**CLASSIFICATION CHANGES**

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DNA ltr dtd 28 Mar 1996; DNA ltr dtd 28 Mar 1996
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Operation REDWING
PACIFIC PROVING GROUNDS
May - July 1956
Project 2.52
NEUTRON-INDUCED SOIL RADIOACTIVITY (U)

Issuance Date: December 16, 1959

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Maynard Cowan, Jr.
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FOREWORD

This report presents the final results of one of the projects participating in the military-effect programs of Operation Redwing. Overall information about this and the other military-effect projects can be obtained from WT-1344, the "Summary Report of the Commander, Task Unit 3." This technical summary includes: (1) tables listing each detonation with its yield, type, environment, meteorological conditions, etc.; (2) maps showing shot locations; (3) discussion of results by programs; (4) summaries of objectives, procedures, results, etc., for all projects; and (5) a listing of project reports for the military-effect programs.
ABSTRACT

Soil samples were exposed to neutron radiation from Shot Cherokee to help establish the importance of neutron-induced residual gamma radiation from a large-yield thermonuclear air burst. After exposure and recovery, the samples had no detectable activity because the slant range to the nearest sample was nearly 3.5 miles, due to an error in bomb drop. After this failure, an experiment was designed in the field for Shot Yuma in order that induced-activity data could be obtained for a soil other than Nevada Test Site (NTS) soil. Samples of sodium, manganese, and coral sand from Site Sally were exposed above and below the surface at a slant range of 120 yards. At this same station, gamma dose rates were measured and neutron detectors were exposed by Project 2.51.

The full-field gamma radiation measured was due to a combination of fission-product and neutron-induced activities, the only important induced activity being due to \( \text{Na}^{22} \left( \sigma \right) \). At 1.1, 3.4, and 10.9 hours after zero time, neutron-induced gamma radiation accounted for 1.2, 1.1, and 0.8 r/hr of the measured 6.0, 2.2, and 1.2 r/hr. These values were found to be within 50 percent of neutron-induced dose rates predicted by the method of References.
1.1 OBJECTIVE

This experiment was designed to provide data to aid the formulation of a method for predicting the gamma-radiation field caused by air bursts. Specific objectives were to measure the induced activity from a large-yield, thermonuclear air burst and to exploit the opportunity offered for an investigation of activity induced in some soil other than the soil of Nevada Test Site (NTS). When a bomb-drop error prevented any data being obtained on Shot Cherokee, an additional experiment was set up on Shot Yuma to obtain some data applicable to the second specific objective.

1.2 REASONS FOR EXPERIMENT

Certain military uses of nuclear weapons demand a minimum of residual contamination. For example, it might be desirable to exploit the first shock of a nuclear burst by having friendly troops occupy or pass through the ground-zero area soon after the detonation. Although for practical purposes there is no local fallout from an air burst, a considerable area around ground zero can be rendered radioactive by neutron bombardment, and the gamma-radiation field thus produced can prove dangerous to persons entering it hours later. The intensity and decay of such a gamma-radiation field will depend on the kind of weapon, its yield and height of burst, and on the elements which make up the soil around ground zero. This experiment was expected to provide data for the evaluation of the military significance of neutron-induced activity.

1.3 BACKGROUND

Tests at the NTS have led to empirical methods for predicting neutron-induced dose rates from air bursts in the kiloton range over Nevada soil (References 1 and 2). These methods attempt to predict 1-hour dose rates over any area within the test site. However, since no attempts have been made to correlate dose rates and decay rates with the chemical composition of a variety of soils, the methods are valueless for any but Nevada-type soil. Furthermore, since in Nevada soil the amount of sodium, the most important contributor to induced activity, varies from area to area within the test site by as much as a factor of four, the accuracy of such methods is severely limited even for NTS.

A method for the prediction of induced activity which does account for chemical differences in soils has been developed by Canu and Dolan (Reference 3). Before predictions are made for a given soil, the amounts of over twenty elements in the soil must be determined. The gold-neutron flux expected is also used to help establish intensities at times of interest. However, this method has the following important shortcomings:

1. No allowance is made for the moderation of high-energy neutrons into the low-
energy region. For most weapons and soils, this is the most important source of ther-
mal neutrons which cause most of the induced activity. Properties of soil elements
pertaining to this most important effect are ignored. Elements are weighted strictly
on the basis of the relative number of thermal neutrons they absorb.
2. No allowance is made for the distribution of activity with depth in soil nor for
properties of the soil constituents which influence this distribution.
3. The neutron spectrum is assumed to be invariant with slant range and weapon
type.
In other respects, too, this method was developed from unrealistic assumptions.
From a practical standpoint, however, the shortcomings above are the most important.

