for passes #13 Temp. at 2100 for passes #1-12, 14-18 Soaking time varied from 5 to 30 min.
85	NASA	462	2000	Same	98 -	83-86	1	6	--	--	Badly Cracked	Original condition of surface, poor
86	Ta-C- Cr-W Modi- fied INCO 713c	457	2200	Same	110 to 111	--	-	6	--	--	1/32" deep edge cracks	Final surface of specimen in good condition, very slightly oxidized
87	NASA	462	2000	Same	99 to 110	75 to 77	-	6	--	--	Com- pletely cracked up	Specimen encapsulated in sheet from melt 459 for passes #1-4; recovered with stain- less steel for rest of test. Stainless steel elongated more than specimen, thereby breaking up sheet.

APPENDIX F (Continued)

ROLL #	ALLOY	MELT #	ROLLING TEMP. (°F)	MILL	INITIAL THICKNESS (.001")	FINAL THICKNESS (.001")	WIDTH (in.)	SEDM THICKNESS PER PASS (.001")	MAXIMUM LOAD (pounds)	MAX. LOAD PER IN. OF WIDTH (pounds)	FINAL SHEET CONDITION	REMARKS
88	Ta-C Cr-W Modified INCO 713c	457	2100	Same	101-114	40 to 47	-	6	--	--	Visible surface cracks	Chilling of sheet caused by drive mechanism slipping.
89	NASA	462	1500	Same	102-115	91	-	3	--	--	1/32" edge cracks	Used for tensile bar.
90	NASA	462	1750	Same	114-115	58 to 60	-	3	--	--	1/32" edge cracks	Cracks widened but did not deepen at end of test, used for tensile test.
91	NASA	462	2000	Same	112-114	90 to	-	6 18	--	--	Completely cracked	.018" shim on pass #2. .006" on others
92	NASA	462	1850	Same	115-125	50	-	3	--	--	1/16" deep edge cracks	Sheet looked in good shape
93	NASA	462	1650	Same	107-113	58 to 60	-	3	--	--	Same as 92	
94	NASA	462	1750	Same	120	45	-	3,6,9	--	--	Same as 92	.003" shim on passes #8-17, .006" shim on passes 3-7, .009 on passes #28 18
95	NASA	462	1750	Same	110	-	-	3	--	--	Badly Cracked	Sheet annealed at 2250 for 10 minutes after 9th pass after which it broke up
96	NASA	425	1750	1-1/2" Dia. 4 High 5 H.P.	95 to 103	85	-	3	--	--	Badly Cracked	Specimen bowed during rolling due to unequal tension on working rolls-caused slipping.
97	NASA	462	1825	Same	97 to 112	84-89	-	6	--	--	Badly Cracked	Specimen bowed badly causing cracking
98 continuation of 90	NASA	462	1825	Same	58 to 60	15 to 16	-	3	--	--	1/64" edge cracks	1/2 of a failed tensile bar surface in good condition
99	INCO 713c	404	2000	Same	108-110	12 to 16	-	3	--	--	3/16" deep edge cracks	Sheet looked good at end, very little bowing, crack widened but not deepened

APPENDIX F (Continued)

ROLL #	ALLOY	MELT #	ROLLING TEMP. (°F)	MILL	INITIAL THICKNESS (.001")	FINAL THICKNESS (.001")	WIDTH (in.)	SEEM THICKNESS PER PASS (.001")	MAXIMUM LOAD (pounds)	MAX. LOAD PER IN. OF WIDTH (pounds)	FINAL SHEET CONDITION	REMARKS
100	Ta-C-Cr-W Modified INCO 713c	457	2100	Same	109 - 112	104	-	3	--	--	Badly Cracked	Working roll size too small for thickness of sheet.
101	Ta-C-Cr-W Modified INCO 713c	457	2100	Same	48 to 49	15	1/4 - 3/4	3	--	--	Badly Cracked	1/2 of failed tensile bar. Specimen in good condition at .030" thickness.
102	Ta-C-Cr-W Modified INCO 713c	457	2100	Same	47 to 48	29 to 32	1/4 - 3/4	3	--	--	1/32" edge cracks	Other 1/2 of tensile bar from roll #88 final surface condition good
103	Ta-C-Cr-W Modified INCO 713c	457	2100	Same	43 to 44	26 to 29	1/4 - 3/4	3	--	--	1/16" edge cracks	Tensile bar that broke while drilling final surface, good
104	NASA	462	1825	Same	58 to 60	15 to 20	1/4 - 3/4	3	--	--	1/32"	Other half of tensile bar that was run on run 98, good surface
105	NASA	418	1825	6" Dia 2 high 5 H.P.	103 - 114	53 to 57	-	3 See re-marks	--	--	1/8" edge cracks	Mill was shimmed every other pass only. Slightly oxidized surface.
106	NASA	425	1825	Same	98 to 102	48 to 56	1 - 1/2	3	--	--	1/8" edge cracks	Same as #105
107	-	-	-	-	-	-	-	-	-	-	-	Non-nickel base rolling
108	NASA	418	1825	Same	96 to 112	-	1 - 1/2	3 See re-marks	--	--	1/8" edge cracks	Same as roll #105
109	NASA	418	1825	Same	96 to 109	49 to 54	-	3	--	--	1/16" deep edge cracks	Same as 105

APPENDIX F (Continued)