1.4 PREDICTION METHOD

A prediction method (Reference 4) has been developed which requires the following:
1. Soil Data. Soil density (gm/cm$^3$) and percentage by weight of Na, Mn, Al.
2. Neutron Data. Gold neutrons versus slant range and total number of 14-Mev
neutrons per kt produced external to the case.

1.4.1 Basic Theory. Briefly, the theory for this method is developed as follows:

\[ N_0 = \text{time-integrated neutron density from an isotropic source in air through} \]
\[ \text{a thin layer of soil at the surface} \]
\[ N = \text{time-integrated neutron density through a thin layer at depth X below the} \]
\[ \text{surface} \]

Then,
\[ N = f(X)N_0 \quad (1.1) \]
where \( f(X) \) depends on the soil and the incident neutron spectrum.

The number of neutrons absorbed by a particular isotope in a layer of unit area \( dx \)
thick at depth \( X \) is
\[ dm_i = N_0 \rho_i k_i(X) f(X) dx \quad (1.2) \]
where \( \rho_i \) is the number of atoms per unit volume of an isotope (considered constant with
depth), and \( k_i(X) \) depends on the capture cross section for this isotope and the neutron
spectrum at \( X \).

The contribution made by the radioactive daughter of this isotope to the residual
gamma-radiation field above the surface is
\[ dl_i = g_i(X) dm_i e^{-\lambda_i t} \quad (1.3) \]
The gamma-ray attenuation function, \( g_i(X) \), depends primarily on the density of the
soil and the energy of gamma radiations from the radioactive daughter. When there
is no gamma-emitting daughter, \( g_i(X) = 0 \).

Radiation of the soil by neutrons is considered to be instantaneous at \( t = 0 \). No
daughter-daughter gamma-emitting products are considered since the neutron-induced
radioisotopes of interest give rise to stable daughters.

From Equations 1.2 and 1.3

\[ I_i = N_0 \rho_i e^{-\lambda_i t} \int g_i(X) f_i(X) k_i(X) \, dx \]  \hspace{1cm} (1.4)

For a particular soil and incident neutron spectrum, the integral is a constant, and

\[ I_i = N_0 K_i \rho_i e^{-\lambda_i t} \]  \hspace{1cm} (1.5)

where \( K_i \) is a constant. When a number of gamma-emitting radioisotopes are formed, the dose rate above the surface is given by

\[ I = N_0 \left[ \rho_1 K_1 e^{-\lambda_1 t} + \rho_2 K_2 e^{-\lambda_2 t} + \ldots + \rho_n K_n e^{-\lambda_n t} \right] \]  \hspace{1cm} (1.6)

This equation describes the dose rate at any time only for a given soil and incident neutron spectrum. Different \( K \) values may be expected for each different soil and spectrum combination.

1.4.2 Neutron Spectra. Although neutron spectra vary both with slant range and weapon type, spectra from all weapons nevertheless seem to approach the same equilibrium spectrum as slant range increases, so that for slant ranges greater than 900 yards there is invariance with slant range and weapon type.

Before this equilibrium spectrum is established, the spectrum is always softer (relatively more gold neutrons). In general, both the degree of softness and the slant range necessary to establish the equilibrium condition are greater for weapons having the largest amounts of high explosive. In this method, corrections are made for changes in spectrum when they occur.

1.4.3 Soil Differences. To establish those characteristics of soils which must be considered to give a reasonable degree of prediction accuracy, extremes in the amounts of important soil elements were considered. The influence of each element on the overall moderating or slowing-down power of soil was assessed, as was the importance of each element for absorption of thermal neutrons. Gold neutron measurements versus depth in Nevada soil served to evaluate the importance of build-up caused by moderation, so that relative dose rates could be calculated for extreme cases.

The moderating power of a soil is determined by its amount of hydrogen, the main source of hydrogen being the moisture content of soil. However, a compensating effect causes dose rates measured above the surface of soil to be fairly insensitive to the actual amount of hydrogen (over the expected range of hydrogen abundance or moisture content). When the hydrogen content is high, build-up of low-energy neutrons occurs closer to the surface, but the diffusion length \( L \) of thermal neutrons thus formed is correspondingly shorter, a factor which causes faster decay with depth. For a low hydrogen content, build-up near the surface is less; but the decay of thermal neutrons with depth is also less (larger \( L \)). However, activity produced near the surface must be considered more important than like amounts at greater depths, since emitted gamma rays must penetrate the soil overburden. Thus, without calculation, it is not obvious to what extent such a compensating effect operates.