ROLL #	ALLOY	MELT #	ROLLING TEMP (°F)	MILL	INITIAL THICKNESS (.001")	FINAL THICKNESS (.001")	WIDTH PER PASS (in.)	SLIM THICKNESS PER PASS (.001")	MAXIMUM LOAD (pounds)	MAX. LOAD PER IN. OF WIDTH (pounds)	FINAL SHEET CONDITION	REMARKS
110	MASA	415	1825	Same	112-117	83	-	0.1-5, 3.0 See re-marks	--	--	1/8" deep edge cracks	.0015" shims for passes #12-15,17,20,22,25 26-28 .003" shims for passes #2-5,8,11,32,34-27. Other passes had no new shims Continued as #114.
111	-	-	-	-	-	-	-	-	-	-	-	Non-nickel base rolling
112 & 113	INCO & 713c	405	2100 See re-marks	Same	98 to 108	45 to 46	2-1/8	0.3 See re-marks	52,000	24,000	1/8" deep edge cracks	.003 shim on passes #2-6, 9-15, 17-23, 25,27,30,36,39. No new shims on other passes (Large glo-bar furnace used from here to end)
114 continuation of 110	MASA	415	2130 See re-marks	See re-marks	83	51 to 53	-	3.6 See re-marks	80,000	--	1/8" deep edge cracks	2 high mill with power increased to 30 HP .003" shim on passes #1-4 .006" shim on passes #5-11
115	MASA	418	2050 See re-marks	See re-marks	90 to 100	15 to 16	See re-marks	3.6 See re-marks	106,000	26,000	1/8" edge cracks	.006" shim on passes 1-18 on 6" Dia., 2 high mill; .003" shim on passes 19-38 on 1-1/2" dia., 4 high mill original width 4" then specimen was cut into 1" wide strips for continuation on 4 high mill. Non-nickel base rolling
116 to 118	-	-	-	-	-	-	-	-	-	-	-	-
119	MASA	387	See re-marks	See re-marks	73-	21-	See re-marks	3	45,000	13,600	1/8" edge cracks	2 high & 4 high mill used. Changed from 2 to 4 high at pass #29. Rolled 5-1/8" width at 1925°F on passes 1-7. Rolled 2 pieces 2-1/2" width at 2015 on passes 8-48, for tensile bars
120	MASA	462	See re-marks	2 high	101-126	68 to 75	3-1/2	3	44,800	12,800	Edge & interior cracks	Rolling temperature 1925°F for passes 1-7, 1875°F for 8-26. Sheet bowed considerably due to variation in thickness
121	MASA	426	1925	Same	82-101	50 to 52	4-3/4	0.3 See re-marks	54,500	11,500	1/8" edge cracks	No new shims on passes #26-28, 32-34, 36, all others .003 shims. 1/3 of sheet completely cracked up by pass #10 .herefore 2/3 of sheet continued
122	INCO 713c	404	See re-marks	See re-marks	98 to 103	24 to 27	2-1/2	3	--	--	3/16" deep edge cracks	Rollled on 2 high mill at 1875°F during passes 1-27. Most of sheet then used for tensile specimens. Remainder continued on 4 high mill at 2025, passes 28-43
123	MASA	425	See re-marks	See re-marks	98 to 106	26 - 28	4-1/2 at start	3	--	--	1/8"-3/16" edge cracks	Rollled on 2 high mill at 1875°F during passes 1-36, then trimmed to 3-3/4" width and continued on 4 high at 2010 for passes 37-43. Temp. raised to 2200 for passes 44-54.

APPENDIX F (Continued)

ROLL #	ALLOY	MELT #	ROLLING TEMP. (°F)	MILL	INITIAL THICKNESS (.001")	FINAL THICKNESS (.001")	WIDTH (in.)	SEDM THICKNESS PER PASS (.001")	MAXIMUM LOAD (pounds)	MAX. LOAD PER IN. OF WIDTH (pounds)	FINAL SHEET CONDITION	REMARKS
124	MASA	418	See re-marks	See re-marks	98 to 112	32 to 34	6 at start	3	--	--	1/8" edge cracks	Rollled on 2 high mill at 1925°F during passes 1-36 then cut into 3 pieces, 2-3/2", 2-3/8" & 3-3/4" wide and continued at 2020°F on 4 high mill
125	MASA	385	See re-marks	See re-marks	95 to 108	31 to 33	2-3/4	3	38,000 on 4 high	13,800	1/8" edge cracks	Two sheets rolled on 2 high mill at 1875°F during passes 1-30 then part cut into tensile bars. Remainder continued at 2015°F on 4 high mill for passes 31-44. Part rolled cold as roll #132.
126	MASA	424	1875	2 high	94 to 108	51 to 53	2-3/4	3	--	--	1/4" deep edge cracks	Sheet cut into tensile bars, rest continued as roll 131
127	INCO 713c	404	2000	Same	98 to 106	48 to 50	2-3/8	3	--	--	3/16" deep edge cracks	Sheet cut into tensile bars
128	MASA	424	2000	See re-marks	95 to 110	21 to 23	3	3	--	--	1/8" deep edge cracks	3" wide sheet rolled on 2 high mill during passes 1-26 then trimmed into several pieces ranging from 1" to 3" wide and continued on 4 high mill.
129	INCO 713c	404	2000	2 high	98-106	48 to 50	2-3/8	3	--	--	3/16" deep edge cracks	Some sheet to tensile tests
130	Fa-C Med.-filed INCO 713c	390	2000	2 high	82 to 105	82 to 92	2-1/4	3	--	--	Badly Cracked	2 sheets, cracking due to variation in thickness
131 126 continued	MASA	424	2000	4 high	51 to 53	48 to 49	2-3/4	3	27,000	9,800	Badly Cracked & Bowed	Bowing caused by different tension in rolling mill drive chain, causing cracking
132 125 continued	MASA	385	Room	4 high	74 to 75	68 to 69	-	3	--	--	Badly Cracked & Bowed	Annealed at 2100°F for 1 hr, air cooled after 11th pass
133	INCO 713c	404	Room	4 high	47 to 48	36 to 38	1/2	3	--	--	Badly Cracked	Sheet from previous hot rolling on 2 high mill
134 120 continued	MASA	462	2100	4 high	64 to 66	21 to 22	1	3	---	---	Cracked	No significant new cracks only ones from 2 high rolling opening up

APPENDIX F (Continued)

ROLL #	ALLOY	MILT #	ROLLING TEMP. (°F)	MILL	INITIAL THICKNESS (.001")	FINAL THICKNESS (.001")	WIDTH (in.)	SEDM THICKNESS PER PASS (.001")	MAXIMUM LOAD (pounds)	MAX. LOAD PER IN. OF WIDTH (pounds)	FINAL SHEET CONDITION	REMARKS
135	NASA	530	2100	4 high	102-112	90	1-1/4	3	--	--	Badly Cracked	All through sheet at grain boundaries
136, 137, 130 continued	NASA	462	2000	4 high	64 to 69	38 to 40	1	3	--	--	Some edge cracks	Unchecked area from 2 high rolling
138 & 139	INCO 713c	404		4 high								Refer- data for 4 high mill, roll 122.
140	NASA	387		4 high								Refer to data for 4 high mill, roll 119
141 continuation of 129	INCO 713c	404	2025	4 high	48 to 52	32	2-3/8	3			1/16" edge cracks	
142 & 143	NASA	424		4 high								Refer- data for 4 high mill, roll 128
144 & 145	NASA	418		4 high								Refer to data for 4 high mill, roll 115
146, 147, 148	NASA	418		4 high								Refer to data for 4 high mill, roll 124
149	NASA	425		4 high								Refer to data for 4 high mill, roll 123
150	NASA	424		4 high								Refer to data for 4 high mill, roll 128
151	NASA	387		4 high								Refer to data for 4 high mill, roll 110
152	NASA	385		4 high								Refer to data for 4 high mill, roll 125
153	NASA	424		4 high								Refer to data for 4 high mill, roll 128
154	NASA	382		4 high								Refer to data for 4 high mill, roll 119
115, 156	NASA	385		4 high								Refer to data for 4 high mill, roll 125
157	NASA	418		4 high								Refer to data for 4 high mill, roll 115
158, 159, 160												Non-nickel base rolling
161	NASA	See remarks	2000	2 high	541-545	504	1-1/2	6, 12, 15, 30	65,000	43,300	Completely broken up	Part of riser scrap from casting surface ground to parallel faces, no syclo defects

APPENDIX F (Continued)

ROLL #	ALLOY	MELT #	ROLLING TEMP. (°F)	MILL	INITIAL THICKNESS (.001")	FINAL THICKNESS (.001")	WIDTH (in.)	SEMI THICKNESS PER PASS (.001")	MAXIMUM LOAD (pounds)	MAX. LOAD PER IN. OF WIDTH (pounds)	FINAL SHEET CONDITION	REMARKS
162	MASA	See re-marks	1980	2 high	442-460	356-358	1-1/4	3,6	30,000	24,000	Completely broken up	Part of a 3" consumably arc-melted ingot, transverse slice; .003" shim on passes #10-33; .006" shim on passes #1-9 Non-nickel base rolling
163, 164, 165												
166	MASA	425	1950	2 high	105-111	58 to 60	2-1/8	3	--	--	1/8" edge cracks	Rolled sheet cut into tensile specimens
167	MASA	385	1950	2 high	102-116	56 to 59	2-1/4	3	--	--	1/16" edge cracks	Sheet cut into tensile specimens Non-nickel base rolling
168, 169, 170												
171	MASA	419	2000	2 high	88 to 122	55 to 57	3	See re-marks 1.5,3			1/8" edge cracks & surface cracks	Surface crack appeared on pass #16 .0016" shim on passes #31-39; .003" on others. Crack due to variation in thickness Non-nickel base rolling
172, 173, 174												
175	INCO 713c	536	2100	2 high	197-223	175-176	3	3	--	--	Completely cracked up	Large grain size sheet plus variation in thickness caused breakup.
176	INCO	535	2100	2 high	184-	175	3	3	50,000	16,600	Completely cracked up	Same as 175
177	INCO 713c	537	2100	2 high	101-105	73 to 74	2	3	--	--	Edge & surface cracks	Cracks appeared at grain boundaries after macro etching.
178	INCO 713c	537 & 538	2000	2 high	99-103 & 58-60	53 to 59	-	3	--	--	1/8" edge cracks	Sheet from 537 started on pass #1 then one from 538 started on pass #20

APPENDIX F (Continued)

ROLL #	ALLOY	MELT #	ROLLING TEMP. (°F)	MILL	INITIAL THICKNESS (.001")	FINAL THICKNESS (.001")	WIDTH (in.)	SEMI THICKNESS PER PASS (.001")	MAXIMUM LOAD (pounds)	MAX. LOAD PER IN. OF WIDTH (pounds)	FINAL SHEET CONDITION	REMARKS
179	IMCO 713c two pieces	537	2000	2 high	105-110	52 to 55	182	3	--	--	Cracked	Initial condition of surfaces was poor
180	IMCO 713c four pieces	537 & 539	2000	2 high	96 to 108, 80 to 82, 75-77, 75-77	-	142	3	--	--	Cracked up	Initial condition of surfaces was poor
181	IMCO 713c three pieces	540 & 541	2000	2 high	94-110, 89-92, 41-90	-	-	3	--	--	Cracked up	Initial condition of surfaces and thickness variation poor
182	IMCO 713c	405	2000 & 2100	2 high	103-107	50 to 54	-	-	--	--	1/8" edge cracks	Rejected sheet (holes) yet rolled well. Temp. at 2000°F during passes 1-18, 2100°F for remainder
183	IMCO 713c	2120 & 2210	2120 & 2210	4 high	50 to 55	45 to 50	10-1/4	3	--	--	1/2" edge cracks	Temp. increased to 2210°F on 5th pass
184	IMCO 713c	405	2200	4 high	54 to 59	46 to 51	13-5/8	6	--	--	1/2" & 1" edge cracks	Since distance between 4 high bearings was 16" some difficulty was encountered in placing sheet in rolls, causing cracks
185 continuation of 179	IMCO 713c four pieces	537	2050	4 high	52 to 58	40 to 43	-	3	--	--	Cracked up	Sheets from roll 179. New cracks appeared in area which was not cracked in roll 179
186 continuation of 180	IMCO 713c	537 & 539	2000	4 high	35 to 40	35 to 40	2	3	--	--	1/8" deep cracks	Considerable bowing throughout run
187	IMCO 713c two pieces	543	2000	4 high	as cast 60 to 65	47-52	-	3	--	--	Broken up	X-ray segregation & surface pores at start
188	IMCO 713c two pieces	544	1900	4 high	50 to 65	40 to 41	-	3	--	--	Surface & edge cracks	Cracks in surface at grain boundaries
189	IMCO 713c two pieces	545	2000	2 high	52-92, 65-106	54-59	2-3/4	3	--	--	Cracked badly	Large thickness variation contributed to sheet bowing throughout run

APPENDIX F (Continued)

ROLL #	ALLOY	MELT #	ROLLING TEMP (°F)	MILL	INITIAL THICKNESS (.001")	FINAL THICKNESS (.001")	WIDTH (in.)	SEMI THICKNESS PER PASS (.001")	MAXIMUM LOAD (pounds)	MAX. LOAD PER IN. OF WIDTH (pounds)	FINAL SHEET CONDITION	REMARKS
190												Non-nickel base rolling
191	INCO 713c	544	2050	2 high	100-107	40-43	1-3/4	3	--	--	See remarks	Part of sheet that cracked up was X-ray rejected area
192	INCO 713c	544	2050	2 high	94-106	43-45	2-1/2	3	--	--	See remarks	Same as 191 (two pieces)
193	INCO 713c	546	2050	2 high	67-102 60-57	46-48	2-1/2	3				Same as 189 & 191 (two pieces)
194	INCO 713c	545	2050	2 high	53-69 as cast	38-	2-1/2	3				Same as 189 & 191 (two pieces)
195	INCO 713c	546	2000	2 high	52-66	52-	2-1/2	3				Same as 189 & 191 (two pieces)
196	INCO 713c	545	2000	2 high	45-60 43-60 45-65	45-55 43-55 45-55	2	3				Same as 189 & 191 (three pieces)
197	INCO 713c	544	2000	2 high	57-64	48-54	4-1/2	3				Same as 189 & 191 plus bowing
198	INCO 713c	645	2000	2 high	49-64 52-62	48-52	3	3				Same as 189 & 191
199												Non-nickel base rolling
200	INCO 713c	544	2000	2 high	92-110	89-92	8	3	50,000	6,200	Completely Cracked	Annealed at 2200°F for 15 hours after #7 pass; thickness variation caused cracking
201	INCO 713c	545	2000	2 high	87-109	89-92	8	3	43,000	5,400		Same as 200
202	INCO 713c	546	2000	2 high	94-113	45-48	5-1/2	3	100,000	18,200	Cracked completely	Same as 200
203	INCO 713c	543	See re-marks	2 high	140-172	83-87	4-1/2	3	61,000	13,600	Completely cracked	Rolled at 2050 for passes #1-34, 1900 for remainder, thickness variation caused cracking
204 to 209												Non-nickel base rolling
210	INCO 713c	548	See re-	2 high	101-107	53-55	6	3	87,000	14,500	Surface, edge cracks	Rolled at 1900°F for pass 1-33 then temp. increased in 2000°F