Figure 1.1, reproduced from Reference 4, shows the extremes expected in relative dose rates for soils with differing amounts of hydrogen. These extremes were calcu-
lated by choosing quantities for other elements found in soil which would maximize or minimize the value of $L$ but would not exceed the expected range of abundance of any element. The relative standing of NTS soil is also shown in the figure. For the equilibrium spectrum, the figure shows that dose rates for soils containing equal amounts of those elements which lead to gamma emitters on $(n, \gamma)$ reaction could be, at most, 30 percent higher or 56 percent lower than those at NTS.

Since neutron spectra from weapons can be softer than the equilibrium spectrum, a similar evaluation of extremes for a 100-percent thermal neutron source was made. It was shown that dose rates for a straight thermal neutron source would be, at most, 20 percent higher or 35 percent lower than for NTS soil.

From the evaluation of these extreme situations, it was concluded that neutron-induced dose rates can be predicted with reasonable accuracy with only a knowledge of those elements which produce gamma-emitting radioisotopes. That is, for a given spectrum, one set of $K$'s may be used in Equation 1.6 for all soils.

**1.5 SUMMARY OF PREDICTION METHOD**

Only three terms will be needed in Equation 1.6, one each for sodium, manganese, and aluminum. Assuming that capture cross sections for these three elements have similar behavior with neutron energy, for the equilibrium spectrum Equation 1.6 may be written:

$$I = \rho n_0 K \left[ (\% \text{ Na}) e^{-\lambda_1 t} + 24.7 (\% \text{ Mn}) e^{-\lambda_2 t} + 60.3 (\% \text{ Al}) e^{-\lambda_3 t} \right] (1.7)$$

Where: $\rho$ = density of soil (gm/cm$^3$)

$n_0$ = gold neutron measurement or estimate ($n_0 \sim N_0$ for a given spectrum)

$K = 3.93 \pm 0.15 \times 10^{-13}$ (evaluated from NTS results)

---

**Figure 1.1** Dose rate versus hydrogen content.
(\%) = percentage by weight of element
\[ \lambda_1 = 0.046/\text{hr} \]
\[ \lambda_2 = 0.266/\text{hr} \]
\[ \lambda_3 = 18/\text{hr} \]

The numbers 24.7 and 60.3 in the equation give added weight to the manganese and aluminum relative to sodium on the basis of capture cross section, half-lives, atomic weights, and gamma energies per disintegration.

Then, for some other slant range, \( r \),
\[ I_r = Y(r)I_{r_1} \frac{r_1}{r} e^{(r_1-r)/200} \] (1.8)
where \( Y(r) \geq 1 \). The factor \( Y(r) \) accounts for the effect of a spectrum which is softer than the equilibrium spectrum. \( Y(r) \) is a function of \( X(r) \) where
\[ X(r) = \frac{(r_n0)_{r}}{(r_1n0)_{r_1}} e^{(r-r_1)/200} \] (1.9)

The factor \( X(r) \) is the ratio of the surface thermal-neutron density actually measured or predicted for slant range, \( r \), to that which would exist at \( r \) if the equilibrium spectrum obtained between \( r \) and \( r_1 \). Because of the importance of build-up of thermal neutrons below the surface,
\[ X(r) \geq Y(r) \]

Figure 1.2 is a plot of \( Y(r) \) versus \( X(r) \) calculated by assuming that the relative importance of build-up is inversely proportional to \( X(r) \). That is,

\[ X(r) q(x)r = q(x)r_1 \]

where \( q(x)r_1 \) is the build-up for equilibrium spectrum. The correction factor shown is for a low H content and is therefore conservative. That is, errors will occur from overestimating, not underestimating, dose rates. For modern weapons with small amounts of high explosive, \( X(r) \) will probably never exceed three.
Chapter 2

PROCEDURE

Shot Cherokee, a 3.5-Mt air burst to be detonated 5,000 feet over Site Charlie, was chosen to investigate induced activity from large-yield, thermonuclear weapons. The ten types of soil used on Shot 5 of Operation Teapot (Reference 4) were exposed to the neutron flux from Shot Cherokee near each of three neutron-detector stations of Project 2.51, located at 80, 1,250, and 2,500 feet from the intended ground zero. Also exposed at each station were a sample of coral sand taken from the island and one sample each of sodium and manganese in the form of salt (NaCl) and manganese dioxide (MnO₂). The latter were included so that a long delay in sample recovery would not result in a loss of all data.

All samples were exposed in watertight steel containers attached by 5/16-inch aircraft cable to large eyebolts set in concrete. The soil sample container was covered with coral soil so that the neutron spectrum incident on the soil samples would be typical of that 1 to 4 inches below the surface. This was thought to be the depth range for maximum thermal neutron density.

Recovery was made at H+6 hours by helicopter, and all samples were taken to Site Elmer where the soil samples were measured with a gamma-ray spectrometer. Dose-rate measurements were made at each of the three stations by the recovery crew.