APPENDIX F (Continued)

ROLL #	ALLOY	MELT #	ROLLING TEMP. (°F)	MILL	INITIAL THICKNESS (.001")	FINAL THICKNESS (.001")	WIDTH (in.)	SDRM THICKNESS PER PASS (.001")	MAXIMUM LOAD (pounds)	MAX. LOAD PER IN. OF WIDTH (pounds)	FINAL SHEET CONDITION	REMARKS
211	INCO 713c	547	1950	2 high	101-110	55-56	6		100,000	16,700	Surface edge cracks	Jammed mill on 37th pass, thickness .068"
212 & 213	INCO 713c	548	1950	2 high	93-120	53-55	6	3	135,000	22,500	Surface edge cracks	Mill jammed several times. Thickness variation caused cracks.
214	INCO 713c two pieces	547	1950	2 high	100-127 101-121	66-67	6	3	61,000 See re-marks	10,200	Surface & edge cracks	Strain recorder failed after pass 18. Thickness variation caused cracks.
215	INCO 713c	549	2050	2 high	55-63 as cast	43-44	3 1/4	3	67,000	20,600	Surface & Edge cracks	
216	INCO 713c	550	1950	2 high	105-119	55-57	5 7/8	3	104,000	17,600	Edge cracks	One large internal crack due to thickness variation. Piece cut in two after 104,000# load and continued
217	INCO 713c two sheets	550	See re-marks	2 high	106-110 94-104	45-50 82-85	5 3/4	3	89,000	31,000 See re-marks	Edge cracks	Rolled at 1950 for passes 1-22 & 29-48, 2000°F on others. Sheet cut in two at pass #28-part not continued. Maximum load per width based on narrower (2 7/8) sheet.
218	INCO 713c two sheets	548, 549	1950	2 high	103-119 128-131	80-81	7 3/4	3	154,000	19,800	Edge & surface cracks	
219 & 220												Non-nickel base rolling
217 con- tinued	INCO 713c See re-marks	550	1950	4 high	45-50	19-20	2 3/4	See re-marks	30,500	11,100	1/8" edge cracks	One of sheets from 217 cut transversely in half, .003" shims used for passes #1-10, then tapered shim blocks with tightening screws employed
222 & 214 con- tinued	INCO 713c See re-marks	547	1950	4 high	66-76	49-50	10 3/8	Same as #221	82,000	7,800	Broken up	One of two sheets from #214
223 & 218 con- tinued	INCO 713c	540	1950	4 high	80-81	22-23	10	Same as #221	102,000	10,200	See remarks	Old cracks opened up. Mill preloaded to 20,000# with tapered shim blocks.

APPENDIX F (Continued)

ROLL #	ALLOY	MELT #	ROLLING TEMP. (°F)	MILL	INITIAL THICKNESS (.001")	FINAL THICKNESS (.001")	WIDTH (in.)	SEMI THICKNESS PER PASS (.001")	MAXIMUM LOAD (pounds)	MAX. LOAD PER IN. OF WIDTH (pounds)	FINAL SHEET CONDITION	REMARKS
224	INCO 713c	551	1950	4 high	71-81 as cast	70-72	1 3/4	3	12,000	6,900	Completely cracked	Roll diameter too small.
225	INCO 713c	552	1950	4 high	106-116	95-97	2	3	30,100	10,100	Completely cracked	Roll diameter too small.
226	INCC	553	1950	4 high	165-111	75-77	1 3/4	3	19,000	10,800	Completely cracked	Both sheets broken. Roll diameter too small.
227	713c two sheets				103-106							
228	INCC	552	1950	4 high	104-115	94	3	3	26,000	8,700	Completely cracked	Roll diameter too small.
229	INCO 713c	551	1950	4 high	71-77	62	2 1/2	3	30,100	12,000	Completely cracked	Roll diameter too small.
230	INCO	550	1950	4 high	55-57	25	3	Same as 221	22,000	7,300	1/16" edge cracks	Cut into tensile bars.
231	713c											
232	INCC	550	1950	4 high	45-50	19-20	3	Same as 221	---	---	1/16" edge cracks	Same as 230.
233	INCC	550	1900	4 high	82-85	30-31	6 1/2	6	80,000	12,300	See remarks	Edge cracks and surface cracks at one end, due to X-ray
234	INCO	550	See re-marks	4 high	17-15	5-6	1 1/2	See re-marks	20,000	13,300	1/16" edge cracks	Rolled two passes at 2150 with mill preloaded to 20,000 Rolled 10 to 20 passes at 2150 with mill preloaded to 40,000 Maximum load recorder was that above 40,000. Preload
235	INCO 713c	550	R.T.	4 high	16-20	7-7 1/2	3	See re-marks	Same as 234	6,700	1/16" edge cracks	Rolled 20 to 20 passes at R.T. with mill preloaded to 40,000 Further preloading caused mill difficulty in starting
236	INCO 713c	550	R.T.	4 high	18-20	3.5-4	3	Same as 235	Same as 234	6,700	1/16" edge cracks	Specimen annealed at 2150 for 5 min. between every 5 to 7 passes

APPENDIX G

TENSILE DATA OF NICKEL ALLOYS
ROOM & 1900°
LOADING RATE 16,000 #/in² per minute

ALLOY	MELT #	CONDITION	TEST TEMP. (°F)	THICKNESS (.001")	AREA (square inches)	ULTIMATE YIELD STRENGTH (PSI)		ULTIMATE TENSILE LOAD #		ULTIMATE TENSILE STRENGTH (FSI)		ELONGATION (PERCENT)	
						Room Temperature	1900° F	Room Temperature	1900° F	Room Temperature	1900° F	Room Temperature	1900° F
NASA	385	as cast	Room	102.5	.02591	112,700	3010	116,200	44,500	2.0			
NASA	385	as cast	1900	100.0	.02596	120,900	2830	121,100	44,800	1.0	4.0		
NASA	386	as cast	Room	92.3	.02337	120,900							
NASA	386	as cast	1900	105.1	.02544	111,100	3590	118,800	54,800	2.0			
NASA	387	as cast	1900	121.1	.03007								
NASA	389	as cast	Room	120.1	.03023								
NASA	389	as cast	1900	102.1	.02821								
NASA	393	as cast	Room	106.4	.02686								
NASA	393	as cast	1900	103.4	.02608								
NASA	394	as cast	Room	94.3	.02382								
NASA	394	as cast	1900	102.9	.02574								
NASA	393	(a.)	Room	54.6	.01431		1325	52,600	53,300	1.0	2.0		
NASA	393	(b.)	1900	53.7	.01397		1730	137,600	56,700	1.0	3.0		
NASA	394	(a.)	Room	51.9	.01334	129,700			60,100	1.0	3.0		
NASA	394	(b.)	1900	59.6	.01765					1.0			
NASA	462	as cast	1900	114.3	.02531		1435	56,700					
NASA	462	as cast	1900	116.2	.02504		1505	60,100					
Cr modified NASA	426	as cast	ROOM	78.1	.01994		2225	111,600	20,600	1.0			
Cr modified NASA	426	as cast	1900	84.5	.02202				19,400	1.0	11.5		
Cr-Al modified NASA	427	as cast	ROOM	105.1	.02642		3070	116,200					
Cr-Al modified NASA	427	as cast	1900	80.3	.01999								
Modified NASA	529	as cast	ROOM	103.6	.0516	83,700	5420	105,000	37,300	2.0			
Modified NASA	529	as cast	ROOM	80.2	.0407	115,500	5310	130,000	28,500	1.7	1.0		
Modified NASA	529	as cast	1900(x)	69.7	.0299								
Modified NASA	529	as cast	1900	111.9	.0478								
Modified NASA	530	as cast	ROOM	98.1	.04741	116,100	6030	127,000	37,000	4.0			
Modified NASA	530	as cast	ROOM	98.8	.0483	---	4070	85,400	32,900	1.0	2.0		
Modified NASA	530	as cast	1900(x)	104.8	.0405								
Modified NASA	530	as cast	1900	96.9	.0377								
Modified NASA	531	as cast	ROOM	102.2	.0506	87,800	5430	107,000	38,800	4.0			
Modified NASA	531	as cast	ROOM	108.1	.0537	85,400	5140	95,600	38,800	2.0	1.5		
Modified NASA	531	as cast	1900	98.2	.0370								
Modified NASA	531	as cast	1900(x)	105.4	.0414								

APPENDIX G (Continued)

ALLOY & ROLL NUMBER	MELT NUMBER	CONDITION	TEST TEMPERATURE (°F)	THICKNESS (.001")	AREA (sq. in.)	ULTIMATE YIELD STRENGTH (PSI) ROOM TEMPERATURE	ULTIMATE TENSILE LOAD (#) ROOM TEMPERATURE	ULTIMATE TENSILE STRENGTH (PSI) ROOM TEMPERATURE	ELONGATION (PERCENT) ROOM TEMPERATURE
Modified NASA	532	as cast	ROOM	84.6	.0309	51,300	2151	67,700	1.0
Modified NASA	532	as cast	ROOM	90.3	.0327	51,300	1780	54,400	1.7
Modified NASA	532	as cast	1900	97.9	.0372			33,100	1.5
Modified NASA	532	as cast	1900(x)	75.8	.0293			25,900 (x)	1.0
NASA		rolled at 1500°F	1900	93.8	.02085			50,839	0.0
NASA		rolled at 1750°F	1900	58.0	.01206			8,800	3.0
NASA		as rolled	1900	55.0	.02035			34,400	5.0
NASA		as rolled	1900	54.0	.0207			37,700	5.0
NASA	143	rolled + (U.)	ROOM	13.8(S)	.006839	189,400	1550	226,600	2.0
NASA	424	rolled + (U.)	ROOM	22.3(T)	.01105	117,200		140,300	
NASA	148	rolled + (U.)	ROOM	24.2(S)	.008134	150,000	1390	170,900	1.0
NASA	418	rolled + (U.)	ROOM	26.0(T)	.008739	136,500		159,000	
NASA	148	rolled + (U.)	1900	24.5(S)	.01378			57,700	4.5
NASA	418	rolled + (U.)	1900	29.5(T)	.01658			48,000	
NASA	142	rolled + (U.)	1900	16.0(S)	.01014			46,800	4.0
NASA	424	rolled + (U.)	1900	19.0(T)	.01204			32,400	
NASA	142	rolled + (U.)	1900	11.0(S)	.006925			49,600	0.0
NASA	424	rolled + (U.)	1900	18.2(T)	.01134			30,400	

APPENDIX G (Continued)

ALLOY & ROLL NUMBER	MELT NUMBER	CONDITION	TEST TEMPERATURE (°F)	THICKNESS (inches)	AREA (sq. inches)	ULTIMATE YIELD STRENGTH (PSI)		ULTIMATE TENSILE LOAD (%)		ULTIMATE TENSILE STRENGTH (PSI)		ELONGATION (PERCENT)	
						ROOM TEMPERATURE	1900°F	ROOM TEMPERATURE	1900°F	ROOM TEMPERATURE	1900°F	ROOM TEMPERATURE	1900°F
NASA	154 387	rolled + (V.)	ROOM	14.1(S)	.007731	129,000	116,600	1290	166,900	151,700	4.0	4.0	
NASA	154 387	rolled + (V.)	ROOM	13.5(S)	.006635	135,200	98,700	1310	144,100	197,400	5.0	5.0	
NASA	154 387	rolled + (V.)	ROOM	10.0(S)	.003236	143,100	105,900	565	174,600	123,300	2.0	2.0	
NASA	154 387	rolled + (V.)	1900	14.0(S)	.00981			465	47,400	42,800	4.0	4.0	
NASA	155 385	rolled + (V.)	1900	15.5(T)	.01086			765	42,000	49,500	2.0	2.0	
NASA	155 385	rolled + (V.)	1900	21.9(S)	.01545			575	46,500	43,300	2.5	2.5	
NASA	152 385	rolled + (W.)	ROOM	18.6(S)	.00653	128,300	92,500	1145	175,200	126,400	3.0	3.0	
NASA	152 385	rolled + (W.)	1900	24.0(S)	.01675			815	48,700	43,600	7.0	7.0	
NASA	153 424	rolled + (W.)	1900	19.0(S)	.01074			505	47,000	37,200	---	---	
				24.0(T)	.01356								