When the results expected from Shot Cherokee were not obtained, Shot Yuma, a 0.19-kt burst on a 200-foot tower at Site Sally, was chosen to investigate the effect of neutron-induced gamma-radiation fields on a soil different from Nevada soil. Samples of NaCl, MnO₂, and coral soil from the island were exposed at 100 yards ground range near Station 253.08 of Project 2.51. Six samples were exposed, each of the three materials in a steel container about 3 inches above the ground and each in a thin aluminum container about 1 inch below the soil surface. The activity of the recovered samples was measured in a gamma spectrometer, and the resulting spectra were determined as functions of time for all samples. The dose rate near Station 253.08 was measured by a Rad-Safe helicopter hovering at 25 feet at H+1.1 hours and by ground crews at H+3.4 and H+10.9 hours.

The gamma spectrometer used was the 20-channel differential pulse-height analyzer manufactured by the Atomic Instrument Company, equipped with their Model 810 scintillator head, which contained a 2-inch diameter NaI well crystal. The primary calibration was made with chemically pure NaCl and MnO₂ activated in the Los Alamos Laboratory's water-boiler reactor. Properties of the sample holder were minimized by substitution; the same kind were used to obtain the final data and for the calibration runs. Calibrations were made with the pure materials and with a mixture of pure materials and coral soil. Calibrations involved about 2,000 counts per channel; final data runs, about 1,000. It should be noted that the spectrometer required too much maintenance to be entirely satisfactory.
Chapter 3

RESULTS

Because of a bombing error, the estimated slant range from Shot Cherokee burst point to the nearest station on Site Charlie was 3.5 miles. There was no detectable activity in any of the samples when measured at H + 7 hours. Gamma dose rates at Site Charlie were background only.

Chemical analysis of two Site Charlie soil samples indicated their sodium content to be 0.11 and 0.16 percent by weight, while one sample from Site Sally indicated 0.15 percent sodium by weight. In Table 3.1, these values are compared to several values determined for NTS soils.

The two samples of coral soil exposed to Shot Yuma showed only Na\textsuperscript{24} activity when first measured at H + 5 hours. In Table 3.2, a summary of data on activity induced in the six samples is given. Activities are given in dis/min-gm of sample, extrapolated to zero time. The kinds of activity present were identified by the energies and half lives of gamma radiation emitted. The probable error in these activation data is ± 5 percent. Figures 3.1 through 3.3 show sample records for coral soil, NaCl and MnO\textsubscript{2}.

Dose rates measured after Shot Yuma near Station 253.08 are shown in Table 3.3. Gold neutron measurements made on Shot ma are summarized in Figure 3.4 (Reference 5).

<table>
<thead>
<tr>
<th>Location</th>
<th>Sodium</th>
<th>Manganese</th>
</tr>
</thead>
<tbody>
<tr>
<td>Center, Site Charlie</td>
<td>0.16</td>
<td>—</td>
</tr>
<tr>
<td>Shore, Site Charlie</td>
<td>0.11</td>
<td>—</td>
</tr>
<tr>
<td>Station 253.09, Site Sally</td>
<td>0.15</td>
<td>—</td>
</tr>
<tr>
<td>Area 1, NTS</td>
<td>1.25</td>
<td>0.158</td>
</tr>
<tr>
<td>Area 2, NTS</td>
<td>0.43</td>
<td>0.027</td>
</tr>
<tr>
<td>Area 3, NTS</td>
<td>1.60</td>
<td>0.024</td>
</tr>
<tr>
<td>Area 4, NTS</td>
<td>0.95</td>
<td>0.032</td>
</tr>
<tr>
<td>Area 7, NTS</td>
<td>1.84</td>
<td>0.045</td>
</tr>
<tr>
<td>Area 7, NTS</td>
<td>1.81</td>
<td>0.047</td>
</tr>
<tr>
<td>Area 7, NTS</td>
<td>1.33</td>
<td>0.030</td>
</tr>
<tr>
<td>Area F, NTS</td>
<td>0.62</td>
<td>0.023</td>
</tr>
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TABLE 3.2 SAMPLE ACTIVATION DATA, SHOT YUMA

<table>
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<tr>
<th>Sample</th>
<th>Kind of Activity</th>
<th>Above Surface dis/min-gm</th>
<th>Below Surface dis/min-gm</th>
<th>Ratio Below/Above</th>
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<tr>
<td>Coral</td>
<td>Na$^{24}$</td>
<td>$5.09 \times 10^5$</td>
<td>$9.81 \times 10^5$</td>
<td>1.93</td>
</tr>
<tr>
<td>NaCl</td>
<td>Na$^{24}$</td>
<td>$5.73 \times 10^7$</td>
<td>$1.34 \times 10^8$</td>
<td>2.34</td>
</tr>
<tr>
<td>MnO$_2$</td>
<td>Mn$^{56}$</td>
<td>$5.32 \times 10^8$</td>
<td>$1.18 \times 10^{10}$</td>
<td>2.22</td>
</tr>
</tbody>
</table>