APPENDIX G (Continued)

ALLOY & ROLL NUMBER	MELT NUMBER	CONDITION	TEST TEMPERATURE (*F)	THICKNESS (.001")	AREA (sq. inches)	ULTIMATE YIELD STRENGTH (SI)		ULTIMATE TENSILE LOAD (F)		ULTIMATE TENSILE STRENGTH (PSI)		ELONGATION (PERCENT)	
						ROOM TEMPERATURE	1900°F	ROOM TEMPERATURE	1900°F	ROOM TEMPERATURE	1900°F	ROOM TEMPERATURE	1900°F
NASA 109	418	rolled + (F.)	ROOM	50.7	.01325	153,600	2165	163,400	1.0				
NASA		rolled + (F.)	1900	47.0	.0179		580	32,400	10.3				
NASA		rolled + (F.)	1900	50.0	.0200		955	47,700	11.5				
NASA		rolled + (G.)	1900	46.0	.0150		535	35,700	24.0				
NASA		rolled + (G.)	1900	49.0	.01656		715	43,200	14.0				
NASA 125	385	rolled + (D.)	1900	41.2	.01482		540	36,400	8.0				
NASA 115	418	rolled + (D.)	1900	50.5	.01758		522	29,700	15.0				
NASA 123		rolled + (E.)	ROOM	46.0	.01598	149,200	2590	162,100	2.0				
NASA 115	418	rolled + (E.)	1900	50.8	.01369		388	28,300	15.0				
NASA 125	385	rolled + (E.)	1900	41.9	.01453		482	33,200	6.0				
NASA 123		rolled + (C.)	ROOM	42.6	.01374	138,600	2260	164,500	3.0				
NASA 125	385	rolled + (C.)	1900	40.9	.01513		522	34,500	3.0				
NASA 115	418	rolled + (C.)	1900	49.4	.01857		608	32,700	9.0				
NASA 123		rolled + (D.)	ROOM	42.5	.01481		1900	128,300	2.0				
NASA 171	419	rolled + (O.)	1900	54.0	.02180		875	40,100	12.0				
NASA 166	425	rolled + (O.)	1900	54.0	.01912		502	26,300	---				
NASA 120	462	rolled + (F.)	ROOM	54.6	.02005	120,200	2675	134,400	1.0				
NASA 171	419	rolled + (P.)	1900	52.5	.02225		866	38,500	16.0				
NASA 122	425	rolled + (P.)	1900	48.2	.01591		582	36,600	7.0				
NASA 125	385	rolled + (H.)	ROOM	41.6	.01083		1568	144,800	6.0				
NASA 125	385	rolled + (H.)	ROOM	39.1	.01025	123,400	1495	145,900	3.0				

APPENDIX G (Continued)

ALLOY & ROLL NUMBER	MELT NUMBER	CONDITION	TEST TEMPERATURE (°F)	THICKNESS (.001")	AREA (sq. in.)	ULTIMATE YIELD STRENGTH (PSI)	ULTIMATE TENSILE LOAD (#)		ULTIMATE TENSILE STRENGTH (PSI)		ELONGATION (PERCENT)
							ROOM TEMPERATURE	1900°F TEMPERATURE	ROOM TEMPERATURE	1900°F TEMPERATURE	
NASA	120 462	rolled + (Q.)	ROOM	58.3	.01849	134,400	2950	161,700	1.0	1.0	
NASA	166 425	rolled + (Q.)	ROOM	51.1	.01918	126,400	2765	144,200	1.0	1.0	
NASA	166 425	rolled + (Q.)	ROOM	55.1	.02021	124,700	2750	136,100	1.0	1.0	
NASA	150 424	rolled + (Q.)	ROOM	15.8(S)	.007573	207,000	1730	217,000	0	0	
NASA	150 424	rolled + (Q.)	ROOM	23.3(T)	.01176	140,300	1450	147,100	0	0	
NASA	149 418	rolled + (Q.)	ROOM	17.0(S)	.008434	165,000	320	172,000	5.0	5.0	
NASA	148 418	rolled + (Q.)	ROOM	15.1(T)	.00548	147,500	555	153,000	4.0	4.0	
NASA	148 416	rolled + (Q.)	ROOM	30.6	.00320	34,800	740	34,800	3.5	3.5	
NASA	149 425	rolled + (Q.)	ROOM	44.5	.01368	40,000	415	40,000	2.0	2.0	
NASA	148 416	rolled + (Q.)	ROOM	40.1	.01664	33,700	1775	33,700	6.0	6.0	
NASA	149 425	rolled + (Q.)	ROOM	18.5(S)	.00916	45,300	1775	45,300	12.0	12.0	
NASA	149 425	rolled + (Q.)	ROOM	19.5(S)	.00965	43,000	362	43,000	3.0	3.0	
NASA	149 425	rolled + (Q.)	ROOM	18.8(S)	.01324	50,600	670	50,600	3.0	3.0	
NASA	126 424	rolled + (N.)	ROOM	23.2(T)	.01633	121,000	1775	121,000	2.0	2.0	
NASA	114 415	rolled + (M.)	ROOM	41.2	.01467	33,200	510	33,200	8.0	8.0	
NASA	126 424	rolled + (M.)	ROOM	45.1	.01538	24,800	362	24,800	12.0	12.0	
NASA	147 413	rolled + (AA)	ROOM	42.0	.01461	41,900	272	41,900	3.0	3.0	
NASA	147 418	rolled + (AA)	ROOM	19.3	.00649	24,400 (Y)	172	24,400 (Y)	3.0	3.0	
NASA	146 418	rolled + (AA)	ROOM	20.7	.00706	29,400 (Y)	272	29,400 (Y)	2.0	2.0	

APPENDIX G (Continued)