TABLE 3.3 DOSE RATES NEAR STATION 253.08, SHOT YUMA

<table>
<thead>
<tr>
<th>Time</th>
<th>Dose Rate 3 ft Above Surface r/hr</th>
<th>Method</th>
</tr>
</thead>
<tbody>
<tr>
<td>H + 1.1 hr</td>
<td>6</td>
<td>Inferred from 25-ft helicopter measurement</td>
</tr>
<tr>
<td>H + 3.4 hr</td>
<td>2.2</td>
<td>Ground monitor reading</td>
</tr>
<tr>
<td>H + 10.9 hr</td>
<td>1.2</td>
<td>Ground monitor reading using AN/PDR-T1B</td>
</tr>
</tbody>
</table>

Figure 3.1 Sample of coral soil data.

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Figure 3.2 Sample of NaCl data.

Figure 3.3 Sample of MnO₂ data.
Figure 3.4 Shot Yuma gold neutrons versus slant range.
4.1 DISCUSSION

By comparing Na$^{24}$ activities in the coral and NaCl samples (Table 3.2), the percent by weight of Na in coral was determined to be 0.31 ± 0.03. This is twice the amount determined by chemical analysis (Table 3.1). The Na$^{24}$ activity must be assumed to be due to Na$^{23}$ (n, $\gamma$) Na$^{24}$, since neither coral nor NaCl contains enough Al or Mg to produce competitive amounts of Na$^{24}$ by (n, $\alpha$) or (n, p).

For such small amounts of sodium, it is believed that the comparative activation method is the more accurate. This could imply a consistent chemical error, because two other samples of coral from Site Charlie (Namu Island, Bikini) show 0.16 percent and 0.11 percent by chemical analysis. It is also possible that the sodium content in coral may vary by as much as a factor of two within a small area.

From Equations 1.7, 1.8, and 1.9 and the neutron measurements of Project 2.51, dose rates were calculated at 1.1, 3.4, and 10.9 hours after zero time. Results are shown in Table 4.1.

The greatest slant range at which a gold neutron measurement was made for Shot Yuma was 604 yards. In order to make a dose rate calculation by the method of Reference 4, it must be assumed that the equilibrium spectrum was established at 604 yards slant range. On the basis of experience with devices containing small amounts of high explosive, this assumption is reasonable. The dose rate was calculated (Equation 1.7) for 604 yards slant range with $n_0 = 1.05 \times 10^{11}$ and $p = 1.6 \text{ gm/cm}^3$. For dry coral, $p = 1.3$, while for coral completely saturated with water, $p = 1.8 \text{ gm/cm}^3$. The value 1.6 is an estimate of conditions at shot time. $X(r)$ was calculated (Equation 1.9) to be 2 for 120 yards slant range; hence $Y(r) = 1.5$ from Figure 1.2. Equation 1.8 was then used to calculate dose rate at 120 yards slant range. As seen in Figure 4.1, measured dose rates do not show the decay with time characteristic of Na$^{24}$. Since Na$^{24}$ was the only important induced activity found in coral, it was concluded that measured values were the result of superimposition of the 15-hour half-life Na$^{24}$ and $t^{-1.2}$ fission product activities. A least-squares fit applied to the dose rate versus time data implied fission product contributions of 4.7, 1.2, and 0.3 r/hr, and Na$^{24}$ contributions of 1.2, 1.1, and 0.8 r/hr at H + 1.1, 3.4, and 10.9 hours (Figure 4.1). This analysis seems consistent with the measured values.

When the amount of sodium used for coral is that determined by the comparative activation method, calculated values range from 1.33 to 1.25 times the values inferred from field measurements. When the chemically determined value for sodium is used, inferred dose rates are 1.50 to 1.6 times the calculated dose rates. This suggests that the correct average sodium content for coral lies somewhere between 0.15 and 0.31 percent. However, errors implicit in the prediction method and in field measurements of dose rates do not allow confidence in this deduction.

Measured dose rates are probably subject to ± 25-percent error, and the estimated probable error in prediction method is ± 30 percent. Assuming that these are the major sources of error, agreement between calculated and measured dose rates is subject to probable error of ± 40 percent.
### TABLE 4.1 COMPARISON OF CALCULATED AND MEASURED DOSE RATES

<table>
<thead>
<tr>
<th>Time After Zero Time</th>
<th>Measured Dose Rate</th>
<th>Contribution of Neutron-Induced Radiation to Measured Dose Rate</th>
<th>Calculated Dose Rate 0.31 pct Na</th>
<th>Calculated Dose Rate 0.15 pct Na</th>
</tr>
</thead>
<tbody>
<tr>
<td>hour</td>
<td>r/hr</td>
<td>r/hr</td>
<td>r/hr</td>
<td>r/hr</td>
</tr>
<tr>
<td>1.1</td>
<td>6.0</td>
<td>1.2</td>
<td>1.6</td>
<td>0.8</td>
</tr>
<tr>
<td>3.4</td>
<td>2.2</td>
<td>1.1</td>
<td>1.4</td>
<td>0.7</td>
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<tr>
<td>10.9</td>
<td>1.2</td>
<td>0.8</td>
<td>1.0</td>
<td>0.5</td>
</tr>
</tbody>
</table>

Figure 4.1 Least-square values compared to measurements.

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The build-up of thermal neutrons with depth in soil is indicated by the relatively high activity of buried samples; however, activity differences between buried and unburied samples are greater than expected. The samples which were exposed above the surface may have been shielded somewhat by the sandbags which were used by Project 2.51 to hold their detectors off the ground. Although the samples were between the sandbags and the device, shielding could still be significant because low-energy neutron flux is nearly isotropic.

4.2 CONCLUSIONS

The difference between the effects of pure fission and fission-fusion neutron spectra on induced activity in soil was not measured, since the soil samples on Shot Cherokee were not activated. However, a method for predicting neutron-induced gamma-radiation intensities (Reference 4) was tested for coral soil on Shot Yuma. Predicted values were within ±50 percent of induced dose rates inferred from field measurements.

2. T. B. Cook and M. Cowan; "Residual Contamination from Nuclear Bursts"; SC-3466(TR), February 1955; Sandia Corporation, Albuquerque, New Mexico; Secret.


4. M. Cowan; "Neutron-Induced Gamma Radiation from Nuclear Air Bursts"; SC-3800(TR), November 1957; Sandia Corporation, Albuquerque, New Mexico; Secret Restricted Data.

5. C. W. Luke, Capt, USA, and others; "Neutron-Flux Measurements"; Project 2.51, Operation Redwing, ITR-1313, December 1956; Chemical Warfare Laboratories, Army Chemical Center, Maryland; Secret Restricted Data.
ARMY ACTIVITIES

1. Deputy Chief of Staff for Military Operations, D/A, Washington 25, D.C. ATTN: Dir. of MW
3. Assistant Chief of Staff, Intelligence, D/A, Washington 25, D.C.
5. Chief of Engineers, D/A, Washington 25, D.C. ATTN: Eng
9. Chief of Transportation, D/A, Office of Planning and Int., Washington 25, D.C.
13. President, U.S. Army Artillery Board, Ft. Bliss, Ohio
17. Commanding General, First United States Army, Governor's Island, New York 4, N.Y.
20. Commanding General, Fourth United States Army, Ft. Sam Houston, Tex. ATTN: G-3 Section
22. Commanding General, Sixth United States Army, Prudential Bldg., San Francisco, Calif. ATTN: AMOCT-4
23. Commandant, U.S. Army Command & General Staff College, Ft. Leavenworth, Kansas. ATTN: ANCHETS
29. The Superintendent, U.S. Military Academy, West Point, N.Y. ATTN: Prof. of Ordnance
34. Commandant, USA Signal School, Ft. Monroe, N.J.
37. Commanding General, Medical Services School, Brooks Army Medical Center, Ft. Sam Houston, Tex.
38. Commanding General, Medical Department Institute of Pathology, Walter Reed Army Med. Center, 625 16th St., N.W., Washington, D.C.
40. Commandant, Walter Reed Army Inst. of Res., Walter Reed Army Medical Center, Washington 25, D.C.
41. Commanding General, ORAB, ONB DCTR, N.Y. ATTN: ORAB-ST
42. Commanding General, U.S. Army Chemical Corps, Research and Development Command, Washington 25, D.C.
43. Commanding General, Chemical Warfare Lab., Army Chemical Center, Md. ATTN: Tech. Library
44. Commanding General, Engineer Research and Dev. Lab., Ft. Belvoir, Va. ATTN: Chief, Tech. Support Branch
45. Director, Waterways Experiment Station, P.O. Box 631, Vicksburg, Miss. ATTN: Library
46. Commanding Officer, Office of Ordnance Research, Box 05, Duke Station, Durham, North Carolina
47. Commanding Officer, Firing Areal, Dover, N.J. ATTN: ORAB-EN
48. Commanding Officer, Director, Ordnance, Ft. Leavenworth, Kans. ATTN: CM, ORDBB
49. Commanding Officer, Aberdeen Proving Grounds, Md. ATTN: Director, Ballistics Research Laboratory
50. Commanding Officer, Frankford Arsenal, Bridge and Tacony St., Philadelphia, Pa.
51. Commanding General, U.S. Army Ordnance Command, Redstone Arsenal, Ala.
52. Commanding Officer, Army Rocket and Guided Missile Agency, Redstone Arsenal, Ala. ATTN: Tech Library
53. Commanding General, White Sands Proving Ground, Las Cruces, N.M. ATTN: ORAB-EN
54. Commanding Officer, Ballistic Missile Agency, Redstone Arsenal, Ala. ATTN: OMBAB-7O
55. Commanding Officer, Ordnance Technical Support Command, Detroit Arsenal, Centerline, Mich. ATTN: ORAB-BO
56. Commanding General, Ordnance Ammunition Command, Fort, Ill.
57. Commanding Officer, USA Signal Radar Laboratory, Ft. Monmouth, N.J.
61. Commanding Officer, USA Transportation Combat Development Group, Ft. Rucker, Ala.
63. Commandant, U.S. Army Chemical Corps, Ordnance Department, Fort, Ky.
64. Commanding General, U. S. GEN Special Weapons-Ammunitions Command, Dover, N.J.
65. Commanding-Chief, U.S. Army Europe, APO 09, New York, N.Y. ATTN: OMDIV, APO
67. Commanding General, Eighth B.S. Army, APO 091, San Francisco, Calif. ATTN: ADJUD-03-3
70. Commander-in-Chief, U.S. Army Pacific, APO 950, San Francisco, Calif. ATTN: Ordnance Officer
71. Commanding General, EWARMANT & KEF, APO, Puerto Rico
72. Commanding Officer, 9th Hospital Center, APO 110, New York, N.Y, ATTN: CC, US Army Nuclear Medicine Research Detachment, Europe
NAVAL ACTIVITIES