ALLOY & ROLL NUMBER	MELT NUMBER	CONDITION	TEST TEMPERATURE (*F)	THICKNESS (.001")	AREA (sq. in.)	ULTIMATE YIELD STRENGTH (PSI)	ULTIMATE TENSILE LOAD (#)	ULTIMATE TENSILE STRENGTH (PSI)	ELONGATION (PERCENT)	
										ROOM TEMPERATURE
NASA	114	rolled + (N.)	ROOM	41.2	.01612	121,900	2200	136,500	---	
NASA	122	rolled + (N.)	ROOM	49.5	.01999	120,600	2800	143,700	2.0	
NASA	122	rolled + (N.)	ROOM	41.0	.01618	123,300	2345	144,900	---	
NASA	122	rolled + (N.)	1900	40.3	.01579		494	31,300	8.0	
NASA	122	rolled + (N.)	1900	48.3	.01899		652	34,300	10.0	
NASA	122	rolled + (N.)	1900	45.1	.01756		534	30,400	3.0	
NASA	119	rolled + (O.)	ROOM	47.2	.01621	123,500	2130	131,500	1.0	
NASA	123	rolled + (I.)	ROOM	44.6	.01143		1560	136,500	0	
NASA	125	rolled + (I.)	1900	37.3	.0119		434	36,500	1.0	
NASA	109	rolled + (F.)	ROOM	47.2	.01499		2035	135,800	0	
INCO 713C	404	as cast	Room	97.5	.02488	125,400	3400	136,700	2.0	
2(Ta-C) Modified INCO 713C { 388 388 M. Modified 390 INCO 713C: 390		as cast	Room	110.6	.02778	110,700	3280	119,300	2.0	
			1900	108.4	.02758		1740	63,600	40,700	2.0
			Room	106.5	.02557			1040		
			1900							
2(Ta-C) Modified INCO 713C		as cast	Room	102.3	.02680	112,700	3160	117,900	2.0	
			1900	104.2	.02622		2145	103,600	41,900	2.0
Ta-C-Mo modified INCO 713C	428	as cast	Room	82.1	.02071	103,600	2145	103,600	2.0	
			1900	85.4	.02162		2750	109,000	28,400	2.0
Ta-C-Mo-W modified INCO 713C	429	as cast	Room	103.0	.02524		756	71,600	4.5	
			1900	108.8	.02663		1900	53,200		
Zr-C modified INCO 713C	432	as cast	Room	106.3	.02654		1155	91,600	1.5	
			1900	80.8	.02172		2390	27,100	24,300	2.0
Ti-C modified INCO 713C	433	as cast	ROOM	104.1	.02609		725	82,900	1.0	
Cr-W-C modified INCO 713C	435	as cast	ROOM	105.0	.02675		680	37,000	2.5	
			1900	110.0	.02724		2030			

APPENDIX G (Continued)

ALLOY & ROLL NUMBER	MELT NUMBER	CONDITION	TEST TEMPERATURE (*F)	THICKNESS (.001")	AREA (sq. in.)	ULTIMATE YIELD STRENGTH (PSI) ROOM TEMPERATURE	ULTIMATE TENSILE LOAD (#) ROOM TEMPERATURE	ULTIMATE TENSILE STRENGTH (PSI)		ELONGATION (PERCENT)
								ROOM TEMPERATURE	1900°F	
Cr-Ti-C modified INCO 713C	436	as cast	ROOM	118.3	.02840			50,500		1.0
	436	as cast	1900	111.3	.02929		2570	1155	39,400	1.5
Cb-Ta-W-C modified INCO 713C	437	as cast	ROOM	112.2	.02797			116,400		1.0
	437	as cast	1900	93.8	.02457		3255	845	34,400	2.5
Al-Ti modified INCO 713C	438	as cast	ROOM	113.1	.02737		2940		107,400	1.0
Ta-C-Mo-W modified INCO	457	as cast	ROOM	100.4	.02155			1260	58,500	0.0
	457	as cast	ROOM	100.5	.02162			1350	62,400	0.0
Ta-C modified INCO 713C	388	rolled at R.T. 20%	1900	86.7	.01932			500	23,900	4.0
		rolled at 1800°F 60%	1900	40.3	.00951			272	28,500	4.0
INCO 713C	404	as rolled	1900	44.0	.0150			625	41,778	3.0
		as rolled	1900	41.0	.0140			615	43,991	3.0
INCO 713C	405	rolled at R.T. 20%	1900	82.8	.01942			930	47,900	2.0
		rolled at R.T. 20%	1900	82.5	.01948			810	41,600	3.0
Ta-C-Mo-W Modified INCO 713C	457	rolled 60% at 2100°F	1900	47.8	.01024			130	12,700	5.0
INCO 713C	216	rolled + (J.)	1900	46.6	.02299			1165	50,700	5.0
		rolled + (J.)	1900	46.0	.02585			1290	49,900	2.0
INCO 713C	216	rolled + (J.)	1900	45.1	.02589			1265	48,900	5.0
		rolled + (J.)	1900	51.7	.03108			1370	44,100	8.0
INCO 713C	210	rolled + (J.)	1900	52.0	.03161			1325	41,900	9.0
		rolled + (J.)	1900	50.1	.02967			1270	42,800	10.0

APPENDIX G (Continued)

ALLOY & ROLL NUMBER	MELT NUMBER	CONDITION	TEST TEMPERATURE (°F)	THICKNESS (.001")	AREA (sq. inches)	ULTIMATE YIELD STRENGTH (PSI)		ULTIMATE TENSILE LOAD (#)		ULTIMATE TENSILE STRENGTH (PSI)		ELONGATION (PERCENT)	
						ROOM TEMPERATURE	1900°F	ROOM TEMPERATURE	1900°F	ROOM TEMPERATURE	1900°F	ROOM TEMPERATURE	1900°F
INCO 713C 210	548	rolled + (K.)	1900	50.6	.02969								
INCO 713C 210	548	rolled + (K.)	1900	46.5	.02649								
INCO 713C 216	550	rolled + (L.)	1900	47.4	.02699								
INCO 713C 216	550	rolled + (L.)	1900	47.2	.02660								
INCO 713C 216	550	rolled + (L.)	1900	45.9	.02576								
INCO 713C 231	550	rolled + (U.)	ROOM	18.1	.00903	141,200	1410	156,100				3.0	
INCO 713C 231	550	rolled + (U.)	1900	17.8	.01333				420		31,500(Y)		6.0
INCO 713C 230	550	rolled + (U.)	1900	18.5	.01439				356		24,700(Y)		13.0
INCO 713C 230	550	rolled + (U.)	1900	20.8	.01643				504		30,700(Y)		12.0
INCO 713C 211	547	rolled + (U.)	1900	55.7	.03703				1692		45,700		10.0
INCO 713C 211	547	rolled + (U.)	1900	55.6	.03684				2100		57,000(Z)		5.0
INCO 713C 211	547	rolled + (U.)	1900	55.2	.03699				2580		69,700(Z) no less than		15.0
INCO 713C 231	550	rolled + (U.)	1900	19.2	.01065						*60,900(Z)		7.0
INCO 713C 231	550	rolled + (U.)	1900	19.3	.01084						*load dial was moved before reading recorded		8.0
INCO 713C 211	547	rolled + (BB)	1900	55.8	.03725				1728		45,400		8.0
INCO 713C 211	547	rolled + (BB)	1900	56.8	.03769				1800		47,800		9.0
INCO 713C -	-	rolled + (F.)	1900	44.0	.0178				805		45,300		4.0
INCO 713C		rolled + (G.)	1900	42.0	.0170				840		45,400		12.0
INCO 713C 122	404	rolled + (D.)	1900	51.8	.01762				754				
INCO 713C 122	404	rolled + (N)	1900	50.8	.01667				480				
INCO 713C 122	404	rolled + (D.)	ROOM	46.2	.01756				3130		178,200		12.0
INCO 713C 122	404	rolled + (E.)	ROOM	50.6	.01224				2010		124,600		10.0
INCO 713C 129	404	rolled + (E.)	1900	46.1	.01544				590		28,800		6.0
											164,200		14.0
											33,200		6.0