87 Chief of Naval Operations, D/8, Washington 25, D.C.
ATTN: OP-4542
88 Chief of Naval Operations, D/8, Washington 25, D.C.
ATTN: OP-73
89 Chief of Naval Operations, D/8, Washington 25, D.C.
ATTN: OP-2000
90 Chief of Naval Personnel, D/8, Washington 25, D.C.
92 Chief of Naval Research, D/8, Washington 25, D.C.
ATTN: Code 611
93 Chief, Bureau of Aeronautics, D/8, Washington 25, D.C.
ATTN: Code 953
95 Chief, Bureau of Medicine and Surgery, D/8, Washington 25, D.C.
96 Chief, Bureau of Ordnance, D/8, Washington 25, D.C.
97 Chief, Bureau of Ordnance, D/8, Washington 25, D.C.
ATTN: B.P.
98 Chief, Bureau of Ships, D/8, Washington 25, D.C.
ATTN: Code 423
99 Chief, Bureau of Yards and Docks, D/8, Washington 25, D.C.
ATTN: B-460
101-102 Commander, U.S. Naval Ordnance Laboratory, White Oak, Silver Spring 19, Md.
103 Director, Material Lab. (Code 500), New York Naval Shipyard, Brooklyn 1, N.Y.
104 Commanding Officer and Director, Navy Electronics Laboratory, San Diego 86, Calif.
109-111 Commanding Officer and Director, U.S. Naval Civil Engineering Laboratory, Fort Ransom, Calif. ATTN: Code 631
112 Superintendent, U.S. Naval Academy, Annapolis, Md.
113 Commanding Officer, U.S. Naval Schools Command, U.S. Naval Base, Treasure Island, San Francisco, Calif.
114 Superintendent, U.S. Naval Postgraduate School, Monterey, Calif.
116 Commanding Officer, Nuclear Weapons Training Center, Atlantic City, N.J. ATTN: Naval Reactor Dept.
117 Commanding Officer, Naval Weapons Training Center, Pacific, Naval Station, San Diego, Calif.
118 Commanding Officer, U.S. Naval Damage Control School, Center, Naval Base, Philadelphia 12, Pa. ATTN: ANC Defense Course
119 Commanding Officer, Air Development Squadron 5, VT-5, China Lake, Calif.
120 Commander, Officer O.G., U.S. Naval Air Development Center, Johnsville, Pa. ATTN: NAS, Librarian
121 Commanding Officer, U.S. Naval Medical Research Institute, National Naval Medical Center, Bethesda, Md.
122 Commanding Officer and Director, Dr. W. Taylor Nodal Basin, Washington 7, D.C. ATTN: Library
123 Officer-in-Charge, U.S. Naval Supply Research and Development Facility, Naval Supply Center, Bayonne, N.J.
125 Commandant, U.S. Marine Corps, Washington 25, D.C.
ATTN: COMUS
126 Director, Marine Corps Landing Force, Development Center, MDS, Quantico, Va.
127 Chief, Bureau of Ships, D/8, Washington 25, D.C. ATTN: Code 372
128 Commanding Officer, U.S. Naval CIC School, U.S. Naval Air Station, Glynn, Brunswick, Ga.
129 Chief of Naval Operations, Department of the Navy, Washington 25, D.C. ATTN: OP-25

AIR FORCE ACTIVITIES

131 Assistant Chief of Staff, Intelligence, Sr. U.SAF, Washington 25, D.C.
ATTN: APOE
132 Deputy Chief of Staff, Operations, Sr. U.SAF, Washington 25, D.C.
ATTN: APOE
133 Maj, U.SAF, ATTN: Operations Analysis Office, Office, Vice Chief of Staff, Washington 25, D.C.
134 Director of Civil Engineering, Sr. U.SAF, Washington 25, D.C.
ATTN: APOE

135-136 Assistant Chief of Staff, Intelligence, Sr. U.SAF, Washington 25, D.C.
ATTN: APOE
138 The Surgeon General, Sr. U.SAF, Washington 25, D.C.
139 Commander, Tactical Air Command, Langley AFB, Va. ATTN: Defense Security, Branch
140 Commander, Air Defense Command, Bethesda Md. ATTN: HQ
141 Commander, Air Research and Development Command, Andrews AFB, Washington 25, D.C. ATTN: HQ
143-144 Commander, AF Cambridge Research Center, L. O. Nason Field, Bedford, Mass. ATTN: CRMC-2
150-151 Director, Air University Library, Maxwell AFB, Ala.
152 Commander, Lowery Technical Training Center (TW), Lowery AFB, Denver, Colorado.
153 Commandant, School of Aviation Medicine, U.SAF, Randolph AFB, Tex. ATTN: Research Secretariat
154 Commander, 100th Sp. Ums. Squadron, Sr. U.SAF, Washington 25, D.C.
155-157 Commander, Wright Air Development Center, Wright-Patterson AFB, Dayton, Ohio. ATTN: WMC
158-159 Director, U.SAF Project RAND, N.Y. ATTN: U.SAF Laison Office, THE RAND CORP., 1700 Main St., Santa Monica, Calif.
161 Chief, Ballistic Missile Early Warning Project Office, 250 Church St., New York 13, N.Y. ATTN: Col. Leo V. Skinner, USAF
162 Commander, Rose Air Development Center, MAD, Griffiss AFB, N.Y. ATTN: Documents Library, NCMC-1
163 Commander, Air Technical Intelligence Center, U.SAF, Wright-Patterson AFB, Ohio. ATTN: APOE-RAL, Library
164 Assistant Chief of Staff, Intelligence, Sr. U.SAF, APO 633, New York, N.Y. ATTN: Directorates of Air Targets
165 Commander, Alaska Air Command, APO 942, Seattle, Washington. ATTN: AUCQ
166 Commander-in-Chief, Pacific Air Forces, APO 953, San Francisco, Calif. ATTN: FPICE-HB, Base Recovery

OTHER DEPARTMENT OF DEFENSE ACTIVITIES

168 Director, Weapons Systems Evaluation Group, Room 19500, The Pentagon, Washington 25, D.C.
169 Commandant, The Industrial College of the Armed Forces, Ft. McNair, Washington 25, D.C.
170 Commander, Armed Forces Staff College, Norfolk 11, Va. ATTN: Library
173 Commander, Field Command, DADA, Sandia Base, Albuquerque, N.M. ATTN: DADA
174 Commander, Field Command, DADA, Sandia Base, Albuquerque, N.M. ATTN: DG
175 Commander, Field Command, DADA, Sandia Base, Albuquerque, N.M. ATTN: FVCF
176 Commander, JTF-7, Arlington Hall Station, Arlington 12, Va.
177-178 Chief of Staff, Defense Atomic Support Agency, DADA, Sandia Base, Albuquerque, N.M. ATTN: DG
179-180 Chief of Staff, Defense Atomic Support Agency, DADA, Sandia Base, Albuquerque, N.M. ATTN: DG
181 Assistant Chief of Staff, Defense Atomic Support Agency, DADA, Sandia Base, Albuquerque, N.M. ATTN: DG

OMIC ENERGY COMMISSION ACTIVITIES

190-199 Los Alamos Scientific Laboratory, Report Library, P.O. Box 1663, Los Alamos, N. Mex. ATTN: Helen Redman

SECRET