APPENDIX G (Continued)

ALLOY & ROLL NUMBER	MELT NUMBER	CONDITION	TEMPERATURE (°F)	THICKNESS (.001")	AREA (sq. inches)	ULTIMATE YIELD STRENGTH (PST)	ULTIMATE TENSILE LOAD (%)	ULTIMATE TENSILE STRENGTH (PSI)	ELONGATION (PERCENT)
INCO 713C 129	404	as rolled + (C.)	ROOM	42.5	.01455	135,700	2725	187,300	18.0
INCO 713C 129	404	as rolled + (C.)	1900°F	45.5	.01638		842	51,400	10.0
INCO 713C		(CC.)	ROOM	31.2	.009348	105,700	1352	144,600	14.0
INCO 713C 182		rolled + (C.)	ROOM	30.1	.009335	143,300	1840	156,800	12.0
INCO 713C 193		rolled + (C.)	ROOM	47.6	.02054	123,700	3500	170,400	14.0
INCO 713C 193		rolled + (C.)	ROOM	44.3	.01830	121,900	3220	176,000	15.0
INCO 713C 191		rolled + (H.)	ROOM	41.2	.01387	122,600	2360	170,200	20.0
INCO 713C 191		rolled + (H.)	ROOM	40.0	.01348	123,500	2300	170,500	19.0
INCO 713C 191		rolled + (H.)	ROOM	42.0	.01386	123,700	2455	177,100	17.0

- (A.) HT 16 HRS at 2000°, water quenched.
 (B.) HT 16 HRS at 2200°, water quenched.
 (C.) HT 24 HRS at 2200° F air cool.
 (D.) HT 3 HRS at 2200° F air cool.
 (E.) HT 3 HRS at 2200° F, quenched in 170° F oil.
 (F.) HT 1/2 HR at 2200° F air cool.
 (G.) HT 1/2 HR at 2200° F air cool + 24 HRS at 1600° F air cool.
 (H.) HT 64 HR at 2200° F, air cool.
 (I.) HT 3 HR at 2250° F, air cool.
 (J.) HT 41 HR at 2150° F air cool.
 (K.) HT 41 HR at 2150° F air cool, + 24 HR at 1600, air cool.
 (L.) HT 41 HR at 2150° F air cool, + 61 HR at 1600, air cool.
 (M.) HT 64 HRS at 2200° F air cool, + 24 HRS at 1450° F air cool.
 (N.) HT 64 HRS at 2200° F air cool + 24 HRS at 1900° F air cool.
 (O.) HT 24 HRS at 2200° F air cool + 64 HRS at 1600° F air cool, + 24 HRS at 1750, air cool.
 (P.) HT 24 HRS at 2200° F, air cool + 64 HRS at 1500° F air cool, + 24 HRS at 1750, air cool + 2 1/2 HRS at 2000° F air cool.
 (Q.) 64 HRS at 2200, retort cool.
 (R.) 64 HRS at 2150° F air cool + 64 HRS at 1450 air cool.
 (S.) Thickness measured with pointed micrometer.
 (T.) Thickness measured with flat micrometer.
 (U.) HT 61 HRS at 2150° F air cool + 24 HR at 1600, air cool.
 (V.) HT 61 HR at 2150° F, air cool + 24 HR at 1750, air cool.
 (W.) HT 61 HR at 2150° F air cool + 64 HR at 1950° F, air cool.
 (X.) Radiant heating method.
 (Y.) Minimum load rate, head travel about .01"/minute, length of test 8-12 minutes.
 (Z.) Maximum load rate, head travel about 5.0 /minute, length of test 2.0 seconds.
 (AA.) HT 64 HR at 2200° F + 24 HRS at 1575° F.
 (BB.) HT 61 HRS at 2150° F air cool + 24 HRS at 1600° F, air cool + 1/2 HR at 1900° F.
 (CC.) After bell-arcing. Weld in gage length.

1. Nickel base alloys
2. High temperature research
- I. AFSC Project 7351, Task 735105
- II. Contract AF 33(616) -7999
- III. Chance Vought Corp Dallas 22, Texas
- IV. H. Greenwald, Jr. I. J. Riley
- V. Aval fr OTS
- VI. In ASTIA Collection

Aeronautical Systems Division, Dir/Materials & Processes, Metals & Ceramics Lab, Wright-Patterson AFB, Ohio
 Rpt No. ASD-TDR-62-869. DEVELOPMENT OF A NICKEL BASE ALLOY SHEET FOR HIGH TEMPERATURE APPLICATION. Final report, Apr 63, 149p incl illus., tables.

Unclassified report
 The objective of this contract was to develop 15 to 30 mil nickel alloy sheet having 50,000 psi tensile strength at 1900°F, having good corrosion (oxidation) resistance, and good ductility. This objective was essentially attained by developing a new process of directly rolling thin cast slabs of nickel base alloy into sheet on a specially designed rigid rolling mill. Two pre-existing nickel base casting alloys and a series of experimental com-

(over)

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positions obtained by modifying the two starting alloys were initially investigated in the as cast condition in this program. The two starting alloys were Inco 713c and MSA's Ta28 alloy. Of the new experimental compositions, alloy No. 429 (a Ta-W-Cr modified Inco 713c) has 1900°F tensile strength exceeding 50,000 psi in the as cast condition, as does the Ta28 alloy. Inco 713c has a tensile strength of about 40,000 psi in the as cast condition. Hot rolled Inco 713c indicated good room temperature ductility. The oxidation resistance of Ta28 alloy is adequate for limited times at 1900°F; that of No. 429 alloy is substantially better; and that of Inco 713c is best of the three.

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